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Development of colour centre lasers for spectroscopy applications

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Résumé. — On présente la description de deux lasers à centres colorés mis au point par nous-mêmes. Les détails techniques des solutions sont donnés surtout au regard de la formation des centres F
(II) dans les cristaux de KCl : Li. Les propriétés de fluorescence de ces centres sont examinées brièvement et l'on présente un exemple d'application à la spectroscopie moléculaire.

Abstract. — The description of two home-built colour centre lasers is presented. Technical details of the solutions are given with special concern to the production of the F
(II) centres in KCl : Li crystals. The fluorescence properties of these centres are briefly investigated and an example of application to molecular spectroscopy is shown.

1. Introduction. — Soon after their first realization [1] the cw colour centre lasers have attracted a considerable interest owing to their potential usefulness both in basic spectroscopic research and in applied research. At a building cost not much larger than an ordinary dye laser, they can deliver tunable narrow (single-mode) emission up to several mW in a spectral region (near I.R. up to and beyond 3 µm) where the only other coherent tunable continuous source is some form of semiconductor laser, with considerably lower spectral properties.

After extensive research carried out in U.S.A. and in Germany in the last few years, it is now possible, by using different kinds of colour centres, to cover almost entirely the spectral range between 0.85 and 4 µm [2-4]. In the 1 µm region, an important application is foreseen in the field of fiber-optic communications [5], while for the whole range 1 to 4 µm they have a strong spectroscopic interest : many transitions between excited atomic levels [6-9] and many vibrational lines of organic molecules and radicals such as OH, HF, NH, CH, NH2, CH3 stretches [10-14], lie in this region.

In metrological applications, they may provide a link in a multiplication chain between the 10 µm radiation of the stabilized CO2 laser and visible radiation by, e.g., the He-Ne laser [15]. Further spectroscopic applications can be found in the high sensitivity detection of atmospheric pollutants, real-time study of the evolution of a chemical reaction, study of decay times in the picosecond range [16], and so on.

This wide variety of possible uses has prompted several laboratories to design their own colour centre lasers, in relation to the individual needs. But the practical realization of this device involves quite a few technological problems, which have prevented up to now a wide spread diffusion of this type of laser not with standing the vast possibilities we shortly outlined above.

In our laboratories we have designed and built
two working lasers operating in the wavelength range 2.5-2.85 μm, these lasers use the same colour centre (F₃(II) in KCl : Li) but are different in design, way of optical pumping, and potential uses. In this paper we shall describe the essential features of these lasers, and how did we overcome with simple means the less common technological problems, in order to ease the way to other researchers who may wish to build their own colour centre laser for their own purpose.

2. Basic features of the laser. — The active medium of this kind of laser is an alkali-halide crystal, appropriately doped and coloured. Several colour centres have been found to lase : F₃(II), F₉(II), F₂⁺, F₂⁻-like and recently Ti⁺-vacancy centres [2, 3]. Laser action has also been obtained in other host crystals [17] and, in pulsed operation only, with F₂ and F₂⁻ centres [18]. General reviews on the colour centre lasers can be found in the literature [2, 3], and therefore we shall review here only briefly the essential details that apply to our own case.

When a KCl crystal is heated in a potassium atmosphere, neutral lattice defects are produced by the following mechanism. Potassium atoms stick to the surface of the crystal at the regular lattice positions where they become trapped as K⁺ because of the thermal ionization. Chlorine ions diffuse to the surface, building new atomic planes and leaving behind anion vacancies. The electron produced by the ionization of the potassium atom become eventually trapped in the potential well produced by the anion vacancy that is electrically equivalent to a positive charge. The system vacancy + electron is called F centre. Upon irradiating such a centre with light of appropriate wavelength, the electron is excited into the conduction band and the vacancy is free to migrate in the crystal. However in a Li doped crystal the vacancies become easily trapped by the Li ions because of the positive binding energy between vacancy and impurity. As the electron recombines with the vacancy a new centre is formed : the F₃(II) centre [19] (see Fig. 1). The energy level diagram of this centre is also shown in the same figure together with the typical optical cycle of a four-levels laser. After optically pumping from the ground state |g⟩ to the excited state |e⟩, the system quickly (≈ 10⁻¹² s) relaxes to the |re⟩ state, then may radiate to the |rg⟩ state, that again quickly relaxes to the ground state. It is therefore relatively easy to obtain population inversion between the |re⟩ and |rg⟩ states, and thus laser action may start. The absorption transition (|g⟩ → |e⟩) falls in the visible region while the fluorescence (|re⟩ → |rg⟩) occurs in the I.R. Optical pumping may thus be accomplished by means of a fixed frequency visible laser, while infrared emission may be tuned over the fluorescence band, exactly like a dye laser.

These crystals cannot operate at room tempe-
Van Doorn technique [22], with an oven modified [23] in order to gain a precise control on the K vapour pressure and prevent as far as possible any damage of the polished surfaces. The coloration process lasts from 20 to 80 min. at temperatures between 570 and 590 °C at a vapour pressure between 1.3 and 4 kPa. Thereafter the crystal is suddenly cooled to below 200 °C, in order to prevent the formation of aggregates of colour centres, by means of a flow of cool dry nitrogen, in a separate area of the oven. The crystal is checked and it is thereafter loaded into the laser cryostat (to be described below), cooled to −30 °C and irradiated with green light (from a tungsten lamp with a filter) for 30 min. With this process we obtain the aggregation of F centres and Li impurities, giving rise to the FA(II) centres as described above. At −30 °C the ion mobility is still high enough for the aggregation to occur, without risk of getting more complex aggregate that would absorb laser and pump radiation. After the aggregation is complete, the crystal is cooled to liquid nitrogen temperature, where the ion mobility is negligible. At this temperature the colour centres are stable even under the strong optical pumping necessary for the laser action.

4. Mechanical set-up. — The optical cavity is a folded, astigmatically compensated cavity [24] of the type commonly used in cw dye lasers. The mechanical set-up of the mirrors is of course complicated by the need to work at cryogenic temperature and therefore under vacuum.

The theory of the optical cavity has been given in reference [24]; we shall therefore not give here the details of its design: the dimensions have been calculated by keeping in mind the crystal size, the required mode spacing and the stability range of the system. Details of this computations can be found in reference [25]. The most important difference between the two lasers we built is the method of pumping. In one case (in Frascati, Roma) the source was an Ar+ laser, and we decided to pump the colour centre laser through the end mirror of the optical cavity. This mirror has a dichroic coating transmitting the Ar+ green radiation and reflecting the infrared radiation. A general view of this laser is shown in figure 2. The end mirror and the folding mirror are inside the cryostat; the optical alignment must therefore be carried out at room temperature with a dummy crystal in place of the coloured crystal. After the cryostat has been evacuated, the only external adjustments are those of the output coupler and the lens L2 for properly focussing the pump light to the mode waist of the laser. This approach is very simple and gave satisfactory results; it requires however a rather uncommon dichroic coating for the end mirror. The schematic of the optical cavity is shown in figure 3 with details of the output coupling and tuning.

In the other laser (in Pisa) the pump source was a Kr+ laser, and we decided to avoid the dichroic mirror by pumping the crystal through an uncoated ZnSe beamsplitter, located between the folding mirror and the output coupler (see Fig. 4). It is positioned at Brewster's angle for the IR radiation, and it reflects about 50% of the visible, vertically polarized, radiation. This arrangement has the advantage of being usable without modifications with several pump sources and with several active crystals, since it is essentially wavelength independent, the end mirror and the folding mirror both having a broadband metallic coating. The positioning of all mirrors is accessible under vacuum by means of appropriate
Fine adjustment of the pump focussing is accomplished by changing the input beam direction. In addition, in order to limit power fluctuations due mechanical instabilities, all mirrors have been fastened to a single superinvar plate, whose size is 115 × 17 × 1.5 cm.

In both lasers the output coupler is outside the vacuum chamber, and with this kind of mounting the output mirror can be replaced by a tuning element (grating or prism plus mirror) and an absorption cell can be easily placed intracavity. Because of the atmospheric water vapour absorption around 2.8 μm, the region between the CaF₂ window and the output mirror is sealed in an airtight enclosure flushed with dry nitrogen.

5. Fluorescence measurements. — There is an optimum for the density of F centres in the crystal. If their number is small, not enough FA(II) centres are formed during the aggregation process in order to absorb most of the pump light, thus increasing the effective threshold. If there are too many F centres, not all of them will form FA(II) centres, and the remaining F centres will absorb uselessly pump radiation, and might even absorb infrared radiation. In addition, in an optically thick crystal, part of the crystal has no useful purpose and rather increases the threshold. It is however difficult to state precisely the optimum F centre density, and only vague indications about it are given in the literature. Therefore we decided to measure, under the same conditions, the fluorescence from the FA(II) centres of variously coloured crystals with the assumption that the crystals that give the highest fluorescence are those with the highest visible absorption and the lowest infrared self-absorption.

The results are given in figure 5: the fluorescence intensity normalized to the incident power is plotted versus the measured FA centres density. Two sets of results are given for vertical and horizontal polarizations of the pump. A maximum can clearly be identified, and experiments showed that the crystal with the FA centres density near this maximum consistently had lower threshold than the other ones. This is therefore an easy way for identifying the most likely candidates for lasing with low thresholds. The luminescence decrease at high FA centres concentration is probably analogous to the quenching of the F centre emission found by Miehlich [26], but it occurs at much lower centres concentration.

Of considerable significance is the difference between the two sets of data with different pump polarization. The crystal is installed into the cavity at Brewster's angle and with the \langle 110 \rangle axis vertical. The FA centre axis (anion vacancy-cation impurity direction) can be oriented along all three crystal axes, but in the absorption process the dipole moments of the transitions depend upon the direction of the centre and the polarization of the light. The high energy band in the green (FA₂ band in Fig. 1) absorbs light polarized perpendicularly to the centre axis, while the lower energy band in the red (FA₁ band) absorbs light polarized parallel to the centre axis.

Moreover, after excitation, the centre can change its orientation due to a jumping of the vacancy around the impurity during the relaxation process. Therefore if one pumps the system with polarized light, reorientation of the centres (along directions such that they can no longer be excited) leads to a progressive decrease of the absorption and the system becomes transparent to the pumping light. With a crystal oriented at Brewster angle to the incident pumping beam, all centres absorb horizontally polarized light in the FA₂ band (green) thus preventing the alignment of the centres in a particular direction. On the other hand, on pumping in the FA₁ band (red) with a vertically polarized beam one could align the centres and bleach the absorption if the « vertical direction » coincides with the \langle 001 \rangle direction of the crystal.

Consequently there are no problems if one uses, like in the Rome-Frascati laser, a horizontally polarized pump through the end mirror. However in a laser that is pumped by means of a beamsplitter, like the Pisa laser, where it is essential to use a vertical polarization (unless the beamsplitter has very excep-
tional properties), one has to take care of the alignment of the crystal in order to avoid bleaching. Therefore it was important to test in this case whether the fluorescence efficiency with vertically polarized pump (at $\lambda = 6.471 \text{ Å}$) was comparable to that with horizontal polarization. Figure 5 shows that with KCl : Li at 77 K the two fluorescence intensities are comparable, and that one can therefore use the uncoated beamsplitter without too large a loss.

6. Wavelength tuning. — In order to change the emission wavelength within the fluorescence band, we used both a prism [27] and a grating [28]. As a prism we use a 60° sapphire prism, whose surfaces lie almost exactly at Brewster’s angle when the prism is placed in the symmetrical position; tuning can be accomplished by rotating the output mirror around a vertical axis. As a grating we use a Bausch & Lomb replica with 420 grooves/mm, blazed at 2.15 µm, in a Littrow mount, the zeroth order reflection being the output beam; tuning is accomplished by rotating the grating. We use a further mirror in front of the grating and parallel to it (see Fig. 4) to steer the output beam. Both this mirror and the grating are mounted on a same rotatory stage, so that, on changing the wavelength, the output beam is neither translated nor changed in direction.

Both prism and grating work satisfactory, but operation of the grating proved to be easier. In order to operate the laser in a single mode, we use an intracavity etalon with special coating; this etalon has a low finesse (about 2) but very low losses in the infrared.

7. Laser performances. — At the threshold the pumping power reaching the crystal was about 100 mW for the free-running laser equipped with a 95 % reflectivity output mirror. With ageing, the lasing threshold increases (Fig. 6) but fairly slowly, and all crystals were still usable one month after preparation.

With the grating as a tuning element, the threshold increases to 230 mW at the emission peak, with dry nitrogen flushing the arm inside the vacuum chamber. The tunability range is from 2.50 to 2.85 µm in the absence of water vapour absorption. The maximum power is about 4 mW with 850 mW pump power on the crystal from the Kr+ laser, and the overall emission width is about 10 GHz (without the mode selecting etalon).

Even though the work on the lasers is still in progress to reduce line-width and increase the spectral range by use of other crystals, we already obtained some spectra that show usefulness of this kind of source in molecular spectroscopy. A spectrum of water has been obtained intracavity and spectra of CH$_3$OH, HCOOH and CH$_3$COOH have been studied by means of optoacoustic detection with the apparatus described in reference [29]. The results of these investigations will be discussed in a forthcoming paper; a spectrum is shown in figure 7, just as an example of the versatility and capabilities of these sources.

Fig. 6. — Increase of the laser threshold of the colour centre laser as function of ageing time.

Fig. 7. — Optoacoustic spectrum of CH$_3$OH molecules (A) and laser output (B). The amount of noise can be deduced from the interval at the centre of the spectrum where the laser was blanked.

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