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### FORMATION AND PROPERTIES OF MBE GROWN Alsb-Gasb (100) INTERFACES

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Ce travail porte sur l'élaboration par jets moléculaires des interfaces GaSb-AlSb et AlSb-GaSb, et sur l'étude pas à pas par diffraction d'électrons rapides, par effet Auger, par mesures du travail de sortie de la surface de départ et de l'interface en formation. On précise ainsi la qualité électronique des surfaces, interfaces, et, met en évidence des diffusions d'Al pour l'interface GaSb/AlSb et pas de diffusion notable de Ga pour AlSb/GaSb.

In this paper, we present results, we have obtained on GaSb-AlSb or AlSb-GaSb interfaces grown by MBE. We first focuss on the surface properties obtained by AES, RHEED, work function topographies. Then, we describe the formation of the interface and the evolution of its properties during growth. We detail AES and work function measurements and show Al migration at the GaSb/AlSb interface and no noticeable Ga diffusion at the AlSb/GaSb interface.

# I - INTRODUCTION.

For a few years, research on GaSb and alloys have been increasing continuously. Type I (GaSb-GaAlSb) or type II (GaSb - InAs) superlattices present interesting properties resulting particularly from strain at the interface and from relevant effects. However, this strain is not too high and these interfaces are good enough to be used in electronic devices and in quantum well structure. Photodetector and lasers integrating these interfaces and optical fibers achieved with these materials are promising. New developments can be expected in the next future.

MBE is now intensively used, to grown these materials (1-5).

In this paper, we present results relevant to the elaboration's conditions, to the physico-chemistry and to the electronic properties of GaSb-AlSb interfaces we have grown by this technique.

# **II - GROWTH APPARATUS AND CONDITIONS**

The samples were grown in a MBE system which has been previously described (6). This system which was home made, is composed of three chambers.

-One evaporation chamber with 7 Riber 110 PBN/cells, with a RHEED, apparatus and a mass spectrometer.

-One chamber for Auger and work function studies.

-One chamber to keep in stock several samples.

The vacuum in the three chamber is in the  $10^{-11}$  Torr range. The samples can be transfered from one chamber to another without breaking the vacuum. Under these conditions, we are able to characterize all the phases of the interface formation. Furthermore, by using a Kelvin probe which is moved in front of the surface of the substrate or in front of the layer, we deduce information on the uniformity of the layer, on the work function  $\varphi$  and on the photovoltage SPV. This last technique is very useful because it does not affect the properties of the interface during measurements. Its sensitivity is high and variations as small as 1.5 meV can be detected.

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# **III - SURFACE PREPARATION**

### 1 - Substrate surface

The samples 100 oriented, obtained from MCP wafer  $(350 \mu)$  were first mechanically polished and etched in different chemical solutions. Studies of the surface composition having shown that the best results (small concentration of C, O) were obtained with Br-Methanol, we have used this solution.

The samples were then inserted into the chamber and heated at  $300^{\circ}$ C. The surfaces prepared in this way show reproducible properties. The carbon concentration is low but they are covered by a thin oxyde layer. In order to eliminate it, substrates were annealed at  $580^{\circ}$ C. Under these conditions,  $Ga_2O_3$ ,  $Sb_2O_5$ , were craked and

desorbed. But at this temperature, Ga and Sb do not evaporate congruently and the surface becomes Sb poor. In order, to avoid this depletion in Sb, we annealed the sample under a Sb flux  $(10^{-9} \text{ Torr})$ .

The surfaces prepared in this way were clean and ordered, and depending on the flux presented the following reconstruction : (1x3) for Ga rich surfaces, (2x3) for Sb rich surfaces.

Similar experiments with an As flux lead to the conclusion that the surface reconstruction is determined by the nature of the vapor flux. The reconstruction was characteristic of GaAs for an As flux, (on GaSb as well as on GaAs) and of GaSb for Sb flux on GaAs or GaSb.

Besides AES, we performed EELS and work function measurements. EELS showed the typical peaks corresponding to valence-band-conduction-band transitions, to Sb (4d), Ga (3d) core level excitation. Work function  $(\Phi)$  and surface photovoltage (SPV) topographies obtained by moving a metallic probe in front of the sample and measuring the contact potential difference and the surface photovoltage are presented in Fig. 1. For GaSb, SPV # 0 and  $\Phi$  variation along the surface are smaller than 100 meV. The surface barrier if present, is small (it is not the case for GaAs substrate) and the variations in  $\Phi$  probably result from variations of the electronic affinity.

2 - Layer surfaces.

Starting from the previous substrates, we have grown GaSb or AlSb layer (growth rate 1.5. A°/s for GaSb , 0.5 - 5 A° / s for AlSb - T substrate # + 580°C, vacuum

#  $10^{-9}$ Torr. These surface are clean and ordered but the work function and surface photovoltage topographies obtained immediatly after the growth, show noticeable variations : the surfaces are not uniform (Fig. 1). Their electronic properties, very sensitive to small modifications, vary from point to point.

# III - INTERFACE FORMATION

We have grown the following interface : AlSb/GaSb substrates, AlSb/GaSb MBE grown/GaSb substrate, GaSb/AlSb MBE grown/GaSb.

In the case of the AlSb/GaSb substrate, the interface was disturbed by the substrate morphology.

Therefore, we have grown AlSb or GaSb layers on a MCP substrate covered by a lum MBE grown GaSb buffer. Some results we have got by AES, work function measurements are given in Fig. 2. When growing AlSb on GaSb, Ga peaks decrease and Al peaks increase but 50 A° of AlSb are enough to attenuate Ga peaks. Furthermore, the ratio of Ga (55 eV) peaks before and during AlSb growth, fastly decrases. Ga atoms may diffuse into AlSb but this diffusion is not significant.

In the case of the growth of GaSb over AlSb Al peaks and Al (68 eV) / Al (1395 eV) ratios progressively decrease when increasing the top layer thickness but Al is still detected for 150 A° of GaSb. Simultaneously the ratio Al (68 eV/Sb (26 eV) goes through a maximum for 1-2 A° of GaSb and this illustrate an Al transfer between the last layer of AlSb and the first layer of GaSb.

Therefore, Al atoms diffuse into GaSb and this diffusion is significative and result in noticeable modification. Indeed measurements of work function variations shown in Fig.2 emphasize high modifications in  $\phi$  during the first monolayer growth.  $\phi$  first increases subsequently reaches a maximum value for half a monolayer, then decreases and finally becomes equal to GaSb work function

### DISCUSSIONS

Diffusion of Al into GaSb cannot be explained by taking into account the heat of formation of GaSb and AlSb; the heat of formation of AlSb is higher than that of GaSb and the situation is not favorable to an Al exchange from AlSb to GaSb. However, the values of the heat of formation found in the literature do not exactly correspond to the conditions of these experiments and to interface reaction.

However, the values of the heat of formation found in the literature do not exactly correspond to the conditions of these experiments and to interface reaction. Therefore from a thermodynamical point of view, may be this exchange cannot be excluded. Furthermore, we can also envisage the formation of Ga clusters in top of AISb. The energy needed to promote the exchange could be supplied by the formation of these clusters as it is the case for metal deposits on III-V compounds (7). However, in view of thermodynamical data, the hypothesis of the formation of small clusters of AISb) supplying Al atoms for the diffusion into GaSb seems more probable.

The work function variations can be associated with this exchange. The last layer of A1Sb lose A1 atoms and the first layer of GaSb seems Sb poor. As A1 and Sb vacancies induce, respectively donor and acceptor states (8),  $\phi$  variations can be explained by an electron transfer from the last layer of A1Sb to the first layer of GaSb and by a resulting increase in electronic affinity **X**. When the thickness of the layer increases,  $\phi$  varies in order to get the value characteristic of the thick layer. The 50-70 A° correspond to the depth of the interface.

#### IV - CONCLUSION

Surfaces prepared obtain by MBE present electronic properties (affinity, barrier) much more uniform than other surfaces (cleaved or chemically cleaned). The growth of AlSb/GaSb and GaSb/AlSb brings to the fore the diffusion of Al which cannot be simply explained by considering the heats of formation. Possibly this diffusion is assisted by strains in the layer (9).

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