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CORRELATION BETWEEN MICROSTRUCTURE OF NiO GRAIN-BOUNDARIES AND INTERGRANULAR DIFFUSION

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Résumé - Une diffusion intergranulaire préférentielle de Ni et de Co a été signalée dans de l'oxyde de nickel polycristallin obtenu par oxydation thermique du nickel. Ces résultats ne sont pas confirmés sur du NiO obtenu par d'autres procédés. La structure des joints est sans doute responsable de la différence de comportement.

Abstract - An enhanced intergranular diffusion of Ni and Co was observed in polycrystalline NiO prepared by thermal oxidation of nickel. These results are not confirmed in nickel oxide prepared by other methods. The structure of the grain-boundaries is believed to be responsible for this different behaviour.

I - INTRODUCTION

An enhanced intergranular diffusion of Ni and Co has been observed in the temperature range 500-800°C in nickel oxide samples prepared by thermal oxidation of nickel at 1100°C /1/. This result, one of the few obtained on non-metallic materials, is not surprising since it is well accepted that diffusion proceeds more rapidly along grain-boundaries (GB's) than through the lattice because, in particular, of the atomic jump frequency much greater than in the lattice /2-5/. Numerous experiments, mainly carried out on metals, support this opinion and one frequently finds that grain-boundary diffusion coefficients D' are several orders of magnitude greater than the corresponding lattice diffusion coefficients D /6/. It is also well accepted that the dislocation structure of the GB's, that is, the type of dislocation (edge or screw), its dissociation, and its possible spreading, can influence the boundary diffusivity. To illustrate this point, one of the most fruitful models remains the pipe mechanism of diffusion of Turnbull and Hoffman based on the dislocation description of sub grain-boundaries (SGB's) /7,8/. The model predicts an influence of the misorientation between adjacent crystals as well as of the type of dislocation on the diffusion process ; it also predicts an anisotropy of the intergranular diffusion.

Within this frame we have tried to refine the statistical results previously obtained on polycrystalline NiO (called hereafter "oxidized Ni") by considering intergranular diffusion experiments on well defined GB's. This led us to carry out a large number of experiments on various $\langle 011 \rangle$ or $\langle 001 \rangle$ tilt axis bicrystalline samples grown from the melt (flame fusion technique), which indicated no preferential grain boundary diffusion with respect to lattice diffusion. In order to explain these contradictory results a statistical study of the grain boundary features was performed by transmission electron microscopy. The orientation relationships, the coincidence index, the orientation texture of the GB' planes and their detailed microgeometry as well as the oxidation texture, when necessary, were studied in the bicrystals grown from the melt and in the "oxidized Ni". The resulting data were compared with those determined on well annealed NiO specimen. As a result it appears that the nickel oxide obtained by oxidation is far from thermodynamic equilibrium and it is believed that this may be the origin of the difference in diffusional behaviour.

II - ORIGIN OF THE SAMPLES UNDER STUDY

The origin and characteristics of the three types of materials which were studied are described elsewhere /9/. Let us remind briefly what they are :

- "1100°C oxidized Ni" obtained by oxidizing a 30 μm thick nickel foil at 1100°C (this polycrystalline material was provided by Atkinson who observed on it an enhanced intergranular diffusion /1/). According to the authors publications the total amount of metallic impurities contained into the original nickel is between 100 and 1200 p.p.m.
- "1300°C oxidized Ni" obtained by oxidizing a 600 μm thick foil at 1300°C. This foil was then annealed during 240 hours at 1600°C leading to an "annealed NiO". The original nickel was a high purity metal purified by ion exchange. The amount of metallic impurities, except Al and Si, was less than 5 p.p.m.
- crystals and bicrystals grown from the melt by the flame fusion technique /10/, which were previously investigated in order to determine their intergranular relative energy (fig. 1) /11/. These bicrystals were made from Johnson Matthey, grade 1 powder (metallic impurities < 20 p.p.m.). Activation analysis realized on NiO crystals after growing have confirmed the Johnson Matthey analysis (except for Al and Si which are difficult to analyse by activation in a Ni matrix).

During the numerous TEM observations and diffraction experiments not any precipitate was observed, either in the GB region or in the bulk. Moreover several bicrystals grown from the melt were broken under high vacuum ($P \simeq 10^{-8}$ Pa) along GB's and the latter analyzed by Auger spectrometry. The resulting analysis showed that the GB's in the bicrystals were free from segregation.

Diffusion experiments were carried out on the three types of materials (by Atkinson et al. in the first case) together with extensive geometrical and morphological studies performed by TEM.

III - DIFFUSION STUDY RESULTS

As shown in fig. 1, the relative energy of the two series of symmetrical GB's is strongly dependent on misorientation, especially in the case of <011> tilt boundaries for which three cusps were evidenced, corresponding to {221}, {111} and {311} boundary planes (the respective coincidence indices Σ are 9, 3 and 11)⁽¹⁾. Although the absolute value of the intergranular energy remains unknown, the experimental evidence of the cusps is well established and a cautious examination of the experimental conditions showed that the cusps correspond to GB energy minima. Now, Herbeuval et al., in particular, have shown on aluminium <011> tilt GB's that there exists a close

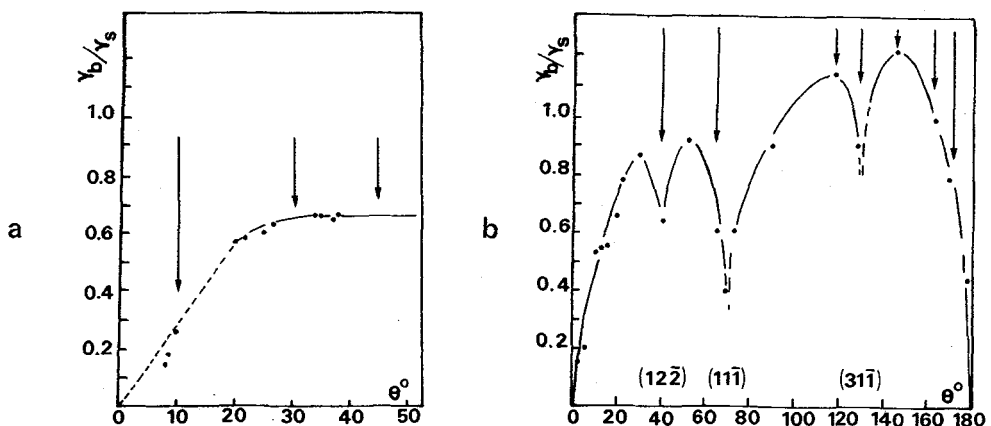


Fig. 1 - Influence of the misorientation θ upon the relative energy of [001] (a) and [011] (b) symmetric tilt grain boundaries at 1520°C /11/. γ_b is the GB energy and γ_s is the surface energy. The low energy boundaries correspond to SGB's and to the (122), (111) and (311) GB planes. The symmetrical tilt boundaries which were used for diffusion experiments are indicated by arrows.

relationship between the energy of a GB and the corresponding intergranular diffusion coefficient D' /12/. By analogy we thought that the $\langle 011 \rangle$ GB's of our NiO samples were well suited for a verification of a similar energy-diffusion correlation in an ionic material. Several misorientations, indicated by arrows in fig.1 and corresponding either to high energy misorientations, or to low energy misorientations, were selected. In the first case, it was expected to observe an important intergranular diffusion (the intergranular diffusion coefficient of Ni determined on "1100° oxidized Ni" ($P_{O_2}=1$ atm) is $D'\delta = 3 \cdot 10^{-8} \exp(-1.78 \text{ (eV)/kT})$, i.e. $D'/D \sim 10^6$, if we assume as Atkinson for the GB width $\delta \simeq 7 \text{ \AA}$ /1/. In the second case, corresponding to CSL misorientations of small misfit between adjacent grains, no diffusion or very limited preferential GB diffusion was expected to be observed ⁽¹⁾ /13/.

Two types of diffusion experiments were undertaken : Self-diffusion using ^{63}Ni (quantitative experiments were carried out by abrasion and counting of the β activity of the samples ; qualitative experiments used auto-radiography), and hetero-diffusion of homo-valent cobalt (quantitative measurements were made using the electron microprobe technique). No enhanced self-diffusion or hetero-diffusion were detected at the GB's. In order to complete our information, additional Ni self-diffusion experiments were undertaken on polycrystalline "1300° oxidized Ni" samples, after polishing the as-oxidized surface, but before further high temperature annealing. Again, no preferential GB diffusion was evidenced on these polycrystalline samples.

IV - GRAIN BOUNDARY MICROSTRUCTURE

In order to find a possible explanation to these apparent diffusion discrepancies, the GB's were subjected to crystallographic characterization, morphological studies and dislocation structure determinations.

IV.1. Crystallographic description of the GB's : such a description includes, the 5 macroscopic parameters needed to define the orientation relationship existing between the adjacent crystals and the indexation of the GB plane in relation to both crystals /14/. This determination was carried out using the crystallographic data collected from TEM micrographs and diffraction patterns, according to methods detailed in several papers /14-16/. Briefly the results can be summarized as follows (see ref. 9 and 14 for details) :

- Orientation relationships : these relations, described by a rotation θ around an axis R, were computed for each interface and the results used to look for a possible coincidence index Σ . In the "flame fusion crystals" some specimen are closely approximating low values of Σ , as indicated in fig. 1. In the "1100°C oxidized Ni" and in the "annealed NiO" a coincidence index (within the limit of $\Sigma \leq 41$) can be allotted to about 12 % of the couples of crystals adjacent to each GB ⁽²⁾. However no preferential relationship appears, whatever the thermal treatment is.
- Orientation of the GB planes : this characteristic was determined for 113 couples of grains for the "1100°C oxidized Ni" and for 23 couples for the "annealed NiO". The results are reported on standard [001] stereographic triangles (fig. 2). These data are quite interesting because of the GB clearcut tendency, in both types of specimen to avoid the $\{011\}$ orientations defined with respect to one of the crystals, this tendency being strongly enhanced by annealing, since we do not observe GB planes within 15° of (011) . It must be emphasized that this cannot result from the texture which develops either during the oxidation, or the annealing processes ⁽³⁾. Hence, geometrical considerations, linked with the columnar oxidation growth of the grains cannot be evoked to explain the absence of $\{011\}$ planes.

⁽¹⁾ - Σ is the ratio of the volume of the Coincidence Site Lattice (CSL) unit cell to that of the crystal lattice cell.

- The CSL misorientations are singular orientation relationships corresponding to a high density of sites common to both lattices, which form the CSL lattice. According to theory, the higher the density of common sites, the lower the energy of an interface passing through a low index CSL lattice plane.

⁽²⁾ - 12 % instead of the previously reported 50 % by ref. 14. The disagreement is due to the erroneous method which was first used to determine the deviation of the bicrystals from the CSL coincidence misorientation /21/.

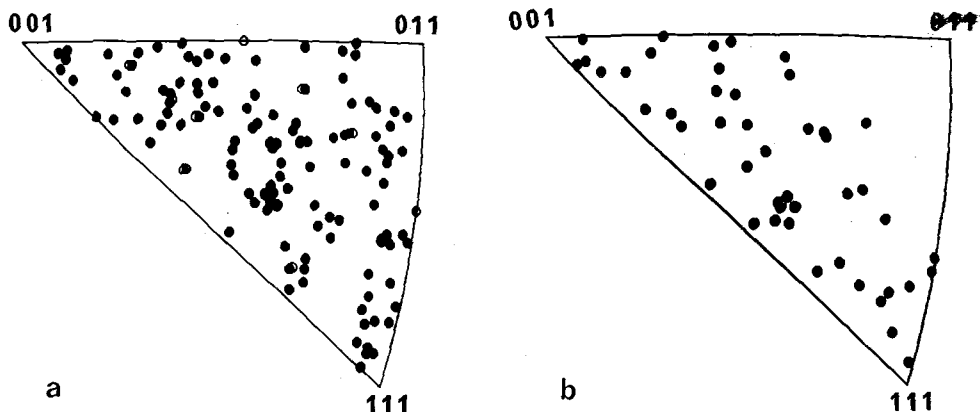


Fig. 2 - Distribution of the GB plane orientation in "oxidized NiO" (a) and "annealed NiO" (b). The absence of {011} GB plane, already noticeable in (a) is strongly enhanced after annealing (b).

As a conclusion it seems that an interface corresponding to a high angle GB, close to a {011} plane, indexed with respect to one of the adjacent crystals, is particularly unstable, as shown by the evolution of the distribution of the GB's during annealing. Although the reason of this tendency is not yet understood it appears clearly that the GB's, in the "1100°C oxidized Ni", are far from being in the equilibrium state.

IV.2. Grain boundary morphology : the morphology of the various examined interfaces depends on the origin of the NiO and on the thermal treatment to which it was subjected. Facetting does not occur in the high angle misorientation GB's, whatever the origin and the thermal treatment of the NiO is. Conversely, the morphology of the SGB's is strongly dependent on the "history" of the material. In the "1100°C oxidized Ni", SGB's are not faceted, but in the "annealed NiO", as well as in the flame fusion crystals, facets are systematically observed. They are mostly made of large portions of {011} and {112} planes and of {113} and {122} ledges. It can be noted that these latter planes correspond to cusps in fig. 1.b, that is, to energetic minima of some GB's.

These observations can be interpreted as follows. A GB can be either plane (on a large scale), or faceted. Facetting has either a kinetic origin (e.g. when the GB is far from thermodynamical equilibrium, during a fast grain growth) or an energetic origin (e.g. when it is closely approximating an equilibrium state) /17/. Close to equilibrium, which occurs for example after a long annealing at high temperature ("annealed NiO"), a GB can reduce its surface energy by facetting along the lowest energy crystallographic planes compatible with its average orientation. Hence, the possible evolution of the morphology is another criterion which may suggest that GB's are not in an intrinsic structural equilibrium. However facetting will take place only if the surface energy of the new facets is low enough to compensate for the increased surface. This is the reason why high angle GB's do not facet easily.

V - CONCLUSION

Data relative to intergranular diffusion studies in polycrystalline oxides are not in good agreement /1,18/ and the mechanism of GB diffusion is not yet well understood. Wuench and Wasilos /19/ and Osenbach and Stubican /20/, for example, suggest that enhanced diffusion of cations along GB's in MgO may be attributed to impurity segregation at the boundary. The diffusion discrepancies could be explained in a similar way since the "1100°C oxidized Ni" is likely the less pure of the materials under investigation, though Atkinson et al. have pointed out that there was no precipitate at the GB's.

(3) - The sample texture, determined on a limited number of crystals by TEM, was shown after an additional X-Ray study, to be representative.

However, the comparison of the characteristics of the GB's shows that the "1100°C oxidized Ni" is far from equilibrium and conversely, the "annealed NiO" looks quite similar to the flame-fusion crystals, where the GB's seem to be almost at equilibrium. Since the properties of a GB are closely related to its structure, it is not unreasonable to think that the preferential GB diffusion observed in the "1100°C oxidized Ni" is largely due to the fact that the grain boundaries are far from equilibrium.

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