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Fine-grained magnetoelectric $\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6\text{-CoFe}_2\text{O}_4$ composites synthesized by a straightforward one-pot method

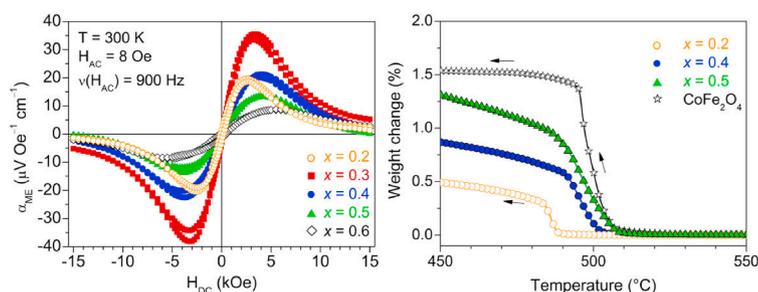
Roberto Köferstein^{*}, Florian Oehler, Stefan G. Ebbinghaus

Institute of Chemistry, Martin Luther University Halle-Wittenberg, Kurt-Mothes-Straße 2, 06120, Halle, Germany

HIGHLIGHTS

- Straightforward synthesis for 0–3 $\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6\text{-CoFe}_2\text{O}_4$ composites.
- Monitoring the phase evolution during sintering to ceramic bodies.
- Determination of Curie temperature.
- Impedance spectroscopy up to 200 °C and 10 MHz.
- Investigations of the magnetoelectric coefficient in dependence on magnetic DC field, frequency, and temperature.

GRAPHICAL ABSTRACT



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ABSTRACT

Magnetoelectric $(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{1-x}\text{-(CoFe}_2\text{O}_4)_x$ ($x = 0.2\text{--}0.6$) composites were prepared by a one-pot soft-chemistry synthesis using PEG400. Calcining at 700 °C resulted in nanocrystalline composite powders ($d_{\text{cryst.}} = 24\text{--}30$ nm) which were sintered between 1050 and 1200 °C to ceramic bodies with relative densities up to 98%. SEM investigations confirm the formation of composite ceramics with a 0–3 connectivity and variable grain sizes from 0.2 to 3.6 μm for sintering up to 1150 °C, while sintering at 1200 °C leads both to a change in the microstructure and a considerable grain growth. Magnetic measurements at 300 K reveal ferrimagnetic behaviour with saturation magnetization values smaller than bulk CoFe_2O_4 and coercivities between 790 and 160 Oe. Temperature-dependent impedance spectroscopy showed that the relative permittivities decrease both with rising frequency and CoFe_2O_4 fraction. The frequency dependence of the impedance can be well described using a single RC circuit. Magnetoelectric measurements show the presence of pronounced field hystereses. The maximum magnetoelectric coefficient (α_{ME}) depends both on the CoFe_2O_4 fraction (x) and sintering temperature. The composite with $x = 0.3$ exhibits the largest α_{ME} value of $37 \mu\text{V Oe}^{-1} \text{cm}^{-1}$ (@ 900 Hz). With rising frequency of the AC driving field α_{ME} increases up to 300–400 Hz and is nearly constant until 1 kHz.

1. Introduction

Multiferroics, i.e. materials, which exhibit at least two ferroic orders (e.g. ferromagnetism, ferroelectricity, and ferroelasticity) are of special

interest because of their potential applications as e.g. memories, actuators, spintronics, and sensors [1,2], because the coupling of order parameters results in new functionalities. In particular, the coupling of ferro-/ferrimagnetic and ferroelectric phases leads to the so-called

^{*} Corresponding author.

E-mail address: roberto.koefenstein@chemie.uni-halle.de (R. Köferstein).

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magnetoelectric (ME) effect. The ME effect, first experimentally verified by Astrov in 1960 [3], allows to change the electrical polarization by applying a magnetic field (direct ME effect), or the magnetization by an electric field (converse ME effect). Single-phase ME materials like Cr_2O_3 , BiFeO_3 , or BiMnO_3 often show a rather small ME effect and/or it appears only at low temperatures [4–6]. In contrast, multiferroic composites, consisting of a ferroelectric and a ferrimagnetic compound show strong ME effects at room temperatures which is mandatory for practical applications [7]. In such composite materials the coupling between the ferroic phases is mediated through the interface. Magnetoelectric composites can be assembled in various connectivities such as 0–3, 2–2, 1–3, and 3–3 geometries, in which the numbers reflect the dimension in which each phase is self-connected [8]. Composites based on e.g. BaTiO_3 or PbTiO_3 as ferroelectric phase and ferrites as ferrimagnetic one have been extensively investigated [9–15]. In contrast, magnetoelectric composites containing $\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6$ (SBN) are by far less well examined [16–26]. $\text{Sr}_x\text{Ba}_{1-x}\text{Nb}_2\text{O}_6$ ($0.25 \leq x \leq 0.75$) is a lead-free relaxor ferroelectric with useful piezoelectric properties and crystallizes in the open tungsten-bronze structure [27–31]. The Curie temperature strongly depends on the Sr/Ba ratio and is about 120 °C for the composition $\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6$ [28]. To prepare $\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6$ - MFe_2O_4 composites mainly the conventional mixed oxide method has been used [16–25], whereas Hoyos et al. [26] synthesized $\text{Sr}_{0.61}\text{Ba}_{0.39}\text{Nb}_2\text{O}_6$ - NiFe_2O_4 composites by a Pechini-like soft-chemistry route. Usually, soft-chemistry routes, e.g. one-pot syntheses, lead to fine-grained powders which enable lower sintering temperatures resulting in ceramic bodies with tunable grain sizes [32,33].

In this paper, we describe a straightforward one-pot soft-chemistry synthesis route using PEG400 to synthesize $\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6$ - CoFe_2O_4 composites with a 0–3 connectivity and with different CoFe_2O_4 molar fractions. The first step of the synthesis leads to nanoscaled powders, which are sintered to fine-grained composite ceramics. Phase formation during calcination and sintering were monitored by XRD. Measurements of the magnetoelectric coupling were carried out depending on the magnetic DC field, frequency of the superimposed AC driving field, and temperature. Furthermore, the composite samples were characterized by magnetic measurements, impedance spectroscopy, and SEM.

2. Experimental

2.1. Material preparation

$(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{1-x}-(\text{CoFe}_2\text{O}_4)_x$ (x = molar fraction) composites with $x = 0.2, 0.3, 0.4, 0.5$, and 0.6 were synthesized according to a polymerization method as described elsewhere [28].

For each composition 0.01 mol NbCl_5 (Alfa Aesar, 99%) was dissolved in 15 ml 2-methoxyethanol. Afterwards, 30 ml polyethylene glycol (PEG400) and 26 g citric acid were added and the mixture was stirred on a heating plate at 150 °C until it became a clear solution. $\text{SrCl}_2 \cdot 6\text{H}_2\text{O}$ (Merck, $\geq 99\%$) and $\text{BaCl}_2 \cdot 2\text{H}_2\text{O}$ (Fluka, $\geq 99\%$) (0.0025 mol each) were dissolved together in 7 ml 1,2-ethanediol and added to the hot Nb-solution. The resulting solution was stirred on the heating plate at 180 °C until it turned into a white suspension. According to the desired composition, stoichiometric amounts of $\text{Co}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$ (Sigma-Aldrich, ACS) and $\text{Fe}(\text{NO}_3)_3 \cdot 9\text{H}_2\text{O}$ (Merck, ACS) ($n_{\text{Co}}/n_{\text{Fe}} = 1/2$) were dissolved together in 5–10 ml 1,2-ethanediol and added to the (Sr,Ba,Nb)-suspension. The resulting brown reaction mixture was further stirred on the heating plate at 180 °C for about 3 h until it turned into a green, highly viscous suspension. The resulting (Sr,Ba,Nb,Co,Fe)-mixtures were calcined at 700 °C for 2 h (heating rate 10 K min^{-1}) in a muffle furnace in static air leading to $(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{1-x}-(\text{CoFe}_2\text{O}_4)_x$ nano-powders. Ceramic bodies were obtained from these powders after mixing with 10 wt% of a saturated aqueous polyvinyl alcohol (PVA) solution as pressing aid and uniaxial pressing into pellets at about 85 MPa (green density: 2.2 g cm^{-3}). These pellets were placed on a ZrO_2 fibre mat and sintered to ceramic bodies between 1050 and 1200 °C

(heating-/cooling rate 5 K min^{-1}) with a soaking time of 1 h.

The chemical compositions of the composite powders, checked by XRF, agreed well with the expected values (Tab. S1, supporting information). In all samples, we found traces of 0.12–0.16 wt% CaO, which stem mainly from the strontium source.

For comparative purposes, pure CoFe_2O_4 was also prepared according to the described synthesis route.

2.2. Characterization

X-ray powder diffraction patterns were recorded at room temperature on a Bruker D8-Advance diffractometer, equipped with a one-dimensional silicon strip detector (LynxEye™) using Cu-K_α radiation and a counting time of 1 s per data point. Crystallite sizes and strain parameters were calculated from the XRD line broadening (integral peak breadth) using the software suite WinXPOW [34] applying the Scherrer and Wilson equations. Thermal analyses were performed with a heating-/cooling rate of 10 K min^{-1} in flowing nitrogen (75 ml min^{-1}) using a TA Instruments TGA 550 (weighing precision 0.01%). To determine the magnet transition temperature of CoFe_2O_4 , a bar magnet was placed underneath the balance [35]. The specific surface area (BET) was determined by nitrogen five-point gasphysisorption (Nova touch 2LX, Quantachrome Corporation). The equivalent BET particle diameter was calculated assuming a spherical or cubic particle shape. XRF measurements were carried out with a Malvern Panalytical Epsilon 4. Scanning electron microscope images were recorded with a Phenom ProX SEM in the backscattered electron mode (BSE). Magnetic and magnetoelectric measurements were carried out using a Quantum Design PPMS9. Magnetic hysteresis loops were taken at 300 K with magnetic DC field cycling between –90 and +90 kOe. The samples were enclosed in gel capsules whose very small diamagnetic contribution was subtracted before data evaluation. For magnetoelectric and impedance investigations, ceramic bodies with a thickness of 1.2–1.4 mm were sputtered on both sides with 100 nm thick gold electrodes using a Cressington Sputter Coater 108auto. The samples for magnetoelectric measurements were electrically poled for 18 h at room temperature applying an electric field of about 6 kV cm^{-1} with a current limit of 0.1 mA [36]. Afterwards, the poled samples were short-circuited for 5 min. The magnetoelectric measurements were performed using a self-made setup [37] with the magnetic DC field parallel to the electrical polarization and a small AC driving field of about 8 Oe was superimposed collinear to the static field by a solenoid. The in-phase voltage (U_{ME}) was recorded using a lock-in technique. The magnetoelectric coefficient (α_{ME}) was calculated as $\alpha_{\text{ME}} = U_{\text{ME}} \cdot (\text{H}_{\text{AC}} \cdot d)^{-1}$ with d being the sample thickness. The magnetoelectric behavior was investigated at 300 K using a DC field cycling between ± 15 kOe and $\nu(\text{H}_{\text{AC}}) = 900$ Hz. The dependence of the magnetoelectric coefficient on the AC field was measured between 20 and 1000 Hz. The temperature dependence of α_{ME} was measured between 50 and 300 K at $\nu(\text{H}_{\text{AC}}) = 900$ Hz. Both the frequency- and temperature-dependent measurements were done at the DC field at which the maximum of α_{ME} was found. An Impedance Analyzer 4192A (Hewlett Packard) was used for frequency- and temperature-dependent impedance measurements in ranges of 25–220 °C and 5 Hz – 10 MHz applying a voltage of 1 V.

3. Results and discussion

3.1. Synthesis and powder characterization

$(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{1-x}-(\text{CoFe}_2\text{O}_4)_x$ composites with $x = 0.2, 0.3, 0.4, 0.5$, and 0.6 , which corresponds to a CoFe_2O_4 weight fraction between 13.0 and 47.1%, were successfully prepared by the one-pot synthesis as described above. To obtain brown nanoscaled composite powders, the green viscous (Sr,Ba,Nb,Co,Fe)-solutions were calcined at 700 °C in a muffle furnace in static air (heating rate 5 K min^{-1} , soaking time 2 h). The XRD patterns (Fig. 1) of all samples show reflections corresponding

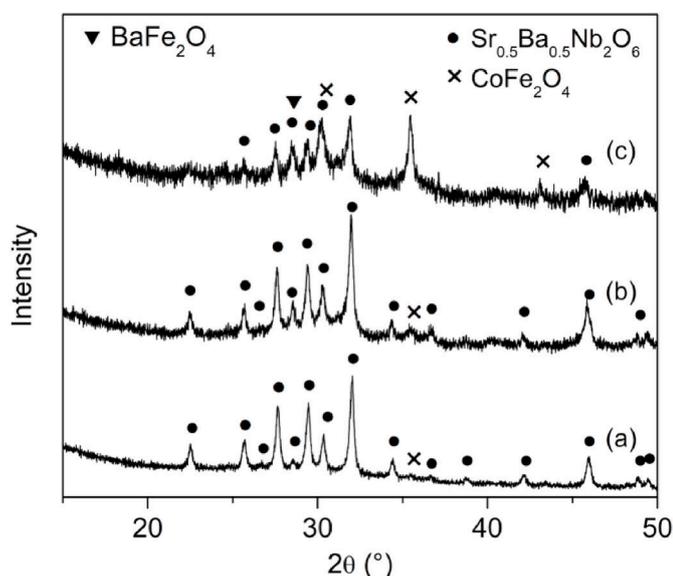


Fig. 1. Room-temperature XRD patterns of selected $(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{1-x}-(\text{CoFe}_2\text{O}_4)_x$ composite powders after calcining at 700°C for 2 h (heating rate 10 K min^{-1}): a) $x = 0.2$, b) $x = 0.4$, c) $x = 0.6$.

to tetragonal $\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6$ (JCPDS #01-074-6520). XRDs for $x = 0.2$, 0.3 , and 0.4 show only a weak reflection of cubic CoFe_2O_4 (JCPDS #00-022-1086), whereas for $x = 0.5$ and 0.6 the formation of crystalline CoFe_2O_4 can be clearly seen. Additionally, an increasing peak intensity at $2\theta \approx 28.5^\circ$ for composite powders with $x = 0.5$ and 0.6 points to the formation of orthorhombic BaFe_2O_4 (JCPDS #00-025-1191) as an intermediate phase, which disappears after calcining at 800°C (Fig. S1, supporting information). A slightly increasing background level primarily between $2\theta = 20\text{--}30^\circ$ suggests an amorphous fraction in all samples.

The specific surface areas of the $(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{1-x}-(\text{CoFe}_2\text{O}_4)_x$ powders, derived from BET measurements, decrease with increasing CoFe_2O_4 content from $31(3)$ to $21(2)\text{ m}^2\text{ g}^{-1}$ corresponding to equivalent primary particle sizes of $36(3)$ and $53(2)\text{ nm}$ for $x = 0.2$ and $x = 0.6$, respectively (see Fig. 2). The volume-weighted average crystallite size

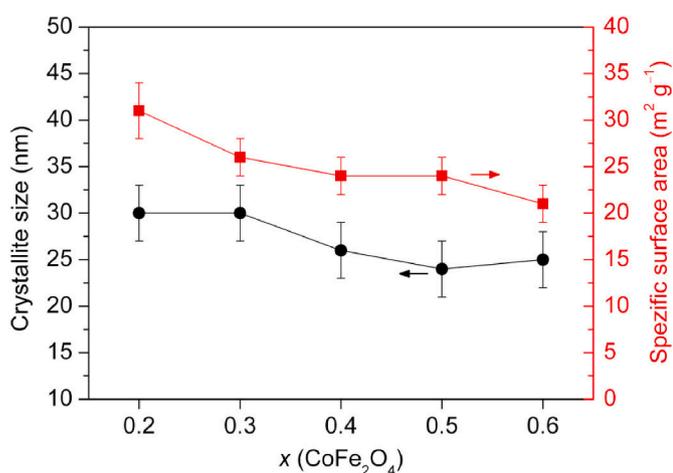


Fig. 2. Development of the crystallite size of $\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6$ and specific surface area depending on the fraction of CoFe_2O_4 of $(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{1-x}-(\text{CoFe}_2\text{O}_4)_x$ composite powders calcined at 700°C for 2 h (heating rate 10 K min^{-1}).

(size of a coherent scattering domain) obtained from the XRD peaks of the $\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6$ phase is between $30(3)$ and $24(3)\text{ nm}$. The difference between the crystallite size and the primary particle size is most likely caused by the formation of closely joined crystallites with surface areas unavailable for nitrogen adsorption [38]. Furthermore, each primary particle may consist of more than one crystallite domain.

3.2. Sintering, microstructure, and phase composition of ceramic bodies

After pressing the nanocrystalline composite powders into pellets and sintering at temperatures between 1050 and 1200°C for 1 h (heating-/cooling rate: 5 K min^{-1}) black-brown ceramic bodies were obtained. Their bulk densities were calculated from the weight and geometric dimension and related to the single crystal densities of $\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6$ and CoFe_2O_4 considering their nominal molar fractions [39]. As shown in Fig. 3, the densities increase between 1050 and 1150°C . Dense ceramic bodies (relative density $\geq 90\%$) were obtained after sintering above 1075°C . Firing up to 1150°C leads to ceramic bodies with relative densities of $98(1)\%$. Upon further increasing the sintering temperature to 1200°C the densities of most samples did not change, however, for $x = 0.5$ and 0.6 the densities slightly decreased to $92(1)\%$ due to the change in the microstructure as described below.

Representative microstructures of the surface of the composite ceramics are shown in Fig. 4. In the applied BSE mode, the dark grains correspond to CoFe_2O_4 while the light grey ones are $\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6$. Upon sintering at 1150°C the $\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6$ phase forms globular- or irregular-shaped particles, while the CoFe_2O_4 phase consists of pyramidal- and octahedral-like grains as shown in Fig. 4a and b for $x = 0.3$ and 0.6 , respectively. Sintering at 1200°C leads to a change of the microstructure as can be seen in Fig. 4c and d. At this temperature $\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6$ (SBN) forms predominantly small pillars besides some globular-like grains. The fraction of the globular-like grains of SBN decreases with increasing CoFe_2O_4 fraction (x) which is most-likely the reason for reduced bulk densities at $x = 0.5$ and 0.6 . Similar observations were done in pure SBN ceramics [28]. The CoFe_2O_4 (CFO) particles retain their octahedron-/pyramidal-like shape. The grain diameters range between about 0.2 and $1.3\text{ }\mu\text{m}$ (SBN) and $0.3\text{--}1.1\text{ }\mu\text{m}$ (CFO) after sintering at 1075°C and increase to $0.3\text{--}1.6\text{ }\mu\text{m}$ (SBN)/ $0.6\text{--}2.6\text{ }\mu\text{m}$ (CFO) and $0.5\text{--}3.6\text{ }\mu\text{m}$ (SBN)/ $0.6\text{--}3.4\text{ }\mu\text{m}$ (CFO) after 1100°C and 1150°C ,

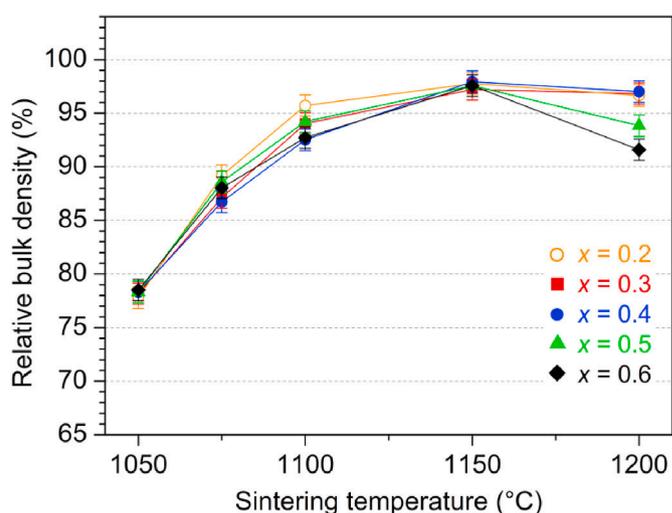


Fig. 3. Relative bulk densities of $(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{1-x}-(\text{CoFe}_2\text{O}_4)_x$ composite ceramic bodies versus sintering temperatures (soaking time 1 h, heating rate 5 K min^{-1}).

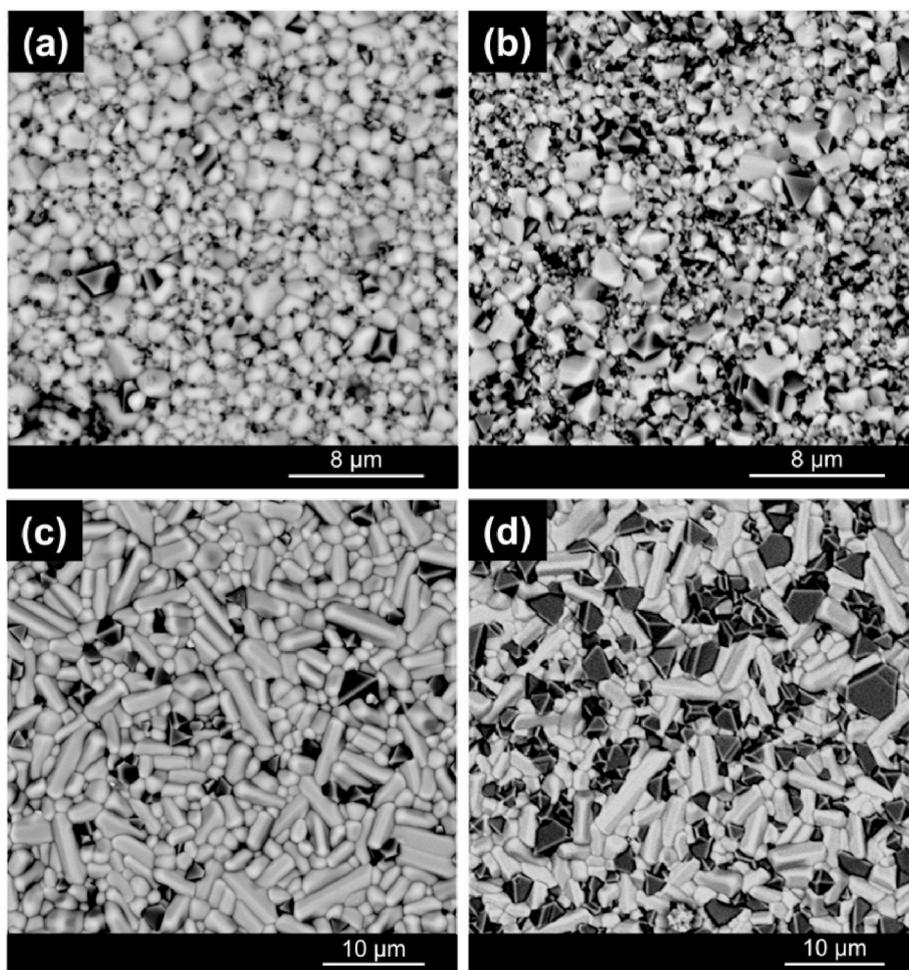


Fig. 4. SEM-BSE images of selected $(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{1-x}-(\text{CoFe}_2\text{O}_4)_x$ composite ceramics sintered at different temperatures (T_s) for 1 h. (a) $x = 0.3$, $T_s = 1150$ °C, (b) $x = 0.6$, $T_s = 1150$ °C, (c) $x = 0.3$, $T_s = 1200$ °C, (d) $x = 0.6$, $T_s = 1200$ °C.

respectively. Firing at 1200 °C SBN forms pillars with lengths in the range of about 2–14 μm and widths between 1.5 and 3 μm corresponding to an aspect ratio between 2 and 6. Besides the pillars, the sizes of the globular-like grains are between 1 and 5 μm and the octahedral-like CoFe_2O_4 grains are between 1.5 and 6 μm . The grain sizes of all samples are listed in Tab. S2 (supporting information). The grain sizes of both components considerably rise with sintering temperature. With increasing CoFe_2O_4 fraction (x) we observed a tendency of increasing sizes of the CoFe_2O_4 grains while the SBN particles are not affected by x .

Powder XRD patterns of selected sintered composite ceramics are shown in Fig. 5. All patterns show strong and sharp reflections of the target phases $\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6$ and CoFe_2O_4 . Sintering between 1050 and 1150 °C leads to small amounts of monoclinic FeNbO_4 (JCPDS #01-071-1849) with a fraction of about 3–4 wt%. In contrast, after sintering at 1200 °C, the FeNbO_4 phase vanished, while the formation of 2–3 wt% tetragonal CoNb_2O_6 (JCPDS #01-075-2340) was observed. The appearance of secondary phases is caused by interdiffusion processes between $\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6$ and CoFe_2O_4 . The amounts of both secondary phases do not change significantly with the fraction of CoFe_2O_4 . In contrast, a Pechini-like synthesis by Hoyos et al. [26] with 1,2-ethanediol at pH = 9 led to the formation of secondary phases up to 17 wt%. Thus, our synthesis route leads to composite ceramics with much higher purity.

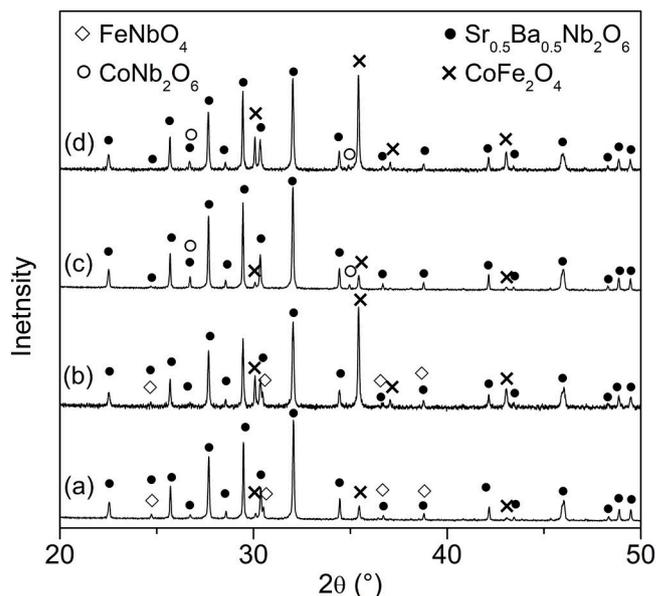


Fig. 5. Room-temperature powder XRD patterns of crushed $(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{1-x}-(\text{CoFe}_2\text{O}_4)_x$ composite ceramics after sintering at different temperatures (T_s) for 1 h (heating-/cooling rate 5 K min^{-1}): a) $x = 0.2$, $T_s = 1150$ °C, b) $x = 0.6$, $T_s = 1150$ °C, c) $x = 0.2$, $T_s = 1200$ °C, d) $x = 0.6$, $T_s = 1200$ °C.

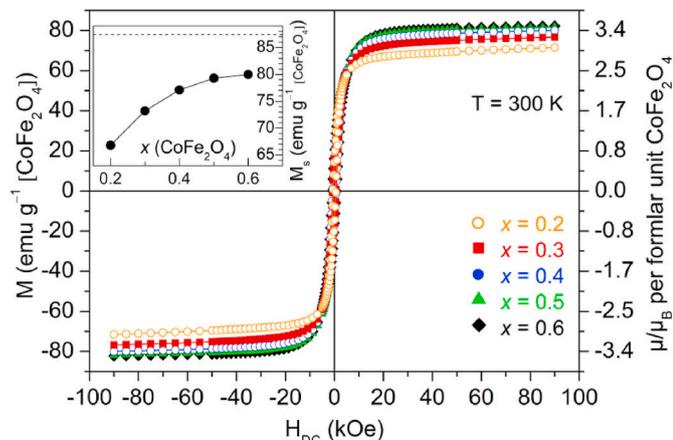


Fig. 6. Field dependence of the magnetization at 300 K for composite ceramics sintered at 1100 °C for 1 h. The inset shows the saturation magnetization (M_s) depending on the CoFe_2O_4 fraction (x). The horizontal line represents the value of M_s measured for pure CoFe_2O_4 . The uncertainties of M_s ($\leq 0.06 \text{ emu g}^{-1}$) are smaller than the symbol sizes. The magnetization is given with respect to the nominal CoFe_2O_4 mass.

3.3. Magnetic behavior

Fig. 6 exemplarily shows the field-dependent magnetization at 300 K of ceramics sintered at 1100 °C. Data have been normalized with respect to the CoFe_2O_4 fraction. All samples show hystereses due to the ferrimagnetic nature of CoFe_2O_4 . The saturation magnetizations (M_s) for the composite samples were calculated by extrapolating the magnetization at high field to $H = 0$ because of the presence of small amounts of paramagnetic secondary phases (FeNbO_4 , CoNb_2O_6), as aforementioned [40–43]. The M_s values of the $(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{1-x}(\text{CoFe}_2\text{O}_4)_x$ ceramics sintered at 1100 °C increase with rising CoFe_2O_4 content from 66.8(6) for $x = 0.2$ to 80.0(2) emu g^{-1} for $x = 0.6$ (inset in Fig. 6). The same trend was found for composite ceramics sintered at other temperatures (see Fig. S2, supporting information).

The difference between the calculated M_s values of the composites and the M_s value of 87.4(1) emu g^{-1} (3.67(1) $\mu_B \text{ f.u.}^{-1}$) of pure CoFe_2O_4 (sintered at 1100 °C) is most likely due to the formation of nonferro-/nonferrimagnetic FeNbO_4 and CoNb_2O_6 secondary phases [41–43] and the observed grain growth of the CoFe_2O_4 grains with increasing x . Additionally, the shape of the grains may affect M_s . As mentioned above, the CoFe_2O_4 grains in the composites have a pyramidal-/octahedral-like shape, while the grains in the CoFe_2O_4 bulk ceramic consists of globular-like grains. Shen et al [44] observed in micro-sized NiFe_2O_4 with octahedral grains a reduced saturation magnetization compared to the bulk value. The decrease of M_s with decreasing CoFe_2O_4 content can be explained on basis of the mentioned impurities, because the fraction of these paramagnetic, secondary phases is nearly independent on the nominal CoFe_2O_4 content. Its relative effect is larger for lower values of x . The M_s values (Fig. 7) of the composites vary slightly with sintering temperature due to increasing CoFe_2O_4 grain sizes as demonstrated for the sample with $x = 0.3$ (inset I in Fig. 7). In contrast, the coercivities (H_c) strongly decrease with sintering temperature, because of grain growth of the ferrite phase [45–47], whereas the dependence on the CoFe_2O_4 content is marginal (inset II in Fig. 7).

To determine the paramagnetic \rightleftharpoons ferrimagnetic transition temperature T_C (Curie-temperature) of the CoFe_2O_4 phase in the composites we performed thermogravimetric measurements in which a magnet was attached under the furnace of the thermobalance. In the ferrimagnetic temperature region, the magnet applies an extra attractive force on the hanging sample, resulting in an apparent higher weight. Thus, during cooling down from 650 °C (paramagnetic state) the weight of the samples remains almost constant until T_C is reached, where a sudden

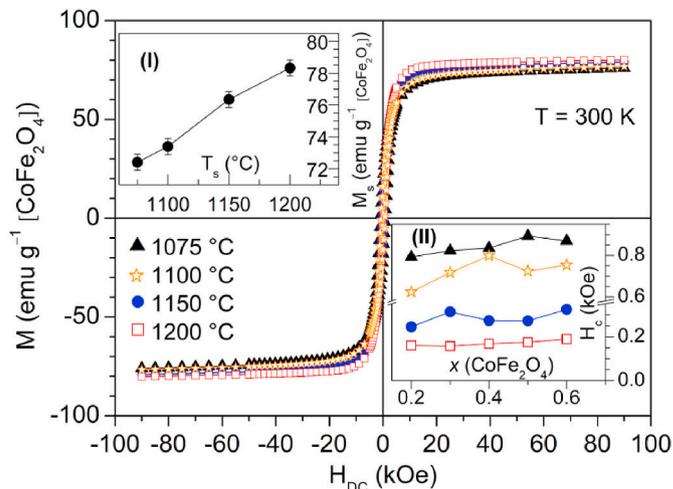


Fig. 7. Magnetization (M) versus applied field (H_{DC}) at 300 K of $(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{0.7}(\text{CoFe}_2\text{O}_4)_{0.3}$ composite bodies sintered at the indicated temperatures for 1 h. Inset (I) shows the saturation magnetization (M_s) depending on the sintering temperature (T_s). Inset (II) shows the coercivities as a function of the CoFe_2O_4 fraction (x) for different sintering temperatures. Uncertainties of the coercivities values ($\leq 20 \text{ Oe}$) are smaller than the symbol sizes. The magnetization is given with respect to the nominal CoFe_2O_4 mass.

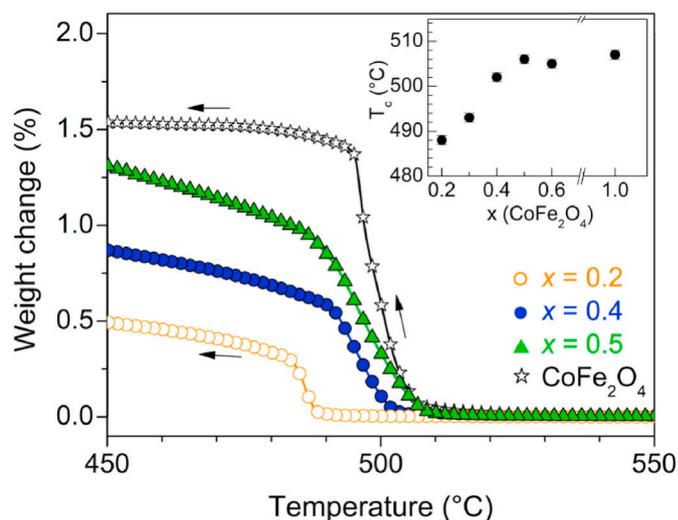


Fig. 8. Thermogravimetric investigations in flowing nitrogen (cooling rate 10 K min^{-1}) in a magnetic field for selected $(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{1-x}(\text{CoFe}_2\text{O}_4)_x$ and pure CoFe_2O_4 ceramic bodies sintered at 1100 °C for 1 h. For the sake of clarity, only every 20th data point is shown as a symbol. The inset shows T_c depending on the CoFe_2O_4 fraction (x).

increase in weight is observed. T_C was determined from the onset of the weight increase. As demonstrated for selected samples in Fig. 8, the Curie temperature increases slightly with increasing CoFe_2O_4 content and amount to 488(1), 493(1), 502(1), 506(1), and 505(1) °C for $x = 0.2, 0.3, 0.4, 0.5, 0.6$, respectively, while for pure CoFe_2O_4 ceramics T_C was determined as 507(1) °C. We suppose, the decrease of T_C at lower CoFe_2O_4 fraction is due to interdiffusion processes resulting in the formation of CoFe_2O_4 grain boundaries containing Sr/Ba/Nb-ions and/or vacancies [48,49]. Gashemi [48] found a reduced Curie temperature in Sr-doped CoFe_2O_4 with rising doping level. As mentioned above, related to the whole CoFe_2O_4 fraction the amount of such doped grain boundaries increases with decreasing ferrite content.

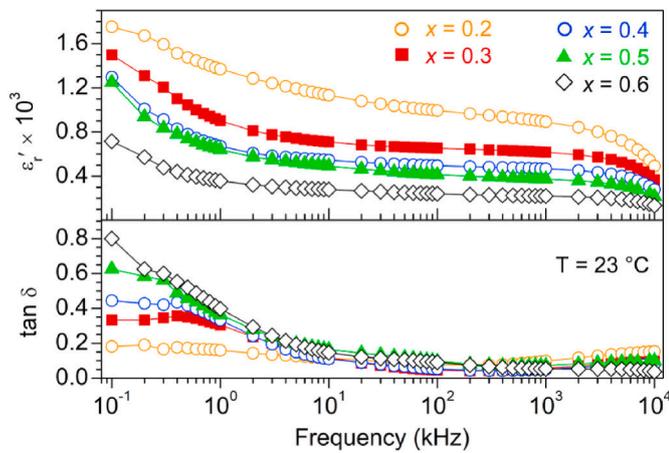


Fig. 9. Frequency dependence of the real part of the permittivity (ϵ_r') and $\tan \delta$ at room temperature of $(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{1-x}-(\text{CoFe}_2\text{O}_4)_x$ ceramic bodies sintered at 1100 °C.

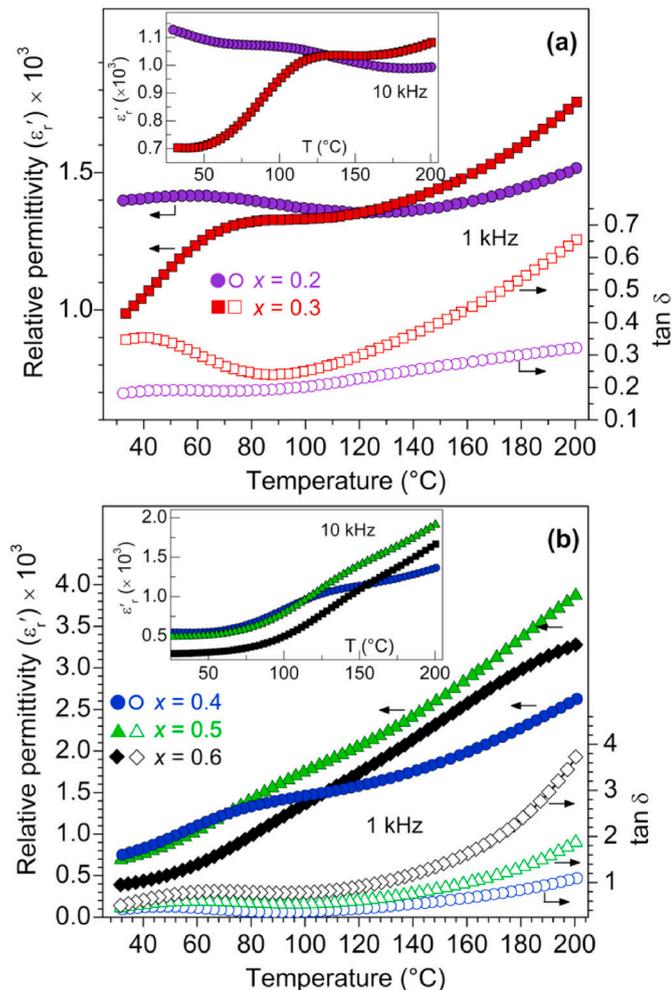


Fig. 10. Temperature dependence of the real part of the permittivity (ϵ_r' , closed symbols) and $\tan \delta$ (open symbols) at 1 kHz of $(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{1-x}-(\text{CoFe}_2\text{O}_4)_x$ ceramic bodies sintered at 1100 °C. (a) $x = 0.2$ and 0.3 , (b) $x = 0.4, 0.5, 0.6$. The insets show the temperature dependence of ϵ_r' at 10 kHz. For the sake of clarity only every third data point is represented by a symbol.

3.4. Impedance spectroscopy

Impedance spectroscopy was carried out on composite ceramics sintered at 1100 °C. The development of the real part of the relative permittivity (ϵ_r') and dissipation factor ($\tan \delta$) with the applied frequency at room temperature is shown in Fig. 9.

All composites show slightly decreasing permittivities and $\tan \delta$ values with rising frequency, with a strong decrease of ϵ_r' above about 2 MHz. This behavior is most likely due to the Maxwell–Wagner interface polarization caused by electrically conductive particles (CoFe_2O_4) embedded in an insulating matrix ($\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6$) [50,51]. Additionally, ϵ_r' becomes smaller with rising CoFe_2O_4 fraction which is typical for ferrite–ferroelectric composites [52–55]. Simultaneously, $\tan \delta$ increases with rising CoFe_2O_4 content reflecting the electrically leaky nature of ferrites (see Fig. 9). In Fig. 10 the temperature dependence of ϵ_r' and $\tan \delta$ at a frequency of 1 kHz are shown. The permittivity values of $(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{0.8}-(\text{CoFe}_2\text{O}_4)_{0.2}$, which is the most insulating sample even at high temperatures, vary only slightly up to a temperature of 200 °C. The samples with CoFe_2O_4 contents of $x = 0.3$ and 0.4 show a broad shoulder roughly between 80 and 130 °C, whereas the permittivity values of composites with $x = 0.5$ and 0.6 increase continuously with temperature and reach values of 3300 and 3900 at 200 °C, respectively. Rising permittivity values at higher temperatures go along with an increase in the loss tangents, most likely due to an increasing conductivity of these samples. A clear ferroelectric \rightleftharpoons paraelectric phase transition maxima of the $\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6$ phase could not be observed. However, at 1 kHz the samples with ferrite contents of $x = 0.2, 0.3$, and 0.4 show the formation of a weak shoulder below 130 °C, which may be attributed to the ferroelectric \rightleftharpoons paraelectric phase transition. The shoulder is more pronounced at higher frequencies (see insets in Fig. 10). In contrast, single $\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6$ shows a clear phase transition at about 120 °C (see Fig. S3, supporting information) [28,56]. However, Chen et al. [17] and Lin et al. [18] found a broad phase transition temperature below 25 °C in $\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6$ – CoFe_2O_4 composites prepared by the conventional mixed-oxide method. We suppose that the considerable broadening of the phase transition maxima is caused by doping of $\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6$ with Co^{2+} and/or Fe^{3+} . Kim et al. [57] and Kshirsagar et al. [58] observed a broadening of the phase transition in iron/cobalt-doped $\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6$. Our findings are in accordance with investigations by Rathore and Vitta [25] on $\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6$ –ferrite composites who also observed a broadening and reduction of the phase transition temperature.

The conductivity of the $(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{1-x}-(\text{CoFe}_2\text{O}_4)_x$ ceramics is rather low ($\sigma_{\text{DC}} \ll 10^{-7} \text{ S cm}^{-1}$) at room temperatures. While the ceramic with $x = 0.2$ remains insulating even at high temperatures, the composite ceramics with $x = 0.3$ – 0.6 exhibit considerable DC conductivities at elevated temperatures suggesting a semiconducting nature. Therefore, the high-temperature impedance data were modelled using an equivalent circuit consisting of one resistance-capacitor (RC) element including a constant phase shift element. The specific complex impedance (ρ^*) for a single RC element is described by:

$$\rho^* = \frac{\rho_{\text{DC}}}{1 + (i\omega\tau)^\beta} \quad (1)$$

where, β is the constant phase shift (CPE) coefficient and $\tau = \rho_{\text{DC}}\epsilon\epsilon_0$. Details of this approach have been reported elsewhere [59]. The resulting Cole-Cole plots (Fig. 11) reveal single semicircular arcs up to 220 °C, indicating that the impedance data can be well described by the applied model. Hence, bulk and grain boundary contributions on impedance cannot be distinguished, which may be the result of the high density and large grain sizes of our ceramics, reducing the effect of grain boundaries. The calculated DC resistivities (ρ_{DC}) decrease continuously with rising CoFe_2O_4 content, e.g. from $1.63(5) \cdot 10^5 \text{ k}\Omega \text{ cm}$ ($x = 0.3$) to $83(3) \text{ k}\Omega \text{ cm}$ ($x = 0.6$) at 220 °C.

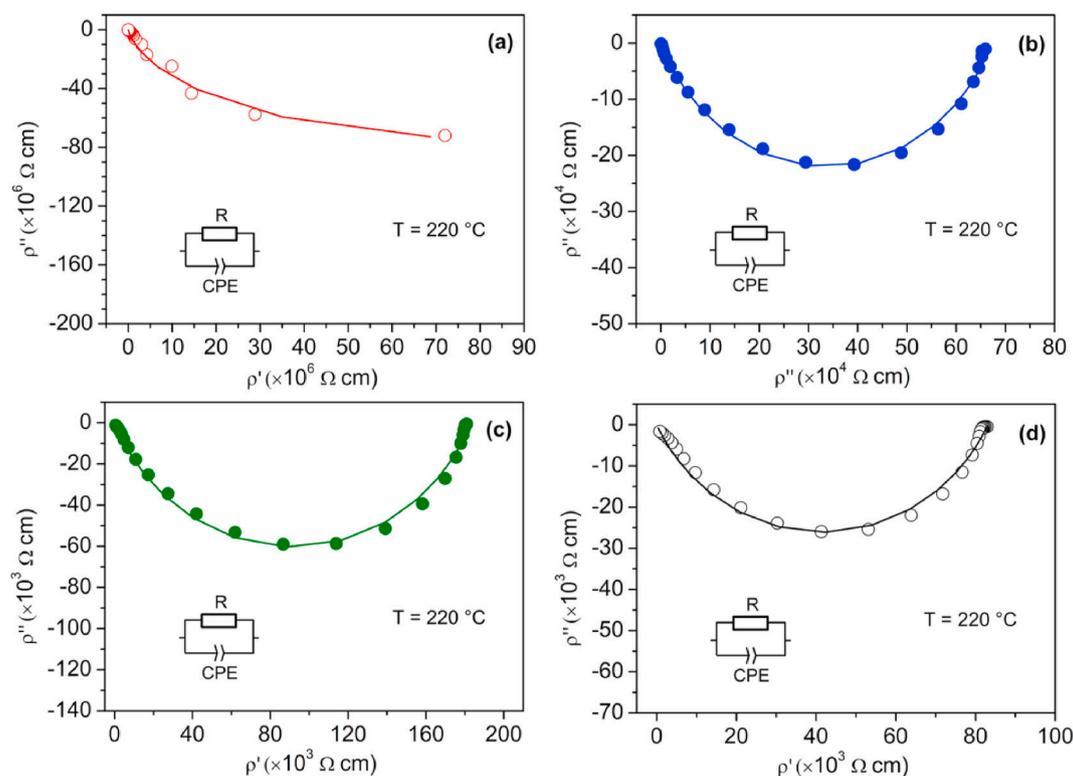


Fig. 11. Cole–Cole plots of $(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{1-x}-(\text{CoFe}_2\text{O}_4)_x$ ceramics sintered at $1100\text{ }^\circ\text{C}$. a) $x = 0.3$, b) $x = 0.4$, c) $x = 0.5$, d) $x = 0.6$.

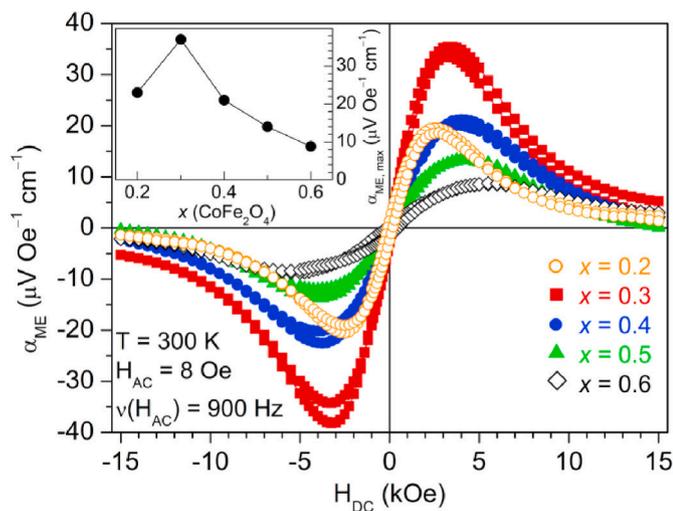


Fig. 12. Magnetolectric coefficient (α_{ME}) vs. magnetic DC field for various composite ceramics sintered at $1100\text{ }^\circ\text{C}$ for 1 h. The inset shows the maximum magnetolectric coefficient ($\alpha_{\text{ME,max}}$) depending on the CoFe_2O_4 fraction (x). The uncertainty of the data ($1\text{ }\mu\text{V Oe}^{-1}\text{ cm}^{-1}$) is smaller than the symbol size.

3.5. Magnetolectric properties

Fig. 12 exhibits the dependence of the magnetolectric coefficient (α_{ME}) at 300 K on the static magnetic field (H_{DC}) for composite ceramics sintered at $1100\text{ }^\circ\text{C}$. Composites sintered at other temperatures show analogous behaviors (Fig. S4 and S5, supporting information).

Starting from the maximum field of 15 kOe , α_{ME} increases with decreasing DC field up to a maximum ($\alpha_{\text{ME,max}}$) and almost vanishes at zero field. For negative fields an inverse behavior is observed. Upon reversing sweep direction of the H_{DC} field small hystereses are found with coercivities up to 250 Oe . The magnetic DC field at which α_{ME}

reaches its maximum ($H_{\alpha_{\text{ME,max}}}$) increases with the CoFe_2O_4 fraction in the composites, e.g. for $x = 0.2$ the maximum for α_{ME} appears at 3.0 kOe , whereas for $x = 0.6$ the maximum α_{ME} value is found at 5.5 kOe (Fig. S6, supporting information). Also, the magnetolectric coefficient depends on the CoFe_2O_4 content and shows a maximum of $\alpha_{\text{ME}} = 37(1)\text{ }\mu\text{V Oe}^{-1}\text{ cm}^{-1}$ ($\nu(H_{\text{AC}}) = 900\text{ Hz}$) at $x = 0.3$ while further increasing of the CoFe_2O_4 content leads to decreasing α_{ME} values (inset in Fig. 12). This behaviour is somewhat different from our results for $\text{BaTiO}_3\text{-CoFe}_2\text{O}_4$ or $\text{BaTiO}_3\text{-Ni}$ for which the maxima were found around $x = 0.4$ [36,37,60]. We assume that the decrease of $\alpha_{\text{ME,max}}$ with higher ferrite content is due to an increase of the conductivity leading to an inner discharging (see impedance section). The maximum α_{ME} value was further found to depend on the sintering temperature, e.g.

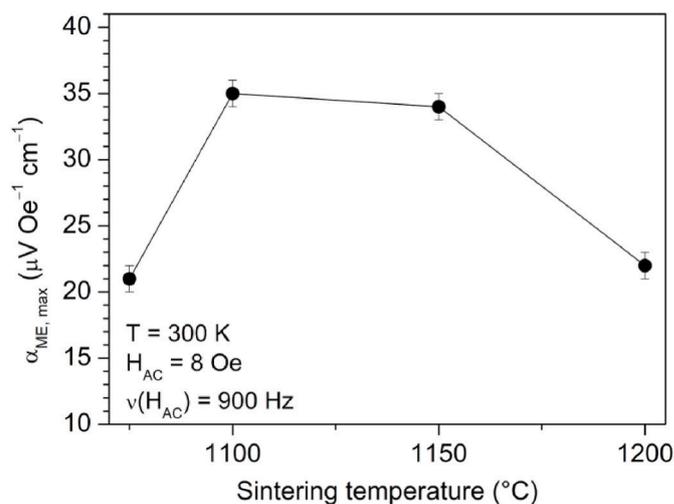


Fig. 13. Maximum magnetolectric coefficient ($\alpha_{\text{ME,max}}$) depending on the sintering temperature for $(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{0.7}-(\text{CoFe}_2\text{O}_4)_{0.3}$ composite ceramics.

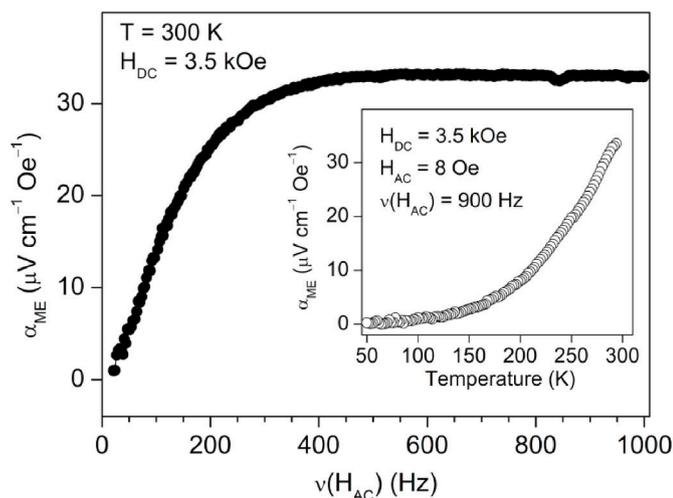


Fig. 14. Frequency dependence (H_{AC}) of α_{ME} for $(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{0.7}-(\text{CoFe}_2\text{O}_4)_{0.3}$ sintered at 1100 °C for 1 h. The inset shows the temperature dependence of α_{ME} .

$(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{0.8}-(\text{CoFe}_2\text{O}_4)_{0.2}$ shows its highest α_{ME} value after sintering at 1150 °C, whereas for composites with $x \geq 0.3$ the optimal sintering temperature is 1100 °C as demonstrated for $(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{0.7}-(\text{CoFe}_2\text{O}_4)_{0.3}$ in Fig. 13. Decreasing α_{ME} values after sintering at 1200 °C get along with an increase of conductivity of at least one order of magnitude.

The evolution of α_{ME} at H_{max} at 300 K with the frequency of the AC driving field is exemplarily shown for $x = 0.3$ in Fig. 14. For all composites we observe a strong increase of the magnetoelectric coefficient up to 300–400 Hz, while from about 400 Hz to 1000 Hz α_{ME} does not change significantly. The increase of α_{ME} at low frequencies can be explained by a partial internal discharge due to the conductivity of CoFe_2O_4 [61]. Furthermore, Bichurin and Petrov [62] showed that in 0–3 composites charge accumulation lead to an inverse Maxwell-Wagner-type relaxation leading to a charging of the capacitance which increases with frequency and saturates at higher frequencies. A similar behaviour was also found in $\text{BaTiO}_3-\text{CoFe}_2\text{O}_4$, BaTiO_3-Ni and $\text{Pb}(\text{Mg}_{1/3}\text{Nb}_{2/3})-\text{PbTiO}_3-\text{CoFe}_2\text{O}_4$ composites [36,37,60,63]. This finding is in contrast to the one of Sutar et al. [64] and Jigajeni et al. [65] who observed a continuous increase of α_{ME} up to about 5 kHz in ferrite–strontium barium niobate composites. As seen in the inset in Fig. 14 α_{ME} becomes continuously lower with decreasing temperature and below about 100 K the magnetoelectric coupling nearly vanishes. This temperature dependence of α_{ME} reflects the decreasing piezoelectric and ferroelectric character of strontium barium niobate with decreasing temperature [27,66–69] and supports the strain-mediated mechanism of the magnetoelectric effect as a tensor product property of the magnetostrictive (CoFe_2O_4) and the piezoelectric ($\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6$) components [2,14,70]. A decreasing of the magnetoelectric coupling with declining temperature has also recently been observed in $\text{BaTiO}_3-\text{CoFe}_2\text{O}_4$ and BaTiO_3-Ni composites [36,60]. For the latter system the low temperature phase transitions of BaTiO_3 could be detected in the ME signal. As $\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6$ does not possess such transitions no anomalies were observed here.

4. Conclusion

Magnetoelectric $(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{1-x}-(\text{CoFe}_2\text{O}_4)_x$ composites were synthesized by a one-pot polymerization method with PEG400. Calcining at 700 °C resulted in brown nanocrystalline composite powders with specific surface areas between 31(3) and 21(2) $\text{m}^2 \text{g}^{-1}$ and crystallite sizes of 24(3) to 30(3) nm. Sintering of the nano-powders up to 1150 °C led to ceramic bodies with relative densities up to 98(1) %

and grain sizes between 0.2 and 3.6 μm . At 1200 °C the microstructure changed significantly with the formation of pillar-like $\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6$ particles. XRD investigations showed the appearance of only small amounts (2–4 wt%) of secondary phases besides the main components $\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6$ and CoFe_2O_4 . The relative permittivities of the composites decrease with increasing CoFe_2O_4 content and frequency. Temperature-dependent measurements show increasing permittivities and $\tan \delta$ values, the impedance of the samples can be described with a single RC-equivalent circuit and the DC conductivity increases with higher CoFe_2O_4 content. Field-dependent measurements of the magnetoelectric coupling show a hysteresis of the magnetoelectric coefficient (α_{ME}) with maxima/minima at DC magnetic fields between ± 3.0 kOe and ± 5.5 kOe depending on the CoFe_2O_4 fraction. The maximum magnetoelectric coefficient of 37(1) $\mu\text{V Oe}^{-1} \text{cm}^{-1}$ (@ 900 Hz) was found in $(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{0.7}-(\text{CoFe}_2\text{O}_4)_{0.3}$ composites. For all samples, the magnetoelectric coefficient decreases at lower temperatures. Furthermore, the variation of the H_{AC} frequency results in almost constant α_{ME} values between 400 and 1000 Hz, while below 400 Hz α_{ME} becomes continuously smaller. The facile synthesis described in this article leads to the formation of magnetoelectric composite ceramics with only minor traces of secondary phases and a fine-grained microstructure compared to the conventional mixed oxide synthesis. The microstructure and thus the magnetoelectric coefficient may be tuned using sintering additives which is a subject of further investigations.

CRediT authorship contribution statement

Roberto Köferstein: Conceptualization, Data curation, Formal analysis, Investigation, Project administration, Resources, Validation, Visualization, Writing – original draft, Writing – review & editing. **Florian Oehler:** Data curation, Formal analysis, Writing – original draft. **Stefan G. Ebbinghaus:** Conceptualization, Project administration, Resources, Supervision, Validation, Writing – original draft, Funding acquisition.

Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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Appendix A. Supplementary data

Supplementary data related to this article can be found at <https://doi.org/10.1016/j.matchemphys.2021.125616>.

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Supporting Information

Fine-grained magnetoelectric $\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6\text{--CoFe}_2\text{O}_4$ composites synthesized by a straightforward one-pot method

Roberto Köferstein*, Florian Oehler, and Stefan G. Ebbinghaus
*Institute of Chemistry, Martin Luther University Halle-Wittenberg,
Kurt-Mothes-Straße 2, 06120 Halle, Germany.*

* Corresponding author

Tel.: +49-345-5525630; Fax: +49-345-5527028.

E-mail address: roberto.koefenstein@chemie.uni-halle.de

$(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{1-x}-(\text{CoFe}_2\text{O}_4)_x$ composites (x = molar fraction)

Table S1: XRF results of $(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{1-x}-(\text{CoFe}_2\text{O}_4)_x$ composite powders

molar ratio	Fe/Co	Sr/Ba	Nb/(Sr+Ba)	Nb/Fe	(Sr+Ba+Nb)/ (Co+Fe)
$(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{0.8}-(\text{CoFe}_2\text{O}_4)_{0.2}$					
expected	2	1	2	4	4
found	2.083(3)	1.071(4)	1.996(4)	3.961(4)	4.017(6)
CaO: 0.158(1) wt%					
$(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{0.7}-(\text{CoFe}_2\text{O}_4)_{0.3}$					
expected	2	1	2	2.333	2.333
found	2.074(1)	1.053(4)	1.983(4)	2.282(4)	2.316(4)
CaO: 0.143(3) wt%					
$(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{0.6}-(\text{CoFe}_2\text{O}_4)_{0.4}$					
expected	2	1	2	1.5	1.5
found	2.102(6)	1.050(4)	1.989(5)	1.503(2)	1.530(4)
CaO: 0.138(2) wt%					
$(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{0.5}-(\text{CoFe}_2\text{O}_4)_{0.5}$					
expected	2	1	2	1	1
found	2.087(2)	1.029(2)	1.967(3)	0.983(2)	1.003(2)
CaO: 0.128(7) wt%					
$(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{0.4}-(\text{CoFe}_2\text{O}_4)_{0.6}$					
expected	2	1	2	0.667	0.667
found	2.106(4)	1.035(1)	2.003(3)	0.664(1)	0.675(1)
CaO: 0.116(2) wt%					

Table S2: Grain sizes of $(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{1-x}-(\text{CoFe}_2\text{O}_4)_x$ ceramic bodies

Composition	Sintering temperature (°C)	Grain size range (μm)
x = 0.2	1075	CFO: octahedral-like: 0.3–1.1 SBN: globular-like: 0.2–1.3
	1100	CFO: octahedral-like: 0.6–1.3 SBN: globular-like: 0.3–1.3
	1150	CFO: octahedral-like: 0.6–2.5 SBN: globular-like: 0.5–3.2
	1200	CFO: octahedral-like: 1.1–4.0 SBN: globular-like: 0.9–4.5/ pillar: 1.5–3.0 × 2.0–9.0
	x = 0.3	1075
1100		CFO: octahedral-like: 0.4–1.6 SBN: globular-like: 0.3–1.3
1150		CFO: octahedral-like: 0.5–2.7 SBN: globular-like: 0.5–3.3
1200		CFO: octahedral-like: 1.0–4.2 SBN: globular-like: 1–4.5 / pillar: 1.5–3.0 × 2.0–13
x = 0.4		1075
	1100	CFO: octahedral-like: 0.7–2.1 SBN: globular-like: 0.5–1.6
	1150	CFO: octahedral-like: 0.5–3.2 SBN: globular-like: 0.7–3.3
	1200	CFO: octahedral-like: 1.4–5.1 SBN: globular-like: 1.0–5.0/ pillar: 1.2–2.9 × 2.0–14
	x = 0.5	1075
1100		CFO: octahedral-like: 0.6–2.3 SBN: globular-like: 0.4–1.5
1150		CFO: octahedral-like: 0.6–3.4 SBN: globular-like: 0.5–3.6
1200		CFO: octahedral-like: 1.5–6.0 SBN: globular-like: 1.1–5.0 / pillar: 1.4–3.0 × 2.1–14
x = 0.6		1075
	1100	CFO: octahedral-like: 0.7–2.6 SBN: globular-like: 0.5–1.6
	1150	CFO: octahedral-like: 0.9–3.4 SBN: globular-like: 0.6–3.6
	1200	CFO: octahedral-like: 1.4–6.0 SBN: globular-like: 1.3–5.0 / pillar: 1.5–3.0 × 2.0–14

CFO = CoFe_2O_4 ; SBN = $\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6$

The uncertainties of the data is about 10 % (at least 0.1 μm)

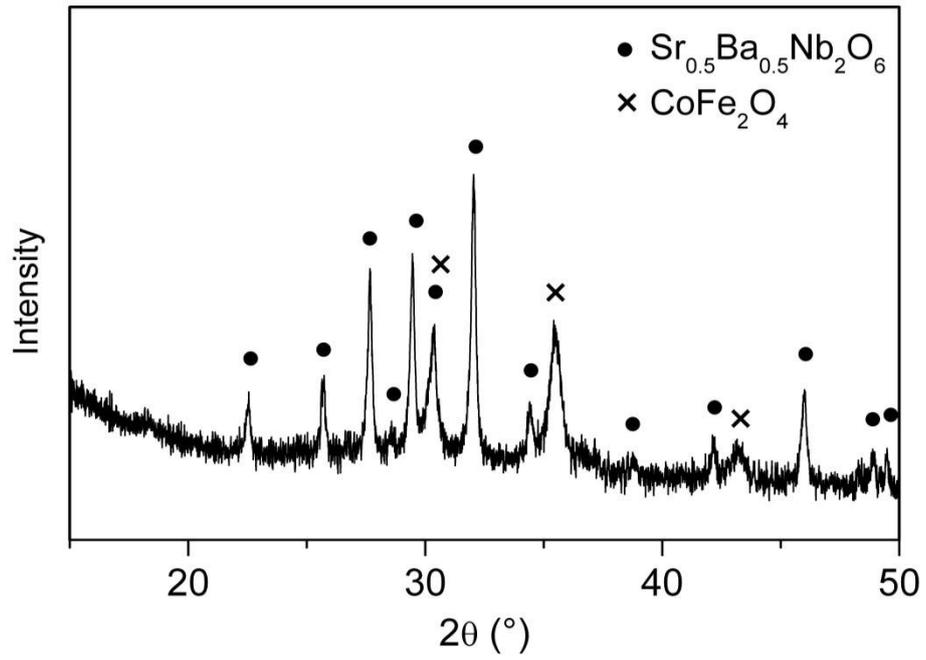


Fig. S1: Room-temperature XRD pattern of $(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{0.4}-(\text{CoFe}_2\text{O}_4)_{0.6}$ composite powder after calcining at 800 °C for 2 h (heating rate 10 K min⁻¹).

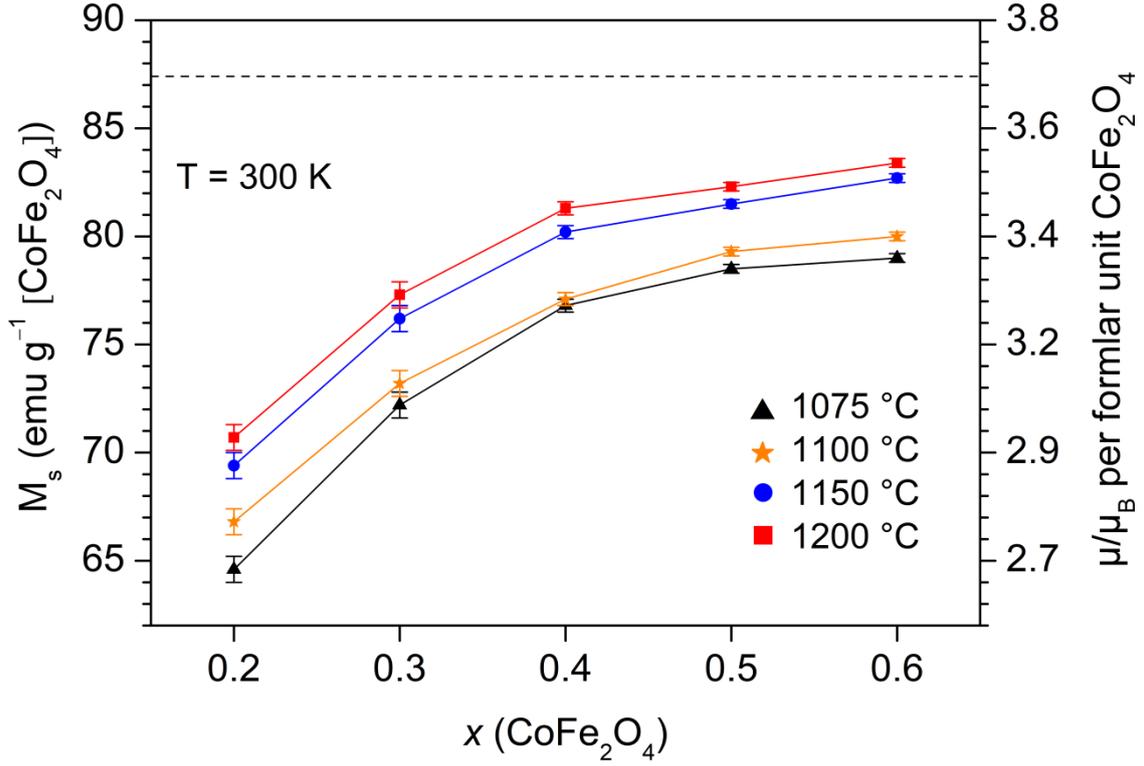


Fig. S2: Saturation magnetization (M_s) depending on the CoFe_2O_4 fraction (x) in $(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{1-x}-(\text{CoFe}_2\text{O}_4)_x$ composite ceramics sintered at the indicated temperatures for 1 h. M_s values for the composite samples were calculated by extrapolating of the magnetization at high field to $H = 0$ [1] and are given with respect to the nominal CoFe_2O_4 content. The black dotted line represents the M_s value of pure bulk CoFe_2O_4 , synthesized under the same conditions.

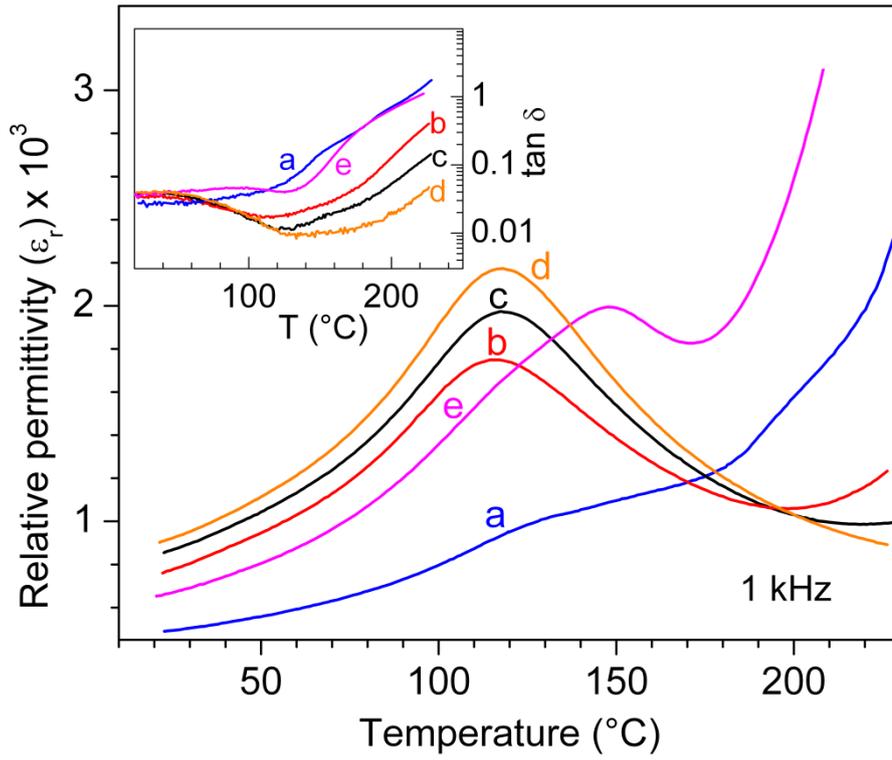


Fig. S3: Dielectric behavior of $\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6$ ceramic bodies sintered at different temperatures for 10 h. a) 1200 °C, b) 1250 °C, c) 1300 °C, d) 1325 °C, e) 1350 °C (from ref. [2])

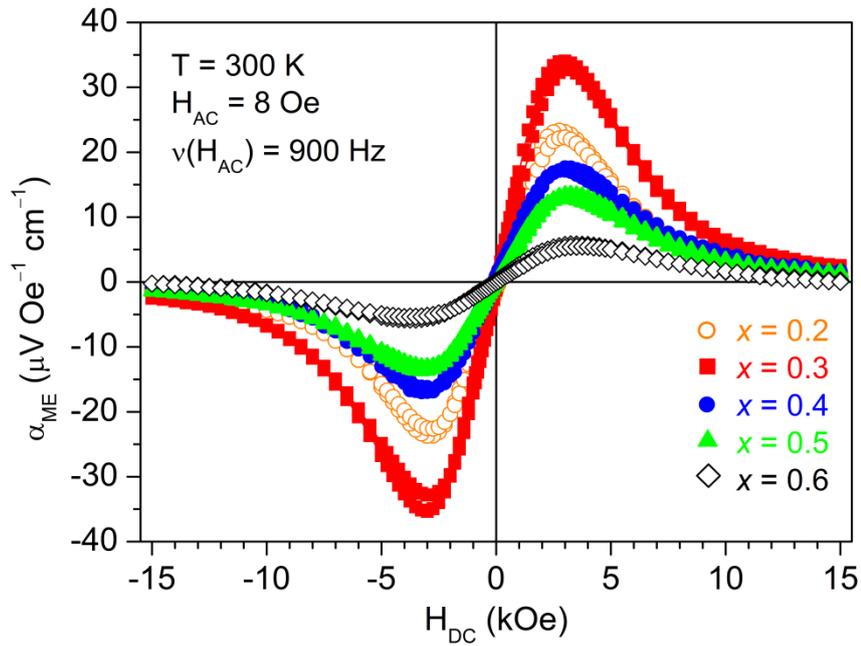


Fig. S4: Magnetoelectric coefficient (α_{ME}) vs. magnetic DC field for various $(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{1-x}-(\text{CoFe}_2\text{O}_4)_x$ composite ceramics sintered at 1150 °C for 1 h.

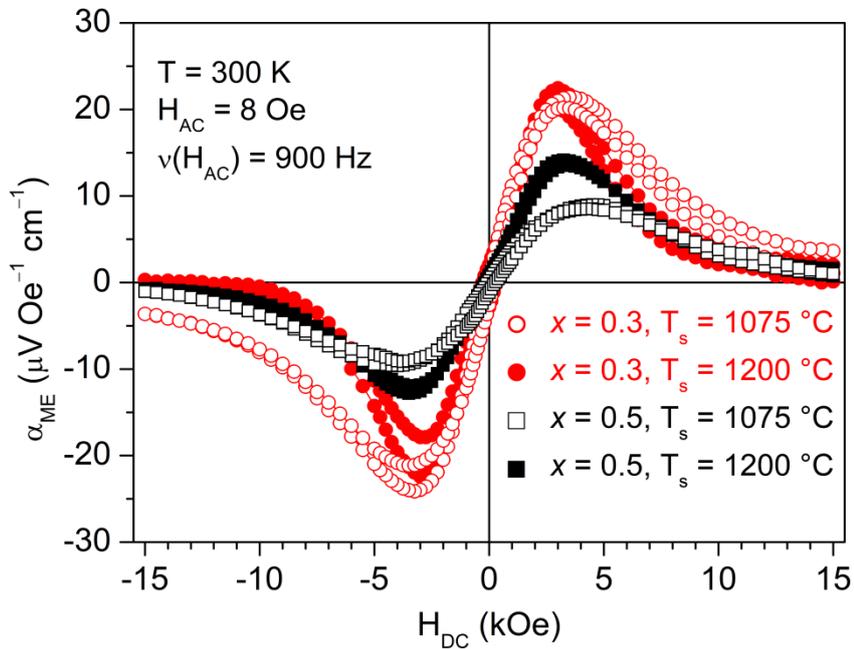


Fig. S5: Magnetoelectric coefficient (α_{ME}) vs. magnetic DC field for $(\text{Sr}_{0.5}\text{Ba}_{0.5}\text{Nb}_2\text{O}_6)_{1-x}-(\text{CoFe}_2\text{O}_4)_x$ ($x = 0.3$ and 0.5) composites sintered at the indicated temperatures (T_s) for 1 h.

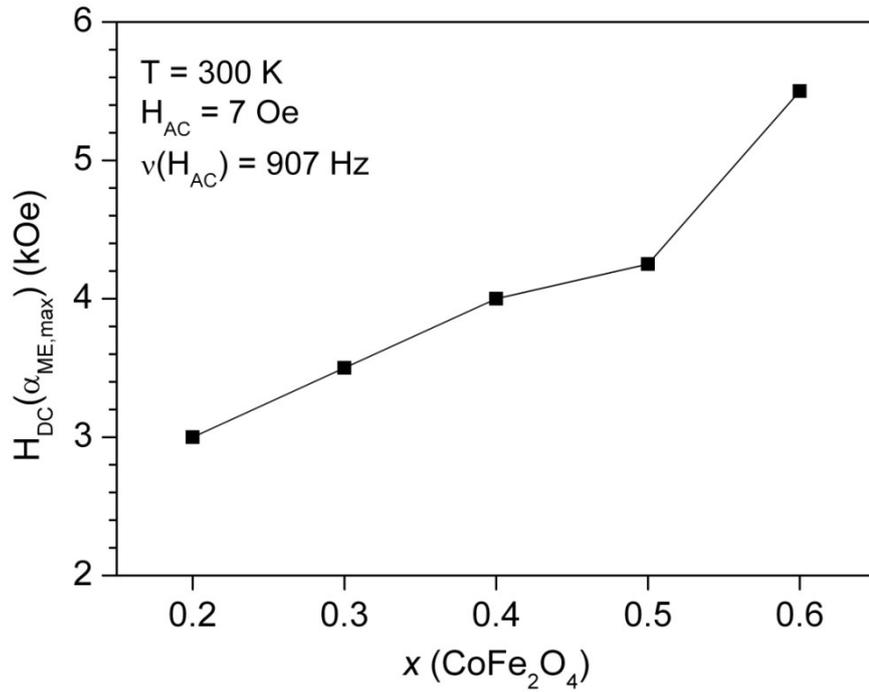


Fig. S6: Magnetic DC field ($H_{\alpha_{max}}$) at which the magnetoelectric coefficient ($\alpha_{ME,max}$) reaches its maximum with respect to the CoFe₂O₄ fraction (x) in $(Sr_{0.5}Ba_{0.5}Nb_2O_6)_{1-x}-(CoFe_2O_4)_x$ composites sintered at 1100 °C for 1 h.

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