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1 **Recycling of nitrogen and light noble gases in the Central American subduction zone: constraints from**

2 $^{15}\text{N}^{15}\text{N}$

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14 **Abstract**

15 How much nitrogen and light noble gases are recycled in modern subduction zones is
16 unclear. Fumaroles act as a means for passive degassing in arcs. They receive variable
17 contributions of volatiles from arc magmas, themselves sourced from the mantle wedge.
18 The gas compositions reflect the extent of volatile enrichment in sub-arc mantle sources
19 and constrain slab dehydration. However, contributions from atmospheric components in
20 fumaroles are unavoidable. For N₂, neon and argon, the atmospheric components are
21 challenging to discern from slab-derived components. Here, we report $^{15}\text{N}^{15}\text{N}$
22 measurements from eight fumaroles and seven bubbling springs, along the Central
23 American arc. Our new $^{15}\text{N}^{15}\text{N}$ data are coupled with noble gases measurements and show
24 that air-derived components in volcanic gas discharges can easily be underestimated, in
25 both fumaroles and springs, using conventional stable isotope or noble gases methods. We

show that, in the absence of $^{15}\text{N}^{15}\text{N}$ data, previously used tracers for air (e.g., $\delta^{15}\text{N}$, N_2/Ar , N_2/He , among others) may lead to erroneous conclusions regarding the origin of volatiles in mixed gases. In contrast, $^{15}\text{N}^{15}\text{N}$ data provide quantitative constraints on the nature and contributions of both atmospheric and magmatic components. Most springs are heavily dominated by air-derived N_2 , while fumaroles show substantial contributions of volcanic endmembers. Based on the fumarole data, we show that magma sources beneath the central American arc are enriched in all volatiles relative to ^3He , by two to three orders of magnitude compared to the MORB source. We use new $^{15}\text{N}^{15}\text{N}$ data to obtain source $\text{N}_2/^3\text{He}$, $^3\text{He}/^{36}\text{Ar}$ and $^3\text{He}/^{22}\text{Ne}$ ratios which we then use to compute volcanic N_2 , Ar and Ne degassing fluxes. Using this approach, we show that outgassing fluxes appear to match subduction fluxes in the Central America subduction zone. We determine an N_2 outgassing flux of between 4.0×10^8 and 1.0×10^9 mol N_2/y , comparable to the subduction flux of 5.7×10^8 mol N_2/yr determined previously. We obtain a similar conclusion for ^{22}Ne and ^{36}Ar . Overall, the volatile fluxes in the central American subduction zone no longer seem to require net transfer of N_2 , Ar, and Ne, to the deep mantle.

1. Introduction

The origin of volatiles emitted from convergent margins provides fundamental constraints on how plate tectonics redistributes volatiles between terrestrial reservoirs (Bekaert et al., 2020; Hilton et al., 2002; Plank et al., 2013). The nitrogen cycle is under-constrained, partly because quantifying degassing N_2 fluxes in arcs is challenging. For instance, basalt glasses are virtually absent in subduction zones, impeding quantification of nitrogen elemental abundances in the underlying mantle wedge. The systematic study of

metamorphic rocks has suggested nitrogen is quantitatively retained within minerals in downgoing slabs (Bebout et al., 2013; Busigny et al., 2003). The latter is based on the composition of rocks from the European alps, where sediments underwent metamorphism in a cold subduction zone (630 °C at 100 km, Busigny et al., 2003). In those rocks, nitrogen is hosted in the structures of clay minerals including micas and illites, as NH_4^+ substituting for potassium (Bebout et al., 1992, Busigny et al., 2003; Nieder et al., 2011). Experimental work confirms that the fluid/rock nitrogen partition coefficient in a cold P-T pathway is in favor of N retention in minerals (Jackson et al., 2021). However, under warmer conditions, fluid/rock partitioning favors N accumulation in fluids, potentially limiting nitrogen subduction to the mantle (Jackson et al., 2021). Subduction temperature gradients were likely steeper in most of the Proterozoic and Archean (Martin and Moyen, 2002). Therefore, it is unclear how modern fluxes determined in cold subduction zones should be extrapolated back into deep time.

The Central American Volcanic Arc (CAVA) is relatively well-characterized with evidence for sedimentary and igneous components from the slab variably contributing to mantle sources across the arc (Patino et al., 2000). It is a “warm” subduction zone, with predicted slab interface temperatures of ~ 800 °C at 100 km depth beneath Costa Rica (Peacock et al., 2005), resulting in sporadic slab melting (Hoernle et al., 2008). Thus, it may be considered an analog for subduction zones at a time when subduction temperature gradients were steeper. In Central America, metamorphic rocks cannot be used to constrain net subduction fluxes because no section of metasediments having undergone the subduction P-T pathway is known to occur in the geological record. Instead, net subduction fluxes of volatiles are estimated from comparing their concentrations and isotopic

compositions in the offshore altered oceanic crust (Li and Bebout, 2005, Busigny et al., 2019) with their concentrations and isotopic compositions in arc fumaroles that represent volcanic outgassing fluxes. This mass balance calculation can be hampered by ubiquitous infiltration of air into most fumaroles, as evidenced by noble gases systematics: although fumaroles typically show mantle-like helium isotope ratios, neon and argon budgets are overwhelmed by atmospheric components (Hilton et al., 2002; Snyder et al., 2003). This is also a problem for nitrogen. Historically, SO_2/N_2 ratios from fumaroles and the overall SO_2 outgassing flux has been used to determine a total N_2 volcanic outgassing flux for the central American arc, yielding a value of 1.7×10^9 mol N_2/yr , or 3.4×10^9 mol N/yr (Hilton et al., 2002, Fischer et al., 2002). This is comparable to an estimate of the central American subduction nitrogen flux of $\sim 1.1 \times 10^9$ mol N/yr (equivalent to $\sim 5.7 \times 10^8$ mol N_2/yr) from Busigny et al. (2019). These flux estimates are given with no uncertainties, but they appear comparable within a factor of 2. Taken at face value, this would support inefficient N recycling from the surface to the deep mantle. However, fumaroles incorporate air-derived N_2 , after the infiltration of meteoric water within subsurface hydrothermal systems and/or because of sampling techniques (Fischer et al., 2002). Thus, the outgassing N_2 fluxes based on raw N_2/SO_2 ratios in fumaroles are overestimates of the outgassing N_2 flux. As a remedy, the volcanic fraction of N_2 in fumaroles was quantified on the basis of N_2/Ar ratios: air-saturated waters have a known N_2/Ar ratio of ~ 40 at STP but most hydrothermal gases in the CAVA have ratios > 80 (Hilton et al., 2002; Fischer et al., 2002, Elkins et al., 2006, Snyder et al., 2003; Zimmer et al., 2004). The N_2 amount in excess of air-saturated water, termed N_2^* , was suggested to reflect the volcanic N_2 fraction (Hilton et al., 2002, Fischer et al., 2002). A volcanic nitrogen degassing flux of 2.9×10^8 mol N_2/year was estimated for the

central American arc (Fischer et al., 2002). This degassing flux estimate is lower by a factor of 2 than the subduction nitrogen flux of $\sim 5.7 \times 10^8$ mol N₂/year (or $\sim 1.1 \times 10^9$ mol N/yr), suggesting N₂ sequestration into the deep mantle by subduction, even in a warm subduction zone (Busigny et al., 2019; Li and Bebout, 2005).

Here, we take a new approach to constrain the origin of N₂ as well as light noble gases in CAVA fumaroles and springs. We use the newly-developed ¹⁵N¹⁵N tracer of atmospheric contamination (Labidi et al., 2020; Yeung et al., 2017; Young et al., 2016). Specifically, we use Δ_{30} , the ¹⁵N¹⁵N concentration relative to a random distribution of ¹⁴N and ¹⁵N atoms among N₂ molecules, as a tracer of surficial atmospheric contamination. The Δ_{30} tracer is defined as $\Delta_{30} = {}^{30}\text{R}/({}^{15}\text{R})^2 - 1$ (‰), where ${}^{30}\text{R} = {}^{15}\text{N}^{15}\text{N}/{}^{14}\text{N}^{14}\text{N}$ and ${}^{15}\text{R} = {}^{15}\text{N}/{}^{14}\text{N}$ for the gas of interest. At relevant temperatures ranging from 200 to 1000 °C, equilibrium among N₂ isotopologues results in Δ_{30} values from 0.5 to 0.1‰, respectively (Yeung et al., 2017). This applies for any magmatic and crustal N₂, whether it is mantle-derived, inherited from the slab, or shallow crustal reservoirs (Labidi et al., 2020). In contrast, air has a pronounced disequilibrium ¹⁵N¹⁵N enrichment, leading to an atmospheric Δ_{30} value of 19.1 ± 0.3 ‰ (2 σ) (Yeung et al., 2017). We make use of this disequilibrium as a tracer for air contributions in natural fluids and identify the compositions of mantle sources for N₂ and light noble gases for 15 fumaroles and gases bubbling from springs in Costa Rica, Panama, Nicaragua, and El Salvador. Our ¹⁵N¹⁵N data allows determination of the N₂/³He, ³He/³⁶Ar and ³He/²²Ne ratios for the CAVA mantle sources. Using an updated ³He outgassing flux for the region, we provide a new range of estimates for the N₂ outgassing flux along the Central American arc. Our result has implications for the net subduction flux of nitrogen, suggesting that the net sequestration of nitrogen to the mantle is within error of zero.

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119 **2. Geological context and samples**

120 The CAVA results from the eastward subduction of the Cocos Plate beneath the
121 Caribbean Plate. Abundant literature describes the spectrum of sub-arc mantle sources,
122 incorporating slab sedimentary components beneath Nicaragua, to volcanic seamounts
123 derived from Galapagos underneath Costa Rica and Panama (Carr et al., 2003; Gazel et al.,
124 2009; Hoernle et al., 2008; Patino et al., 2000; Ranero and von Huene, 2000;
125 Schwarzenbach et al., 2016).

126 Eight gas samples from fumaroles were collected from Poás and Momotombo
127 volcanoes during several field expeditions in the early to mid-2000's with gas chemistry
128 and isotope data published in Zimmer et al. (2004), Elkins et al. (2006), De Leeuw et al.,
129 (2007) and Fischer et al. (2015). Fumaroles had outlet temperatures ranging from 98°C to
130 747°C. Samples were collected in pre-evacuated Giggenbach bottles filled with 5N NaOH
131 solution (Giggenbach and Goguel, 1989). Gas splits were taken from the headspace of the
132 bottles with sealed glass tubes shortly after sample collection and stored until analyzed for
133 this work. Five gases from the Poás crater were collected between 2003 and 2006, from the
134 fumarolic sites "Official" and "Naranja", at temperatures between 98°C and 158°C. For
135 these, we report new N₂ isotopologue and ⁴⁰Ar/³⁶Ar data. ³He/⁴He of these samples are
136 7.0±0.2 R_A (Hilton et al., 2010) where R_A is air ³He/⁴He or 1.384 x 10⁻⁶. Gas chemistry is
137 available from previous work (Fischer et al., 2015). One fumarole sample was collected at
138 Santa Ana Volcano (Salvador), with a vent temperature of 400°C. It has a helium isotope
139 ratio of 7.5±0.1 R_A (De Leeuw et al., 2007). The He-Ne-Ar abundances are known (De
140 Leeuw et al., 2007). No ⁴⁰Ar/³⁶Ar is available but other gases from El Salvador all have

near-atmospheric $^{40}\text{Ar}/^{36}\text{Ar}$ ratios (Snyder et al., 2003). We additionally discuss two fumarole gas samples collected at Momotombo in Nicaragua, at an outlet temperature of $\sim 750^\circ\text{C}$. For these two samples, we report new $^{40}\text{Ar}/^{36}\text{Ar}$ isotope measurements. N_2 isotopologues, helium isotopes, He-Ne-Ar and major gas chemistry data for these samples are available from previous work (Elkins et al., 2006; Yeung et al., 2017). Hot springs found in the flanks of volcanic arcs are the most accessible samples, and hence are routinely used to characterize volcanic endmembers (e.g. Snyder et al., 2003). As a means to compare these to our fumaroles, we also report data from seven gases from bubbling springs in Costa Rica and Panama, collected in 2018 as part of the *Biology Meets Subduction* initiative. Sample collection sites are shown on [figure 1](#). Samples were collected in 15 cm^3 copper tubes following standard procedures. Waters in the springs had temperatures between 29°C and 55°C . For these, we report major element gas compositions, N_2 isotopologues and He-Ne-Ar systematic data.

3. Methods

Gas aliquots were split up to three ways and processed for (1) gas chemistry, (2) N_2 , and (3) noble gases systematics. Inorganic gas components (such as N_2 , He, Ar and O_2) and methane were separated and quantified on a TCD gas chromatograph coupled with a quadrupole mass spectrometer as in earlier work (Fischer et al., 2015). This allowed determining the N_2/He and N_2/Ar ratios of each gas splits. Nitrogen isotopologues were determined via high-resolution mass spectrometry at the University of California, Los Angeles (Young et al., 2016). Noble gas analyses were conducted at the University of Oxford (UK) and at the University of New Mexico using standard methods (Barry et al., 2016; Lee et al., 2017). Details may be found in the [supplementary online file](#).

4. Results

We present an integrated dataset that includes concentrations of He, Ne, Ar, N₂, O₂, and CH₄, elemental ratios of those gases, noble gas isotope compositions, and N₂ isotopologue ratios. Data for the fumaroles (n=8) are in table 1 and data for the springs (n=7) are in table 2.

4.1. $\delta^{15}\text{N}$ and Δ_{30} relationships

Newly determined Δ_{30} values vary between $19.3 \pm 0.4\text{‰}$ and $3.5 \pm 1.0\text{‰}$ (1 σ uncertainty, Fig. 2). Two previously published data points from Momotombo extend the range down to $1.5 \pm 0.6\text{‰}$ and are plotted on Fig. 2 (Yeung et al. 2017). The highest Δ_{30} is similar to the air value of $19.1 \pm 0.3\text{‰}$ (2 σ). The range in Δ_{30} values suggests samples incorporate variable amounts of atmospheric nitrogen, contributing between ~8 and ~100% of the total N₂. Fumaroles from Poás and Momotombo have the lowest Δ_{30} values, indicating that these fumaroles have the lowest air-derived N₂ contributions. A range of $\delta^{15}\text{N}$ values is observed, with values of between $-3.7 \pm 0.3\text{‰}$ and $+4.2 \pm 0.3\text{‰}$ (Fig. 2). For Poás and Momotombo, $\delta^{15}\text{N}$ values are $+0.4 \pm 0.3\text{‰}$ and $+5.4 \pm 0.3\text{‰}$ respectively, where air contributions are at their lowest values (minimum Δ_{30}). The $\delta^{15}\text{N}$ values as a whole (with both low and high Δ_{30}) are within the range of $\delta^{15}\text{N}$ values of between -3.0 ± 0.6 and $6.3 \pm 0.3\text{‰}$ (n=73) reported previously for central American gases (Elkins et al., 2006; Fischer et al., 2015, 2002; Snyder et al., 2003; Zimmer et al., 2004). The $\delta^{15}\text{N}$ values exhibit a negative correlation with Δ_{30} values (Fig. 2).

4.2. Noble gas isotopes

Concentrations of He, Ne and Ar vary over 4 orders of magnitude and are correlated (Table 1 and 2). We observe $^{40}\text{Ar}/^{36}\text{Ar}$ ratios between 297 ± 5 and 342 ± 5 for our samples (Table 1 and 2), far from the upper mantle value of $\sim 25,000$ (Moreira et al. 1998) but close to air at 298.5 (Lee et al., 2006). This is comparable to the known range of values for Central American rocks and gases of between 292 ± 7 and 310 ± 5 (Fischer et al., 2005; Kennedy et al., 1991; Snyder et al., 2003; Staudacher and Allègre, 1988). The $^{20}\text{Ne}/^{22}\text{Ne}$ and $^{21}\text{Ne}/^{22}\text{Ne}$ ratios are known only for the hot springs, and data appear indistinguishable from air (Table 2).

The $^3\text{He}/^4\text{He}$ ratios vary substantially, between 0.3 and $7.6\pm0.1 R_A$. Unlike Ne and Ar, the He budget is not significantly affected by atmospheric components and instead reflect appreciable deep contributions. The $^3\text{He}/^4\text{He}$ ratios are broadly correlated with $^4\text{He}/^{20}\text{Ne}$ ratios (Fig. 3). Most of the data are explained by two-component mixing between air ($^3\text{He}/^4\text{He} = 1 R_A$ by definition, $^4\text{He}/^{20}\text{Ne} = 0.3188$) and a magmatic endmember. The magmatic component has a $^3\text{He}/^4\text{He}$ of ~ 7 (Fig. 3). The $^4\text{He}/^{20}\text{Ne}$ ratio of the magmatic component must be at least as high as the highest value from our dataset, ~ 270 . Three Costa Rican springs show a clear offset from the two-component mixing line, showing much lower $^3\text{He}/^4\text{He}$ ratios at a given $^4\text{He}/^{20}\text{Ne}$ value. Similar features in other Central American gases were interpreted to result from mixing with gases from crustal fluids (De Leeuw et al., 2007).

4.3. Nitrogen, oxygen and methane concentrations

Nitrogen concentrations vary from 0.2 vol% to 93 vol% in the springs, and from 0.6 vol% to 1.9 vol% in the fumaroles (Table 1 and 2). Oxygen concentrations are variable,

resulting in O_2/N_2 ratios ranging over two orders of magnitude, between 3.8×10^{-4} to 1.1×10^{-1} (Supplementary Fig. 1). The O_2/N_2 ratios are not directly correlated with Δ_{30} values (supplementary Fig. 1), suggesting that O_2/N_2 is not a reliable indicator of the presence of nitrogen from air. Methane concentrations are below 0.1 vol % for most samples, with the exception of two of the Costa Rica hot springs, where CH_4 concentrations are as high as ~ 80 vol %, suggesting contributions from shallow crustal gases (Snyder et al. 2003). N_2/CH_4 varies over 5 orders of magnitude, between $\sim 1.2 \times 10^{-1}$ to $\sim 2.1 \times 10^4$, but remains uncorrelated with Δ_{30} values (Supplementary Fig. 1).

4.4. Nitrogen - noble gas ratios

We compute N_2/Ar , $N_2/^{36}Ar$, N_2/He and $N_2/^3He$ ratios, with a $\pm 20\%$ relative uncertainty on individual samples. N_2/Ar ratios are between 21 and 120 for most samples, with one sample from Momotombo at 507 (Fig. 4). For comparison, the air value is ~ 84 , air-saturated water (ASW) at STP is ~ 41 , and MORB gases have N_2/Ar ratios of 55 ± 5 (Javoy and Pineau, 1991), although a higher estimate of $\sim 120 \pm 40$ (Marty and Dauphas, 2003) was derived from MORB samples with unusually radiogenic $^{40}Ar/^{36}Ar$ ratios (data from Marty et al., 1999). For samples where $^{40}Ar/^{36}Ar$ was measured, $N_2/^{36}Ar$ ratios vary between 1.2×10^4 and 4.2×10^4 . Assuming the $^{40}Ar/^{36}Ar$ is atmospheric as for other Central American gas samples (Snyder et al., 2003), the $N_2/^{36}Ar$ range extends to between 6.5×10^3 and 1.5×10^5 (Fig. 4). This includes values below and above both air (2.5×10^4) and air-saturated water at STP (1.3×10^4). The highest $N_2/^{36}Ar$ value remains well below estimates of the convective mantle of $\sim 2 \times 10^6$ (Labidi et al., 2020; Marty and Humbert, 1997) even for samples where Δ_{30} is low.

The N_2/He ratios vary between 1.5×10^2 and 4.2×10^4 , below the value of air of 1.5×10^5 or air-saturated water $\sim 3.0 \times 10^5$ (Fig. 5). Similarly, $N_2/^3He$ ratios vary between 1.6×10^7 and 1.1×10^{11} (Fig. 5), mostly below the value of air and ASW of 1.0×10^{11} and $\sim 2.0 \times 10^{11}$ respectively (Ballentine et al., 2002). The N_2/He (and $N_2/^3He$ ratios) are correlated with Δ_{30} values and are described by two-component mixing hyperbolae between air (or air-saturated waters) and high-temperature components with elevated N_2/He (and $N_2/^3He$ ratios) relative to MORB gases (Fig. 5).

4.5. Noble gas ratios

The $^3He/^22Ne$ ratios range over 4 orders of magnitude, between 8.9×10^{-6} and 2.7×10^{-2} (Fig. 6). These values are between air, at 4.5×10^{-6} , and upper-mantle gases at $\sim 5.5 \times 10^{-1}$ (Moreira et al., 1998). The $^4He/^20Ne$ ratios are also ranging between air and the upper-mantle value (Supplementary Figure 2). $^3He/^22Ne$ (and $^4He/^20Ne$) ratios are correlated with Δ_{30} values (Fig. 6, Supp Fig. 2). At $\Delta_{30} = 0$ the high-temperature component appears to have He/Ne ratios lower than MORB gases by about four orders of magnitude (Fig. 6, Supp Fig. 2). The $^3He/^36Ar$ ratio ranges between 1.6×10^{-6} and 1.7×10^{-4} (Fig. 6). This is higher than the air value of 2.5×10^{-7} , but considerably lower than the upper-mantle value of $\sim 5.0 \times 10^{-1}$ (Moreira et al., 1998; Raquin et al., 2008). A similar observation can be made of the $^4He/^40Ar$ ratio (Supp Fig. 2). The variations in $^3He/^36Ar$ (and $^4He/^40Ar$) ratios are correlated with Δ_{30} values and at $\Delta_{30} = 0$, the high-T component shows $^3He/^36Ar$ ratios lower than MORB gases by about three orders of magnitude (Fig. 6, Supp Fig. 2). Most $^{22}Ne/^36Ar$ ratios in fumaroles and springs range between 1.3×10^{-1} and 3.1×10^{-3} (with one outlier at $\sim 6.5 \times 10^{-1}$, the Poás gas with the Δ_{30} value closest to air). The higher values are similar to the air and MORB values of 5.0×10^{-2} and 1.0×10^{-1} ,

respectively (Mukhopadhyay, 2012) and are observed for samples with air-like Δ_{30} values. At $\Delta_{30} = 0$, much lower $^{22}\text{Ne}/^{36}\text{Ar}$ values are observed, lower than both MORB and air by one to two orders of magnitude (Table 1 and 2).

5. Discussion

Our rationale is that correlations between Δ_{30} and various isotope and element ratios can be used to identify and correct for air components and reveal pristine high-temperature endmembers. Air is identified with a Δ_{30} of $19.1 \pm 0.3\text{‰}$, while any N_2 produced by a geologic process such as magmatic degassing has $\Delta_{30} \sim 0\text{‰}$ (Labidi et al., 2020). Arguments against re-ordering and crustal contamination are given in the supplementary discussion.

5.1 Atmospheric N_2 may easily go unnoticed

CAVA gas Δ_{30} values are variable, indicating commensurately variable contributions of air to the N_2 budgets in hydrothermal discharges. Based on Δ_{30} data, air accounts for 40 to 100% of the N_2 in the springs. In fumaroles, air accounts for 8 to 90% of the N_2 . Atmospheric N_2 is therefore ubiquitous. Systematics involving $\delta^{15}\text{N}$, $[\text{O}_2]$, O_2/N_2 , N_2/He and N_2/Ar ratios may disentangle magmatic N_2 from air-derived N_2 in gas discharges (Sano et al., 2001, Fischer et al. 2002, Elkins et al., 2006). The rationale for all of those approaches is that air has known $\delta^{15}\text{N}$, O_2/N_2 , N_2/He and N_2/Ar ratios, that are different from magmatic components. Our Δ_{30} data present a series of challenges to these previous approaches. For example, samples with high $^4\text{He}/^{20}\text{Ne}$ ratios were suggested to illustrate a volatile budget largely uncontaminated by air, especially when a number of other criteria are met, e.g.,

O₂/N₂ ratios < 10⁻³, high outlet temperatures, or non-atmospheric N₂/Ar ratios (Elkins et al., 2006). The Santa Ana volcano fumarole is vented at 400°C, a ⁴He/²⁰Ne ~ 200 times higher than air and a O₂/N₂ ratio of ~10⁻⁴ (Table 1, Fig. 3), which would suggest minimal air contamination. However, the near-air Δ₃₀ of 15.5±0.3‰ requires that ~80% of N₂ is from air in this sample, showing that high ⁴He/²⁰Ne, even in conjunction with high vent temperature and low O₂/N₂, is not necessarily sufficient evidence that N₂ is dominantly magmatic (See supplementary discussion).

5.2 Atmospheric N₂ undergoes isotopic fractionation in hydrothermal systems

Nitrogen isotope ratios could be a direct tracer of atmospheric N₂, since air and magma-derived nitrogen are thought to have distinct δ¹⁵N values (Fischer et al., 2002, Sano et al., 2001). The Δ₃₀ data confirm the veracity of this approach for most, but not all, samples. Gases (except those from Poás) fall on a single two-component mixing trend between air and a high-temperature component with near-zero Δ₃₀ and δ¹⁵N of ~ 5‰ (Fig. 2). For those, the air component appears unfractionated with respect to δ¹⁵N. When mixed with high-temperature volatiles, δ¹⁵N increases and Δ₃₀ values decrease (Fig. 2).

Samples from Poás are different. They form a unique linear mixing trend where both air-derived and high-temperature N₂ are distinct from elsewhere in the arc system, with δ¹⁵N endmember values of -4.0±0.3‰ and +1.0±0.3‰ for air and the high-T component, respectively (Fig. 2). This represents a ~4‰ shift for both endmembers compared to values observed elsewhere on the arc. Our observation creates ambiguities at Poás, since there, and there only: (1) high-temperature δ¹⁵N is only marginally higher than air; and (2)

air-derived N_2 has a $\delta^{15}N$ very similar to the MORB value of $-5 \pm 2\text{‰}$ (Javoy and Pineau, 1991, Marty et al., 1999).

The data from Poás require that atmospheric N_2 experienced a $\sim 4\text{‰}$ mass-dependent $^{15}N/^{14}N$ fractionation in the sub-surface. This may result from the degassing of once air-saturated waters. The isotopic fractionation associated with N_2 dissolved in water was experimentally documented to be $+0.9$ to -0.4‰ from 6 to 60 °C (Lee et al., 2015). The dissolved-gas isotope fractionation was shown to cross over at about 40 °C, from a positive to a negative sign (Lee et al., 2015). Above 40 °C, the isotopic difference between dissolved and gaseous N_2 increases: isotopically light N_2 is increasingly partitioned in the dissolved fraction as temperature increases. This behavior indicates that a kinetic isotope effect attends N_2 dissolution/degassing in geothermal waters (Lee et al., 2015). Using the experimental results of Lee et al., (2015), we calculate the consequences of N isotope fractionation during open-system degassing using Rayleigh fractionation. At 60 °C, with a gas/dissolved-gas $^{15}N/^{14}N$ fractionation factor of 1.0004, the lowest $\delta^{15}N$ values of $\sim -3.5\text{‰}$ with air-like Δ_{30} require that the N_2 sampled by the fumaroles is the residuum left over after degassing of about 99.98 % of the dissolved nitrogen. Extrapolating the temperature dependence of the fractionation factor to 100 °C, 97% degassing of N_2 is required to account for a $\delta^{15}N$ of $\sim -3.5\text{‰}$.

We envision the hydrothermal degassing to be the natural consequence of hydrothermal systems experiencing continuous degassing in the absence of continuous meteoric (air-saturated) water recharge. It is puzzling however that among the sites we sampled, air nitrogen undergoes isotopic fractionation only at Poás. In order to explain this observation, we suggest a restricted set of conditions is met in the Poás subsurface, but not

elsewhere. Gases from Momotombo and Santa Ana are vented at temperatures between 747 °C and 400 °C, respectively. No liquid water remains stable at those temperatures. Thus, isotope partitioning between water and gas is not occurring, and N₂ remains unfractionated. Spring gases are vented at temperatures between 29 °C and 59 °C (Table 2), at temperatures where degassing is probably limited. In contrast, at Poás, the fumaroles are vented at ~100-150 °C (Fischer et al., 2015). At these temperatures, liquid and gaseous water are stable and water/gas partitioning can occur. These conditions are comparable to those observed for hydrothermal gases from Iceland or Yellowstone, where negative $\delta^{15}\text{N}$ values were previously observed in conjunction with air like Δ_{30} values (Labidi et al., 2020). Our conclusion is also consistent with the observation of sporadically negative $\delta^{15}\text{N}$ (with no available Δ_{30} data) in Poás fumaroles sampled in 2001. Values as low as $\sim -3.0\text{‰}$ were observed in fumarole vented at temperature between 76 and 108 °C (Fischer et al., 2002). In other fumaroles (89-101 °C) sampled between 1998 and 2001, while unfractionated air-like $\delta^{15}\text{N}$ and N₂/Ar values were reported (Vaselli et al., 2003). More work is warranted to follow the potential Δ_{30} - $\delta^{15}\text{N}$ evolution of a given fumarole through time, but in the interim, we suggest caution in interpreting $\delta^{15}\text{N}$ data at Poás in the absence of Δ_{30} .

5.3. Is high-temperature nitrogen also fractionated?

Whether high-temperature components with $\Delta_{30} \sim 0\text{‰}$ also experience $\delta^{15}\text{N}$ isotope fractionation in hydrothermal systems is an open question. At Momotombo, fumaroles have a high-temperature endmember with $\delta^{15}\text{N} \sim +5\text{‰}$ (Fig. 2.). There, vent temperatures are > 700 °C. This precludes liquid water to exist even at depths, ruling out the possibility of any water/gas isotope exchange. This is consistent with a previous observation of

unfractionated $\delta^{15}\text{N}$ for fumaroles when they are vented at $> 300\text{ }^{\circ}\text{C}$ (Fischer et al., 2005). At Poás however, the interpretation is not as straightforward, where the high-temperature endmember has a $\delta^{15}\text{N} \sim +1\text{‰}$ (Fig. 2), which is lower than at Momotombo by $\sim 4\text{‰}$, implying that a fractionation similar to that for the air component could have occurred. It is conceivable that high-T volatiles have been delivered to the water table *prior to* hydrothermal degassing, where they would experience $^{15}\text{N}/^{14}\text{N}$ fractionation together with the air-derived gases. In this interpretation, the high-T $\delta^{15}\text{N}$ value of $\sim +1\text{‰}$ obtained by extrapolation to $\Delta_{30} = 0$ would not record the actual magmatic $\delta^{15}\text{N}$ value at Poás, and the Poás trend would be parallel to the main array as the result of a fractionation equally affecting both the atmospheric and high-T volatiles. However, with our dataset alone, we cannot exclude that high-T volatiles had been delivered to the fumaroles *after* hydrothermal degassing. If so, the Poás high-T endmember could be truly unique in $\delta^{15}\text{N}$, and the Poás trend being parallel to the main array (Fig. 2) would be fortuitous. This latter scenario relies on a coincidence, and thus appears unlikely, although it cannot be fully ruled out.

5.4. N_2/Ar ratios place independent constraints on hydrothermal degassing

The solubilities of argon and nitrogen are distinct by about a factor of 2 in geothermal waters at temperatures between 20 and $100\text{ }^{\circ}\text{C}$ (Ballentine et al, 2002). Thus, water/gas processes may be tracked with N_2/Ar ratios, independently of nitrogen isotopes. Low N_2/Ar ratios for air-like Δ_{30} values are observed at Poás, while most other samples have near-air values for N_2/Ar at air-like Δ_{30} (Fig. 4), confirming that at Poás liquid-gas

partitioning occurred in the sub-surface. We can again use a Rayleigh fractionation calculation to infer the amount of degassing implied by the N_2/Ar ratio data, where

$$N_2/Ar = (N_2/Ar)_0 f^{\alpha-1} \quad (1)$$

and

$$\alpha = (K_{N_2}/K_{Ar}), \quad (2)$$

where f is the fraction of remaining Ar in the water, $(N_2/Ar)_0$ the known starting composition of ASW, and α is the fractionation coefficient given for a gas/liquid system given by Equation (2), and K_i is the Henry's-law constant for species i , as compiled in Ballentine et al., (2002). Taking $(N_2/Ar)_0$ to be 41, we find that the lowest N_2/Ar value of ~ 10 implied by our data occurs for about 94% degassing of N_2 (75% degassing of Ar) prior to sampling (Fig. 4). To first order, this degree of degassing is consistent with estimates made on the basis of $^{15}N/^{14}N$ values (section 5.2).

There is a caveat to this interpretation. Data for $\delta^{15}N$ and N_2/Ar appear consistent with fractionation within hydrothermal systems, but no direct correlation is observed between $\delta^{15}N$ and N_2/Ar (Table 1). This is perhaps because degassing may yield variable $^{15}N/^{14}N$ and N_2/Ar fractionations depending on the exact temperature and pressure conditions (Lee et al., 2015; Warr et al., 2015). Under conditions making the gas behavior non-ideal, degassing may change the N_2 -Ar solubility relationship, as suggested for heavier noble gases (Labidi et al., 2020, Warr et al., 2015). Additionally, small contributions of high-temperature volatiles are likely to cause shifts in $\delta^{15}N$ with no obvious increases in N_2/Ar ratios (Fig. 4). This is because high-temperature endmembers with $\Delta_{30} = 0\text{‰}$ appear to have N_2/Ar and $N_2/^{36}Ar$ ratios of ~ 100 and $\sim 10^4$, respectively (Fig. 4). These values are

essentially similar to air, consistent with pioneer work on the CAVA (Hilton et al., 2002; Snyder et al., 2003), but remain lower than volcanic endmembers at other arcs (Taran, 2009; Zelenski and Taran, 2011). We note that one of the Momotombo fumaroles has a high N_2/Ar of ~ 500 , higher than other high-temperature gases by a factor of 5. The significance of variable N_2/Ar ratios in Central American high-temperature endmembers must be systematically investigated in future work, with Δ_{30} data.

5.5. $\delta^{15}N$ - N_2/He relationships constrained by Δ_{30} data

In contrast to argon, the solubility of helium in geothermal waters is nearly indistinguishable from that of nitrogen in the relevant temperature range (Ballentine et al., 2002). Data can be accounted for by mixing between air and an endmember with $N_2/^3He$ ratios $\sim 10^8$ (Fig. 5). This is higher than MORB gases that are characterized by $N_2/^3He$ ratios of about 10^6 (Javoy and Pineau, 1991, Marty and Humbert, 1999). The Poás fumaroles involve a mixing scenario that is similar to the one for Momotombo, involving high-T endmembers with a $N_2/^3He$ of $\sim 10^8$ (Fig. 5). This observation argues against a significant contribution of MORB derived volatiles to Poás volcanic gas discharges. Overall light $\delta^{15}N$ values from Poás were previously suggested to reflect a contribution from MORB-derived volatiles, contrary to the majority of other CAVA gas discharges that showed predominantly slab-derived N (Fischer et al., 2002, Elkins et al., 2006). Our new Δ_{30} data suggest a revision to the interpretation of ^{15}N depletions and show that they rather reflect contributions of fractionated air (Fig. 2) mixed with a predominantly non-MORB magmatic component with elevated N_2/He ratios (Fig. 5).

5.6. Magmatic endmembers throughout the arc are enriched in N₂

In an attempt to characterize subduction volatiles, we constrain the high-temperature endmembers for nitrogen isotopes and N₂/Ar, N₂/He, He/Ne and He/Ar ratios. Endmember compositions are calculated assuming two-component mixing with air constrained by Δ_{30} values, as described above. We focus on fumarole data to constrain endmember compositions, since they show the lowest Δ_{30} and no evidence of crustal contamination (see supplementary discussion).

The derived N₂/He ratios are higher than MORB by two orders of magnitude. In contrast, He/Ne, He/Ar and N₂/³⁶Ar are all lower than MORB by two to three orders of magnitude (Fig. 4-6). The simplest explanation is that slab-derived N₂, Ne and Ar were added to a depleted mantle source with an unmodified helium signature (Fig. 7). A straightforward quantification of slab-derived N₂, Ar and Ne to the mantle wedge is possible using N₂/³He, ³He/²²Ne and ³He/³⁶Ar ratios. This approach is viable if ³He subduction is negligible (Staudacher and Allegre, 1988), notwithstanding that slab components are required to contribute minute ⁴He ingrowth from subducted uranium (Hilton et al., 2002).

The high-T, arc magma endmembers at Poás and Momotombo have N₂/³He ratios two orders of magnitude higher than MORB, requiring nitrogen addition to mantle sources in these arc systems. Sedimentary nitrogen may be a straightforward contributor to the Momotombo source, explaining the high $\delta^{15}\text{N}$ of + 5‰ there (Fig. 7). Sediments on the modern oceanic crust directly offshore of Central America are likely candidates, since they have an average $\delta^{15}\text{N}$ of ~5‰ (Li and Bebout, 2005). Poás volcano is more complicated, because the lower $\delta^{15}\text{N}$ (Fig. 2) may not be a genuine representation of the Poás mantle

source. We note that because $N_2/{}^3He$ are indistinguishable between Poás and Momotombo, they must have received a comparable amount of slab-derived N.

Sub-arc magmatic endmembers have lower ${}^3He/{}^{22}Ne$ and ${}^3He/{}^{36}Ar$ ratios by 2 to 3 orders of magnitude compared to those of MORBs. Like nitrogen, this likely requires substantial addition of slab-derived neon and argon to a mantle source. Samples with near-zero Δ_{30} have ${}^{22}Ne/{}^{36}Ar$ lower than both MORB and air by one to two orders of magnitude (Table 1), with values trending toward $\sim 0.7 \times 10^{-2}$. This is similar to the median ratio of an entire section of altered oceanic crust and sediments of $\sim 1 \times 10^{-2}$ (Chavrit et al., 2016). The similarity in ${}^{22}Ne/{}^{36}Ar$ between the magmatic endmembers at CAVA and the slab is consistent with a mantle wedge being overwhelmed by the addition of slab-derived neon and argon. Samples with near-zero Δ_{30} also have atmospheric ${}^{40}Ar/{}^{36}Ar$ values (Table 1), indicating that Ar from the mantle wedge has retained the noble gas isotopic signature of its subducted atmospheric source. This is likely explained by the contribution of surface-derived argon with atmospheric ${}^{40}Ar/{}^{36}Ar$ ratios to the mantle wedge. This sheds new light on atmospheric ${}^{40}Ar/{}^{36}Ar$ observed systematically in all known Central American fumaroles (Snyder et al. 2003) and essentially all gas discharges sampled to date from subduction zones (Hilton et al., 2002, Sano and Fischer, 2013): A likely origin for the atmospheric Ar in high-T endmembers is devolatilizing subducted protoliths that are otherwise known to have atmospheric Ar isotopic signatures (Chavrit et al., 2016; Holland and Ballentine, 2006; Staudacher and Allègre, 1988). Our high-T data shows that not all of the subducted air-derived Ar is transferred into the deeper mantle and that some of it is released via arc volcanism. Our data also dispels the notion that atmospheric Ar-isotope

signatures of arc-gases are exclusively the result of shallow atmospheric contamination in the hydrothermal system.

5.7 Revisiting volatile fluxes in the central American subduction zone.

The perspective that volatiles from the mantle wedge incorporate recycled argon is problematic, as it limits the use of nitrogen excesses, N_2^* , calculated on the basis of N_2/Ar ratios (Fischer et al. 1998,2002, Hilton et al., 2002). We show here that magmatic volatiles deliver measurable amounts of argon (with atmospheric $^{40}Ar/^{36}Ar$) to the fumaroles. Consequently, assumptions on N_2^* may naturally underestimates the outgassing flux of N_2 . Estimates for the fluxes of N_2 , Ar, and Ne can be obtained from 3He degassing fluxes. Using a combination of S/CO_2 , He/CO_2 and $^3He/^4He$ ratios, and an SO_2 flux of 21.3×10^9 mol/yr, Hilton et al. (2002) derived a 3He flux of 5.4 mol/yr for the entire CAVA. Using the CAVA CO_2 flux from Fischer et al., (2019) of 94×10^9 mol/yr, and a $CO_2/^3He$ ratio of 2×10^{10} (Kagoshima et al., 2015), we obtain a 3He flux 4.7 mol/yr. Here, we take a 3He flux of 5.0 mol/yr for the sake of illustration. We combine our new elemental ratios estimates for the CAVA source (Fig. 4-6) with 3He fluxes to derive new estimates for N_2 , ^{36}Ar , and ^{22}Ne outgassing fluxes in the central American arc.

We derive $N_2/^3He$ ratios between 8×10^7 and 2×10^8 for magmatic volatiles in Central America (Fig. 5), from samples with $\Delta_{30} < 5\text{‰}$. This results in a N_2 flux between 4.0×10^8 and 1.0×10^9 mol N_2/y (or 1.1×10^{10} and 2.8×10^{10} g N/ y). This range for the nitrogen mass flux is higher than previous estimates of 8.2×10^9 g N/ y (Fischer et al., 2002, Hilton et al., 2002). Our derived range of nitrogen fluxes overlaps the estimated flux of subducting nitrogen of 1.6×10^{10} g N/ y (1.1×10^9 mol N/y, or 5.7×10^8 mol N_2/yr), given with no

uncertainty in Busigny et al. (2019) and Li and Bebout (2005). This suggests that in the Central American subduction zone, the N₂ cycle is not required to be out of equilibrium. Instead, the revisited fluxes being identical within uncertainty, they allow N₂ to be quantitatively recycled through the mantle wedge and returned to the surface by degassing rather than being delivered to the deep mantle. An additional source of outgassing underestimation, here, is that we ignore forearc devolatilization for N₂, known to occur in the CAVA (Inguaggiato et al., 2004). Future work will be necessary to constrain this fraction of the N₂ flux that will inevitably increase the outgassing flux further. In any case, our newly defined N₂ outgassing fluxes are high, and in the range of subducting fluxes. Because this subduction zone is relatively “warm” and therefore more similar to subduction zones in the past over geological timescales (Keller and Schoene, 2018; Martin and Moyen, 2002), the finding that nitrogen is recycled by subduction and arc magmatism in the Central American system suggests that nitrogen subduction could have been inefficient through geological time. This conclusion is also in agreement with recent experimental work suggesting warm subduction zones limit nitrogen recycling in the deep mantle (Jackson et al., 2021).

We estimate the ³He/³⁶Ar of magmatic inputs to be $\sim 10^{-4}$ (Fig. 6). A ³He/³⁶Ar ratio of $\sim 10^{-4}$ leads to a degassing ³⁶Ar flux of $\sim 5.0 \times 10^4$ mol/y (1.8×10^6 g ³⁶Ar/y) in Central America. Using 1.4×10^{16} g of subducted crust, depths of ~ 6 km of oceanic crust and ~ 500 m of sediments, and the noble gases abundances in the oceanic crust (Chavrit et al., 2016), we derive a ³⁶Ar subduction flux of 1.1×10^8 cm³ STP/y, or 4.5×10^4 mol ³⁶Ar/y. Because of the large uncertainties in ³⁶Ar concentrations in rocks (Chavrit et al., 2016), this subduction flux estimate is probably associated with a $\sim 50\%$ uncertainty, and is presented here for illustration only. We note however that subduction and outgassing ³⁶Ar flux compare

favorably. Like nitrogen, our flux analysis seems to allow quantitative release of slab-derived ^{36}Ar to the mantle wedge, which would not require substantial ^{36}Ar recycling into the deep mantle in this subduction zone. The same conclusion holds for neon. Using a $^3\text{He}/^{22}\text{Ne}$ of $\sim 2 \times 10^{-2}$ (Fig. 6), we obtain a ^{22}Ne degassing flux of $\sim 2.5 \times 10^2$ mol/y (5.5×10^3 g ^{22}Ne /y). The $^{22}\text{Ne}/^{36}\text{Ar}$ of subducting components is largely variable, around a mode of $\sim 10^{-2}$ (Chavrit et al., 2016). This yields a ^{22}Ne subduction flux of 1.1×10^6 cm³ STP/y, or 4.5×10^2 mol ^{22}Ne /y. Again, although large uncertainties must be taken into consideration, the neon subduction and degassing fluxes appear equivalent, arguing in favor of quantitative ^{22}Ne recycling in the central American subduction zone back to the surface.

The simplest interpretation of the fluxes derived here is that the Central American subduction zone acts as a subduction barrier for N, Ar, and Ne: volcanic fluxes for N₂, Ar, and Ne, determined with the $^{15}\text{N}/^{14}\text{N}$ approach, *no longer require* subduction of these elements past the sub-arc melting region in the central American subduction zone.

6. Conclusion

Gas discharges in Central America are volatile mixtures involving contributions from at least three endmembers: air ($\pm$ fractionated air), crust, and mantle-derived components. Using Δ_{30} as a tracer of nitrogen from air in fumaroles and springs, we show that O₂/N₂ ratios are unreliable tracers of air in these systems. We also show that $\delta^{15}\text{N}$ and N₂/Ar ratios experienced fractionation during water degassing at depth, rendering these unreliable as signatures of air where liquid water is thermodynamically stable. This is problematic since the degassing fractionation may lead to $\delta^{15}\text{N}$ values that insidiously

mimic MORB gases. This does not seem to affect magmatic components, but more work is needed on Poás – where interaction with the hydrothermal system may be significant, before a firm conclusion can be reached. A preliminary conclusion is that without Δ_{30} data, the $\delta^{15}\text{N}$ - N_2 -Ar systematics may be deceptive, and can lead to confusion with regard to the origin of N_2 in mixed gases.

We derive estimates of N_2/He , He/Ar and He/Ne ratios that reliably exclude atmospheric components. N_2 , Ar and Ne enrichments by two or three orders of magnitude compared to a MORB source are observed. They are attributed to mantle wedge sources incorporating slab-derived gases. Using known ^3He degassing fluxes, we calculated N_2 , Ne and Ar degassing fluxes for the Central American arc that could balance subduction fluxes, within uncertainties, no longer requiring the Central American arc to have heavily imbalanced volatile cycles.

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552 **References**

- 553 Barry, P.H., Lawson, M., Meurer, W.P., Warr, O., Mabry, J.C., Byrne, D.J., Ballentine, C.J., 2016. Noble gases solubility models
554 of hydrocarbon charge mechanism in the Sleipner Vest gas field. *Geochim. Cosmochim. Acta* 194, 291–309.
- 555 Bebout, G.E., Agard, P., Kobayashi, K., Moriguti, T., Nakamura, E., 2013. Devolatilization history and trace element mobility
556 in deeply subducted sedimentary rocks: Evidence from Western Alps HP/UHP suites. *Chem. Geol.* 342, 1–20.
- 557 Bekaert, D.V., Turner, S.J., Broadley, M.W., Barnes, J.D., Halldórsson, S.A., Labidi, J., Wade, J., Walowski, K.J., Barry, P.H.,
558 2020. Subduction-Driven Volatile Recycling: A Global Mass Balance. *Annu. Rev. Earth Planet. Sci.* 49.
- 559 Busigny, V., Cartigny, P., Laverne, C., Teagle, D., Bonifacie, M., Agrinier, P., 2019. A re-assessment of the nitrogen
560 geochemical behavior in upper oceanic crust from Hole 504B: Implications for subduction budget in Central
561 America. *Earth Planet. Sci. Lett.* 525, 115735.
- 562 Busigny, V., Cartigny, P., Philippot, P., Ader, M., Javoy, M., 2003. Massive recycling of nitrogen and other fluid-mobile
563 elements (K, Rb, Cs, H) in a cold slab environment: evidence from HP to UHP oceanic metasediments of the Schistes
564 Lustrés nappe (western Alps, Europe). *Earth Planet. Sci. Lett.* 215, 27–42. [https://doi.org/10.1016/s0012-](https://doi.org/10.1016/s0012-821x(03)00453-9)
565 821x(03)00453-9
- 566 Carr, M.J., Feigenson, M.D., Patino, L.C., Walker, J.A., 2003. Volcanism and geochemistry in Central America: Progress and
567 problems. *Geophys. Monogr. Geophys. Union* 138, 153–174.
- 568 Chavrit, D., Burgess, R., Sumino, H., Teagle, D.A.H., Droop, G., Shimizu, A., Ballentine, C.J., 2016. The contribution of
569 hydrothermally altered ocean crust to the mantle halogen and noble gas cycles. *Geochim. Cosmochim. Acta* 183,
570 106–124.
- 571 De Leeuw, G.A.M., Hilton, D.R., Fischer, T.P., Walker, J.A., 2007. The He–CO₂ isotope and relative abundance characteristics
572 of geothermal fluids in el salvador and honduras: New constraints on volatile mass balance of the central american
573 volcanic arc. *Earth Planet. Sci. Lett.* 258, 132–146.
- 574 Elkins, L.J., Fischer, T.P., Hilton, D.R., Sharp, Z.D., McKnight, S., Walker, J., 2006. Tracing nitrogen in volcanic and
575 geothermal volatiles from the Nicaraguan volcanic front. *Geochim. Cosmochim. Acta* 70, 5215–5235.
- 576 Fischer, T.P., Hilton, D.R., Zimmer, M.M., Shaw, A.M., Sharp, Z.D., Walker, J.A., 2002. Subduction and recycling of nitrogen
577 along the Central American margin. *Science* (80-.). 297, 1154–1157.
- 578 Fischer, T.P., Ramírez, C., Mora-Amador, R.A., Hilton, D.R., Barnes, J.D., Sharp, Z.D., Le Brun, M., de Moor, J.M., Barry, P.H.,
579 Füri, E., Shaw, A.M., 2015. Temporal variations in fumarole gas chemistry at Poás volcano, Costa Rica. *J. Volcanol.*
580 *Geotherm. Res.* 294, 56–70. <https://doi.org/10.1016/j.jvolgeores.2015.02.002>

581 Fischer, T.P., Takahata, N., Sano, Y., Sumino, H., Hilton, D.R., 2005. Nitrogen isotopes of the mantle: Insights from mineral
 582 separates. *Geophys. Res. Lett.* 32.
 583 Gazel, E., Carr, M.J., Hoernle, K., Feigenson, M.D., Szymanski, D., Hauff, F., Van Den Bogaard, P., 2009. Galapagos-OIB
 584 signature in southern Central America: Mantle refertilization by arc-hot spot interaction. *Geochemistry, Geophys.*
 585 *Geosystems* 10.
 586 Hilton, D.R., Fischer, T.P., Marty, B., 2002. Noble gases and volatile recycling at subduction zones. *Rev. Mineral.*
 587 *geochemistry* 47, 319–370.
 588 Hilton, D.R., Ramirez, C.J., Mora-Amador, R., Fischer, T.P., Füre, E., Barry, P.H., Shaw, A.M., 2010. Monitoring of temporal and
 589 spatial variations in fumarole helium and carbon dioxide characteristics at Poás and Turrialba volcanoes, Costa Rica
 590 (2001-2009). *Geochem. J.* 44, 431–440.
 591 Hoernle, K., Abt, D.L., Fischer, K.M., Nichols, H., Hauff, F., Abers, G.A., Van Den Bogaard, P., Heydolph, K., Alvarado, G., Protti,
 592 M., 2008. Arc-parallel flow in the mantle wedge beneath Costa Rica and Nicaragua. *Nature* 451, 1094–1097.
 593 Holland, G., Ballentine, C.J., 2006. Seawater subduction controls the heavy noble gas composition of the mantle. *Nature*
 594 441, 186–191. <https://doi.org/10.1038/nature04761>
 595 Inguaggiato, S., Taran, Y., Grassa, F., Capasso, G., Favara, R., Varley, N., Faber, E., 2004. Nitrogen isotopes in thermal fluids
 596 of a forearc region (Jalisco Block, Mexico): Evidence for heavy nitrogen from continental crust. *Geochemistry,*
 597 *Geophys. Geosystems* 5.
 598 Jackson, C.R.M., Cottrell, E., Andrews, B., 2021. Warm and oxidizing slabs limit ingassing efficiency of nitrogen to the
 599 mantle. *Earth Planet. Sci. Lett.* 553, 116615.
 600 Javoy, M., Pineau, F., 1991. The volatiles record of a “popping” rock from the Mid-Atlantic Ridge at 14°N: chemical and
 601 isotopic composition of gas trapped in the vesicles. *Earth Planet. Sci. Lett.* 107, 598–611.
 602 Keller, B., Schoene, B., 2018. Plate tectonics and continental basaltic geochemistry throughout Earth history. *Earth Planet.*
 603 *Sci. Lett.* 481, 290–304.
 604 Kennedy, B.M., Hiyagon, H., Reynolds, J.H., 1991. Noble gases from Honduras geothermal sites. *J. Volcanol. Geotherm. Res.*
 605 45, 29–39.
 606 Labidi, J., Barry, P.H., Bekaert, D. V., Broadley, M.W., Marty, B., Giunta, T., Warr, O., Lollar, B.S., Fischer, T.P., Avicé, G., 2020.
 607 Hydrothermal ^{15}N abundances constrain the origins of mantle nitrogen. *Nature* 580, 367–371.
 608 Lee, H., Fischer, T.P., Muirhead, J.D., Ebinger, C.J., Kattenhorn, S.A., Sharp, Z.D., Kianji, G., Takahata, N., Sano, Y., 2017.
 609 Incipient rifting accompanied by the release of subcontinental lithospheric mantle volatiles in the Magadi and
 610 Natron basin, East Africa. *J. Volcanol. Geotherm. Res.* 346, 118–133.
 611 Lee, H., Sharp, Z.D., Fischer, T.P., 2015. Kinetic nitrogen isotope fractionation between air and dissolved N_2 in water:

612 Implications for hydrothermal systems. *Geochem. J.* 49, 571–573.

613 Lee, J.-Y., Marti, K., Severinghaus, J.P., Kawamura, K., Yoo, H.-S., Lee, J.B., Kim, J.S., 2006. A redetermination of the isotopic
614 abundances of atmospheric Ar. *Geochim. Cosmochim. Acta* 70, 4507–4512.

615 Li, L., Bebout, G.E., 2005. Carbon and nitrogen geochemistry of sediments in the Central American convergent margin:
616 Insights regarding subduction input fluxes, diagenesis, and paleoproductivity. *J. Geophys. Res. Solid Earth* 110.

617 Libourel, G., Marty, B., Humbert, F., 2003. Nitrogen solubility in basaltic melt. Part I. Effect of oxygen fugacity. *Geochim.*
618 *Cosmochim. Acta* 67, 4123–4135. [https://doi.org/10.1016/s0016-7037\(03\)00259-x](https://doi.org/10.1016/s0016-7037(03)00259-x)

619 Martin, H., Moyen, J.-F., 2002. Secular changes in tonalite-trondhjemite-granodiorite composition as markers of the
620 progressive cooling of Earth. *Geology* 30, 319–322.

621 Marty, B., Humbert, F., 1997. Nitrogen and argon isotopes in oceanic basalts. *Earth Planet. Sci. Lett.* 152, 101–112.

622 Mikhail, S., Sverjensky, D.A., 2014. Nitrogen speciation in upper mantle fluids and the origin of Earth's nitrogen-rich
623 atmosphere. *Nat. Geosci.* 7, 816–819. <https://doi.org/10.1038/ngeo2271>
624 <http://www.nature.com/ngeo/journal/v7/n11/abs/ngeo2271.html#supplementary-information>

625 Moreira, M., Kunz, J., Allegre, C., 1998. Rare gas systematics in popping rock: isotopic and elemental compositions in the
626 upper mantle. *Science* (80-.). 279, 1178–1181.

627 Mukhopadhyay, S., 2012. Early differentiation and volatile accretion recorded in deep-mantle neon and xenon. *Nature*
628 486, 101–104. <https://doi.org/10.1038/nature11141>

629 Patino, L.C., Carr, M.J., Feigenson, M.D., 2000. Local and regional variations in Central American arc lavas controlled by
630 variations in subducted sediment input. *Contrib. to Mineral. Petrol.* 138, 265–283.

631 Peacock, S.M., van Keken, P.E., Holloway, S.D., Hacker, B.R., Abers, G.A., Fergason, R.L., 2005. Thermal structure of the Costa
632 Rica–Nicaragua subduction zone. *Phys. Earth Planet. Inter.* 149, 187–200.

633 Plank, T., Kelley, K.A., Zimmer, M.M., Hauri, E.H., Wallace, P.J., 2013. Why do mafic arc magmas contain ~ 4 wt% water on
634 average? *Earth Planet. Sci. Lett.* 364, 168–179.

635 Ranero, C.R., von Huene, R., 2000. Subduction erosion along the Middle America convergent margin. *Nature* 404, 748–752.

636 Raquin, A., Moreira, M.A., Guillon, F., 2008. He, Ne and Ar systematics in single vesicles: mantle isotopic ratios and origin of
637 the air component in basaltic glasses. *Earth Planet. Sci. Lett.* 274, 142–150.

638 Schwarzenbach, E.M., Gill, B.C., Gazel, E., Madrigal, P., 2016. Sulfur and carbon geochemistry of the Santa Elena peridotites:
639 Comparing oceanic and continental processes during peridotite alteration. *Lithos* 252–253, 92–108.
640 <https://doi.org/http://dx.doi.org/10.1016/j.lithos.2016.02.017>

641 Snyder, G., Poreda, R., Fehn, U., Hunt, A., 2003. Sources of nitrogen and methane in Central American geothermal settings:
642 Noble gas and ¹²⁹I evidence for crustal and magmatic volatile components. *Geochemistry, Geophys. Geosystems* 4,

643 1–28.

644 Staudacher, T., Allègre, C.J., 1988. Recycling of oceanic crust and sediments: the noble gas subduction barrier. *Earth*

645 *Planet. Sci. Lett.* 89, 173–183.

646 Taran, Y.A., 2009. Geochemistry of volcanic and hydrothermal fluids and volatile budget of the Kamchatka–Kuril

647 subduction zone. *Geochim. Cosmochim. Acta* 73, 1067–1094.

648 Vaselli, O., Tassi, F., Minissale, A., Montegrossi, G., Duarte, E., Fernandez, E., Bergamaschi, F., 2003. Fumarole migration and

649 fluid geochemistry at Poás volcano (Costa Rica) from 1998 to 2001. *Geol. Soc. London, Spec. Publ.* 213, 247–262.

650 Warr, O., Rochelle, C.A., Masters, A., Ballentine, C.J., 2015. Determining noble gas partitioning within a CO₂–H₂O system at

651 elevated temperatures and pressures. *Geochim. Cosmochim. Acta* 159, 112–125.

652 Yeung, L.Y., Li, S., Kohl, I.E., Haslun, J.A., Ostrom, N.E., Hu, H., Fischer, T.P., Schauble, E.A., Young, E.D., 2017. Extreme

653 enrichment in atmospheric ¹⁵N/¹⁵N. *Sci. Adv.* 3, eaao6741.

654 Young, E.D., Rumble III, D., Freedman, P., Mills, M., 2016. A large-radius high-mass-resolution multiple-collector isotope

655 ratio mass spectrometer for analysis of rare isotopologues of O₂, N₂, CH₄ and other gases. *Int. J. Mass Spectrom.*

656 401, 1–10.

657 Zelenski, M., Taran, Y., 2011. Geochemistry of volcanic and hydrothermal gases of Mutnovsky volcano, Kamchatka:

658 evidence for mantle, slab and atmosphere contributions to fluids of a typical arc volcano. *Bull. Volcanol.* 73, 373–

659 394.

660 Zimmer, M.M., Fischer, T.P., Hilton, D.R., Alvarado, G.E., Sharp, Z.D., Walker, J.A., 2004. Nitrogen systematics and gas fluxes

661 of subduction zones: insights from Costa Rica arc volatiles. *Geochemistry, Geophys. Geosystems* 5.

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663

664 **Captions**

		PO 06-1-3	PO 03-2	Pnar 06-1	PO 06-1-1	PO 06-1-2	ES02 10	Nic-3	Nic-2
		Poas 11/5/06	Poas 3/31/03	Poas 2/24/06	Poas 2/24/06	Poas 2/24/06	Santa Ana	Momotombo 1/5/02	Momotombo 1/5/02
dates of collection									
latitude		10.12	10.12	10.12	10.12	10.12	13.5111	12.2519	12.2519
longitude		-84.1359	-84.1359	-84.1359	-84.1359	-84.1359	-89.3748	-86.3224	-86.3224
temperature (C°)		113	98	153	113	113	400	747	747
quantity of processed N ₂	10 ⁻⁶ mol	2.1	2.2	8.8	8.0	5.8	22.4		
N ₂ vol fraction (ucla)	x10 ⁻²	9.21E-01	8.47E-01	3.10E-01	7.43E-01	6.01E-01	7.14E-01		
d ¹⁵ N	vs. air	-3.69	0.35	-1.38	-1.38	-1.7	0.2	5.41	3.89
D ₃₀	vs. stochastic	17	3.43	10.92	10.25	11	15.5	1.5	3.9
1 se d ¹⁵ N		0.032	0.018	0.02	0.021	0.022	0.008	0.1	0.1
1 se D ₃₀		0.496	0.527	0.288	0.309	0.307	0.138	0.3	0.2
CO ₂		0.992	0.990	0.975	0.993	0.990	0.983	0.859	0.861
He		4.56E-05	5.00E-06	5.59E-06	4.97E-06	8.53E-06	6.54E-06	7.61E-06	1.66E-05
H ₂		1.17E-04	3.25E-04	1.40E-02	6.62E-05	7.05E-05	4.64E-03	1.24E-01	1.20E-01
Ar		2.42E-04	1.12E-04	1.24E-04	3.10E-04	4.21E-04	9.98E-05	1.52E-04	3.70E-05
O ₂		4.19E-04	1.25E-05	1.45E-04	5.28E-04	7.90E-04	4.54E-06	1.14E-05	8.93E-06
N ₂		6.70E-03	1.00E-02	1.06E-02	6.56E-03	8.94E-03	1.20E-02	1.62E-02	1.88E-02
CH ₄		6.44E-06	1.25E-06	1.20E-04	7.24E-06	9.48E-06	1.95E-05	8.88E-06	7.66E-06
He/Ar		1.89E-01	4.44E-02	4.50E-02	1.60E-02	2.03E-02	6.55E-02	5.00E-02	4.48E-01
N ₂ /He		1.47E+02	2.00E+03	1.91E+03	1.32E+03	1.05E+03	1.83E+03	2.13E+03	1.13E+03
N ₂ /Ar		2.77E+01	8.89E+01	8.58E+01	2.11E+01	2.12E+01	1.20E+02	1.07E+02	5.07E+02
O ₂ /N ₂		6.25E-02	1.25E-03	1.36E-02	8.04E-02	8.83E-02	3.80E-04	7.03E-04	4.76E-04
N ₂ /CH ₄		1.04E+03	8.00E+03	8.87E+01	9.06E+02	9.43E+02	6.14E+02	1.83E+03	2.45E+03
⁴⁰ Ar/ ³⁶ Ar					297	293		305	
³ He/ ⁴ He	(R _A)	6.44	7.15	6.96	7.15	7.15	7.56	6.99	6.99
N ₂ / ³ He		1.63E+07	2.00E+08	1.96E+08	1.32E+08	1.05E+08	1.73E+08	2.18E+08	1.16E+08
N ₂ / ³⁶ Ar		8.49E+03	2.72E+04	2.62E+04	6.47E+03	6.49E+03		3.26E+04	1.78E+05
⁴ He/ ²⁰ Ne		8.93E+00	1.90E+02	3.61E+02	1.56E+02	1.56E+02	6.66E+01	2.75E+02	2.75E+02
³ He/ ²² Ne		8.04E-04	1.90E-02	3.52E-02	1.56E-02	1.56E-02	7.05E-03	2.69E-02	2.69E-02
X value		28	596	1133	488	488	209	864	864
³ He/ ³⁶ Ar		5.20E-04	1.36E-04	1.34E-04	4.90E-05	6.20E-05		1.50E-04	
⁴ He/ ⁴⁰ Ar		1.89E-01	4.46E-02	4.51E-02	1.61E-02	2.03E-02		5.02E-02	

665

666 **Table 1. Nitrogen and light noble gases isotope and concentrations for central**
667 **Americans fumaroles**

668

sample ID		LH180406	CL180409	CH180410	PX180416	PS180405	BS180407	HA180403	CW180415	XF180416
spring name		Los Pozos	Colibre	Chiguit Abajo	San Juan Praxair	Playa Sando	Bajo Mended	Hatillo	Cauhita	San Juan Praxair
latitude		8.870954	8.404478	8.705078	10.480755	8.575544	8.66645	9.360224	9.735746	10.485523
longitude		-82.689901	-80.803745	-80.269193	-84.113598	-83.36416	-82.349098	-83.916639	-82.825737	-84.113229
temperature (C°)		55	51	31	29	31	41	33	35	29
quantity of processed N ₂	10 ⁻⁶ mol	18.0	540.0	7.2	38.0	28.0	588.0	556.0	39.2	2.8
N ₂ vol fraction (ucla)	x10 ⁻²	2.89E-02	8.67E-01	1.16E-02	6.10E-02	4.49E-02	9.44E-01	8.92E-01	6.29E-02	4.49E-03
d ¹⁵ N	vs. air	-1.38	0.55	1.60	0.20	0.71	4.19	0.22	0.34	1.30
D ₁₀	vs. stochastic	18.39	15.27	12.52	19.35	18.42	8.73	7.29	18.47	15.09
1 se d ¹⁵ N		0.029	0.004	0.028	0.007	0.008	0.004	0.004	0.004	0.031
1 se D ₁₀		0.825	0.099	0.563	0.22	0.24	0.168	0.077	0.097	0.617
CO ₂		0.997	0.018	0.985	0.944	0.082		0.031	0.054	
He		1.47E-07	4.98E-04	1.54E-05	1.97E-06	1.90E-06		4.19E-04	1.04E-05	
H ₂		9.07E-07	3.92E-06	1.77E-07	9.81E-08	1.51E-05		7.48E-05	1.62E-05	
Ar		4.79E-05	1.12E-02	3.76E-04	3.52E-04	1.07E-03		6.90E-03	1.71E-03	
O ₂		5.75E-04	4.78E-03	1.87E-03	1.19E-02	2.77E-03		3.12E-02	1.10E-02	
N ₂		2.04E-03	9.63E-01	1.32E-02	4.41E-02	5.03E-02		9.30E-01	9.92E-02	
CH ₄		9.62E-08	2.52E-03	4.53E-05	2.66E-06	8.64E-01		3.09E-04	8.34E-01	
He/Ar		3.07E-03	4.46E-02	4.09E-02	5.60E-03	1.77E-03	2.57E-03	6.07E-02	6.09E-03	3.85E-02
N ₂ /He		1.39E+04	1.93E+03	8.54E+02	2.23E+04	2.65E+04	4.23E+04	2.22E+03	9.56E+03	1.21E+03
N ₂ /Ar		4.27E+01	8.62E+01	3.50E+01	1.25E+02	4.69E+01	1.09E+02	1.35E+02	4.64E+01	4.64E+01
O ₂ /N ₂		2.81E-01	4.96E-03	1.42E-01	2.69E-01	5.50E-02		3.36E-02	1.11E-01	
N ₂ /CH ₄		2.12E+04	3.83E+02	2.91E+02	1.65E+04	5.82E-02		3.01E+03	1.19E-01	
⁴ He	x10 ⁻⁹ cc/ccSTP	3.21E+01	4.39E+05	1.28E+04	1.12E+03	1.65E+02	2.23E+04	3.98E+05	4.40E+03	3.72E+03
²⁰ Ne	x10 ⁻⁹ cc/ccSTP	2.24E+00	1.37E+04	1.23E+02	6.81E+02	3.98E+02	1.58E+04	7.07E+03	6.15E+02	5.85E+01
⁴⁰ Ar	x10 ⁻⁶ cc/ccSTP	3.85E+00	1.15E+04	3.59E+02	3.95E+02	4.83E+02	8.67E+03	6.35E+03	7.49E+02	9.65E+01
³⁶ Ar	x10 ⁻⁶ cc/ccSTP	1.28E+01	3.74E+04	1.05E+03	1.29E+03	1.59E+03	2.67E+04	2.06E+04	2.43E+03	3.16E+02
⁴⁰ Ar/ ³⁶ Ar	(R _A)	301	308	342	305	304	325	308	309	306
³ He/ ⁴ He		7.48	8.78	6.37	6.52	1.55	0.27	1.82	2.05	7.59
²⁰ Ne/ ²² Ne		10.00	9.83	9.89	9.82	9.84		9.85	9.87	9.91
²¹ Ne/ ²² Ne		0.023	0.029	0.029	0.029	0.029		0.029	0.029	0.029
N ₂ / ⁴ He		1.33E+09	1.57E+08	9.59E+07	2.45E+09	1.22E+10	1.12E+11	8.72E+08	3.33E+09	1.14E+08
N ₂ / ²⁰ Ne		1.29E+04	2.66E+04	1.20E+04	3.83E+04	1.43E+04	3.53E+04	4.17E+04	1.80E+04	1.42E+04
⁴ He/ ²⁰ Ne		1.43E+01	3.21E+01	1.04E+02	1.65E+00	4.15E-01	1.41E+00	5.63E+01	7.15E+00	6.37E+01
³ He/ ²¹ Ne		1.50E-03	3.87E-03	9.19E-03	1.48E-04	8.85E-06	5.34E-06	1.41E-03	2.02E-04	6.70E-03
X value		45	101	327	5	1	4	177	22	200
³ He/ ²⁰ Ar		9.70E-06	1.69E-04	1.25E-04	1.57E-05	1.17E-06	3.16E-07	4.78E-05	5.41E-06	1.25E-04
⁴ He/ ²⁰ Ar		3.08E-03	4.47E-02	4.10E-02	5.62E-03	1.77E-03	2.50E-03	6.09E-02	6.11E-03	

Table 2 Nitrogen and light noble gases isotope and concentrations for central American springs

Fig. 1: location of all the samples studied here.

Green, red and blue symbols are fumarole locations from Santa Ana, Momotombo and Poás (Data in table 1). Yellow symbols are springs (Data in table 2).

Fig. 2 the nitrogen isotopic composition of volcanic discharges in central America.

Variable Δ_{30} values establish that our samples incorporate variable amounts of atmospheric nitrogen. Fumaroles from the Poás and Momotombo have the lowest Δ_{30} values indicating they have the lowest air-derived N_2 contributions while the hot springs from Costa Rica have the highest air contributions. Variable $\delta^{15}\text{N}$ values are observed, between $-3.7 \pm 0.3\text{‰}$ and $+4.2 \pm 0.3\text{‰}$. At an air Δ_{30} values, variable $\delta^{15}\text{N}$ must be caused by a mass-dependent isotope fractionation, presumably associated with hydrothermal degassing. The high-temperature components have positive $\delta^{15}\text{N}$ values.

686

687 **Fig. 3: Measured $^3\text{He}/^4\text{He}$ ratios versus $^4\text{He}/^{20}\text{Ne}$ ratios of the samples, normalized to air by**
 688 **convention (De Leeuw et al, 2007).** Most samples indicate simple two-component air-magma
 689 mixing. Three Costa Rica hot springs (HS) however show additions of crustal ^4He .

Fig. 4: Measured N_2/Ar and $N_2/^{36}Ar$ ratios versus Δ_{30} values. Mixing lines are shown for mixtures between the high-temperature component observed at both Momotombo and Poás, and variably fractionated atmospheric components, including air, air-saturated water (ASW), and degassed water. The high-temperature endmembers with $\Delta_{30} = 0\text{‰}$ defined by the data from Momotombo and Poás have similar N_2/Ar and $N_2/^{36}Ar$ ratios of ~ 100 and $\sim 10^4$, respectively. Here, we adopt values of ~ 100 and 3×10^4 for N_2/Ar and $N_2/^{36}Ar$ ratios for the high-T endmember at Momotombo. Nitrogen and argon have indistinguishable solubilities in silicate melts and thus are not fractionated by magmatic degassing at $\sim 1200^\circ\text{C}$ (Libourel et al., 2003). Thus, the low $N_2/^{36}Ar$ at Momotombo is not resulting from fractionations during magmatic degassing. Instead, it may be a feature of the Momotombo mantle source. At Poás, fumarole samples are consistent with the same N_2/Ar ratio for the high-temperature component as at Momotombo. If hydrothermal degassing affected the high-T volatiles at Poás, the pristine N_2/Ar and $N_2/^{36}Ar$ for high-T gases at Poás could be higher than the observed values. Note that combining a $^3He/^{36}Ar$ ratio of $\sim 10^{-4}$ (see Fig. 6B) to the observed $N_2/^3He$ of $\sim 10^8$ (Fig. 4) returns to a $N_2/^{36}Ar$ of $\sim 10^4$, consistent with the conclusion of a low N_2/Ar in our sampling of high-temperature endmembers (see section 5.4).

710 **Fig. 5: Measured N_2/He and $N_2/^3He$ ratios versus Δ_{30} values.** Mixing lines are shown for
711 mixtures between air and three high-temperature components: MORBs, and two variably N_2 -
712 enriched endmembers required to fit the data. Springs with additions of crustal 4He are shown
713 in the smallest symbols. They are not used for fluxes calculations – see discussion in the
714 supplementary discussion.

Fig. 6: Measured $^3\text{He}/^{22}\text{Ne}$ and $^3\text{He}/^{36}\text{Ar}$ ratios versus Δ_{30} values. Springs with additions of crustal ^4He are shown in the smallest symbols. They are not used for fluxes calculations – see discussion in the supplementary discussion. Mixing lines are the same as in figure 3 and 4. They are shown for mixtures between air and high-temperature components: MORB, and a N_2 - ^{36}Ar - ^{22}Ne enriched endmember that fits the data. The mixing relationships are curved because of the logarithm scale, but also because of how distinct the N_2/Ne of the mixing endmembers may be. Air has a $\text{N}_2/^{20}\text{Ne}$ ratio of air is 4.5×10^4 . For MORB, we calculate a the $\text{N}_2/^{20}\text{Ne}$ of 1.3×10^6 . This is based on a $\text{N}_2/^3\text{He}$ of 3×10^6 for the MORB mantle (Javoy and Pineau, 1991). We combine this $\text{N}_2/^3\text{He}$ estimate with $^3\text{He}/^4\text{He}$ and $^4\text{He}/^{20}\text{Ne}$ values for the MORB mantle from Moreira et al (1998). Using the $^3\text{He}/^{22}\text{Ne}$ ratio of the MORB mantle (Moreira et al., 1998), we derive a $\text{N}_2/^{22}\text{Ne}$ of 1.7×10^7 . Conversely, we here derive $\text{N}_2/^3\text{He}$, $^3\text{He}/^{22}\text{Ne}$ ratios of the central American high-temperature volatiles. Combining those with $^3\text{He}/^4\text{He}$ of ~ 7 from Fig. 3, we derive $\text{N}_2/^{20}\text{Ne}$ and $\text{N}_2/^{22}\text{Ne}$ ratios of 3.4×10^5 and 3.3×10^6 , respectively. On panel b, the curvatures are function the N_2/Ar of the mixing endmembers, which are estimated on Fig. 4.

Fig. 7 Cartoon representing various processes and reservoirs constrained in this study. Scales are grossly exaggerated, especially for melt conduits and magma chambers. A slab with sediments, basalts, and oceanic lithospheric mantle is shown to enter subduction. The slab composition is not constrained here but is likely enriched in all volatiles relative to ^3He (Busigny et al., 2019, Chavrit et al., 2016). In the slab, nitrogen is fixed as NH_4^+ (Busigny et al., 2019 and

references therein) so no Δ_{30} values are defined; Δ_{30} is only relevant for N_2 molecules. Both slab devolatilization and slab melting are shown for illustration. Any slab melting would be relevant underneath Costa Rica (Hoernle et al., 2008) while slab devolatilization and sediment-derived fluids would contribute to sources underneath Nicaragua and El Salvador (Patino et al., 2000). The melting region is characterized as the high-temperature endmembers constrained in this work (Fig. 7). Melts and super-critical fluids are suggested to host dissolved N_2 (Libourel et al., 2003; Mikhail and Sverjensky, 2014). Upon partial melting of a NH_3 -bearing mantle source, magmatic N_2 would form with a Δ_{30} of 0‰. The high-T N_2 is then contributed to hydrothermal systems in the sub-surface by near-quantitative magmatic degassing. Air circulation in the sub-surface allows air-saturated waters to undergo degassing, and subsequent gas release with fractionated compositions.