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Growth kinetics and origin of residual stress of two-phase crystalline-amorphous nanostructured films

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Composition-driven amorphous-to-crystalline transition is widely observed in sputter-deposited thin films. We have recently shown that peculiar two-phase crystalline-amorphous nanostructures can be obtained for compositions in the transition zone between single-phased amorphous and crystalline films, offering a new possibility to control surface topography and related functional properties. Here, the growth kinetics of the two-phase nanostructures formed in the transition zone is explored and related to residual stress measurements. From the analysis of top-view scanning electron microscopy images, the evolution with the film thickness of relevant parameters describing the growth process was extracted, i.e., surface coverage, volume fraction and perimeter per unit area of the crystalline regions growing in competition with the amorphous ones. We demonstrate, supported in the aforementioned parameters, that residual stress evolution in the transition zone is governed by the nucleation, growth and interaction of the crystalline regions during the competitive growth process. Our results shed new light on the kinetics of the crystalline-amorphous competitive growth phenomenon and demonstrate the relevance of residuals stress measurements for exploring the growth mechanisms of complex film microstructures.

I. INTRODUCTION

Understanding the origin of residual stresses in thin films is of crucial importance in various practical applications. High levels of residual stress can be detrimental to the mechanical stability and adhesion of the film to the substrate, resulting in cracking and delamination¹⁻⁴. In this sense, an interesting strategy for stress reduction consist in the growth of a nanostructured compliant layer of a low-density material sandwiched between the film and the substrate^{5,6}. In addition to the pragmatic problem of preserving film quality, residual stress can directly influence many of the film's properties, such as mechanical^{7,8}, magnetic^{9,10}, optical¹¹ or electrical¹², to cite a few. As in first instance is the microstructure which determines the properties, establishing a link between residual stress and microstructure is of primordial importance and has been the object of many publication through the years. A good example of this are the well-known structure zone models (SZM)¹³⁻¹⁶, which relate the most relevant deposition conditions (working-gas pressure, substrate temperature, or any other parameter causing a transfer of energy and/or momentum to the atoms arriving at the film surface) with the microstructure of the deposited film. Effectively, it has been largely observed that increasing the working gas pressure in sputter-deposited films, a transition from a denser (zone T) to a porous and columnar (zone 1) microstructure occurs which is associated to a change of residual stress from compressive to tensile¹⁷⁻²². More

recently, many studies have focused on the evolution of stress in the early-stages of film growth, greatly advancing the understanding of the atomic-level mechanisms of stress generation²³⁻²⁷. Overall, in most of these studies, mono-phased films are considered, and reports relating the evolution of residual stress during the growth of films with multi-phased microstructures are still scarce. Additional efforts are needed to define parameters at the origin of stress built-up in such films, which is at the core of the present study.

In previous works^{28,29}, we reported on the mechanisms of competitive growth between amorphous and crystalline phases in thin films, leading to original microstructures consisting of a crystalline phase of feather-like morphology embedded in an amorphous matrix. By working at compositions in the transition zone between those needed to obtain single-phased amorphous or crystalline films, the desired microstructure is obtained as a consequence of the competitive growth of the amorphous and the crystalline phases. Thereby, surface-dependent functional properties, such as optical reflectivity and wettability can be manipulated to a large extent. Although these studies primarily focused on the binary Zr-Mo and Zr-W systems, similar microstructures have been observed in Ti-Al³⁰, Ti-O³¹ and Al-N³² films, suggesting the possibility to generalize the concept. Other authors have recently reported the phenomenon in the Zr-W system using different deposition conditions³³. Moreover, we have also shown that it occurs under a wide range of deposition conditions. For instance, the shape of the crystalline domains can be modified by selecting low or high pressure conditions²⁸. In this crystalline-amorphous competitive growth process, the crystalline and amorphous regions can easily be identified and the kinetics of the competitive growth can be modeled through access to

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FIG. 1. Amorphous-to-crystalline transition in sputter-deposited Zr-W thin films. (a) Schematic of the sputtering chamber showing the experimental conditions used to obtain the alloys. By varying the discharge current applied to the cathodes, the chemical composition of the films can be controlled. (b) X-ray diffractograms of the films ($\sim 1000 \text{ nm}$ thick) revealing the structural transition associated to the change in composition. Increasing the W content, the films structure evolves from amorphous (at 58 at.% W) to crystalline (at 80 at.% W). In the transition zone (at 72 at.% W) the X-ray signal is a mixture of the amorphous and crystalline signals. (c) Top view SEM images of the films revealing the separation of the amorphous and crystalline phases occurring in the transition zone. Scale bar: $1 \mu\text{m}$.

the shape of crystalline regions. Hence, this type of films can serve as a platform to understand and model stress-microstructure evolution in nanostructured thin films.

Here, we continue the exploration of these peculiar two-phase crystalline-amorphous nanostructures. Using Zr-W thin films as a model system, new insights into the kinetics of this phenomenon are presented. In particular, the growth kinetics of the obtained two-phase nanostructures is linked with residual stress measurements.

II. EXPERIMENTAL DETAILS

Thin films synthesis: Zr-W thin film alloys were obtained by DC magnetron co-sputtering of Zr and W metallic targets in the presence of argon (see Fig. 1a). The films were deposited on (100) silicon single crystal substrates. The depositions were made with the substrate holder in rotation mode in order to ensure a good lateral homogeneity of the film composition and thick-

ness. The sputtering chamber was pumped down via a mechanical and a turbo-molecular pump allowing a base vacuum of 10^{-6} Pa . The cathodes were mounted in a confocal configuration and the targets to substrate distance was fixed at 9 cm. An argon flow rate of 30 sccm and a working argon pressure of 3 Pa were used. Zirconium and tungsten targets (50.8 mm diameter, 3 mm thick and purity higher than 99.9%) were powered by an Maxim 1000 DC generator. A fixed discharge current of 0.3 A was applied to the Zr target meanwhile the discharge current applied to the W target was varied from 0.3 to 0.7 A between experiments and depending on the desired chemical composition of the films. The films were deposited without external heating. The self-established temperature at the substrate was measured using a thermocouple and did not exceed 55°C during growth.

Thin films characterization: X-ray diffraction (XRD) measurements were conducted in the Bragg-Brentano configuration with $K_\alpha\text{Cu}$ radiation ($\lambda = 1.5406 \text{ \AA}$) using an AXS Bruker D8 Advance diffractometer. Transmis-

sion electron microscopy (TEM) analysis was performed using a cold FEG JEOL ARM200 microscope. Cross-sectional TEM samples of films were prepared using a focused ion beam (FIB)-scanning electron microscope dual beam system (FEI Helios 600). Throughout the FIB process, the time during the ionic cuts was the shortest possible to avoid any heating effect. Top and cross-sectional scanning electron micrographs of the films were taken in a Philips XL-30 S-FEG scanning electron microscope (SEM) using an accelerating voltage of 3 kV.

Residual stress measurements: The residual stress of films deposited on Si(100) substrates of $0.5 \times 2.5 \text{ cm}^2$ was determined using the Stoney's equation³⁴:

$$\sigma = \frac{M_s}{6} \frac{h_s^2}{h_f} \left(\frac{1}{R} - \frac{1}{R_0} \right) \quad (1)$$

where h_f is the film thickness, h_s is the substrate thickness ($200 \text{ }\mu\text{m}$), $M_s = 180.3 \text{ GPa}$ is the biaxial modulus of the substrate, R_0 is the curvature radius of the substrate before the film deposition and R is the curvature radius after the film deposition. Film thicknesses and curvature radii were measured using a DektakXT BRUKER profilometer.

Processing of top view SEM images: Surface coverage by the nanocrystalline phase was determined using the software Image J by processing the top view SEM images. The procedure followed is briefly described here (for more details see²⁹). First, a *Bandpass Filter* was applied to the SEM image, which allows better delimiting the nanocrystalline and amorphous phases. After, a careful selection of a *Threshold* was done, allowing us to obtain a black and white image. Then, the boundaries delimiting neighboring nanocrystalline nuclei in contact were manually marked. This permits to apply a *Fill Holes* function, yet keeping the nanocrystalline nuclei correctly separated. After that, the surface coverage could be calculated straightforwardly using the software.

III. RESULTS AND DISCUSSION

A. Two-phase crystalline-amorphous nanostructures in the Zr-W system: kinetics

In sputter-deposited Zr-W thin film alloys synthesized without thermal assistance, a transition from an amorphous to a nanocrystalline structure is expected upon increasing the W content^{35,36}. By changing the discharge current applied to one of the targets (Fig. 1a), the chemical composition of the alloy can be controlled. This change in composition results in films with different structural features (Fig. 1b). Effectively, the X-ray diffractogram of films presenting 58 at.% W shows a wide hump, characteristic of the amorphous phase (Fig. 1b, left). On the other hand, films with 80 at.% W show well-defined diffraction peaks, which can be ascribed to the bcc phase of W (Fig. 1b, right). This composition-driven transition from a single-phased amorphous to a

FIG. 2. TEM investigation of Zr-W film obtained in the amorphous-to-crystalline transition zone. (a) Bright-field TEM image showing an isolated cone-like crystalline region. Scale bar: 500 nm. (b) Selected area electron diffraction (SAED) patterns of the zones highlighted in (a). (c) Bright-field TEM image showing two crystalline regions in contact. Scale bar: 500 nm.

single-phased crystalline structure, shown here in the Zr-W system, is characteristic of many sputter-deposited binary alloys^{37–42}. As we have recently shown^{28,29}, two-phase crystalline-amorphous nanostructures can be obtained by working in the transition zone between single-phased amorphous and crystalline films of the Zr-Mo and Zr-W systems. The X-ray diffractogram of films grown

in this zone (Fig. 1b, center) shows a mixture of both (amorphous and crystalline) signals, giving a first indication of their two-phase structure. Interestingly, top-view SEM images of the films (Fig. 1c) reveal that in the transition zone, the amorphous (dark contrast) and nanocrystalline (bright contrast) phases are separated in well-defined spatial regions, i.e., amorphous phase acts as a matrix embedding the rounded-shaped nanocrystalline phase.

The bright-field cross-sectional TEM images and selected area electron diffraction patterns for a film grown in the transition zone (Fig. 2) evidence the amorphous and crystalline character of the regions. The crystalline regions are composed by nano-columns. Moreover, it reveals that isolated crystalline regions (Fig. 2a) nucleate in the amorphous matrix and grow as cones with the cone axis oriented along the film growth direction and that these regions terminate by a dome whose base is aligned with the surface of the amorphous matrix. As two crystalline regions contact, an interface is created, truncating the cones (Fig. 2c). The analysis of film cross-section associated and top view SEM images for different thicknesses enables to understand the dynamics of the process. Fig. 3a shows that the film initially grows in an amorphous state and, above a certain thickness (a critical thickness), the nucleation of the nanocrystalline phase starts and both phases continue growing together. In this nanocrystalline-amorphous competitive growth process, the nanocrystalline phase progressively overgrows the amorphous one. An important point in this process is that looking at the surface morphology evolution with thickness (Fig. 3a, top view), the amorphous phase “seems” to be transformed into the nanocrystalline one. This allows to establish an equivalence between the evolution of the surface morphology and a 2-dimensional amorphous-to-nanocrystalline phase transformation²⁹. Thus, using the formalism of phase transformations⁴³, the surface coverage by the nanocrystalline phase, S , can be expressed as a function of the film thickness as follows:

$$S(h) = 1 - (1 + (\xi - 1)\chi_{e(h)})^{-1/\xi - 1} \quad (2)$$

In Eq. (2), ξ is the impingement parameter which takes in consideration the distribution of the nanocrystalline nuclei in the amorphous matrix. When $\xi \rightarrow 1$, Eq. (2) transforms into $S = 1 - e^{-\chi_e}$, corresponding to the classical Kolmogorov-Johnson-Mehl-Avrami (KJMA) approach, in which a random nucleation of the nanocrystalline phase is assumed^{44–48}. For $\xi > 1$, impingement of the nanocrystalline phase is stronger, indicating that nucleation occurs at specific sites, such as grain boundaries. χ_e is the so-called extended transformed fraction^{43,49–51}, representing, in our case, the projected area, perpendicular to the growth direction of the film, of nanocrystalline regions if they could each nucleate and grow in the absence of interference with other growing nuclei. For the nanocrystalline-amorphous competi-

FIG. 3. Morphology evolution in the transition zone. (a) Cross-sectional view (left) and evolution of the surface morphology with thickness (right) of 72 at.% W film. Scale bar: 1 μm . (b) Evolution with the film thickness of the surface coverage by the nanocrystalline phase for three compositions in the transition zone: 67 (blue), 72 (black) and 74 (red) at.% W. The dots represent the experimental data extracted from the analysis of the SEM top views. The lines follow Equation 2 with the parameters shown in Table 1. Top view images of some films are shown in the figure. Scale bar: 2 μm .

tive growth process it follows the analytical expression²⁹:

$$\chi_{e(h)} = \pi \left(\frac{V_G}{V_0} \right)^2 \frac{N_0}{\gamma^2} \Upsilon \Theta[h - h_c], \quad (3)$$

where $\Upsilon = -2e^{-\gamma(h-h_c)} + (\gamma(h-h_c) - 1)^2 + 1$. In Eq. (3) V_G/V_0 is the ratio between the growth velocity of the nanocrystalline phase in the in-plane direction (V_G) and the deposition rate of the film (V_0). It can be shown that $V_G/V_0 = 0.404$ ²⁹. N_0 is the total number of nanocrystalline nuclei per unit area when all of them are already

FIG. 4. (a) Evolution of the volume fraction (calculated from Eq. (4)) with the film thickness for the three compositions studied. The inset shows the dependence of the volume fraction with the surface coverage. (b) Evolution of the perimeter per unit area (calculated from Eq. (5)) with the film thickness for the three compositions studied. The geometrical meaning of this parameter is illustrated in the schematic shown in the figure.

nucleated, γ is a parameter related with the nucleation rate of the nanocrystalline phase and Θ is the unitary-step Heaviside function, reflecting the fact that there is a critical thickness, h_c , from which nucleation starts.

Fig. 3b shows the evolution of S with thickness, extracted from the analysis of the top view SEM images, for three different compositions within the transition zone. SEM images evidence that size and density of nanocrystalline nuclei can be controlled by changing film thickness or composition. Consequently, a variety of surface morphology features can be obtained by changing this two

TABLE I. Fitting parameters. Evolution with W content of the density of nucleation sites N_0 , γ , the critical thickness h_c and the impingement parameter ξ .

Wcontent(at%)	$N_0(\mu\text{m}^{-2})$	$\gamma(10^{-3} \text{ nm}^{-1})$	$h_c(\text{nm})$	ξ
67	0.34	0.41	322	2.15
72	3.74	8.33	120	1.26
74	7.92	7.19	108	1.03

parameters. The lines in Fig. 3b follow Eq. (2) with the fitted parameters shown in Table I. When the W content increases (approaching to the composition for obtaining single-phase nanocrystalline films), N_0 increases and h_c decreases, indicating more favorable conditions for nucleation. The evolution of ξ between 2 and 1 with the increase of W content indicates a transition from nucleation at specific sites when conditions for nucleation are more difficult towards random nucleation when more favorable conditions exist.

From Eq. (2), the volume fraction of the nanocrystalline phase inside the film, V , can be calculated as follows:

$$V(h) = \frac{\int_0^h S(h) dh}{h} \quad (4)$$

Fig. 4a shows the evolution of $V(h)$ for the three compositions studied. The inset reveals that the values of volume fraction are lower than those of surface coverage, which is due to the geometrical (conical-like) shape of the nanocrystalline regions in the competitive growth process. Another parameter which can be determined supported by Eq. (2) is the perimeter per unit area of the nanocrystalline regions, P , whose geometrical meaning is illustrated in the schematic of Fig. 4b. It can be calculated as:

$$P(h) = \left(\frac{V_G}{V_0} \right)^{-1} \frac{dS}{dh} \quad (5)$$

As can be followed from Fig. 4b, once the critical thickness is reached, P starts to increase due to the nucleation and growth of the nanocrystalline phase, reaches a maximum and then decreases until zero. P decreases due to the coalescence of nanocrystalline regions at late growth stages. Thus, P is a measure of the interaction among these regions. With the increases of the W content, the maximum value of P increases and is reached at a lower thickness. This is because the density of nanocrystalline nuclei also increases (see N_0 in Table 1), creating more perimeter per unit area and provoking that the nanocrystalline regions begin to interact earlier. In the following, the evolution of the residual stress in the transition zone will be analyzed supported on the parameters presented so far, i.e. S , V and P .

FIG. 5. Evolution of residual stress with (a) the thickness of the films and (b) the surface coverage by the nanocrystalline phase for three compositions in the transition zone. The arrows in (a) show the position of the most relaxed state (maximum) in the stress curves for the three compositions studied.

B. Origin of residual stress of two-phase crystalline-amorphous nanostructured films

Before nucleation has started, i.e., as the films are still completely amorphous, a compressive state is observed. During deposition of an amorphous film, atoms adding to the growth can pack closer than in the bulk state to increase their local electron density. This creates an excess atomic density compared to the fully coordinated bulk material. As the surface layer becomes buried and becomes bulk coordinated due to further deposition, its expansion is suppressed, leading to establish compressive stress⁵². In the following, we study the evolution of stress during competitive growth between the amorphous and crystalline phases. In Fig. 5a, the evolution of residual stress with thickness for the three compositions studied is shown. When the W content increases, the most re-

FIG. 6. Evolution of (a) the residual stress, (b) the perimeter per unit area, and (c) its derivative with the volume fraction of crystalline region inside the film. In (a) is also shown the stress-volume fraction dependency for Zr-Mo films presenting the two-phase crystalline-amorphous nanostructures. In (c), the derivative dP/dh was normalized to the maximum value.

laxed state of the film is reached at a lower thickness, as indicated by the arrows in Fig. 5a. This shift to the left of the most relaxed state can be associated to the decrease of the critical thickness and increase in density of crystalline nuclei with the W content (see h_c and N_0 in Table 1). Consequently, more crystalline regions appear for the same thickness (see, for example, the top view SEM images of $\sim 1 \mu\text{m}$ thick films in Fig. 3b). This indicates that the stress is driven by a parameter that takes into consideration the ratio between the crystalline and amorphous phases in the film. This parameter could be S , which has the advantage that it can be determined directly from the analysis of the top view SEM images. In Fig. 5b, the stress-coverage curves are shown, evidencing less scattering in the stress behavior compared with the stress-thickness curves. However, considering that residual stress measurements give an “average” of the stress

distribution over the entire film, a better parameter for guiding the stress should be V (calculated from Eq. (4)), which takes in consideration the crystalline/amorphous ratio in the whole film, not just at its surface. Effectively, Fig. 6a shows that the dependency of the stress with V is the same for all the studied compositions in the Zr-W system. Furthermore, from the stress and surface coverage data reported in²⁸ for Zr-Mo films presenting the nanocrystalline-amorphous competitive growth phenomenon, we determined the stress- V dependency in this system. The results are also shown in Fig. 6a. Thus, the stress evolution as a function of the volume fraction follows the same behavior in the Zr-W (for all the studied compositions) and the Zr-Mo systems. We can conclude that the factor determining the level of stress of the two-phase crystalline-amorphous nanostructures presented here is the volume fraction of the crystalline phase inside the film.

So far, the behavior of the stress has been demonstrated to be the same when plotted against V . However, there is another point that needs to be addressed, i.e., which factors determines the increase or decrease of stress. The stress evolution observed during this competitive growth is different from what is usually reported for single-phased thin films. Understanding which factors govern it could lead to the possibility to manipulate it in the future. There are several mechanisms associated to nucleation, growth and impingement of crystalline regions that could act on the stress behavior:

- First, the nucleation and growth of the crystalline phase leads to a tensile component of the stress to develop. This is because the crystalline phase is denser than amorphous one and, consequently, its nucleation induces a volume contraction of the film. A similar mechanism of tensile stress build-up during crystallization has been reported in Zr-Fe⁵³ and Mo-Si^{54,55} alloys.

- Second, a careful observation of Fig. 2 and Fig. 3a reveals that new nano-columns nucleate inside crystalline regions even after they contact. These nano-columns originate from the branching of individual nano-columns of feather-like morphology, as visible in Fig. 2. The increase in the number of nano-columns per unit volume could be the origin of the conical shape of the embedded part of the crystalline regions: the lateral expansion of the crystalline regions is necessary to accommodate the generation of new nano-columns. After crystalline regions contact, this possibility of expansion is suppressed despite new nano-columns tend to be created. Therefore, the number of nano-columns per unit volume tends to increase. This mechanism could lead to gap closure between nano-columns and progressive build-up of a compressive component of the stress as individual nano-columns compete upon film growth.

Thus, when nucleation starts, the first mechanisms is dominating. Consequently, the increase in the volume fraction of the crystalline phase induces the development of a tensile component of the stress (see Fig. 6a for $0 < V < V_1$). At V_1 ($\sim 10\%$), this trend is reversed.

This is likely due to the proposed second mechanism to prevail. The tipping point can be correlated with the evolution of P (Fig. 6b) and its derivative dP/dh (Fig. 6c). As followed from the figures, the tipping point corresponds to the maximum in P (or zero of dP/dh). This supports the idea that P serves as a measure of interaction between crystalline regions. Thus, an increase in P indicates that the nucleation and growth (first mechanism) is dominating. On the contrary, a decrease in P points out that the impingement (second mechanism) is prevailing.

Even though the impingement mechanism is controlling the stress of the films for volume fractions higher than V_1 , a plateau in stress values can be detected from V_2 ($\sim 20\%$) to V_3 ($\sim 50\%$). Interestingly, the minimum of dP/dh for the three compositions is reached at $\sim V_2$ (Fig. 6c). This indicates that this minimum is the signature of the stabilization in the stress behavior. This can be understood taking in consideration that dP/dh represents the rate at which P is increasing (or decreasing). Thus, when the volume fraction is such that the minimum in the dP/dh curves is passed, the rate at which P is decreasing because the coalescence, decrease. This indicates a reduction of interaction from this point, likely reflected in a stabilization of the stress. Finally, from V_3 most of the surface is covered by the crystalline phase (surface coverage is higher than 90% as can be seen in the inset of Fig. 4a) and, consequently, P and dP/dh are almost zero. Therefore, from this volume fraction, the slight increase in the compressive component of the stress detected is associated to the growth of the crystalline phase after the entire film surface is covered.

IV. CONCLUSIONS

In summary, we have delved into the relationship between the growth kinetics and residual stress evolution of the two-phase crystalline-amorphous nanostructures obtained in the amorphous-to-crystalline transition in sputtered deposited Zr-W thin films. Two parameters describing the growth of these nanostructured films were introduced, i.e., the volume fraction of crystalline phase inside the film, V , and the perimeter per unit area of crystalline regions, P . We have demonstrated that V is guiding the evolution of the residual stress in the crystalline-amorphous competitive growth process. Furthermore, P serves as a parameter accounting for the interaction between crystalline regions. An increase in P indicates that nucleation and growth of crystalline regions are dominating and a decrease in P points out that the impingement of crystalline regions is prevailing. Thus, the evolution of P and its derivative were directly linked to residual stress measurements.

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DATA AVAILABILITY

The data that support the findings of this study are available from the corresponding authors upon reasonable request.

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