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NIR-persistent luminescent nanoparticles for bio-imaging, principle and perspectives

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Abstract

The development of nanoparticles for NIR imaging and diagnostics is an area of considerable interest. Among the different imaging modalities, optics emerged has an interesting technique since it is a non-invasive, cheap imaging technique allowing real time imaging. *In-vitro*, this technique is very useful, however *in-vivo* fluorescence imaging suffers from suboptimal signal-to-noise ratio, which is caused by the strong tissue autofluorescence under constant external excitation. To address this limitation, novel types of optical nanoprobe are actually being developed in the deep red/near infrared (NIR) range and among them, persistent luminescence nanoparticles (PLNPs), with long lasting near-infrared luminescence capability. These NPs allow optical imaging to be performed in an excitation-free and consequently autofluorescence-free manner. This chapter will first introduce the physical phenomenon associated to the long luminescence delay of such nanoprobe, from minutes to hours after ceasing the excitation, and will then highlight the tools used in physico-chemistry laboratories to characterize these nanoparticles with a focus on the ZnGa_2O_4 nanoparticles which are widely studied over the world. Then their biocompatibility will be mentioned and finally the evaluation in term of new advances for *in-vivo* bioimaging theranostics nanoprobe will be presented. We will conclude this chapter by envisioning perspectives for such nanomaterials.

Keywords : Persistent luminescence, nanoparticles, deep-red and NIR, imaging, therapy and theranostics.

Introduction

Since the discovery of $\text{SrAl}_2\text{O}_4:\text{Eu}^{2+},\text{Dy}^{3+}$ long persistent luminescence (PersL) by Matsuzawa *et al.* more than 20 years ago ¹, many persistent phosphors have been developed and studied. In the past decade, visible persistent phosphors based on sulfides, aluminates, gallates, and silicates hosts doped with various active ions have been developed ². Several compounds with sufficiently strong and long-lasting (> 10 h) persistent luminescence properties in the green and blue ranges have already been commercialized and widely used in various applications, such as security and emergency route signs, dials and displays. More recently, persistent phosphors in the deep red emission range at nanometric scale have been proposed for *in-vivo* bio-imaging ³. Optical imaging is highly complementary to other imaging methods, such as X-rays or magnetic resonance imaging, in particular due to its potential for data acquisition at high speeds. It allows the visualization of dynamic biological processes, events related to physiology and disease progression ⁴. For that purpose, fluorescent probes enable the study of biological processes in great detail ⁵. Among these probes, semiconductor quantum dots (QDs) exhibiting fluorescence optical properties have emerged as a class of nanoparticles for bioimaging and diagnostics. The possibility to detect and diagnose cancer or other human diseases at earlier stages than with current imaging methods caused a drastic increase of interest in nano-imaging technology. A non-invasive very cheap imaging technique, which is comfortable, portable, highly sensitive and that allows real-time imaging is still to be developed. The field of biomedical optics has matured rapidly during the last ten years, and is expected to continue its maturation in the next years. It provides a rapid, immediate (real-time dynamics) and cheap method for diagnosis. However, in spite of these great advantages, biomedical optics is limited because photons are scattered and absorbed by the tissues. The penetration depth of photons inside a tissue depends strongly on the type of tissue ⁶, but most importantly, it depends on the wavelength (λ) of the photons used. Scattering drastically decreases when λ increases in the so-call tissue transparency window ⁷ (see in figure 1 that mainly the red and near-infrared photons can go through living tissues). *In-vivo* imaging of exogenous fluorescent probes that target diseased tissues has also shown promising results in clinical settings, such as the early detection of breast cancer, the outlining of tumor margins during surgery and endoscopic diagnosis of cancer micrometastasis. However, the method is limited by tissue attenuation (scattering and absorption of the excitation or the emission light) and by tissue autofluorescence ⁸. To minimize tissue attenuation effects, researchers have been concentrated on near-infrared (NIR) fluorophores that are excited and emit in the spectral window between wavelengths of 650–

950 nm. However, tissue autofluorescence still produces a substantial background signal in this spectral range that severely limits the quality of images, especially when very low concentrations of the fluorescent probe accumulate at the target site ⁹.

Persistent luminescence phosphors are materials able to store optical energy and release it gradually by photon emission. This particular property is linked to the presence of trapping levels located in the forbidden gap of the material (see figure 1) ^{2, 10}. Optical excitation of these materials leads to photoluminescence (PL) but can also induce charge trapping ¹¹. The trapped charges (electrons, holes) can then slowly be released following thermal stimulation, leading to their recombination at emitting centers followed by light emission for a long time (typically from minutes to hours ^{1, 12}) after stopping the irradiation. For persistent luminescence, the trap depth should be smaller than 1 eV to enable charging and discharging at room temperature ².

It appeared that metal transition cations such as Mn^{2+} or Cr^{3+} , or rare earths elements are good dopants for *in-vivo* optical bio-imaging based on persistent luminescence in the deep red/NIR range ³ but others compounds listed below have recently demonstrated a lot of interest. Please notice that $ZnGa_2O_4:Cr^{3+}$ (ZGO) is one of the most interesting long lasting phosphor in the gallate family. It shows deep red emission at 696 nm originating from the 2E to 4A_2 transition of Cr^{3+} ions ¹³. A focus on this compound and its main characteristics are presented in the following part of the chapter.

Table 1: Main materials used with metal transition (MT) cations and RE cations for red/ near infrared emission and applications in bioimaging.

Hosts	Dopants	Comments and applications in bioimaging	Year and Refs
Gd ₂ O ₂ S	Eu ³⁺ , Mg ²⁺ , Ti ⁴⁺	NPs regular shape, bimodality optical / MRI	¹⁴ (2015)
Ca ₃ (PO ₄) ₂ /HAP	Mn ²⁺ , Tb ³⁺ , Dy ³⁺	Fully Biocompatible, NPs and <i>in-vivo</i> imaging	¹⁵ (2015)
Ca ₂ Si ₅ N ₈	Eu ²⁺ , Tm ³⁺	Bioimaging applications	^{3f} (2012)
SrAl ₂ O ₄	Eu ²⁺ , Dy ³⁺	NPs, Functionalization; Bioimaging applications, <i>green emission</i>	¹⁶ (2014, 2018)
Ca _{0.2} Zn _{0.9} Mg _{0.9} Si ₂ O ₆	Mn ²⁺ , Eu ²⁺ , Dy ³⁺	NPs, Functionalization, pioneer work for Bioimaging: Cancer cells	¹⁷ (2007, 2011, 2012)
Ca _{1.86} Mg _{0.14} ZnSi ₂ O ₇	Eu ²⁺ , Dy ³⁺	imaging, cell targeting FRET and various bio-sensing applications	¹⁸ (2018)
CaMgSi ₂ O ₆	Mn ²⁺ , Eu ²⁺ , Pr ³⁺	NPs, Functionalization, Bioimaging	¹⁹ (2011)

MAIO ₃ (M=La, Gd)	Mn ⁴⁺ /Ge ⁴⁺	Bioimaging in pork tissue	¹¹ (2014)
GdAlO ₃	Mn ⁴⁺ , Ge ⁴⁺ @ Au	Trimodality imaging	²⁰ (2016)
	Sm ³⁺ , Cr ³⁺	Optical and magnetic dual mode imaging	²¹ (2018)
ZnGa₂O₄	Cr ³⁺	NPs, Functionalization, Bioimaging (cancer cells imaging, cell targeting, cytotoxicity, visible light stimulation NIR Photostimulation, X-rays activation	^{3e} (2014) ²² (2014)
		Oral administration & breast cancer imaging	²³ (2017), (2018)
		Toxicology analysis	²⁴ (2018)
		Protobiotic analysis	²⁵ (2017)
ZnGa ₂ O ₄ in hollow cavity	Cr ³⁺	Photodynamic therapies	²⁶ (2017)
ZnGa ₂ O ₄	Cr ³⁺ , Gd ³⁺	NPs, Functionalization, Bimodality Optical/NMR imaging	²⁷ (2018)
ZnGa ₂ O ₄ /SiO ₂	Cr ³⁺	NPs, core-shell structure, drug delivery	²⁸ (2015)
ZnGa ₂ O ₄ /Fe ₂ O ₃	Cr ³⁺		²⁹ (2014)
ZGOCS@MSNs@Gd ₂ O ₃	Cr ³⁺	Cell labelling and magnetic vectorization	³⁰ (2018)
		Multimodal nanoprobe	³¹ (2017)
Zn _{1.1} Ga _{1.8} Ge _{0.1} O ₄ /SiO ₂	Cr ³⁺ , Eu ³⁺	NPs, core-shell structure, drug delivery	³² (2015)
Zn ₃ Ga ₂ Ge ₂ O ₁₀	Cr ³⁺	Imaging of pork tissue, Photostimulation, cytotoxicity	³³ (2014)
Zn _{1.1} Ga _{1.8} Ge _{0.1} O ₄ @SiO ₂	Cr ³⁺	bioimaging and drug delivery	³⁴ (2018)
Zn _{1.25} Ga _{1.5} Ge _{0.25} O ₄	Cr ³⁺ , Yb ³⁺ , Er ³⁺	Metastasis Tracking and Chemophotodynamic Therapy	³⁵ (2018)
Zn _{1.1} Ga _{1.8} Ge _{0.1} O ₄	Cr ³⁺	Nanothermometry	³⁶ (2017)
Zn ₃ Ga ₂ Sn ₁ O ₁₀	Cr ³⁺	Imaging of goldfish	³⁷ (2014)
Zn _{2.94} Ga _{1.96} Ge ₂ O ₁₀	Cr ³⁺ , Pr ³⁺	NPs, Functionalization	³⁸ (2013)
Zn ₃ Ga ₂ Ge ₂ O ₁₀	Cr ³⁺	Recognition of Breast Cancer Cells	³⁹ (2015)
Zn ₃ Ga ₂ GeO ₈	Cr ³⁺ , Yb ³⁺ , Er ³⁺	Upconversion	⁴⁰ (2014)
LiGa ₅ O ₈	Cr ³⁺	NPs, Functionalization, Bioimaging, Visible light stimulation, photostimulation	⁴¹ (2013, 2014)
Ca ₃ Ga ₂ Ge ₃ O ₁₂	Cr ³⁺ , Yb ³⁺ , Tm ³⁺	NIR Stimulation, Upconversion	⁴² (2014)
	Pr ³⁺ , Yb ³⁺	In-vivo imaging	⁴³ (2017)
mSiO ₂ @Gd ₃ Ga ₅ O ₁₂	Cr ³⁺ , Nd ³⁺	multimodal imaging and cancer therapy	⁴⁴ (2018)
Sr ₂ SnO ₄	Nd ³⁺	Finger image	⁴⁵ (2014)
SiO ₂ /CaMgSi ₂ O ₆	Eu ²⁺ , Pr ³⁺ , Mn ²⁺	Bioimaging, intraperitoneal injection	⁴⁶ (2014)
YAGG (garnet)	Er ³⁺ , Cr ³⁺	Photostimulation imaging of pork tissue	⁴⁷ (2018)
		Imaging in the second biological window	
NaYF ₄ + SrAl ₂ O ₄	Yb ³⁺ , Tm ³⁺ , Eu ²⁺ , Dy ³⁺	Upconversion & photodynamic therapy	⁴⁸ (2018)
Sr ₂ MgSi ₂ O ₇	Eu ^{2+,3+} , Dy ³⁺	Photodynamic activation	⁴⁹ (2016)

		Visualization of abdominal inflammation	⁵⁰ (2018)
La ₃ Ga ₅ GeO ₁₄	Cr ³⁺ , Zn ²⁺	bioimaging-guided in-vivo & drug delivery	⁵¹ (2018)
@SiO ₂ @Van			
CaTiO ₃	Pr ³⁺ , Yb ³⁺ , Tm ³⁺	Upconverting and guided photothermal therapy	⁵² (2017)
ZnSn ₂ O ₄	Cr ³⁺ , Eu ³⁺	Cellular and deep tissue imaging	⁵³ (2017)
Sr ₃ Sn ₂ O ₇	Nd ³⁺	Second window imaging	⁵⁴ (2017)

Persistent luminescence phenomenon, whereby luminescent materials keep emitting light after the excitation has stopped, has intrigued people since several centuries. As this property occurs in inorganic materials when the excitation has been stopped, this is sometimes compared to the bioluminescence mechanism observed in some animals (such as fireflies and jellyfishes). Others fluorescent probes in the NIR range as reported in this book are the semiconductor quantum dots (QDs) or upconversion nanoparticles (UCNPs).

Main characteristics of the persistent luminescence materials

Long persistent luminescence is controlled by the slow liberation of trapped charge carriers at room temperature by a simple thermal de-excitation process. The luminescence can last for several minutes to hours after removal of the excitation source (typically several minutes of ultraviolet light)^{1, 55}. In persistent luminescence materials, energy can be stored by traps/defects that are charged under irradiation. Emergency signage that can be used in case of electricity failure is one of the main applications of these persistent phosphors, but others applications were proposed such as watch dials, when radioactive elements were fully forbidden more than twenty years ago, decorative objects and toys and more recently for bio-imaging as described in this review paper.

Fig. 1. left: Schematic of the energy levels and traps involved in the persistent luminescence mechanisms. Right: Partial transparency of living tissues in deep red range (see light through girl left hand and figure 11 (c)), extracted from Saint Joseph charpentier, G. de La Tour, 1643, Musée du Louvres, Paris, France

For the materials concerned with the persistent luminescence properties, there is in the literature plenty of compounds tested with various successes. Rare earth and metal transition cations have been widely used as dopant for

persistent luminescence materials and in last years. More than two hundreds combinations of host materials and activator ions have been depicted. The reader could see recent reviews and book chapter on the topic ^{2, 56}. For bio-imaging and *in-vivo* application, as seen in figure 1, the past five years have witnessed several major advances to establish deep red/near-infrared emitting persistent luminescence nanomaterials (called in the paper PLNPs as acronym for Persistent Luminescence NanoParticles) as a novel, approach for real-time used in-vivo in small animal.

The main requirements that the materials must fulfil for this application are the following: (i) compounds prepared as nanoparticles (PLNPs with size < 100 nm) and even ultrasmall nanoparticles (< 10 nm) (ii) intensive emission extending in the deep red toward the near infrared range in the first or second windows corresponding to the tissue transparency ⁵⁷, (iii) efficient functionalization, (iv) stability in aqueous solution (v) persistent emission over hours. These points are the reasons why applications using deep red emitting persistent phosphors for bio-applications were barely investigated at the first step of the development of the persistent phosphors in 1996 and the investigation started in 2007 after our pioneer paper ^{17a}. Several materials were further developed for imaging applications (see figure 1 and review papers ⁵⁶). Notice that at first all the researches were focused in the deep-red range (high transmission range of the living tissue, see also figure 1) where the Si detector also reaches its maximal sensitivity. More recently researchers extend the range to the second and third biological windows (BW) even if the results are still very preliminary in that 1000 nm-1350 nm spectral range (BW-2) and above 1500 nm (BW-3) as presented in the following part of the chapter. In the development of the new phosphors with long persistent luminescence for bio-applications, one should keep in mind that various synthesis strategies are used for small size preparation of PLNPs with good crystallinity but also with the optimum of defects at the origin of the persistent luminescence. Hydrothermal, solvothermal, sol-gel, co-precipitation and microwaves methods have been widely used. In addition, effective surface functionalization strategies along with the bio-compatibility issues required attention to make them useful for *in-vivo* bioimaging.

Persistent luminescence mechanisms

The physics behind the phenomenon is however not that simple and intensive research has been carried out the last years. For the trapping processes, which is the primary very important step, two kind of traps are envisioned: (i) *Intrinsic defects and optimization of the traps by composition variation and thermal annealing*. Intrinsic defects have been well identified in lot of materials. ^{10a, 58} In garnet hosts, antisite defects for instance are very likely,

they shorten some Y-O bonds and create shallow electron traps close to the conduction band. Antisite defects in the vicinity of Cr^{3+} ions have been proposed to be involved in the persistent luminescence mechanism of the zinc-gallate spinel host^{13, 59}. One should also notice that thermal annealing under reducing atmosphere is often used to create or enhance intrinsic traps such as oxygen vacancies, which correspond to electron traps and can be responsible for the persistent luminescence⁶⁰.

(ii) *Optimization of the traps by co-doping*. Co-doping is the second strategy that has been intensively tested to enhance the persistent luminescence. Co-doping with one or two lanthanides cations are most of the time used to enhance the property. For instance, Dy^{3+} was used as electrons trap in the first hybrid enstatite/diopside silicate for the proof of concept of the in-vivo bioimaging application with PLNPs^{17a}. Moreover it appeared that Pr^{3+} was indeed better when only the diopside bandgap was considered¹¹. The variation of traps depths strongly depends on the couple host/dopant and corresponds to a so-call bandgap engineering as introduced in previous works⁶¹, with the prediction of the energy level diagram⁶². Such bandgap engineering was proposed in various hosts used for persistent luminescence but was also tested with success in others *Photonic applications* such as scintillation and lighting. It is then currently possible to vary the position of the conduction band and the traps depth by varying the composition playing with cationic and anionic substitutions or a careful choice of the doping cation for an optimal couple composition/dopant.

To go in deeper understanding, knowledge of the energy formation of defects, defects stability and defects energy position are required, through band structure and defects calculations⁶³. The persistent luminescence mechanism has been characterized by lots of different spectroscopies including optical, EPR to control the origin of defects,⁶⁴ X-rays spectroscopies⁶⁵ to give insights to the mechanism, thermoluminescence¹⁹ and photoconductivity⁶⁶ to measure the traps depth^{10a, 67}. In the case of electron traps and hole traps⁶⁸, the stored charges can be released by various processes as expressed in the following part such as thermal⁶⁹, optical⁷⁰ or other physical stimulations⁷¹, resulting in stimulated emissions from the active recombination centers.

ZnGa₂O₄:Cr nanoparticles for persistent luminescence applications

If metal transition doped ZnGa₂O₄ nanoparticles can be obtained by different techniques, our activity recently focus on the hydrothermal synthesis assisted by microwave heating in order to obtain ultrasmall nanoparticles with persistent luminescence properties. To avoid sintering and keep the nanometric size during the following thermal treatment, the nanoparticles are

embedded in a silica layer using a sol-gel chemistry way. A mixture of TEOS:EtOH 1:4 is added to a basic solution of nanoparticles. Doped with trivalent chromium for deep red /NIR emission, the obtained $\text{ZnGa}_2\text{O}_4:\text{Cr}^{3+}, \text{Bi}^{3+}@\text{SiO}_2$ nanoparticles are then calcined in air during 2 h at 1000 °C. The nanoparticles are monodisperse with a sub-10 nm size ⁷². To better characterize the persistent luminescence properties of the $\text{ZnGa}_2\text{O}_4:\text{Cr}^{3+}$ nanocrystals, we also recently proposed with colleagues from Orleans, France ⁷³ to prepare these nanocrystals embedded in a glass phase in the so-called ZGO nano glass-ceramics. The TEM picture of both nano-objects are presented in figure 2. The nanoparticles embedded in glassy matrix present an average diameter about 16 nm while after hydrothermal synthesis assisted by microwave, diameter nanoparticles smaller than 10 nm can be obtained (see figure 2). The nanoparticles are spherical and well dispersed. In both cases after crystallization at 900 °C and thermal treatment at 1000 °C respectively, the obtained nano-objects are well crystallized in the pure spinel phase and present persistent luminescence properties.

Fig 2. TEM pictures of a) $\text{ZnGa}_2\text{O}_4:\text{Cr}^{3+}$ (ZGO) nanoparticles embedded in glass matrix (TC = 900 °C) b) ZGO nanoparticles after hydrothermal synthesis assisted by microwave heating, associated distribution diagram, adapted from ^{74,73}.

Cr^{3+} -doped ZGO nanoparticles photoluminescence and persistent luminescence spectra are presented figure 3. Several contributions of the chromium emission are well identified: the R line corresponding to the zero phonon line. Stokes and anti-Stokes phonon side bands (S-PSB and AS-PSB respectively) were also recognizable. The N2 line is attributed to the emission of a Cr^{3+} ion in the vicinity to an antisite defect ¹³. The persistent luminescent emission spectra is mostly dominated by the N2 line, indicating that these cations, spatially closed to the electron traps are partly responsible of persistent luminescence (see following part). Figure 3 shows the persistent luminescence decay after 2 min of UV irradiation of ZGO: Cr^{3+} nanoparticles embedded in glass. With these ultrasmall nanoparticles, it is possible to detect the signal more than 2 hours after swithing off the UV lamp.

Fig. 3: Left, (top) Emission spectra of photoluminescence and (bottom) persistent luminescence at room temperature, adapted from ¹³. Right: Persistent luminescence decay of $\text{ZnGa}_2\text{O}_4:\text{Cr}^{3+}$ nanoparticules (within more than 2 hours) adapted from ⁷³.

As experimentally determined as well as calculated ^{75, 75b}, in such nano-materials, antisites defects around the Cr^{3+} cations are needed to get persistent luminescence properties. NMR and EPR magnetic spectroscopies can

gives some insights on the disorder, and on the surrounding of the dopant (Cr^{3+}) or the host (Ga^{3+}) while thermoluminescence give insights on the persistent luminescence capability ⁷³. In the particular case of $\text{ZnGa}_2\text{O}_4:\text{Cr}^{3+}$ and derivatives, electron paramagnetic resonance (EPR) is a technique of choice to learn about Cr^{3+} environments. Indeed, there is a splitting of the $^4\text{A}_2$ chromium level into two degenerated states (one $m_s = \pm 3/2$ and one $m_s = \pm 1/2$) due to the crystal field distortion and the spin orbit coupling. An application of a magnetic field can raise the degeneracy of these levels. Depending on the Cr^{3+} vicinity, the splitting of these levels may be different leading to various EPR signals. Several EPR studies have been performed on $\text{ZnGa}_2\text{O}_4:\text{Cr}^{3+}$, resulting in a better understanding of the local material structure and the possibility to distinguish several Cr^{3+} environments using simulations. First, the simulations on powder phosphors have shown that five different Cr^{3+} surroundings may exist in $\text{ZnGa}_2\text{O}_4:\text{Cr}^{3+}$ material ^{75a}. The results also strengthened the assumption about the role of antisite defects on persistent luminescence properties. Indeed, it has been demonstrated that the charge trapping in $\text{ZnGa}_2\text{O}_4:\text{Cr}^{3+}$ may be related to two antisite defects in the Cr^{3+} neighborhood (Cr- γ , Cr- δ , Cr- ϵ). We recently demonstrated that simulations using these five Cr^{3+} also perfectly fits our nano-crystals embedded in glass matrix ⁷³ (see figure 4). An important EPR result in this system is the possible quantification of chromium environment. This quantification enables the dependence on synthesis parameters. Indeed, we can notice a global increase of undistorted Cr^{3+} environment when T_{crist} increases revealing a better symmetry around Cr^{3+} in agreement with the improvement of crystalline quality. However, the persistent luminescence properties of lower crystallization temperature materials are poor, this may be due to the presence of parasitic defects due to poor crystallinity leading to quenching processes at the expense of the radiative processes.

Thermoluminescence (TL) or thermally stimulated luminescence (TSL) is the most commonly used technique to demystify the trap number and their depths. This is a required characterization technique prior to persistent luminescence compounds developemnt. At first, in the thermoluminescence experiments, traps of the the compounds mustbe filled by optical (from X-rays to NIR photons) or others stimulations. Notice that mechanoluminescence ⁷⁶ is also recently an important field of investigation. As mentioned above, in a second step thermal energy or others excitation processes, can release these traps. During the TL experiments, the sample is linearly heated, from 10 K to 600 K in our facility. When the thermal energy brought by the heater is sufficient *i. e.* in the order of the energy difference between a recombination center and the trap, charges start to be de-trapped. After being released, charges can either be trapped again in the same or others traps in

the vicinity or recombine on the emitted centre. In this latest process, thermoluminescence occurs and can be recorded. The result of the experiment, as presented in figure 4, shows luminescent intensity as a function of the temperature. The number and position of the glow curve peaks give insights on the traps depths and distributions. Notice that the trap depth can also be estimated from the thermoluminescence profile. Regarding to the glow curve shape, several equations can link the trap depth to the peak temperature^{77, 78}. In case of broader bands, which appears to be often the case for PLNPs, a simple estimation can be made using⁷⁹ :

$$E = \frac{T_m}{500}$$

Where E is the trap depth and T_m is the temperature of the peak maximum. For the persistent luminescence properties at room temperature, the release should occur at room temperature, in case of *in-vivo* applications it has to be close to the living tissues temperature. This means that materials should present thermoluminescence peaks centred at *ca.* 310 K to be suitable for this application. Another option is to design materials with several trap depths. One kind of traps corresponding to room temperature and deeper traps, that cannot be thermally emptied at room temperature. In this case, a possible *in-vivo* re-activation of the persistent luminescence *via* trap redistribution using optical stimulation (using NIR light) can be envisioned.

The thermoluminescence results for $\text{ZnGa}_2\text{O}_4:\text{Cr}^{3+}$ are displayed on figure 4 for several particles size and various shapes. The bulk material shows a relatively sharp thermoluminescence peak, labelled as the main peak, centred at 318 K, making it suitable for *in-vivo* imaging based on persistent luminescence. It also shows two shallow traps, weaker in intensity around 145 K and 200 K. Going from bulk material elaborated *via* solid state method to nanoparticles elaborated by hydrothermal synthesis assisted by microwave heating, the thermoluminescence glow curve becomes quite different. Indeed, the corresponding profile presents a broad signal that seems to be composed of two major contributions, one around 210 K

Fig. 4: Left : Thermoluminescence glow curves with different sizes and shapes and right: Simulation and experimental EPR spectra for the ZGO:Cr, adapted from⁷³.

Table 2 : EPR parameters and Cr^{3+} surrounding adapted from⁷³

	Cr_α	Cr_β	$\text{Cr}_\gamma + \text{Cr}_\delta + \text{Cr}_\epsilon$	Φ
ZGO-GC 900 °C	5.4 %	59.9 %	24.5 %	10.2 %
ZGO-GC 1000 °C	18.2 %	52.4 %	21.5 %	7.9 %

and one around 300 K. The first one can be related to the shallow traps whereas the second can be related to the previous main peak. As the surface over volume ratio increases when the size is reduced, surface defects may have a major contribution in nanoparticles. With that in mind it has been assumed that the shallow traps can be related to surface defects. The thermoluminescence glow curve of $\text{ZnGa}_2\text{O}_4:\text{Cr}^{3+}$ glass ceramics elaborated at 900 °C, presented on the same figure (see Fig. 4), presents a broad main peak centred at *ca.* 310 K. The two shallow traps can be observed as humps enlarging the main peak at around 145 K and 205 K. In the glass ceramics samples, the surface over volume ratio is very high because of the nanoscale of the crystals. Still, the surface defects quantity may be very low comparing to nanoparticles elaborated *via* microwave assisted hydrothermal method as the nanocrystals are here embedded in a glassy matrix. This can explain the relatively low response of shallow traps, as it is assumed that they are related to surface defects. However, the main response peak appears broader in glass ceramics samples compared to bulk material. This has previously been explained by the crystal quality of the material.

Biocompatibility

To be used *in-vivo*, the nanoprobe should have the lowest toxicity. The main matrices presented above are either NPs incorporated into silica (it can also be mesoporous silica as seen below) or oxide-based NPs. Since they are designed for the sustained release of drugs, the tissue compatibility required a careful investigation. This was done by several groups (see for instance ⁸⁰). No pathological abnormality could be observed in both gross and microscopic histological examinations of various tissues including heart, liver, spleen, lung and kidney up to one month after vein injection into mice, suggesting that the mesoporous silica PLNPs had not caused significant tissue toxicity and inflammation even they had not completely be degraded.

It was confirmed that neither gross nor histopathological abnormalities could be observed in the major organs of mice, such as in liver, spleen, kidneys, heart, intestines, stomach, muscles or lungs⁸¹. Furthermore, at a high vein-injection dose of 50 mg.kg⁻¹ per day, the acute toxicity of PLNPs was almost negligible compared with the blank control according to the monitoring of the body-weight change, the visible and/or palpable dermal infection, the presence of ascites, the grooming or the impaired mobility. At an extremely high dose of 1200 mg.kg⁻¹, no mouse survived after intraperitoneal or intravenous injection of MCM-41-type or SBA-15-type mesoporous silica nanoparticles (MSNs), however the subcutaneous injection did not cause death at an equally high administration dose and doses of up to 200 mg.kg⁻¹, which was high enough for drug loading and delivery owing to their high drug loading capacities, was always safe by both intraperitoneal and intravenous injection⁸².

Concerning persistent luminescence nanoparticles by themselves, two articles have been recently published concerning biocompatibility and toxicity assays *in-vivo*. The first one was reported in 2017 by Ramirez-Garcia *et al.* In this study, mice were injected with a single intravenous administration of either hydroxylated or PEGylated persistent luminescence nanoparticles at different concentrations, from 1 to 8 mg per mice and a set of standard tests were carried out 1 day, 1 month and even 6 months after the administration²⁵. High concentrations of hydroxylated nanoparticles generate structural alterations at histology level, endoplasmic reticulum damage and oxidative stress in liver, as well as rising in white blood cells counts. A mechanism involving the endoplasmic reticulum damage could be responsible of the observed injuries in case of ZGO-OH. On the contrary, no toxicological effects related to PEGylated nanoprobe treatment were noted during the *in-vivo* experiments, denoting the protective effect of PEG-functionalization and thereby, their potential as biocompatible *in-vivo* diagnostic probes. In 2018, Zhang and co-workers reported a 60 days *in-vivo* study of ZGO derivatives (Zn_{1.1}Ga_{1.8}Sn_{0.1}O₄:Cr³⁺) compound. In this work, healthy Balb/c mice were intravenously injected with ZGO derivative at a dose of 10 mg/kg. ZGO pre-irradiated with a 254 nm UV lamp was also set as one of the factors to evaluate the possible effect of NIR persistent luminescence of ZGO *in-vivo*. No signs of apparent weakness, spontaneous animal death and significant body weight gain or loss were observed within 60 days. It was found that all of the parameters were in the normal reference range. In the haematological analysis, various serum biochemical parameters were measured with particular attention paid to liver and kidney function. Indicators of kidney function, including creatinine and urea nitrogen, were also within normal ranges and were similar to these of control mice. These results show

no obvious injury to the liver and kidney with ZGO exposure in mice, even at long exposure times. Based on long-term *in-vivo* biodistribution studies, the major organs from mice were sliced for hematoxylin and eosin staining and histological examination to determine whether or not ZGO exposure caused tissue damage, inflammation or lesions. The structures of the organs exhibited hardly any difference from the control group. No apparent histopathological abnormalities or lesions were found in any of the experimental groups. All of these data suggested that no significant toxicity was induced by ZGO injection, even up to 60 days.

Excitation capabilities and long term *in-vivo* imaging.

Though intended for diagnosis applications in living animals, persistent luminescence nanoparticles suffer from severe limitations. First, as the release of photons is thermally activated, accumulations of the signal (from seconds to minutes) are required during recording and materials with intensive persistent luminescence on one side and low noise detector on the other side should be used. The second point concerns the excitation mechanisms of the PLNPs but recent improvements have been made as described in this part of this chapter. Indeed the first generation of PLNPs has to be excited *ex-vivo* by UV light prior to systemic administration preventing long-term imaging in living animal. Depending on the nanoparticle characteristics, slow accumulation of stealth nano-carriers within malignant stroma by the enhanced permeability and retention effect usually requires from 2 to 24 hours⁸³. This was far too long in relation to the emission from persistent luminescence nanoparticles, which hardly exceeds one hour *in-vivo*.

To overcome this major restriction, developments of new materials and of new modalities have been undertaken. First, efforts to optimized compositions and enhanced optical characteristics have been made. In that sense, compounds based on gallate and gallo-germanate spinels compounds have attracted large attention, due to their bright deep persistent luminescence when doped with trivalent chromium (Cr^{3+}) after UV excitation in comparison with the previously used silicate nanoprobe. Then the second point was to excite the persistent luminescence material at lower energy in the biological window if possible, longer time after the injection. The new modality has been proposed in different ways as seen in Figure 5.

Figure 5 summarizes the various approaches that can be followed for long term *in-vivo* imaging with persistent luminescent nanoparticles: A suspension of PLNPs in a biological buffer can be pre-activated *ex situ*, then injected into the animal and placed under a photon-counting system to detect the emitted persistent luminescence signal (see step 1, Fig. 5). PLNPs can be activated or re-activated *in situ* (after the injection) through animal tissues (step 2, Fig. 5). This breakthrough for *in-vivo* optical imaging allows the most simple and convenient recovery of the persistent luminescence signal, whenever required. The efficiency is lower than for the persistent luminescence under UV excitation (step 1) but is indeed sufficient to be detected and to localize the probe *in-vivo*. Furthermore for some inorganic persistent luminescent NPs, when deeper traps are observed in the thermoluminescence spectra such as in garnet and spinels hosts, deep-red persistent emission can be stimulated by near infrared light, as the traps can be depopulated by low-energy light stimulation in the so-called photostimulated persistent luminescence (PSPL)^{22, 70, 84}. The photostimulation capability of several materials is widely reported⁸⁵ and researchers have focused their attention on the Cr³⁺ doped samples. The photostimulation technique has been used over the years for UV dosimetry, as well as for dating geological and archaeological materials⁸⁶. One can adjust the depth of the traps responsible for the persistent luminescence and therefore to control carefully the composition. In that case, the release of the traps and thus the emission could be *started at the convenience of the user* using a red/near infrared LED for instance which will be in the best transparency range. The first preliminary carried out tests have shown the originality and feasibility of this new modality⁸⁷. The excitation energy for charge detrapping may vary from red to NIR lamps, LED or laser sources. The photostimulation or excitation by low-energy light can also be carried out by additional step, corresponding to wavelength conversion^{40, 42, 88}, or possible charge through second order effects under high excitation power^{73, 89}.

Fig 5: Top: different strategies to perform *in-vivo* optical imaging with persistent luminescence nanoparticles (PLNP). (1) Excitation of a suspension of nanoparticles followed by tail vein injection and optical imaging using a photon counting camera. (2) Once the *in-vivo* persistent luminescence signal has completely disappeared, some PLNP can give further persistent luminescence signal either by photo-or X-rays stimulations possibly giving *n* decay signals. (3) Some PLNP can be excited without preliminary excitation (with lower efficiency) in the animal body using visible, near infrared –for upconversion– or X-rays photons, **Bottom** Traps could be emptied by a near infrared source (energy ~ 1eV) showing the crucial role of charge-storage to deep traps and photo-

transfer to shallow traps leading to persistent luminescence. This demonstrated the photostimulation capability at convenience time. Within this experiment, one observed long-lasting phosphorescence under 977 nm light excitation⁸⁷, making in-vivo excitation of the spinels probes envisioned and possible long term imaging applications.

However, *in-vivo* tracking is still hindered by its limited penetration in tissue even in the near infrared. An alternative way is to use NIR or X-ray as excitation source for imaging techniques. Indeed compared to the traditional excitation source, X-ray possesses competitive advantages of weaker scattering and deeper penetration depth in tissues, as well as simplified image reconstruction for optical tomography. Developing X-ray rechargeable persistent luminescence nanoplatform, by combining the advantages of X-ray excitation and NIR persistent emission in the second biological window, will enable to localize tumors and achieve treatment simultaneously⁹⁰. This opens the door for achieving deeper tissue and higher sensitivity optical bioimaging with better spatial resolution. An example of such modality with ZGO:Cr PLNPS is presented in the following part of this chapter.

Strategies developed to perform long time imaging

The pioneer work using persistent luminescence nanoparticles (PLNPs) for optical *in-vivo* bioimaging was reported in 2007 by Scherman and co-workers^{17a}. As discussed in the previous section, to be used *in-vivo*, the probe should have a nanometric scale and should emit light in the transparency window (> 650 nm). In their work, the authors prepared a silicate doped with Eu^{2+} , Dy^{3+} and Mn^{2+} and *in-vivo* imaging without any background was performed for 30 min to 1 hour without any *in-vivo* excitation, allowing performing *in-vivo* imaging without autofluorescence. Then after these pioneer works, lots of bio-applications and various modalities have been presented as seen Table 1, and some of them are detailed in the following part of the chapter. To overcome this major restriction and to allow longer time imaging, in 2014 Maldiney *et al.* changed the matrix and used zinc gallium oxide doped with Cr^{3+} ^{3e}. As can be seen on Figure 6, contrary to the first silicate – type generation, this material has several excitation peaks, among them one being at the limit of the tissue transparency window (Figure 6A) that can also sensitized the persistence luminescence. On nanosized $\text{ZnGa}_2\text{O}_4\text{:Cr}$, they observed a persistent luminescence decay signal after both UV and activation with visible orange/red light-emitting diodes (LEDs) (Figure 6B). A suspension of this material dispersed into 5% glucose was injected into mice, and for the first time the authors have shown that the probe could be activated *in-vivo* with orange/red light-emitting diodes. The authors have also shown that UV pre-excitation before LED re-activation is favourable

but not necessary (Figure 6C and 6H). In such conditions, there was no more time limit and the probe could be detected *in-vivo* whenever wanted through a simple excitation with the LEDs (Figure 6E-G).

Fig. 6. *In-vivo* imaging using *in situ* excitable PLNPs. (a-c) Optical properties of powder after either UV or visible excitation. (d-h) *In-vivo* excitation of un-functionalized PLNPs (e) or stealth ZGO in healthy (f) on tumor bearing mice (g), adapted from ^{3e}.

It is of great importance to develop deep red/NIR PLNPs with rechargeable nature *in-vivo* by a deep tissue penetrating source (see figure 1, old painting showing only red light through living tissues). Recent years have witnessed the rapid development of utilizing innovative excitation sources for imaging techniques ⁹¹. For instance X-ray photon energies are able to stimulate luminescent centers in phosphors and then generate light. Furthermore, compared with the traditional excitation sources (UV/visible/NIR), X-ray possesses competitive advantages of weaker scattering and deeper penetration depth in tissues and simplified image reconstruction for optical tomography ⁹². As a proof of concept, Xue et al selected $\text{ZnGa}_2\text{O}_4:\text{Cr}^{3+}$ PLNPs as models for X-ray-activated NIR persistent emission ^{23b}. *In-vivo* NIR persistent luminescence bioimaging based on these X-PLNPs was demonstrated (see figure 7). More importantly, these X-PLNPs were also repeatedly charged by X-rays at a deep location up to 20 mm. These results revealed that X-ray could act as a new excitation source for persistent renewable bioimaging with a deep penetration depth based on NIR PLNPs.

Fig. 7. (a) Principle of *in-vivo* imaging using X-ray excitation. (b) Applications *in-vivo*, adapted from ^{23b} and ⁹³.

Multimodal imaging

Multimodal imaging has drawn much attention in biomedical applications because it provides more accurate, complete, and reliable information on diagnosis ⁹⁴. Each imaging modality has its own advantages and disadvantages regarding sensitivity, spatial/temporal resolution, and penetration depth. The NIR optical imaging possesses the advantage of high sensitivity

but limited spatial resolution. In contrast, MRI provides excellent spatial resolution and unlimited depth penetration but remains at an inherently low sensitivity⁹⁵. The combination of NIR emitting PLNPs and MRI contrast agent into one single nanoplatform would offer attractive synergistic advantages in biomedical imaging with high sensitivity, good spatial resolution, high signal over noise ratio, and no ionizing radiation.

Several strategies have been reported using gadolinium to have access to bimodal imaging PLNPs. In 2014, Yan and co-workers reported the surface functionalization of $\text{Zn}_{1.1}\text{Ga}_{1.8}\text{Ge}_{0.1}\text{O}_4$ with DTPA-Gd complexes to get bimodal imaging agents, able to be detected both by optical imaging and as a positive T_1 MRI contrast agent⁹⁶. In 2015, Maldiney *et al.* reported another method to prepare PLNPs-based bimodal imaging agent. In their strategy, instead of using the surface of the PLNPs to link DTPA-Gd complexes, the authors co-doped their NPs by incorporating various amount of Gd^{3+} ions at the beginning of the synthesis, by substituting Ga^{3+} ions²⁸. In such conditions, a negative contrast was obtained, with a r_2 value close to $60 \text{ mM}^{-1} \cdot \text{s}^{-1}$. In 2017, Zou *et al* used a third strategy to prepare bimodal imaging agents. In their work, a novel core-shell structure multimodal imaging probe was prepared, based on the coverage of mesoporous silica nanoparticles (MSNs) loaded $\text{ZnGa}_2\text{O}_4:\text{Cr}^{3+}$, Sn^{4+} (ZGOCS@MSNs) by a Gd_2O_3 shell ($\sim 1.5 \text{ nm}$)³¹. Compared with previously reported Gd^{3+} -based NIR persistent luminescence-based multimodal nanoprobess, the as-prepared nanoparticles enable surface available, no persistent intensity loss and only a slight size increase.

Instead of gadolinium, ultrasmall iron oxide nanoparticles (also called USPIO) are used in clinic as negative (T_2) MRI contrast agents. In 2015, Teston *et al* reported the first synthesis of mesoporous nanohybrids (MPNHs) composed of a combination of PLNPs and USPIOs embedded into a mesoporous silica structure to get a bimodal imaging agent⁹⁷. In addition of its bimodal imaging property, the authors showed that this nanoplatform could be attracted by a magnet. They used this property to label and attract cells *in-vivo*³⁰.

Theranostics nanoprobess

During the past 10 years, it has been shown that diagnosis and therapy could be combined within a single multifunctional nanomaterial, known as theranostic nanoparticles^{98, 3b}. The ideal theranostic nanomaterial should possess several advantages⁹⁹: (i) the ability for highly selective accumula-

tion in the diseased tissue, (ii) the capability of delivering an effective therapeutic action selectively, and (iii) safety concerns and the material should undergo biodegradation into nontoxic byproducts. Such nanomaterials are one of the keys for detection and treatment of early stage cancer in the 21st century¹⁰⁰. Theranostic nanoparticles are still in the very early translational stages, with nearly all efforts devoted to preclinical studies and no clinical trials to date. In this section, we will highlight the recent use of PLNPs for the development of new theranostic nanoprobcs.

Typically, mesoporous silica ($mSiO_2$) can be used as a promising drug carrier because of its stable mesoporous structure, high specific surface area and good biocompatibility¹⁰¹. Using mesoporous silica alone as a drug carrier it appears impossible to effectively locate a target site (i.e., tumor) and monitor in real time due to the lack of detectable signals. In 2014, Maldiney *et al* introduced a multifunctional nanoplatfrom based on persistent luminescence nanoparticles for both *in-vivo* optical imaging and drug delivery²⁹. Taking advantage of the well-known biocompatibility of mesoporous silica and non-toxic ZGO, the authors modelled a novel core-shell structure on the basis of a hybrid chromium-doped zinc gallate/mesoporous silica architecture specifically designed to allow both highly sensitive optical detection through living tissues and concomitant drug delivery. Employing doxorubicin as a drug model, the authors demonstrated that these mesoporous persistent luminescence nanophosphors can be successfully loaded with an anti-cancer agent, and subsequently initiate its progressive release in a pH-sensitive manner. The use of such doxorubicin-loaded theranostic agent is finally shown to induce acute cytotoxicity toward U87MG cells *in vitro*, preserved persistent luminescence properties, and allowed both sensitive and non-invasive localization of the carrier *in-vivo*.

In 2016, Zhang and co-workers developed another core-shell NPs composed of Gd_2O_3 , ZGO and $mSiO_2$ ¹⁰². This multimodal nanoprobe could be detected both by MRI and optical imaging and was used to monitor it's bio-distribution in healthy and tumor bearing mice after loading with doxorubicin (see Figure 8).

Fig. 8. Synthesis procedure of mesoporous ZGO, loading with Dox and *in-vivo* application, adapted from¹⁰².

Photodynamic therapy with PLNPs

A few photosensitizers (PSs), such as Porfimer sodium, have been approved by the United States Food and Drug Administration (FDA) to treat certain cancers. However, most of the available PSs for PDT can only be activated under the UV/visible light (<700 nm) with low tissue penetration. New strategies in order to use such molecules in better conditions are under investigations.

Photodynamic therapy (PDT) is a relatively new modality for cancer treatment¹⁰³. PDT consists of three essential components: light, oxygen, and a photosensitizer¹⁰⁴. Photosensitizers, often pharmacologically inactive without illumination, can be activated by light of a specific wavelength. This activation is followed by transfer of energy to nearby oxygen molecules to generate cytotoxic reactive oxygen species (ROS), most importantly singlet oxygen ($^1\text{O}_2$). PDT is a relatively less invasive treatment modality, inducing low systematic toxicity and causing little intrinsic or acquired resistance. One primary downside of PDT, however, is its inability to treat tumors located deep under the skin due to the short penetration depth of light in tissues¹⁰⁵. This problem can be partially alleviated by advanced light-delivering technologies that allow for illumination of certain internal cavities, such as the bladder, prostate, lung, and oesophagus. Nevertheless, it is considered as challenging or impossible for conventional PDT to treat tumors of large volumes or multiple loci. High X-rays dose (>5.0 Gy) are needed for efficient cancer therapy, and this dose was comparable or even higher than the fraction dose used in clinical radiotherapy¹⁰⁶. Many researches have shown that high-dose X-ray irradiation inevitably causes damage to normal tissues¹⁰⁷.

In this regard, reducing X-ray dose is a major concern for deep-tissue PDT practical applications. Persistent luminescence nanoparticles (PLNPs) are promising luminescent materials that can store excitation energy and then slowly release the trapped charge carriers to emit persistent luminescence without continuous excitation. PLNPs alone have limited impact on cancer cells development¹⁰⁸. In 2015, a novel X-ray inducible photodynamic therapy (X-PDT) approach was reported that allows PDT to be regulated by X-rays. Upon X-ray irradiation, the integrated nanosystem, which corresponds to a core of a nanoscintillator $\text{SrAl}_2\text{O}_4:\text{Eu}^{2+}$ (SAO) and a mesoporous silica coating loaded with photosensitizers (MW540)⁹³, converts X-ray photons to visible photons to activate the photosensitizers and cause efficient tumor shrinkage (see Figure 9).

Fig. 9. Schematic of PLNPs loaded with PDT agent (MC540), adapted from⁹³.

In 2018, Yang and co-workers have shown that Zn (II) phthalocyanine tetrasulfonic acid (ZnPcS4) can be covalently linked to ZGO in order to fabricate PDT nanoplatfrom (ZGO:Cr/W–ZnPcS4). The persistent luminescence continuously excites the coupled PS after the X-ray irradiation has been removed, resulting in significantly reduced X-ray dosage (≈ 0.18 Gy), minimizing the side effects of PDT treatment ^{23a}.

Photothermal therapy with PLNPs

Among all the cancer treatments, near infrared (NIR 700 nm–1000 nm) laser induced photothermal therapy (PTT) is a non-invasive mode that can generate heats at the tumors site ¹⁰⁹. Compared with conventional therapies, it has attracted much interest in tumor therapy as a minimally invasive and non-injurious therapeutic program as well as large penetration into bio-tissues. Photothermal therapy agents are capable of converting light energy into heat inducing a rise in the local temperature beyond 42 °C and killing consequently cancer cells. As a FDA-approved drug, indocyanine green (ICG) is attractive for localized hyperthermia by absorbing near-infrared light (around 800 nm) even in deep tissue to generate heat. To further enhance the fluorescence stability and Enhanced Permeation and Retention effect (EPR) of ICG in-vivo, a variety of nanocarriers have been developed. Therefore, the ICG loading nanoparticle is a brand-new photothermal therapy agent which could be diffusely used in clinic for photothermal therapy. Nevertheless, the traditional ICG loading nanoparticles cannot be tracked when they circulate in the body, therefore it is out of the question to ensure an accurate irradiation with additional near infrared (700 nm–1000 nm) laser. Optical imaging-guided therapy has received great attention recently and become a fine choice of loading the ICG and tracking it for PTT ¹¹⁰.

Fig. 10. PTT with ICG loaded PLNPs. Adapted from ¹¹⁰.

Very recently in 2017, Chang and co-workers used ZGO NPs and ICG co-loaded into mesoporous silica nanoparticles (ICG@mZGO nanoparticles) for imaging-guided PTT ¹¹¹. The ICG@mZGO nanoparticles consisted of two parts: i) The mesoporous $\text{SiO}_2/\text{ZnGa}_2\text{O}_4:\text{Cr}^{3+}$ (mZGO) near-infrared PLNPs were used as near-infrared emitting probe to track NPs in-vivo after excited by light. They found that mesoporous silica nanoparticles could be a fine template to synthesize ZGO phosphors in situ with good morphology

and expected size. Mesoporous silica nanoparticles could also load and protect ICG, which had the ability to avoid particle aggregation effectively during the transportation; ii) Indocyanine green (ICG), a kind of tricarbocyanine dye, which substantially absorbed the near-infrared light (NIR) and in result returned a photothermal response.

In a different approach, upconverting PLNPs and photo-absorbing agents ICG co-loaded into mSiO_2 (UC-PLNPs and ICG co-loaded mSiO_2) nanoparticles were developed for upconverting and persistent luminescent imaging guided PTT⁵². The UCPLNPs and ICG co-loaded mSiO_2 nanoparticles include three parts: i) the upconverting PLNPs, doped with Pr^{3+} , Er^{3+} and Tm^{3+} chosen for upconverting and persistent luminescent imaging. These NPs could be excited by NIR light and emit upconverting luminescence, which can be used for *in-vivo* imaging when the persistent luminescence signal became weak; ii) the ICG (indocyanine green), a water soluble anionic tricarbocyanine dye which is the only NIR agent approved by FDA and which has been widely used clinically for PTT; iii) the mesoporous silica nanoparticles working as a three dimension hard template for loading and protecting both upconverting PLNPs and ICG⁵².

Perspectives of the NIR-persistent luminescent nanoparticles for bio-imaging

Firstly, to move the persistent emission wavelength to the second and third biological windows (see Figure 11-C) and compared the pro and drawbacks of such strategy, different approaches are currently explored: i) Traps can be coupled to various rare earth cations (Yb^{3+} , Nd^{3+} , Er^{3+} , Ho^{3+} , Tm^{3+} or even Pr^{3+}) well known in the case of NIR lasers (see Fig 11-(B))¹¹². ii) Down shifting can also be obtained by coupling different rare earth cations or metal transition (MT) and rare earth cations. The so-call traps redistribution by energy transfer can downshift the persistent emission wavelengths (see Fig 11-(A)). Such down-conversion processes were proposed recently for garnet and perovskite hosts coupling metal transition cation (Cr^{3+}) with rare earth cations to promote by persistent energy transfer the persistent luminescence in the deep red (from Cr^{3+}) but also in the second or third biological windows (results were obtained with Yb^{3+} , Nd^{3+} , Er^{3+} , Ho^{3+} ions)^{47, 113}. iii) Transition metal cations such as Ni^{2+} or Co^{2+} were also recently investigated¹¹⁴ (see Fig 11-(D)) to cover the second biological window and these MT cations have also large potential. Indeed, recently, Qiu and co-workers reported a

strategy for tuning the emission bandwidth and intensity in persistent luminescence phosphors by synthesizing Ni^{2+} -doped $\text{Zn}_{1+y}\text{Sn}_y\text{Ga}_{2-x-2y}\text{O}_4$ phosphors, with an emission band peaking from 1270 to 1430 nm in the second near-infrared (NIR) window ¹¹⁵. Ueda ¹¹⁶ and Tanabe ¹¹⁷ and our groups also reported compositions able to be used in the second transparency window ^{114a}.

Fig. 11. : (A) trap redistribution and persistent energy transfer in metal transition (Cr) and rare earth doped garnet ⁴⁷, (B) Trivalent rare earth cations as efficient emitters for short wave infrared persistent luminescence ¹¹², (C) Optical window in biological tissues; high: attenuation coefficient (log) as a function of the wavelength. Bottom: sensitivity of different cameras (silicon (Si) cameras, indium gallium arsenide (InGaAs) or mercury cadmium telluride (HgCdTe)) ¹¹⁸ and (D): optical features and persistent luminescence of $\text{ZnGa}_2\text{O}_4:\text{Ni}$ and $\text{Zn}(\text{GeGa})_2\text{O}_4:\text{Ni}$. Decay profiles and scheme of excitation, adapted from ^{114a}.

Secondly, moving to the second and third near-infrared transparency windows (1000 nm - 1700 nm) allows further reducing of scattering, absorption and tissue autofluorescence effects. Moreover, if the quantum efficiency values, for Si, InGaAs (and HgCdTe) can be comparable, the acquisition time is also to be taken into account. For a CCD camera the dark current is of the order a few electron/pixel/hour and it is about 100 ke/pixel/s for InGaAs. The readout noise has also to be taken into account (see values in Table 3). By the end, the signal is one to two orders of magnitude smaller for InGaAs detector (and HgCdTe) in the NIR range with respect to the Si detector in BW-1. Imaging platforms then require the use of powerful and cooled cameras.

Table 3: Characteristics of the NIR and deep-red sensors

	1.5 micron NIR camera	Visible camera
Material type	InGaAs	Si/CMOS
Quantum efficiency	80 %	80 %
Readout noise	about 30 e ⁻	about 1 e ⁻
Dark current	100 ke ⁻ /pixels/s	< 0.5 e ⁻ /pixels/s
Pixel size	About 15 micrometers	About 4 micrometers

Then the balance between benefit and drawbacks for these two aspects related to the second and third biological windows remains to be explored. In a decade, persistent luminescence has emerged as a useful property for the development of new optical imaging nanoprobe. Thanks to this property of

long lasting luminescence, imaging in the first transparency window (NIR-I) with high target to background ratio has been reported by several research groups (more than 40 research groups all over the world and more than 200 papers published in the last ten years). Beyond to their uses for imaging and therapy, new applications emerged. We can cite for instance their use for the development of nanothermometers. For this purpose, the idea is to use the luminescence emitted by nanoprobe to detect temperature variations at the nanoscale^{119 36}.

Conclusions

Very recent and promising works on persistent luminescence nanoparticles concern the development of biocompatible nanoprobe for multimodal imaging, theranostics and drug delivery, photodynamic and photothermal therapies, nanothermometry and hyperthermia etc. These PLNPs were even recently proposed as a guide during surgery operation to accurate delineation of hepatocellular carcinoma¹²⁰.

Then, one of the next challenge will be: can these PLNPs be used at wavelengths longer than 1000 nm ?. Indeed, optical imaging in the second near-infrared region (NIR-II, 1.0-1.7 μm) has recently attracted significant attention owing to some advantages : if the ballistic light (without diffusion) and the so-call “snake” light with limited diffusion remain about the same, the diffuse light with all the information is lost can be very limited moving in that NIR range. Photons in the NIR-II region can provide high spatial resolution at deep tissue penetration depths owing to the reduced scattering of long-wavelength photons¹²¹. Furthermore, imaging in the second near-infrared region addresses several challenges simultaneously: minimal autofluorescence of biological tissue leading to increased sensitivity and significantly reduced light attenuation from scattering and from absorption by the blood and other structures enables imaging with high spatiotemporal resolution and penetration depth¹²². Consequently, large organisms such as a whole mouse may be rendered translucent when imaged in this area¹²³. The image reconstruction is also a very important work as was done for instance in the acousto-optic images¹²⁴.

Nevertheless there is still several others challenges for the scientific community such as, the synthesis of long circulating “stealth” nanomaterials, the functionalization with targeting agents prerequisite for PLNPs later use. Others important requirements are to keep a high intensity at the nanoscale, and to well control the surface properties in order to stabilize the PLNPs in

different biological media. Then, with the nano-objects, imagination will be the only limit to the use and applications of these fascinating nanoparticles.

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Fig. 1. Left: Schematic of the energy levels and traps involved in the persistent luminescence mechanisms. Right: Partial transparency of living tissues in deep red range (see light through girl left hand and figure 11 (c)), extracted from *Saint Joseph charpentier*, G. de La Tour, 1643, Musée du Louvres, Paris, France

Fig. 2: TEM pictures of a) $\text{ZnGa}_2\text{O}_4:\text{Cr}^{3+}$ (ZGO) nanoparticles embedded in glass matrix (TC = 900 °C) b) ZGO nanoparticles after hydrothermal synthesis assisted by microwave heating, associated distribution diagram, adapted from E. Glais et al and V. Castaing et al.

Fig. 3: Left, (top) Photoluminescence and (bottom) persistent luminescence spectrum at room temperature, adapted from Bessière et al [2011]. Right: Persistent luminescence decay of $\text{ZnGa}_2\text{O}_4:\text{Cr}^{3+}$ Nanoparticules (within more than 2 hours) adapted from V. Castaing et al.

Fig. 4: Thermoluminescence glow curves with different sizes and shapes and right: Simulation and experimental EPR spectra for the ZGO:Cr extracted from V. Castaing et al.

Fig 5: Top: different strategies to perform in vivo optical imaging with persistent luminescence nanoparticles (PLNP). (1) Excitation of a suspension of nanoparticles followed by tail vein injection and optical imaging using a photon counting camera. (2) Once the in vivo persistent luminescence signal has completely disappeared, some PLNP can give further persistent luminescence signal either by photo-or X-rays stimulations possibly giving n decay signals. (3) Some PLNP can be excited without preliminary excitation (with lower efficiency) in the animal body using visible, near infrared –for upconversion- or X-rays photons, Bottom Traps could be emptied by a near infrared source (energy $\sim 1\text{eV}$) showing the crucial role of charge-storage to deep traps and photo-transfer to shallow traps leading to persistent luminescence. This demonstrated the photostimulation capability at convenience time. Within this experiment, one observed long-lasting phosphorescence under 977 nm light excitation⁸⁷, making in vivo excitation of the spinels probes envisioned and possible long term imaging applications

Fig. 6. *In vivo* imaging using in situ excitable PLNPs. (a-c) Optical properties of powder after either UV or visible excitation. (d-h) *In vivo* excitation of un-functionalized PLNPs (e) or stealth ZGO in healthy (f) on tumor bearing mice (g).

Fig. 7. (a) Principle of in vivo imaging using X-ray excitation. (b) Application in vivo, adapted from Xue et al.

Fig. 8. Synthesis of mesoporous ZGO, loading with Dox and in vivo application, adapted from Dai et al. .

Fig. 9. Schematic of PLNPs loaded with PDT agent (MC540) adapted from Chen et al. .

Fig. 10. PTT with ICG loaded PLNPs. Adapted from Chen *et al.*

Fig. 11. : (A) trap redistribution and persistent energy transfer in metal transition (Cr) and rare earth doped garnet ⁴⁷, (B) Trivalent rare earth cations as efficient emitters for short wave infrared persistent luminescence ¹¹², (C) Optical window in biological tissues; high: attenuation coefficient (log) as a function of the wavelength. Bottom: sensitivity of different cameras (silicon (Si) cameras, indium gallium arsenide (InGaAs) or mercury cadmium telluride (HgCdTe)) ¹²⁵ and (D): optical features and persistent luminescence of $\text{ZnGa}_2\text{O}_4:\text{Ni}$ and $\text{Zn}(\text{GeGa})_2\text{O}_4:\text{Ni}$. Decay profiles and scheme of excitation ^{114a}