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Revealing Crystallization Dynamics and Compositional Control Mechanism of 2D Perovskite Film Growth by In-situ Synchrotron-based GIXRD

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Abstract

Quasi-two-dimensional (2D) perovskites promise the intrinsically stable solar cell performance. However, the crystal orientation and phase distribution in 2D solution processed perovskite are difficult to be manipulated, which restricts the device efficiency as well as its reproducibility. Here, we simply incorporate potassium ion (K^+) into *quasi*-2D precursor solution, which can dramatically change the nucleation steps during perovskite films spin-coating process probed by *in-situ* synchrotron-based grazing incident X-ray diffraction (GIXRD). It is notable that a desired vertical oriented 2D phase without intermediate compound can be easily formed after spin-coating, which simultaneously reduces the distribution of low dimensional 2D perovskite phases in association with suppressed trap states. Therefore, the power conversion efficiency of doped 10% K^+ 2D perovskite solar cells can yield up to 11.3% as well as long-term stable performance with high reproducibility. This work paves a key path to control the *quasi*-2D nucleation and crystallization processing via chemical additives.

In recent years, organic-inorganic hybrid lead halide perovskites have drawn wide attentions due to its outstanding photoelectric properties.¹⁻⁵ The perovskite material has widely applied to solar cells,^{1, 6} light-emitting diodes (LED),⁷⁻⁸ and photodetectors.⁹⁻¹⁰ The power conversion efficiency (PCE) of three-dimensional (3D) perovskite solar cell has raised up to 25.2%,¹¹ however, the stability issues still retard the future development including the temperature degradation, humidity degradation and ultraviolet light-induced degradation.¹²⁻¹⁵ In order to improve 3D perovskite material stability, interfacial modification has been considered as an efficient method to prevent the degradation of perovskite solar cells.¹⁶⁻¹⁹ Except for the optimization of interface engineering, the moisture sensitivity of perovskite material can also be crucial for device resultant stability.²⁰⁻²² Dimensionality engineering has drawn consideration attention to improve devices stability by structural manipulation.²³⁻²⁵ Two-dimensional (2D) halide perovskite has been recently considered as more stable materials than 3D ones. Meanwhile, 2D perovskite plays crucial roles on photovoltaic performance, which can be used as pure light absorber,²⁶⁻²⁹ capping layer,³⁰⁻³¹ passivation layer or mixed 2D/3D light absorber.³²⁻³⁶ The application of 2D perovskite can obviously improve the stability and performance of perovskite solar cells.

Among the multiple application, *quasi*-2D Ruddlesden-Popper (RP) perovskite can be formed by incorporating organic long chain molecules,³⁷⁻³⁸ which can retard the 3D perovskite crystal growth. Meanwhile, hydrophobic organic molecules could protect perovskite from moisture intrusion. The 2D RP structure can be regarded as the dimensional reduction of 3D perovskite, which has the chemical formula of A_2B_n .

${}_1M_nX_{3n+1}$, where A represents organic ligand, such as phenethylammonium (PEA⁺), butylammonium (BA⁺); B represents monovalent cation, such as formamidinium (FA⁺), methylammonium (MA⁺); M represents divalent metal ion, such as lead (Pb²⁺), tin (Sn²⁺); and X stands for halide ion, such as I⁻, Br⁻ and Cl⁻. n represents the dimensional thickness of the inorganic layers, which affects the optical and electronic properties of *quasi*-2D RP perovskite. In *quasi*-2D RP perovskite, the exciton binding energy is enhanced due to the dielectric confinement effect between organic layer and inorganic layer, and the charge transport also be suppressed on account of the additional introduction of bulky organic ligands.

Recently, considerable works have found that the *quasi*-2D perovskite at room temperature can always form different phase distribution due to the rapid reorganization process even though the precursor solution is prepared with particular nanoplatelet thickness (n value).³⁹⁻⁴⁰ Transient absorption spectroscopy indicates that the electrons and holes can incur funnel effect, where energy can transfer from low n -number to high n ones.⁴¹⁻⁴² Different nanoplatelet thickness distributions would affect the funnel effect. Beside the phase distribution issue, it has been found that when the crystal growth orientation is parallel to substrate, it would hinder charge transport in the out-of-plane direction due to the insulating organic layers, which affects device performance.⁴³ Therefore, both the crystal orientation and phase distribution are crucial to improve the performance of *quasi*-2D RP perovskite solar cells.

In order to solve the above-mentioned issues of parallel orientation and mixed phase distribution in *quasi*-2D RP perovskite, Tsai *et al.* found that thin film morphology

appears near-single-crystal quality by hot-cast technique, more delightedly, the (001) and (010) planes of inorganic perovskite component display remarkable out-of-plane direction relative to substrate, which facilitates efficient charge transport. The PCE of $\text{BA}_2\text{MA}_3\text{Pb}_4\text{I}_{13}$ composition drastically increased from initial 4.02% to 12.5%.²⁹ Although the efficiency was enhanced obviously, the mechanism and orientation control has not been fully understood until recently Chen *et al.* revealed that the nucleation and vertical crystal growth incurs in the liquid-air interface.⁴³ Unfortunately, hot-cast technique is difficult to be manipulated because the substrate requires a quick transfer from a hot plate to cold spin-coater chuck in order to maintain the substrate temperature, therefore, the device exhibits poor reproducibility. Liu *et al.* utilized mixed solvent of dimethyl formamide (DMF) and dimethyl sulfoxide (DMSO), which can also form desired vertical orientation.³⁹ However, the method tends to form multiple phase distribution due to slow down the crystallization, which results in inferior PCE of 8.54%. Alternatively, chemical additives are likely to control the growth orientation and improve crystallization quality, such as ammonium thiocyanate (NH_4SCN), ammonium chloride (NH_4Cl), methylammonium chloride (MACl).^{28, 44-45} However, the incorporation of chloride ion requires the regulation of anti-solvent, which increase the complexity of processing. Recently, alkali metal doping in 3D perovskite has drawn tremendous attention, it can obviously improve both the performance and stability of solar cells. Our previous work also revealed that incorporating potassium ion (K^+) into 3D precursor solution can form new potassium-rich phase in grain boundary, which

reduced the trap densities.⁴⁶ However, based on our best knowledge, there is no research on small alkali metal doping in *quasi*-2D perovskite.

Here, K^+ was used to align the 2D perovskite layers perpendicular to the substrate as well as tune the nanoplatelet phase distribution. It was found that the doped K^+ *quasi*-2D perovskite film achieved desired vertical orientation *via in-situ* GIXRD and θ -2 θ X-ray diffraction (XRD) characterizations during spin-casting at room temperature without any other further post-treatments. Notably, compared with DMSO solvent method and hot-cast technique, the addition of K^+ can directly convert to large crystallite size 2D phase without intermediate compound and suppress the formation of low n -number phase ($n < 4$), which is favorable for preparing highly efficiency and stability *quasi*-2D perovskite solar cells. Therefore, the optimized 10% K^+ doped solar cell can achieve the PCE of 11.3% with excellent repeatability. In addition, the un-encapsulated devices can sustain 90% of initial performance after 1000 hours in 40% $\pm$ 10% relative humidity at room temperature. Our results provide new insight into morphology, crystallinity, and orientation of *quasi*-2D perovskite targeting highly efficient and stable device.

Following previously reported method,³⁹ *quasi*-2D perovskite with $BA_2MA_3Pb_4I_{13}$ composition was fabricated *via* dissolving 1 M PbI_2 , 0.75 M methylammonium iodide (MAI) and 0.5 M butylammonium iodide (BAI) in DMF as the reference device. The films without and with K^+ are defined as BAMA (0% K^+) and KBAMA, respectively. To fabricate KBAMA *quasi*-2D perovskite films, KI is incorporated into BAMA perovskite precursor solution with different molar ratios ($x\% = M_{KI}/M_{BAMA}$) from 5% to

20%. Meanwhile, in order to compare the properties of *quasi*-2D perovskite films with different processing conditions, we also fabricate the films with DMSO solvent and adopting hot-cast technique.^{29, 39, 47} The detailed experimental information can be found in **Supporting Information ST1**. To investigate the morphology of films under different treatment conditions, we employ the scanning electron microscopy (SEM) characterization, as shown in **Figure 1(a)-(d)**. When DMF is used as solvent and the film is spin-coated at room temperature (DMF RT), there are some pinholes in the film, and the film is discontinuous due to the rapid crystallization.⁴⁸ When DMSO instead of DMF is used as solvent (DMSO RT), the crystal shape turns to be typical flake and the film becomes more continuous due to the reduced crystallization rate. Interestingly, when 10% K⁺ is incorporated, the film morphology becomes smooth without any pinholes in association with few crystalline cracks, which is as similar as that by hot-cast technique. Continuous film would benefit charge transport and reduce trap states, which is favorable to enhanced performance of perovskite solar cells. When different concentration K⁺ is incorporated, the corresponding SEM images are shown in **Figure S1**. There are still some pinholes when only 5% K⁺ is incorporated. Further increasing the concentration of K⁺ beyond 10%, the film morphology quality becomes worse, especially when doped 20% K⁺, there are many large pinholes on the surface. The excess K⁺ likely acts as impurity, which may retard the perovskite crystallization. In order to investigate the distribution of K⁺ in *quasi*-2D perovskite film, we conduct energy-dispersive X-ray spectroscopy (EDX) mapping of 10% KBAMA film, as shown in **Figure S2**. K⁺ is homogeneously distributed in the *quasi*-2D perovskite film, which

is as similar as Pb^{2+} and I⁻. The phenomenon is different from our previous work in doped K^+ 3D perovskite, where K^+ is only located at grain boundary.⁴⁶

In order to investigate the crystal growth orientation of *quasi*-2D perovskite films with different processing conditions, the GIXRD characterization is conducted, as shown in **Figure 1(e)-(h)**. The crystal growth orientation with varied processing conditions are different, which is obviously showed in **Figure 1(i)-(j)**. The DMF RT film shows powder-like rings in the diffraction pattern indicating the grains are randomly oriented, which is as similar as 3D perovskite crystal. The DMSO RT film shows sharp and discrete Bragg spots, which indicates the *quasi*-2D crystals predominately orient vertically. Simultaneously, sharp (0*k*0) peaks appear at small angle, which belong to a mixture of low *n*-number phases.⁴⁹⁻⁵⁰ The film fabricated with 10% K^+ additive exhibits the sharp, sporadic Bragg spots, suggesting highly vertical oriented growth, being consistent with hot-cast technique. In addition, the peaks for low *n*-number phases disappear in the GIXRD pattern, which would indicate that K^+ incorporation hinders the formation of lower *n*-number nanoplatelet phases in association with promoting the desired oriented *quasi*-2D phase. This result is astonishing, as vertical orientation and organized phase in *quasi*-2D perovskite film are widely convinced to be favorable for solar cell performance.

The θ -2 θ XRD pattern is utilized to analyze different processing conditions affecting on the formation of *quasi*-2D perovskite crystal phase and crystallinity, as shown in **Figure 1(k)** and **Figure S3**. The DMF RT film shows weak diffraction peak at 13.95 ° and 27.9 °, indicating its poor crystallinity. The diffraction peak intensity of DMSO RT

film is stronger than that of DMF RT film distinctly. However, accordingly to the enlarged XRD pattern shown in **Figure 1(k)**, the diffraction peaks are located at 6.7° , 8.9° and 10.2° , which should be assigned to different lower n -number phases ($n = 2$ & $n = 3$)⁴⁹. As it is expected, the film with 10% K^+ additive exhibits two dominant peaks at 14° and 28° that correspond to (111) and (222) lattice planes of *quasi*-2D perovskite, respectively, and the diffraction peak intensity is dramatically enhanced, which is consistent with hot-cast technique. Meanwhile, the enlarged XRD pattern shows no extra diffraction peaks at small angle, indicating that the phase distribution becomes narrow. When doping different K^+ concentration, all the diffraction peak intensities are enhanced compared with that of reference one. However, especially doped 20% K^+ , the film exhibits weak series of crystal diffraction of (0*k*0) lattice plane of *quasi*-2D perovskite, which manifests that the superfluous K^+ prompts the formation of lower n -number phases. Besides, there is no shift for the diffraction peak of (111), as shown in **Figure S3 (b)**, indicating that K^+ incorporation could not change the crystal structure. **Figure 1(l)** shows the full-width at half-maximum (FWHM) and crystallite size of *quasi*-2D perovskite films for the (111) lattice plane with different processing conditions, the corresponding parameters are plotted in **Table S1**. The FWHM of DMSO RT, hot-cast technique and doped 10% K^+ films all decrease from 0.66° , 0.17° to 0.16° , respectively, indicating the film with K^+ displays superior crystallinity compared with reference one. In order to further investigate the crystal quality, we calculate the crystallite size of specific lattice plane according to Scherrer formula,⁵¹ the detailed calculation of average crystallite size is discussed in **Supporting**

Information ST2. Regarding to DMSO RT film, crystallite size increases from 12 nm (DMF RT) to 35 nm, and the crystallite size of hot-cast film even reaches to 48 nm. When doped 10% K⁺, the crystallite size is as large as 51 nm. The larger crystallite size decreases the formation of grain boundaries, which can reduce non-radiative recombination and is beneficial to the solar cell performance.

To further investigate the vertical oriented formation mechanism for KBAMA film, we particularly investigate the crystallinity process of *quasi*-2D perovskite films. KI aggregates, which show much lower solubility in DMF than either MAI-PbI₂ or BAI-PbI₂ colloids, would likely generate early precipitate out of the solution during spin-coating process. In this case, we suspect that any generated KI aggregates would act as crystal nucleus at the air-liquid interface in the early stage of spin-coating, leading to the further heterogeneous crystallization with vertical distribution of perovskite materials. Similar mechanism of nucleating at air-liquid interface has already been conducted in the former work.⁴³ Therefore, we speculate that the crystallization for doped 10% KBAMA film has already incurred along vertical crystallization during the spin-coating process, which is as similar as hot-cast process. To verify the above speculation, the XRD patterns for *quasi*-2D perovskite films before and after the annealing are shown in **Figure S4**. The 0% KBAMA film displays different XRD spectra before and after the thermal annealing. Before the annealing, there are intermediate phases in the film where crystallization process has not finished yet. However, there is no difference between the XRD spectra of doped 10% KBAMA film before and after thermal annealing process, which could also indicate that the film has

already finished the crystallization. Doped 10% KBAMA film thickness has nearly no change before and after the annealing, as shown in **Table S3**. On the contrary, there is large film thickness variation of 30-60 nm for 0% KBAMA film before and after the annealing. This observation is consistent with the above XRD measurement.

Besides, we observed the crystallization dynamics and compositional control mechanism of doped 0% and 10% K⁺ *quasi*-2D perovskite films during spin-coating process by *in-situ* synchrotron-based GIXRD characterization. The testing schematic diagram is shown in **Figure 2(a), (b)**. The *in-situ* GIXRD signals for one-minute formation process of 0% and 10% K⁺ *quasi*-2D perovskite films are recorded by a supporting video (SV) of SV01 and SV02, respectively. From **Figure 2(c)-(g)**, diffraction signals of perovskite grains of doped 0% KBAMA film appear at approximately 12 s, and signals for low *n*-number intermediate phases could be clearly observed after 20 s. When further increasing spin-coating process time, the signals of crystal grains enhanced with random orientation, which is consistent with previous results. By comparison for the doped 10% KBAMA film, as shown in **Figure 2(h)-(l)**, the crystallization signals appeared at 15 s and enhanced significantly without any intermediate phases when further increasing spinning coating time. And the vertical orientation has formed even without annealing process. **Figure S5** shows the azimuthally integrated intensity profiles for these two films derived from **Figure 2(c)-(g)** (0% K⁺) and **Figure 2(h)-(l)** (10% K⁺), respectively. The diffraction peak (111) of perovskite films becomes very sharp after ~15 s spin-coating elapsed process. The sharp and strong diffraction peak (111) indicates that the film crystallization incurs in

this stage. And the crystallization process would finish during spin-coating method. Besides, the corresponding radially integrated intensity of the diffraction rings at $q \approx 10 \text{ nm}^{-1}$ at different five time points (0, 12, 20, 40, 60 s) for 0% K^+ film and (0, 15, 20, 40, 60 s) for 10% K^+ film are showed in **Figure S6**. The film with 0% K^+ shows several small peaks at different azimuth angles, which could not be found in 10% K^+ film, indicating that the film with 0% K^+ crystallizes with random orientations. The results further verify the above hypothesis that the incorporation of K^+ can directly convert to vertical oriented *quasi*-2D perovskite phase. Besides, according to the perovskite diffraction peaks (111) for 0% K^+ and 10% K^+ films with different spin-coating time in **Figure S7(a)-(b)**, we could clearly find that the intensity of diffraction peak (111) of film with 0% K^+ reaches its saturation point at approximately 20 s. However, the intensity of 10% K^+ film reaches similar value at 20 s compared with 0% K^+ film, and further increases till the end of spin-coating process. This result indicates the better crystallinity of K^+ doped film. **Figure S7(c)-(e)** plot the diffraction peak (111) position, intensity (peak area) and FWHM from films with different spin-coating time, which reveal the more symmetric crystallization structure and improved crystallinity for K^+ incorporated film.

To probe the phase distribution and light absorption property of *quasi*-2D perovskite films with different processing conditions, ultraviolet visible (UV-vis) absorption and photoluminescence (PL) characterizations are conducted, as shown in **Figure 3** and **Figure S8, S9**. According to the UV-vis absorption spectra, it is observed that the films display multiple phases containing different n value although all precursor solutions are

prepared according to formula of $\text{BA}_2\text{MA}_3\text{Pb}_4\text{I}_{13}$. When DMF is used as solvent, the absorption peaks representing 2D phases are almost invisible due to the prominent 3D perovskite phase. The slowing crystallinity prompts to the formation of different 2D phases when adopting DMSO as solvent. For the *quasi*-2D perovskite processed with hot-cast technique, excitonic peaks at ~ 570 nm, ~ 606 nm, ~ 642 nm are observed, corresponding to different *quasi*-2D phases ($n=2, 3, 4$). Once K^+ is incorporated, the excitonic peaks representing $n=2, 3, 4$ remain un-shifted. Dramatically, the absorption peak for $n=3$ are prominently enhanced while other peaks are weakened. In other hand, the incorporation of K^+ can enable the formation of *quasi*-2D perovskite phases, being as similar as the signal observed from the 2D perovskite film processed with hot-cast technique and DMSO solvent.

The PL peak position can be different in *quasi*-2D perovskite when pumping film along two different directions either from the front side (surface of the perovskite film) or the rear one (substrate) if *quasi*-2D perovskite containing large phase distribution⁵². PL measurement is conducted to further probe phase distribution of *quasi*-2D perovskite films fabricated under different processing conditions, as shown in **Figure 3(c), (d)** and **Figure S6**. When probed from the front side, the films with different processing conditions all display one peak. However, the PL peaks of DMSO RT (770 nm), hot-cast technique (777 nm) and doped 10% K^+ films (777 nm) all shift to longer wavelength approaching to 778 nm, being consistent with the peak position of 3D perovskite. When probed from the back side, the film processed with hot-cast technique exhibits obvious extra peaks at 577 nm, 617 nm, 652 nm, as shown enlarged PL peak

in **Figure 3(d)**, representing different low n -number phases ($n=2, 3, 4$), which is consistent with the absorption exciton peak. In contrast there is still one PL peak for doped 10% K^+ film when pumping rear side. It can be safely concluded that the doped 10% K^+ film reduces the distribution of different phases according UV-vis absorption and PL spectra measurement. When further increasing the K^+ concentration to 20%, the low n -number phases ($n<4$) increase again according to UV-vis absorption (**Figure S5**) and PL spectra measurement (**Figure S6**), which is consistent with XRD result (**Figure 1k**). In addition, the PL intensity of doped 10% K^+ film reaches the peak value. Further increasing the concentration of K^+ results in deteriorating PL, which is likely correlated with the existence of extra multiple 2D phases.⁵³ Beside the difference of PL intensity, the main PL peak begins to red shift when doped 5% K^+ , and the peak position remains no obvious shift when further increasing the concentration of K^+ . Combined with the PL spectra of the film fabricated with different processing conditions, it is inferred that the low- n phase is favored at the bottom and high- n phase is inclined at the top, the schematic diagram is shown in **Figure 3(b)**.

The suitable orientation and organized phase in *quasi*-2D perovskite are both very critical for optoelectronic device application²⁸⁻²⁹. LED depends on the quantum well confinement existed between multiple different vertical oriented phases, which is favorable to efficient electron-hole recombination and carrier transport. Similarly, the *quasi*-2D perovskite solar cell performance is affected by multiple factors, mainly including phase distribution, vertical orientation, crystallization and grain size, which can suppress the carrier recombination and accelerate charge transport. To investigate

how crystal orientation and phase distribution impact the performance of *quasi*-2D perovskite solar cells, we prepare conventional devices with different processing conditions, the device structure is shown in **Figure S10(a)**. The structure is fluorine-doped tin oxide glass (FTO) /TiO_x/perovskite/2,2',7,7'-Tetrakis[N,N-di(4-methoxyphenyl)amino]-9,9'-spirobifluorene (Spiro-OMeTAD) /Ag. **Figure 4(a)** displays the current density-voltage (J - V) curves of *quasi*-2D perovskite solar cells with different processing conditions under simulated air mass (AM) 1.5G one sun illumination. The electrical output characteristics of all the devices are summarized in **Table 1 and Table S2**. The PCE of KBAMA devices gradually increases with increasing K⁺ concentration. The PCE of 10% KBAMA device reaches a peak value of 11.3% with the open-circuit voltage (V_{oc}) of 1.00 V, short current density (J_{sc}) of 18.9 mA cm⁻², and fill factor (FF) of 52%. The PCE is obviously superior to BAMA device, especially the J_{sc} and FF values, which is correlated with the vertical orientation and organized phase. The hot-cast technique fabricated device with PCE of 9.82% is lower than that of 10% KBAMA device, but the J_{sc} is slightly higher and FF is poorer because thickness of 10% KBAMA film (~180 nm) is thinner than that of hot-cast technique (~250 nm). With further increasing K⁺ concentration (>10%), PCE begins reduced (**Figure S11**), which is consistent with the above poor film morphology and inferior UV-vis and PL characteristics. Particularly doped 20% K⁺ perovskite solar cell, the PCE reduces from 11.3% to 8.2%. The hot-cast technique film can achieve the highly crystallization and vertical orientation^{29, 54}, however, the poor controllability retards further development due to difficult manipulation the whole film fabrication process.

To investigate the 10% KBAMA device repeatability compared with the hot-cast technique, the distribution of 25 devices with hot-cast technique and doped 10% K⁺ KBAMA one is plotted in **Figure 4(b)**. The PCE of BAMA devices with hot-cast technique distributes between 6.6% and 9.8%, however, the 10% KBAMA devices distribute between 10.5% and 11.3%, which indicates that the latter device performance not only surpasses the former, but also has decent reproducibility. Besides, according to previous report,⁴⁰ vertical oriented *quasi*-2D perovskite film can not only benefit charge transport but also improve the moisture stability. We also investigated the long-term stability of un-encapsulated 10% KBAMA and 3D MAPbI₃ devices stored in air under dark for 1000 h in 40%±10% relative humidity, details are shown in **Supporting Information ST3**.

The effect on the performance of *quasi*-2D perovskite devices is not only ascribed to vertical oriented crystallinity, but also correlated with the trap states in perovskite films.⁵⁵ To further evaluate the trap densities of doped 0% K⁺ and 10% K⁺ *quasi*-2D perovskite films, space-charge-limited-current (SCLC) method by fabricating electron-only devices was conducted.⁵⁶⁻⁵⁷ **Figure S13** shows dark *J-V* curves of electron-only devices with different perovskite films. The primary linear region represents the Ohmic contact when the bias voltage is low enough. The intermediate region can be regarded as trap filling process when increasing the current injection. Further increasing the bias, the trap states are constantly filled until reaching the trap-filling limit voltage (V_{TFL}). The formula detail can be expressed in **Supporting Information ST4**. Therefore, lower V_{TFL} reflects the low concentration of defect states. The DMF RT *quasi*-2D BAMA

device shows higher V_{TFL} (0.88 V), however, upon doped 10% K^+ , the V_{TFL} value is obviously lower to 0.26 V. Therefore, the corresponding calculated trap density for KBAMA film reduced from $9.6 \times 10^{16} \text{ cm}^{-3}$ (BAMA) to $2.8 \times 10^{16} \text{ cm}^{-3}$. The calculated trap density for KBAMA film reduced from $9.6 \times 10^{16} \text{ cm}^{-3}$ (BAMA) to $2.8 \times 10^{16} \text{ cm}^{-3}$. The excellent crystallinity and vertical orientation for *quasi*-2D 10% KBAMA perovskite dramatically suppresses the defect states.

In summary, we have developed a simple method to achieve the desirable vertical oriented *quasi*-2D perovskite film for high efficiency and stable solar cell *via* incorporation of K^+ . Compared with the reference film, it has been found that the K^+ doping perovskite precursor wet film converts directly to desirable 2D phase with excellent crystallinity along vertical orientation even in spin-coating process monitored by real time *in-situ* synchrotron-based GIXRD. Meanwhile, the vertical oriented phase could occupy the majority of crystallized film with the incorporation of K^+ . The easier crystallization step at air-liquid interface can well interpret the preferred oriented crystallization process. Accordingly, the optimal 10% KBAMA device achieves the PCE of 11.3% and the un-encapsulated device can maintain 90% of initial performance after 1000 h in $40\% \pm 10\%$ relative humidity at room temperature. In addition, the devices simultaneously show high repeatability. Our work provides one new method for the orientation regulation of *quasi*-2D perovskite apart from the popular hot-cast technique.

Supporting Information

Supporting Information is available from the ACS publication or from the authors. The

videos of *in-situ* GIXRD characterization for perovskite films could also be found in Supporting Information.

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Conflict of Interest

The authors declare no conflict of interest.

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Figures and Captions

Figure 1 SEM images of *quasi*-2D perovskite thin films with different processing conditions in a) DMF RT, b) DMSO RT, c) Hot-cast technique and d) doped 10% K⁺. All the scale bars are 1 μm. GIXRD patterns of *quasi*-2D perovskite thin films with different processing conditions in e) DMF RT, f) DMSO RT, g) Hot-cast technique, h) Doped 10% K⁺. i) Schematic diagram of parallel orientation for *quasi*-2D perovskite. j) Schematic diagram of vertical orientation for *quasi*-2D perovskite. k) θ -2 θ XRD

patterns of *quasi*-2D perovskite thin films with different processing conditions. Inset shows their amplified XRD pattern at small diffraction angle. 1) FWHM and crystallite size of *quasi*-2D perovskite thin films with different processing conditions.

Figure 2 a) Schematic diagram of *in-situ* synchrotron-based GIXRD characterization for doped 0% K⁺ *quasi*-2D perovskite thin films. b) Schematic diagram of *in-situ* GIXRD characterization for doped 10% K⁺ *quasi*-2D perovskite thin films. *In-situ* GIXRD patterns of doped 0% K⁺ *quasi*-2D perovskite thin films in spinning coating process from 0 s to 60 s. c) 0 s, d) 12 s, e) 20 s, f) 40 s, g) 60 s. Red arrows represents low *n*-number intermediate phases. *In-situ* GIXRD patterns of doped 10% K⁺ *quasi*-2D perovskite thin films in spinning coating process from 0 s to 60 s. h) 0 s, i) 15 s, j) 20 s, k) 40 s, l) 60 s.

Figure 3 a) UV-vis absorption spectra for *quasi*-2D perovskite thin films with different processing conditions. b) Schematic diagram of energy transfer between different *quasi*-2D perovskite phase and 3D phase. c) Steady-state front side PL spectra of *quasi*-2D perovskite thin films with different processing conditions. Inset shows the test direction from the front perovskite side. d) Steady-state rear side PL spectra of *quasi*-2D perovskite thin films with different processing conditions. Right inset shows the test direction from the rear side. Left inset shows their amplified PL pattern at short wavelength.

Figure 4 a) Typical J - V curves of *quasi*-2D perovskite devices with different processing conditions under simulated AM 1.5 G light illumination. b) PCE distribution of solar cells with 10% K⁺ and hot-cast technique.

Table 1 Solar cell parameters for the *quasi*-2D perovskite devices with different processing conditions.

Sample	V_{oc} (V)	J_{sc} (mA cm ⁻²)	FF (%)	PCE (%)
DMF RT	1.01	11.8	36.1	4.24
DMSO RT	0.99	14.8	57.2	8.46
Hot-cast	1.00	18.9	52.1	9.82
10% K ⁺	1.08	17.1	62.3	11.3

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We discover that the incorporation of potassium ion into *quasi*-2D precursor solution can change the nucleation process *via in-situ* synchrotron-based grazing incident X-ray diffraction, which forms vertical oriented 2D phase without intermediate compound and reduces the distribution of low *n*-number phases.

Revealing Crystallization Dynamics and Compositional Control Mechanism of 2D Perovskite Film Growth by In-situ Synchrotron-based GIXRD

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TOC