Atmospheric input of trace metals to the western Mediterranean: uncertainties in modelling dry deposition from cascade impactor data

By FRANÇOIS DULAC¹, PATRICK BUAT-MÉNARD¹, ULLAH EZAT¹, SAMI MELKI² and GILLES BERGAMETTI³, ¹*Centre des Faibles Radioactivités, Laboratoire Mixte CNRS-CEA, Domaine du CNRS, B.P. 1, F-91198 Gif-sur-Yvette Cedex, France*; ²*Faculté des Sciences, Université Libanaise, Hadath-Beyrouth, Lebanon*; ³*Laboratoire de Physico-Chimie de l'Atmosphère, Université Paris 7, 2 Place Jussieu, F-75251 Paris Cedex 05, France*

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ABSTRACT

The mass-particle size distributions (MSDs) of Na, Al, Cd, and Pb were determined from 17, 1–5 day, high-volume cascade impactor samples collected throughout the Western Mediterranean atmosphere between 1980 and 1983. As expected, the mass median diameter (MMD) was the largest for Na, representative of sea-salt aerosol, with a median value of 5.9 μm . The median value for the MMD for Al, representative of mineral aerosol, was 2.8 μm . The smallest values of the MMD were found for pollution-derived elements, Cd and Pb: 0.7 μm . In most cases, the MSDs, for each of the elements, were log-normal. Total dry deposition velocities were calculated from the two-layer deposition model of Slinn and Slinn (1980) using three approaches: (i) by characterizing the distribution as a MMD, (ii) by considering the size distribution as directly given by the cascade impactor, (iii) by fitting the assumed log-normal distribution and dividing it into 100 successive intervals. The first approach appeared to give underestimates. The two other approaches yielded similar results for Cd and Pb, of the order of 0.05 cm s^{-1} . For these elements, however, more than 20% of the total dry deposition flux was due to particles with diameters of 7.2 μm or greater (collected by impactor stage 1). For Na and Al, the third approach yielded values at least one order of magnitude higher than when using the two others. This clearly underlines the major role played by large particles in controlling the dry deposition of sea-salt and mineral aerosol particles. Direct measurements of Al dry deposition, made in 1985–1986 on the northwestern coast of Corsica, agree best with the values predicted by the third approach (mean calculated and measured values of 1.8 and 3.0 cm s^{-1} respectively). Scanning electron microscopy examinations of samples confirm that mineral aerosol particles with diameters of 10 μm or greater dominate the dry deposition flux. Preliminary data indicate that for mineral aerosol particles, this flux is, on a yearly basis, about half of the wet deposition flux. However cascade impactor data are not adequate to retrieve true mass-particle size distribution in the size range which controls dry deposition ($D > 7 \mu\text{m}$), and this is shown to be the source for one order of magnitude uncertainties in dry deposition calculations.

1. Introduction

In a previous paper, we have investigated the factors influencing the variability of atmospheric concentrations in the Western Mediterranean, using geochemical data on elemental tracers of natural and anthropic aerosol sources (Al, Na,

Br, Mn, V, Pb, and Cd), and 3-dimensional air-mass trajectories (Dulac et al., 1987). The determination of the biogeochemical fluxes of these elements to the marine environment requires a quantitative estimate of the total atmospheric input, and the knowledge of its temporal variability. This requires an assessment

of factors influencing wet and dry deposition removal rates. Wet deposition is generally the major contributor to the total deposition fluxes of trace elements in oceanic regions (Arimoto et al., 1985; Uematsu et al., 1985; Dedeurwaerder et al., 1986). Over the Western Mediterranean Sea, the total deposition of mineral aerosol particles appears also quantitatively dominated by sporadic red rains and dust storms (Loÿe-Pilot et al., 1986; Bergametti et al., 1989). However, the almost continuous dry deposition flux must also be known in order to assess the impact of atmospheric inputs on the Mediterranean marine system. Indeed, the climate in this region is characterized by long periods without precipitation (GESAMP, 1985). Further, residence times of some reactive trace elements in surface sea water may be short enough (a few days or weeks; Coale and Bruland, 1985) so that dry deposition can affect their surface marine biogeochemical cycles. Moreover, the vicinity of African aerosol sources should induce higher dry deposition velocities than over remote oceanic regions, due to the presence of a significant coarse fraction in the mineral aerosol.

Elemental deposition fluxes are generally calculated from atmospheric concentrations because of the difficulties of a direct assessment

of atmospheric inputs over the sea (Hicks et al., 1980). Although models of dry deposition of particles on natural water surfaces are already available, our knowledge of deposition processes is still not complete (Slinn, 1983). In this paper, we present some calculations and measurements of dry deposition velocities and dry deposition fluxes of trace elements to the Western Mediterranean, in order to focus on the sensitivity of the results to the mass-particle size distributions, and on related uncertainties. Al, Cd, Na, and Pb dry deposition fluxes have been calculated using the dry deposition velocity model of Slinn and Slinn (1980), which has been applied to our data on atmospheric concentrations as a function of particle size (Dulac et al., 1987). Comparisons have been made with results from Al dry deposition samples collected in Corsica.

2. Material and methods

2.1. Sampling and analytical procedures

17, 1–5 day size-separated aerosol samples have been collected throughout the Western Mediterranean during 4 cruises between September 1980 and October 1983. Sampling locations are given in Fig. 1. The samples were collected

Fig. 1. Sampling locations in the Western Mediterranean Sea.

from a bow tower, on acid washed Whatman® 41 filters, using a 5-stage high volume slotted cascade impactor, model Sierra® 235. At the operational flow rate of $68 \text{ m}^3 \text{ h}^{-1}$, the cutoff diameters are as follows from the stage 1 to 5: 7.2, 3, 1.5, 0.95, and $0.49 \text{ }\mu\text{m}$ (Willeke, 1975). Curves of collection efficiency of the different stages as a function of particle size are often taken into account to retrieve continuous mass-size distributions from impactor data (Puttock, 1981). Their accuracy is a critical point. As pointed out by Jaenicke and Blifford (1974), calibration artifacts, such as the use of non true monodisperse calibration aerosols, are likely to produce the non-ideal, S-shaped curves observed. Thus, the cut-off diameter was considered as the lower limit size for particle collection on a given stage. A back-up filter collects the smallest particles.

Sampled air volumes ranged between 1000 and 4000 m^3 , and average wind speeds between 0.1 and 9.9 m s^{-1} . Total elemental concentrations from this sampling system were found to be statistically comparable to concentrations from bulk filter samples. 13 impactor samples were also collected between February 1985 and April 1986, from a 10 m high meteorological tower at Cape Cavallo, on the Northwestern coast of Corsica (Fig. 1), using a 5-stage low volume, circular monojet cascade impactor, model EGAI 80 (flow rate: $0.78 \text{ m}^3 \text{ h}^{-1}$; cutoff diameters: 8.9, 3.55, 2.1, 1.4, and $0.67 \text{ }\mu\text{m}$ (Bergametti et al., 1982)). The sampling site has an unimpeded view of the sea, is 700 m from the shore and at an altitude of about 300 m. 30, 9- to 21-day samples of total deposition were also collected there, using a CRAPAL sampler (Lambert, 1963; Lambert and Nezami, 1965). This is a hemispheric plexiglass collection device, 1 m^2 in surface, with a 10 cm-high neck, covered by a 1 mm nylon mesh. Its base is connected to an acid cleaned polyethylene bucket, containing 0.5 l of deionized Milli-Q® water acidified to pH 1 with suprapure HNO_3 . The buckets were kept in darkness during sampling, and changed about every 10 days. Before removal, the collector was carefully rinsed with 0.5 l of the HNO_3 solution. The collected solutions were filtered in a clean atmosphere, on $0.4 \text{ }\mu\text{m}$ porosity Nuclepore® filters, and the particulate and dissolved fractions were analysed for Al concentrations. Five of these samples of

total deposition occurred during periods with no precipitation (in July and September 1985, and March 1986).

Flame and flameless atomic absorption spectroscopy (Na, Al, Pb and Cd) and instrumental neutron activation analysis (Na and Al) were performed on samples from the cruises. Analytical procedures have been published in Dulac et al. (1987). X-ray fluorescence (Elichegaray et al., 1981; Losno et al., 1987) was used for Al analyses of impactor collections from Corsica and of the undissolved fraction of deposition samples. Dissolved Al was measured in the dry deposition samples by flameless atomic absorption spectrometry. Analytical precision was typically about $\pm 10\%$ for atomic absorption analyses, and $\pm 5\%$ for activation and fluorescence analyses.

2.2. Elemental mass-size distributions of aerosol particles

Total dry deposition rates were often calculated in the past using only the deposition velocity corresponding to the mass-median diameter (MMD) of particles carrying the element considered. Now, it is recognized as essential to consider the whole mass-size distributions (MSDs) of the elements in order to calculate deposition fluxes as a sum of partial fluxes from several particle size classes (McDonald et al., 1982; Meszaros, 1982; Dedeurwaerder et al., 1986; Arimoto et al., 1985; Arimoto and Duce, 1986). As elemental MSDs cannot be assessed from direct measurements of dry deposition to surrogate surfaces, they are generally obtained from cascade impactor sampling.

We have used three different approaches to take the MSDs into account in the dry deposition flux calculations, and to test their influence on the results. The simplest approach is to consider the discrete distribution as directly given by the cascade impactor ("6-step" method) (McDonald et al., 1982; Dedeurwaerder et al., 1986). The other two approaches depend on fitting of this discrete distribution to a continuous one by a regression technique. In a crude approach, the whole distribution was characterized only by its MMD ("1-step" method), while in a refined treatment the fitted distribution was divided into 100 discrete successive size intervals ("100-step" method), each one corresponding to 1% of the total elemental mass, and being characterized by

the diameter of particles at its center $D_{0.5\%}$, $D_{1.5\%}$, ... $D_{99.5\%}$; Arimoto et al., 1985). We assumed monomodal lognormal MSDs for fitting. This has been approximately verified, using either impactor or other data, for sea-salt (Na, Cl) (Woodcock, 1953), soil derived (Al) (Patterson and Gillette, 1977; Buat-Ménard et al., 1983), sulfate (S) (Whitby, 1978) and Pb-210 (Arimoto and Duce, 1986) aerosol particles.

In fact, it is now considered acceptable to approximate aerosols by the sum of three modes, each characterized by a lognormal size distribution (Jaenicke, 1985; Giorgi, 1988). These are the so called nucleation, accumulation, and coarse mode, with respective average MMD of about 0.03, 0.5, and 5 μm (Whitby, 1978). The backup filter of the impactor sampler has reasonable collection efficiencies for all particle diameters, and we verified by microscopic examinations that particles of a few tenths of microns in diameter are collected by impactor stage 1 (see Subsection 4.2). However, the limited number of stages and the range of the cut-off diameters unfortunately gives no mean to assess the MSDs in the nucleation size range ($D < 0.1 \mu\text{m}$) nor in the coarse size range over 10 μm . But we may suppose that the small particles carrying Pb or Cd in our samples come exclusively from the accumulation mode of anthropogenic aerosols, even though Pb and Cd emissions occur mainly in the nucleation mode. This is because, when they reach the sampled area, these pollutants have already been transported far from their continental sources. On the other side, the coarse mode of the atmospheric aerosol is mainly of natural origin, i.e., sea-salt and soil derived particles, and we may consider that Na and Al, tracers of these two types of aerosol particles, are only attached to coarse particles. Patterson and Gillette (1977) have shown the existence of another, coarser mode for soil-derived aerosol particles collected close to the source, during periods of high dust loadings. This mode, with its MMD ranging between 20 and 200 μm , would be characteristic of the parent soil size distribution. But no such extra coarse mode is likely to occur in our atmospheric samples. Indeed the atmospheric loadings of soil derived particles are generally less than 15 $\mu\text{g m}^{-3}$ (maximum observed Al concentration: 905 ng m^{-3}), and are thus much lower than the mg m^{-3} loadings for which

Patterson and Gillette reported the presence of an extra coarse mode in the aerosol size distribution. Indeed Schütz' (1979) calculations on aerosol particle transport show that such a separate coarse mode cannot exist far from the sources of particles.

Fitting was performed by least-square linear regression between t and $\ln D$, t being the parameter of the integrated normal distribution $\Phi(t)$, as given by tables (Van der Waerden, 1967) from the cumulative experimental MSDs, and $\ln D$ the natural logarithm of the particle diameter. The parameters, MMD and geometric standard deviation σ , of the fitted lognormal distribution are obtained from the regression equation

$$t = A \ln D + B, \quad (1)$$

as

$$A = 1/\sigma, \quad (2)$$

$$B = -\ln(\text{MMD})/\sigma. \quad (3)$$

2.3. Calculation of dry deposition fluxes

The dry deposition flux (F) of an element can be calculated from the product of its airborne concentration (C) and a dry deposition velocity (V):

$$F = CV. \quad (4)$$

Some dry deposition velocities measurements have been obtained from wind-tunnel experiments (Möller and Schumann, 1970; Sehmel and Sutter, 1974; Little and Wiffen, 1977) but direct measurements over natural water surfaces are still controversial (Wesely and Williams, 1981; Sievering et al., 1982). Therefore, these velocities are generally calculated using mathematical models (Slinn et al., 1978; Sehmel, 1980; Slinn and Slinn, 1980; Williams, 1982), although large uncertainties still exist, as discussed by Slinn (1983) and Sievering (1984), especially for the smallest aerosol particles, e.g., particles smaller than 1 μm in diameter (Arimoto and Duce, 1986).

In this work, dry deposition velocities were calculated using the model of Slinn and Slinn (1980) for deposition of particles on natural waters. Briefly, this model is based on the assumption of two layers over the sea surface. In the upper layer (about 10 m high), the flux of particles is governed by gravitational settling and turbulent transfer. Deposition at the air-sea inter-

face takes place in a viscous sublayer where turbulent transfer is negligible. In this layer, hygroscopic particles can grow by humidity absorption. This model assumes a stationary state (constant flux and immediate growth to the equilibrium size for hygroscopic particles entering the lower layer), and spherical particles. It does not take into account the re-emission of particles by the sea surface (Fairall and Larsen, 1984). Corrections for particle scavenging by wave breaking and spray formation have already been proposed (Williams, 1982) and discussed (Slinn, 1983). They can be significant for submicrometer size particles (Arimoto and Duce, 1986). However, we have not introduced them here, since they do not affect the dry deposition rate of large particles. Indeed, large particles will be shown to be a much more important factor controlling the total deposition rate of trace elements. The detailed equations that have been used are reported in the appendix.

The mean wind velocity was considered for each impactor sample, and we used a constant drag coefficient of 0.0013, which corresponds to a steady wind at 10 m over open waters, near neutral stability of the atmosphere (Kraus, 1972). We assumed a null rate of water evaporation, a density of 2.5 for constant size hydrophobic mineral particles, and a relative humidity of 75 and 90% in the upper and lower deposition layer respectively. Under such conditions, NaCl particle density is respectively 2.1 (dry state) and 1.1 (Eriksson, 1959). Their equilibrium radius (D_w) is related to their dry radius D_d by

$$D_w = 2.3489 D_d^{1.00638}, \quad (5)$$

after Fitzgerald (1975; see appendix). Calculations have been performed for Na, Al, Pb and Cd, assuming Al was present in mineral hydrophobic particles, Na in hygroscopic NaCl particles, and Pb and Cd in particles having the same specific gravity and hygroscopic properties than NaCl particles. It might be more realistic to treat Pb- and Cd-rich particles as having properties of $(\text{NH}_4)_2\text{SO}_4$ particles. However, results of dry deposition would not be very different (see Subsection 4.4).

Using the calculated dry deposition velocities (V_d) and the measured elemental concentrations from samples collected at sea, dry deposition fluxes were calculated as follows:

(i) *1-step method*:

$$F_1 = C K(\sigma) V_d(\text{MMD}) \quad (6)$$

where

$$K(\sigma) = \sigma^{2 \ln \sigma}. \quad (7)$$

$K(\sigma) \geq 1$, and the product $K(\sigma) V_d(\text{MMD})$ is the flux-mean deposition velocity for a log-normal size distribution (Slinn, 1983, eq. (160)).

(ii) *6-step method*:

$$F_6 = \sum_{i=1-6} C_i V_d(D_i), \quad (8)$$

where $D_1 = 10 \mu\text{m}$, $D_2 = 5 \mu\text{m}$, $D_3 = 2.5 \mu\text{m}$, $D_4 = 1.25 \mu\text{m}$, $D_5 = 0.75 \mu\text{m}$ and $D_6 = 0.25 \mu\text{m}$ (for impactor stage 1 to 5 and final filter respectively).

(iii) *100-step method*:

$$F_{100} = (C/100) \sum_{i=1-100} V_d(D_{(i-0.5)\%}), \quad (9)$$

where $D_{(i-0.5)\%}$ is the diameter at the center of the size interval i , which particles carry the i th% of cumulated mass of the log-normal MSD, and $C/100 = C_i$ by definition (see Subsection 2.2). A consequence of this discrete representation of a lognormal distribution is that we end up dealing with a distribution of particles divided into a finite number of size classes, these sizes ranging between the diameters corresponding to 0.5 and 99.5% of the cumulated mass.

3. Results

Results of elemental mass-size distributions (MSDs) from the 17 impactor samples collected at sea have been summarized for Al, Na, Cd and Pb in Dulac et al. (1987). Mean atmospheric concentrations are given in Table 1 and mean

Table 1. *Mean atmospheric concentrations and dry deposition fluxes*

Element	Mean concentration (ng m ⁻³)	Mean dry deposition (μg m ⁻² d ⁻¹)	No. of samples
Na	1,400	30,000	8
Al	340	870	13
Pb	44	1.45	11
Cd	0.66	0.064	8

Fig. 2. Average influence of mass-particle size distributions on the atmospheric load and on the calculated dry deposition flux (refer to Table 1 for respective mean values). Particle size classes from stage 1 to 6 are respectively: $>7.2 \mu\text{m}$; $3-7.2 \mu\text{m}$; $1.5-3 \mu\text{m}$; $0.95-1.5 \mu\text{m}$; $0.49-0.95 \mu\text{m}$; $<0.49 \mu\text{m}$.

Table 2. Elemental mass-median diameters (MMDs) in the Western Mediterranean and at Enewetak Atoll

Element	Western Mediterranean				Enewetak*	
	Range		No. of values	MMD, median (μm)	MMD, geometric mean (μm)	No. of values
	MMD (μm)	$\sigma^\dagger$				
Na	4.6-10.8	0.6-1.9	8	5.9	6.3	4
Al	1.9- 5.2	0.6-1.6	13	2.8	2.9	4
Cd	0.5- 2.1	0.5-1.6	8	0.68	0.82	—
Pb	0.4- 0.9	0.6-1.2	11	0.69	0.61	$<0.8^*$

* Tropical North Pacific Ocean; Arimoto et al. (1985).

$^\dagger \sigma$ is the geometric standard deviation of the fitted mass-particle size distributions.

distribution profiles in Fig. 2. As elements are present on each impactor stage, and according to the curves of impaction efficiency (Willeke, 1975; Knuth, 1979), it can be said that elements are present on particles showing wide continuous granulometric distributions. According to the correlation coefficients of the least square linear regression, the monomodal lognormal fitting of elemental MSDs was satisfactory at the 99% probability level for 40 size distributions (8 for

Na, 13 for Al, 11 for Pb, and 8 for Cd) over the 47 available ones. This was also the case for the 13 spare Al distributions obtained from the impactor samples collected on the coast of Corsica. A summary of results for the parameters (MMD and σ) of those fitted distributions is given in Table 2. Table 3 gives a summary of calculated dry deposition rates using the three methods described above, and results are plotted in Fig. 3.

Table 3. Dry deposition fluxes calculated by the three different size step methods for mass-size distribution characterization (see text)

Element	No. of samples	Mean MMD (μm)	No. of size steps	Dry deposition flux				
				unit	min.	max.	mean	std. dev.
Na	8	6.5	100	$\mu\text{g m}^{-2} \text{d}^{-1}$	390	210,000	30,000	240%
			6		480	1,000	700	29%
			1		395	2,460	915	75%
Al	8	2.8	100	$\mu\text{g m}^{-2} \text{d}^{-1}$	17	3,200	585	186%
			6		12	255	79	119%
			1		2.5	155	37	155%
Cd	2	0.72	100	$\text{ng m}^{-2} \text{d}^{-1}$	21	150	85	106%
			6		30	110	70	80%
			1		7	32	20	90%
Pb	3	0.66	100	$\text{ng m}^{-2} \text{d}^{-1}$	1,050	2,380	1,645	41%
			6		1,520	2,990	2,055	39%
			1		225	1,160	575	89%

Fig. 3. Comparative results of calculated dry deposition rates (○: maximum; ■: mean; ●: minimum; refer to Table 3 for detailed values). Two stages of several impactor samples have been analysed together for a detection limit purpose, so that the 6-step method of flux calculation could not be applied to these samples.

4. Discussion

4.1. Mass-median diameter

The summary of results given in Table 2 shows that we observed the lowest variability of the MMD for Na (factor 2.3) and the highest for Cd (factor 4.2). The MMDs which were observed at Enewetak Atoll by Arimoto et al. (1985) appeared to be less variable, and this could be related to the fact that elemental concentrations are less variable over the remote tropical North Pacific (Buat-Ménard et al., 1983; Arimoto et al., 1985) than over the Western Mediterranean Sea (Dulac et al., 1987).

It appears also from the comparison between MMDs found over the Western Mediterranean and at Enewetak Atoll (Table 2), that the MMD of Na is greater in the Pacific. This may be explained by the higher mean wind speed observed in that area, 9–10 m s⁻¹ (Arimoto et al., 1985), compared to about 4 m s⁻¹ for the Mediterranean samples.

On the other hand, the MMD of Al is 50% greater in the Mediterranean samples than in the Pacific samples. This can be ascribed to the proximity of continental sources, which have been shown to influence atmospheric elemental concentration levels over the Western Mediterranean (Dulac et al., 1987). Indeed it can

be seen in Fig. 4 that samples showing a MMD of Al greater than 3.5 μm were generally obtained close to the African coast. They also correspond to the highest Al atmospheric concentrations ($\geq 500 \text{ ng m}^{-3}$). Air mass trajectory analyses (Dulac et al., 1987) have indicated that during all these sampling periods, and only during these periods, did transport events from Africa occur. However, the MMDs of Al observed in Corsica during two similar dust events were respectively equal to 2.0 and 2.3 μm , and thus not different from values usually observed there ($2.3 \pm 0.4 \mu\text{m}$). This observation suggests a decrease of the MMD during the transport of mineral aerosol from Africa over the Mediterranean Sea (Fig. 4), probably due to gravitational settling (en route) of the coarse particle fraction. This argument is supported by the observation that the mean MMD for Al in Corsica (2.3 μm), about 1000 km away from African coasts, is hardly greater than the mean MMD observed at Enewetak Atoll (2.0 μm), about 5000 km away from the nearest continental sources. Such experimental results are in agreement with calculations of Schütz (1979), which predicted that the coarse fraction of aerosol particles during wind-blown dust transport events over the tropical North Atlantic should settle during the first 1500 km of transport, and that a further decrease of MMD should

Fig. 4. Geographical distribution of observed mass-median diameters of Al.

be rather limited. Nevertheless, scanning electron microscopic examinations of Corsican aerosol samples have confirmed the presence of large mineral particles which are a few tens of μm in diameter, and whose contribution to the dry deposition flux is likely to be predominant (Fig. 5).

Considering the proximity of continental sources, it may seem paradoxical that the MMD of Pb is lower in the Mediterranean than at Enewetak (Table 2). However this can be explained as follows: half of Pb dry deposition fallout at Enewetak is due to the recycling of Pb from the sea surface microlayer by sea-salt particles larger than $7.2\ \mu\text{m}$ (Settle and Patterson, 1982; Arimoto et al., 1985). This recycled component was found to be negligible in the Western Mediterranean (Dulac et al., 1987), where the concentrations of submicrometer-sized Pb-rich particles are two orders of magnitude higher than over the tropical North Pacific.

4.2. Dry deposition fluxes

The comparison between dry deposition fluxes calculated by the 3 different approaches shows that the 100-step method generally gives the highest value, and the 1-step method the lowest one (Fig. 3). Moreover, the observed ranges and standard deviations (Table 3) indicate that the 100-step method gives more variable results than the other two methods, especially for Na and Al which have the highest deposition fluxes. The results are much less variable for the pollutant elements, Pb and Cd, which are primarily associated with much smaller aerosol particles.

The highest values obtained using the 100-step method suggest the importance of the largest size fraction. Indeed, the last 1% of cumulated mass is generally composed of particles larger than $10\ \mu\text{m}$ in diameter, and sometimes larger than $100\ \mu\text{m}$ for Al and Na (Table 4), which have very high dry deposition velocities due to gravitational settling. This is illustrated by our results for the

Fig. 5. Picture (scale bar: $10\ \mu\text{m}$) and X-ray spectrum of a large aluminosilicate particle (scanning electron microscopy) from a filter collected on the coast of Corsica during an African dust transport event.

Table 4. Contributions (%) of the largest aerosol particles to the total atmospheric mass and dry deposition flux of Na, Al, Cd and Pb

Element	Particle size class	Contribution to the atmospheric load		Contribution to the dry deposition flux ^o	
		range	mean	range	mean
Na	* $D > 7.2 \mu\text{m}$	22–59	43	60–100	90
	† $D > D_{95\%}$ (68 μm)	—	5	23– 92	57
	‡ $D > D_{99\%}$ (273 μm)	—	1	7– 65	31
Al	$D > 7.2 \mu\text{m}$	10–42	24	49–100	87
	$D > D_{95\%}$ (25 μm)	—	5	32– 83	66
	$D > D_{99\%}$ (81 μm)	—	1	11– 53	35
Cd	$D > 7.2 \mu\text{m}$	0– 8	3.5	0– 84	32
	$D > D_{95\%}$ (5.7 μm)	—	5	19– 76	48
	$D > D_{99\%}$ (16 μm)	—	1	6– 53	23
Pb	$D > 7.2 \mu\text{m}$	0– 2	1	0– 40	23
	$D > D_{95\%}$ (3.7 μm)	—	5	26– 64	52
	$D > D_{99\%}$ (9.5 μm)	—	1	8– 30	22

^o Calculated using the 100-step granulometric approach (see text).

* Corresponds to particles collected by cascade impactor stage 1.

† Particles of the last 5% of cumulated mass (mean $D_{95,5\%}$).

‡ Particles of the last % of cumulated mass (mean $D_{99,5\%}$).

contributions from the largest particles to atmospheric concentrations and dry deposition fluxes (calculated using the 100-step method) presented in Table 4 and Fig. 2. For Al and Na, particles larger than 7.2 μm represent respectively 10 to 40 and 20 to 60% of the total mass, and appear to account for about 90% of the dry deposition fluxes. Although less than 10% of Pb and Cd are present in this size range, these large particles may be responsible for an average of more than 20% of Pb and 30% of Cd dry deposition fluxes (Fig. 2). These results are in agreement with those found at Enewetak Atoll (Arimoto et al., 1985; Arimoto and Duce, 1986). Moreover, the largest particles accounting for the last 1% of the cumulated MSDs are responsible for 20 to 35% of the dry deposition fluxes whatever the element (Table 4). Clearly, this emphasizes the need for a precise knowledge of MSDs for any accurate flux prediction.

For Na and Al, however, Arimoto et al. (1985) have verified that measured dry deposition fluxes over a plastic plate are within a factor 3 comparable to the fluxes calculated after the 100-step method, even for MMDs of more than 10 μm . Such an agreement appears to validate the 100-step granulometric approach. For the present

work, the results of Al dry deposition fluxes from dry deposition collections on the coast of Corsica also confirm the results from the 100-step granulometric approach (Table 5) and thus the influence of coarse particles on the dry deposition rate. The presence of particles larger than 10 μm in diameter during dust transport from Africa (Fig. 5) has already been observed in Barbados

Table 5. Elemental dry deposition fluxes in the Western Mediterranean

Element	Units	Range	Mean	No. of values
Na*	$\text{mg m}^{-2} \text{d}^{-1}$	0.4– 208	30	8
		0.4– 15†	4.6	7†
Al*	$\mu\text{g m}^{-2} \text{d}^{-1}$	17 –6550	870	13
Al ^o	$\mu\text{g m}^{-2} \text{d}^{-1}$	390 –2700	1090	5
Cd*	$\text{ng m}^{-2} \text{d}^{-1}$	0.2– 175	64	8
Pb*	$\text{ng m}^{-2} \text{d}^{-1}$	460 –2400	1450	11

^o Measured from sampling on the Northwestern coast of Corsica.

* Calculated using the Slinn and Slinn (1980) deposition model, applied to 100 size intervals stepped lognormal mass-size distributions.

† The high value of 208 $\text{mg m}^{-2} \text{d}^{-1}$ discarded from above results.

(Prospero et al., 1970). Moreover, microscopic investigations of the collection surface of impactor stage 1 of two different samples (11 and 17 (see Fig. 1), a case of pollution aerosol and a case of Saharan dust respectively) have also shown the presence of large aluminosilicate particles, with equivalent diameters up to 90 μm .

These considerations demonstrate that the use of the single dry deposition velocity for the MMD (1-step method) is definitely not accurate enough to calculate realistic dry deposition rates, and that the whole mass-size distribution has to be considered. For pollutant elements Pb and Cd, the deposition fluxes simply calculated using the 6-step method without fitting are roughly the same as the fluxes obtained using the 100-step method.

4.3. Dry deposition velocities

In Table 6 are summarized the results of mean elemental dry deposition velocities (V_d^*) integrated over the entire MSDs, deduced from flux calculations using the 100-step method, by the reciprocal relationship

$$V_d^* = F_{100}/C. \quad (10)$$

The mean dry deposition velocity of Na ranges from 1 to 9 cm s^{-1} , except for a very high value of 190 cm s^{-1} which corresponds to an unusual MMD of 10.8 μm . Such a value is not unrealistic, but in this case the lognormal fitting, which involves atmospheric particles of more than

1 mm in diameter, is questionable, although the probability of the fit is more than 99.9%.

The mean dry deposition velocity of Al is also generally about 1 cm s^{-1} , and ranges from 0.2 to 10 cm s^{-1} . In addition to these very large variations, there is a clear separation between low and high dust concentration periods, the averages being 0.87 and 3.9 cm s^{-1} respectively. Such variations are correlated with large changes in the MMD of Al: corresponding means $\pm$ standard deviations are 2.5 ± 0.5 and 4.5 ± 0.7 μm respectively. This variability suggests that the influence of African dust episodes on mean Al atmospheric concentrations over the basin is even greater on mean Al dry deposition rates. Moreover this shows again the necessity to have a good assessment of MSDs for both Al and Na. Dry deposition velocities measured from the Corsican samples are also presented in Table 6. The observed values range from 1.3 to 6.9 cm s^{-1} , and thus calculated and measured values are consistent.

Despite the presence of a non-negligible component of Pb and Cd in the coarse fraction of the aerosol, the calculated dry deposition velocities of these elements are one order of magnitude lower than those of Al and Na, generally lower than 0.1 cm s^{-1} (Table 6). The size of the largest particles of the MSDs for Pb and Cd is much smaller than those of Al and Na (Table 4). This probably explains the differences found between the calculated mean dry deposition velocities of these elements.

Table 6. *Elemental dry deposition velocities (cm s^{-1}) in the Western Mediterranean*

Element	Range	Median	Sampling-time weighted mean	Standard deviation	No. of values
Na*	0.93 – 190	2.50	20.2	267%	8
	0.93 – 8.7†	2.26	3.2	89%	7†
Al*	0.18 – 9.7	0.69	1.8	140%	13
Al $\leq 260 \text{ ng m}^{-3}$	0.18 – 2.6	0.49	0.87	100%	9
Al $\geq 500 \text{ ng m}^{-3}$	1.07 – 9.7	3.1	3.9	91%	4
Al°	1.3 – 6.9	3.8	3.0	59%	5
Cd*	0.0074 – 0.33	0.053	0.13	100%	8
Pb*	0.020 – 0.068	0.041	0.043	40%	11

° Measured from sampling on the Northwestern coast of Corsica.

* Calculated using the Slinn and Sinn (1980) deposition model, applied to 100 size intervals stepped lognormal mass-size distributions.

† The high value of 190 cm s^{-1} discarded from above results.

4.4. Uncertainties

Despite the apparent agreement between observations and calculations using the 100-step method, in this work for Al (Tables 5 and 6), and in Arimoto et al. (1985) for a few elements such as Al, Na, Mn, V, ... it is crucial to assess the uncertainties of such a dry deposition modelling approach. First it is clear from (9) that the calculated total dry deposition flux F_{100} (or from (10), the integrated dry deposition velocity V^*) is just the mean of the 100 different values calculated respectively for the 100 different size steps. As noticed in 2.3, although normal distributions are mathematically infinite, the range of particle size contributing to the calculated dry deposition is in fact limited by $D_{0.5\%}$ and $D_{99.5\%}$, due to the stepping of the MSDs. But these limits are simply due to the choice of a 100 size interval stepping of the MSDs. Indeed the limits would be $D_{1\%}$ and $D_{99\%}$ with a 50 size interval stepping, in which each size step would correspond to particles carrying no longer 1, but now 2% of the total mass of the element. These limits would be $D_{0.25\%}$ and $D_{99.75\%}$ with a 200 size interval stepping, each size step corresponding to particles carrying 0.5% of the total mass. It could seem at first glance that $D_{99.5\%}$ is not very different from $D_{99\%}$ or $D_{99.75\%}$. However, significant differences are found which can influence the integrated calculated deposition rate. This is because such percentages of the cumulated mass correspond to the large size tail of the lognormal distribution, where the slope of the distribution as a function of particle size is very low. Moreover, the related changes in dry deposition velocities are proportional to the square of the changes in particle diameter, as we are in the size range for which gravitational settling controls dry deposition. So if the number of intervals for the stepping is changed, this will modify even more the largest deposition velocities taken into account for individual size steps, and so their average will change. For instance let us consider a reference Al case with the following mean conditions: concentration = 340 ng m^{-3} ; MMD = $3.1 \text{ }\mu\text{m}$; $\sigma = 1.2$; specific gravity = 2.5; wind speed = 4 m s^{-1} . The particle diameters corresponding to particles at the ends of the lognormal MSD are the following: $D_{0.5\%} = 0.141 \text{ }\mu\text{m}$; $D_{1\%} = 0.190 \text{ }\mu\text{m}$; $D_{99\%} = 50.57 \text{ }\mu\text{m}$; and $D_{99.5\%} = 68.13 \text{ }\mu\text{m}$. If we apply a 50 size

interval stepping to this distribution for dry deposition calculation, i.e., if the size range of the distribution is reduced from $0.14\text{--}68.1 \text{ }\mu\text{m}$ to $0.19\text{--}50.6 \text{ }\mu\text{m}$, we average deposition velocities between 0.0055 and 19.4 cm s^{-1} , instead of values between 0.0067 and 35.2 cm s^{-1} . Finally, the resulting total flux or apparent velocity for dry deposition is reduced by a factor of about 2.2. So this is an important source of uncertainty, as we actually cannot decide where to put the cut-off on the large size tail of the lognormal MSDs of the elements, nor whether they are really lognormal over $10 \text{ }\mu\text{m}$ in diameter.

Unfortunately, geochemists have no other practical means than using cascade impactor samplers to retrieve elemental MSDs, although it is recognized that these devices are far from being perfect samplers. For instance, sampling artifacts such as mineral particle bouncing on the impaction surfaces have been seriously considered (Walsh et al., 1978; Knuth, 1979; Buat-Ménard et al., 1983). This can produce a shift in the apparent particle sizes and in the spreading of size distributions, which would lead to an underestimation of MMDs and to an overestimation of standard deviations of the MSDs. Also the true collection efficiencies of the different impactor stages as a function of particle size are generally poorly known because of calibration artifacts (Jaenicke and Blifford, 1974), because they depend on the aerodynamic size and density of particles and because atmospheric particles are very inhomogeneous. Moreover the sampling is not isokinetic. This may have important consequences on the overall collection efficiency of the sampler, which can be variable as a function of particle size. Unfortunately, we are not able to assess the MSD uncertainties resulting from these experimental biases, in order to deduce uncertainties in dry deposition calculations. However, we have performed several tests in order to discuss these final uncertainties. We shall mainly focus on Al, since we can compare calculated dry deposition rates to those measured using a surrogate surface.

Considering that the cascade impactor was inadequate for assessing Al MSDs, due to bounce-off effects of mineral particles, we have assumed that the MSD of Al was constant throughout the samples. Further, Al was assumed to be attached to particles from the coarse mode

of the atmospheric aerosol particles, characterized by $MMD = 6.8 \mu m$ and $\sigma = 0.8$, following Giorgi's (1988) hypothesis. The calculated Al dry deposition fluxes F_{100} ranged from 55 to $935 \mu g m^{-2} d^{-1}$, with a mean value of $350 \mu g m^{-2} d^{-1}$. This is roughly comparable to values calculated from impactor data (Table 5). However, for each sample, the ratio between the flux calculated with a constant MSD and the flux calculated from impactor data ranges from 0.12 to 6.4. Although individual results can be very different, it seems however that, on the average, the supposed spreading of the distributions (higher observed σ) compensates the supposed shift towards a lower MMD.

Finally, this constant MSD hypothesis gives a very stable apparent Al dry deposition velocity of $1.18\text{--}1.26 cm s^{-1}$. This suggests that wind speed and concentration have a minor influence on the model outputs compared to granulometric parameters (MMD and σ). To test the sensitivity of the results to the different parameters used in the model calculations (see Appendix), we considered the dry deposition flux for the above characterized reference Al case, and compared the calculated fluxes assuming a variation of $\pm 50\%$ for each of the 5 listed parameters (atmospheric concentration, MMD, σ , specific gravity, and

wind speed) plus the drag coefficient. The respective induced changes in the calculated fluxes are plotted in Fig. 6. They show that variations in the drag coefficient and wind speed have a negligible influence on the result (changes $< \pm 3\%$). Variations in the calculated Al flux have been found to be proportional to the changes in Al concentration and in the specific gravity of aluminosilicate particles, and roughly proportional to the square of the changes in MMD. This is because gravitational settling is the dominant deposition process for Al. Nevertheless, it is clear that the $\pm 50\%$ variation in σ has by far the greatest effect on the calculated flux. Indeed, the ratio found between the maximum and minimum calculated flux is about 85, i.e., one order of magnitude higher than the ratio found for a variation of $\pm 50\%$ in MMD.

For Pb, uncertainties due to the $\pm 50\%$ variation in drag coefficient and wind speed are more significant ($< \pm 20\%$), but still negligible compared to the effect from variation in σ . We have also investigated the influence of considering Pb to be attached to hygroscopic particles with properties of $(NH_4)_2SO_4$ particles (dry specific gravity = 1.8; wet specific gravity = 1.1; see Fitzgerald (1975) for the equilibrium size of particles as a function of the relative humidity). Dry

Fig. 6. Range of variation of the Al dry deposition flux F_{100} (see text) assuming $\pm 50\%$ variations of the six main parameters of the model successively. Reference case: drag coefficient = 0.013 ; wind speed = $4 m s^{-1}$; specific gravity = 2.5 ; Al concentration = $340 ng m^{-3}$; MMD = $3.1 \mu m$; $\sigma = 1.2$.

deposition fluxes of Pb were found to be lower by only 20–40% than when properties of NaCl particles were considered for Pb-rich particles.

This demonstrates that σ is definitely the most sensitive parameter of this dry deposition model, whatever the element considered. This will be particularly critical for Na and Al, as there is only one impactor stage sampling particles larger than $7.2 \mu\text{m}$ in diameter, making it tricky to characterize the MSDs in the size range which controls dry deposition.

Given such an influence, it is worth discussing if a $\pm 50\%$ error in σ is realistic. We verified that analytical uncertainties in the elemental concentrations from each impactor stage may induce changes of more than 10% in the σ of the fitted distribution (with equivalent probability of fitting), while related changes in MMD are much smaller. Moreover the fitting itself has a statistical uncertainty. According to York (1969), the standard error in the regression coefficient A from (1) may be approximated by:

$$S(A) = \left\{ \left(\sum_{i=1-n} [t_i - A \ln D_i - B]^2 \right) / (n-2) \left(\sum_{i=1-n} [\ln D_i - \overline{\ln D}]^2 \right) \right\}^{1/2}, \quad (11)$$

where n is the number of points to fit, and A and B the coefficients of the regression equation (1) (see Subsection 2.2). After (2), σ and A have the same relative errors:

$$S(\sigma)/\sigma = S(A)/A. \quad (12)$$

Using (11) and (12), it appeared from our data that the average relative statistical errors in σ , resulting only from the MSD fitting, are about $10 \pm 5\%$ for all of Na, Al, Pb and Cd. Thus, cumulated with analytical uncertainties, the use of a perfect impactor sampler may still result in an error of $\pm 25\%$ in the σ of the recovered distributions. Further, it seems quite plausible that the non-ideal characteristics of the sampler may lead to an equivalent error, in particular for dust particles because of possible bouncing.

5. Conclusions

Three different methods to characterize elemental mass-particle size distributions (MSDs)

have been used for calculations of elemental dry deposition fluxes in the Western Mediterranean. The comparison between results has shown the major role played by large particles ($D > 7.2 \mu\text{m}$) in controlling the dry deposition of atmospheric particles. Direct measurements of Al dry deposition agree best with the values predicted by the third approach, in which the total deposition fluxes are calculated as the sum of the partial fluxes due to particles from 100 successive size intervals.

Despite large uncertainties due to the imperfect knowledge of elemental mass-size distributions, and given the rather good agreement between direct measurements of deposition to a surrogate surface and model calculations, we used average dry deposition rates calculated using the 100-step method (Table 1) to estimate the elemental dry deposition fluxes to the Western Mediterranean Sea on a yearly time scale. On the basis of 360 days of dry deposition, yearly inputs per m^2 to the Western Mediterranean would be 320 mg of Al, 0.52 mg of Pb and 0.023 mg of Cd. The total amount of Al collected during the year of total deposition sampling in Corsica was 970 mg m^{-2} . By comparison, it appears that dry deposition would account for a third of the total Al deposition in the Western Mediterranean Sea. This fraction is two times higher than the one found over the remote tropical North Pacific (Arimoto et al., 1985), and would confirm the influence of African sources on the dry deposition of Al in the Western Mediterranean.

However, it appears that the model calculations may have uncertainties of more than one order of magnitude, due to the high sensitivity of the dry deposition rate to the largest particles, and to the use of imperfect cascade impactor samplers. This work points out that geochemists do not have available samplers adequate to characterize atmospheric mass-particle size distributions of elements in order to assess precisely their dry deposition rates to the sea from atmospheric measurements. An ideal sampling device should allow the fitting of elemental size distributions in the nucleation, accumulation and coarse modes, i.e., over four orders of magnitudes in particle size. It should also allow the determination of low atmospheric concentrations, as are often found over remote marine areas, on sampling time scales of a day or less.

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7. Appendix

The formulation used for our dry deposition velocity calculations, after the two layer model of dry deposition of particles on natural waters of Slinn and Slinn (1980) is given in this appendix. The CGS unit system is used throughout. In the basic equation, the overall resistance to dry deposition is considered as a combination in series of the resistance to deposition in each of the two layers of the model, minus a term due to gravitational settling:

$$1/V_d = 1/K_c + 1/K_d + (V_s - V_g(D_d))/(K_c K_d), \quad (13)$$

where

- V_d : dry deposition velocity of the particle;
- $1/K_c$: resistance to deposition in the constant flux layer;
- $1/K_d$: resistance to deposition in the deposition layer;
- V_s : Stefan velocity of the particle, due to the upward flux of evaporated water

(Waldmann and Schmitt, 1966; Vittori and Prodi, 1967);

- V_g : gravitational settling velocity of the particle, calculated after the Stokes approximation;

- D_d : diameter of the dry particle.

The terms K_c , K_d , V_s and V_g are given by eqs. (14) to (16) respectively:

$$K_c = C_d U/(1 - k) + V_g(D_d) - V_s, \quad (14)$$

where the three terms account respectively for turbulence, gravity, and evaporation;

$$K_d = C_d U(S_c^{-1/2} + 10^{-3/St})/k + V_g(D_w) - V_s, \quad (15)$$

where the four terms account respectively for Brownian diffusion, impaction, gravity, and evaporation;

$$V_s = E/d_a \quad (16)$$

$$V_g(D_x) = 2g D_x^2(d_x - d_a)/(9v d_a) \quad (17)$$

$$\approx 1.211 \cdot 10^6 D_x^2 dx, \quad (17a)$$

with

$k = 0.4$: Von Karman constant;

C_d : drag coefficient;

U : mean wind speed at the reference height (10 m);

Sc : Schmidt number;

St : Stokes number;

E : rate of water evaporation;

d_a : air specific gravity;

$g = 981 \text{ cm s}^{-2}$: acceleration of gravity;

$v = 0.15 \text{ cm}^2 \text{ s}^{-1}$: kinematic viscosity of air at about 20°C and 1 atm;

D_x : diameter of the particle in the state x (see after);

d_x : specific gravity of the particle in the state x (see after);

x : index of state of the particle; all particles are dry ($x = d$) in the upper layer; in the deposition layer, hygroscopic particles become wet ($x = w$), while hydrophobic particles remain dry.

The value of the Schmidt number is given by

$$Sc = v/B_d, \quad (18)$$

where

B_d : Brownian diffusivity of the particle in the deposition layer; it may be expressed as a

function of the particle diameter (p. 63 in Davies, 1966):

$$B_d = \{0.238/D_w\} \{1 + 163 \cdot 10^3/D_w + 548 \cdot 10^2/[D_w \exp(6.66 \cdot 10^6 D_w)]\}. \quad (19)$$

The Stokes number is calculated by

$$St = C_d U^2 V_g(D_w)/g v. \quad (20)$$

Finally, the wet particle diameter in the deposition layer depends on the relative humidity of the air. For soluble NaCl particles, the equilib-

rium size D_w is calculated as a function of the dry diameter of the particle D_d , using the relation given by Fitzgerald (1975):

$$D_w = 1.62 a (D_d/2)^b \quad (21)$$

where

$$a = \exp[66 \cdot 10^{-3} R_h/(1.058 - R_h)] \quad (22)$$

$$b = \exp[77 \cdot 10^{-5} R_h/(1.009 - R_h)]. \quad (23)$$

Rh: relative humidity, ranging from 0.81 to 0.97.

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