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Charged defects during alpha-irradiation of actinide oxides as revealed by Raman and luminescence spectroscopy

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Abstract

We have recently evidenced an original Raman signature of alpha irradiation-induced defects in UO₂. In this study, we aim to determine whether the same signature also exists in different actinide oxides, namely ThO₂ and PuO₂. Sintered UO₂ and ThO₂ were initially irradiated with 21 MeV He²⁺ ions using a cyclotron device and were subjected to an *in situ* luminescence experiment followed by Raman analysis. In addition, a PuO₂ sample that has previously accumulated self-irradiation damage due to alpha particles was investigated only by Raman measurement. Results obtained for the initially white ThO₂ showed that a blue color appeared in the irradiated areas as well as luminescence signals during irradiation. However, Raman spectroscopic analysis showed the absence of Raman signature in ThO₂. In contrast, the irradiated UO₂ and PuO₂ confirmed the presence of the Raman signature but no luminescence peaks were observed. These findings lead to propose a mechanism involving electronic defects in ThO₂ while a coupling between electronic defects and phonons is needed to explain the Raman signal in UO₂ and PuO₂.

Keywords: Alpha irradiation; Raman signature; Luminescence; Electronic defects

1. Introduction

Accurate determination of nuclear fuel behaviors in normal or accidental condition is a challenge for the nuclear industry in order to ensure its safety and profitability. Irradiation damages have a direct impact on the physical properties of the nuclear fuels and several theoretical works are currently in progress aiming to determine the stable point defect in UO₂. Unfortunately, limited experimental data is available to characterize irradiation defects in nuclear fuels. X-Ray diffraction and Transmission Electron Microscope evidenced changes in the crystalline structure and grain microstructure during irradiation [1-2]; however these methods are not well suited to evidence point defects. Recently [3] evidenced a specific signature of irradiation in UO₂ using the Raman spectroscopy method. This signal corresponds to three additional peaks in the Raman spectrum of UO₂ implanted with alpha ions and can be related to a single defect that is annealed between 300 and 525 °C [4]. According to [5] its creation kinetics can be modelled by a single impact model and has previously been used to describe UO₂ cell parameter behavior under irradiation. These results observed by Raman spectroscopy has characteristics similar to irradiation-induced point defects, and therefore deserved an in-depth characterization and understanding.

A striking feature of the Raman defect arises from its creation process. According to SRIM simulation code, a 25 MeV Helium ion beam will primarily lose its energy through electronic stopping. However, the electronic energy due to the alpha ions ($dE/dx = 1\text{keV/nm}$) is much weaker than the value reported in literature for creation of tracks by swift heavy ions [6]. In that range of deposited energy, only self-trapped excitons or colored defects could be formed [7]. To date the existence of such defect types has not been reported about irradiated UO₂.

This paper is devoted to understand the behavior of three actinide oxides namely UO₂, ThO₂ and PuO₂ since they all share the same crystallographic structure, with similar mass, density and number of electrons. The study aims to examine the existence of electronic processes in these oxides. Moreover the energy deposited profile in all these compounds, as determined by SRIM, shows same trend and this suggests that they should have similar behavior under irradiation.

2. Experimental Procedures

2.1 Samples

Sintered UO₂ and ThO₂ samples were prepared at CEA-Cadarache and ICSM respectively. UO₂ discs were heat treated at 1700°C under dry Ar/H₂ while ThO₂ was sintered at 1600°C in air. The discs were all mirror-polished at one side and then annealed at 1000°C to maintain stoichiometry and also remove polishing damages. In this work, the pellets were sectioned into two-half discs along their cylindrical axis as it allows to correlate the penetration profile of the ions along the depth of the samples with simulated data as described in [3]. The irradiation was performed at the Cyclotron facility, CEMHTI Orléans (France) and it was possible to carry out both *in situ* (Ion-Beam Induced Luminescence) and *ex situ* (Visual and Raman) experiments.

A PuO₂ sample was initially prepared at CEA-Cadarache and heat treated at 1700°C under reducing environment for 2 hours back in 2011. The pellet showed a high alpha activity due to the presence of alpha emitters. To consider the effect of alpha ions, the PuO₂ was left to decay for 2.5 years. The auto-irradiated pellet was later characterized with Raman spectroscopy in 2013.

2.2 Irradiations

The UO₂ and ThO₂ half-disks were all irradiated with 21 MeV He²⁺ during 2 hours and an ion flow of 50 nA using the cyclotron device. The energy chosen was able to deliver more electronic energy ($(dE/dx)_E = 132 \text{ keV}/\mu\text{m}$) near the surface of the pellets. All irradiations were carried out at room temperature and under vacuum condition so as to prevent any risk of pellet oxidation.

2.3 Ion-Beam Induced Luminescence Experiment (*In situ*)

The luminescence set-up used to monitor the behavior of UO₂ and ThO₂ during irradiation consisted of an optical set-up of a pyrometer probe implemented on the CEMHTI cyclotron beam line and positioned in the direction of the pellet's surface exposed to the irradiation beam. An optical fiber coming from the pyrometer probe-head was connected to a one-grating monochromator (SR 163 Shamrock Andor) and an ICCD multichannel detector (Andor iStar DH720) all located in a safety room 20m away from the irradiation cell, by means of optical fibers and electrical cables.

2.4 Raman measurement (*Ex situ*)

A Renishaw RA-100 Raman Analyzer was used to characterize fuel oxides before and after irradiation. A 633 nm He-Ne was initially employed and each Raman spectrum was acquired through an exposure time of 40s for a wavelength ranging from 300 to 700 cm⁻¹. A second step consisted of analyzing the irradiated pellets using a 532 nm green laser so as to confirm the presence of Raman signals.

A similar procedure was applied for PuO₂; however with a different experimental procedure. This set-up variance was necessary since the sample was within a glove box to prevent radiological contamination. Hence, the Raman head was placed outside the box and the laser was focused onto the sample passing through the glass of the glove compartment (Mitutoyo x20 objective, 30mm front distance).

3. Results

3.1 Ion-Beam Induced Luminescence

The luminescence spectrum of ThO₂ was recorded for a wavelength ranges from 200 to 800 nm during irradiation as shown in Fig. 1. Before irradiation ($t = 0 \text{ min}$), the ThO₂ sample emits no luminescence signals. The spectrum is largely modified given the number of peaks appearing during the initial stage of the He²⁺ beam. It can be seen that more emission occurs between 300 nm to 620 nm which could be related to the formation of electronic defects in the material as discussed by [8]. However under the same irradiation conditions, no luminescence peaks were observed in UO₂ and the spectrum remained unchanged even when the ion flow was increased from 50 nA up to 200 nA. This result provides evidence that a different defect mechanism is involved in UO₂ compared to ThO₂.

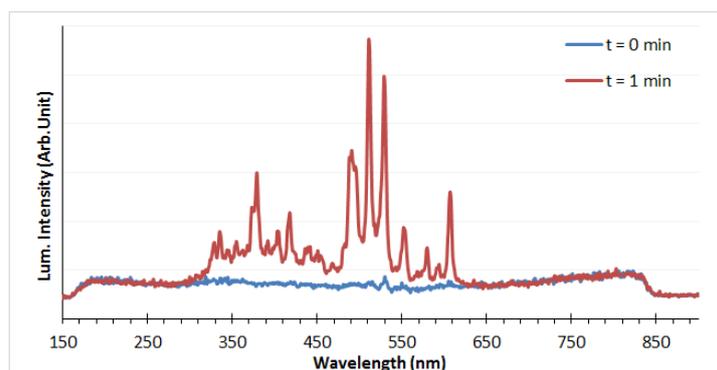


Fig. 1. Luminescence spectrum of ThO₂ during 21 MeV He²⁺ and 50nA ion flow

3.2 Visual Inspection

The UO₂ and ThO₂ samples were both visually compared. Although UO₂ exhibited no significant change in its physical appearance, the sections of the initially white ThO₂ exposed to the irradiation beam turned into a deep blue color (Fig. 2). The coloration of ThO₂ has previously been observed [9, 10, 11-12] and [13] reported that it take its origin from colored defects with swallow energy which can be annealed out at high temperature.

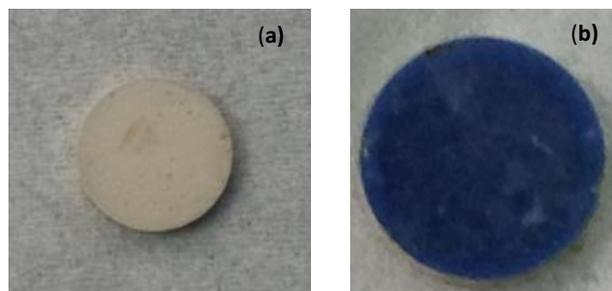


Fig. 2. ThO₂ pellet (a) before and (b) after irradiation respectively

3.3 Raman Spectroscopy

3.3.1 Before Irradiation

As described in section 2.4, the 633 nm was initially used for the acquisition of the Raman spectra of the virgin pellets (Fig. 3). According to its phonon dispersion curves [14-15], UO₂ has a single Raman active mode which corresponds to the symmetric T_{2g} vibration mode occurring at 445cm⁻¹ [16-17]. The T_{2g} peak is also signature of oxides exhibiting the fluorite structure sharing the Fm3m space group. Another peak with a small intensity appears at 575 cm⁻¹ which is attributed to the LO phonon [18]. It is Raman forbidden in virgin UO₂ and can be due to the presence of oxygen vacancies in the FCC structure [5]. ThO₂, sharing the same fluorite structure as UO₂, has its T_{2g} peak occurring at 465cm⁻¹ as shown in Fig. 3(b).

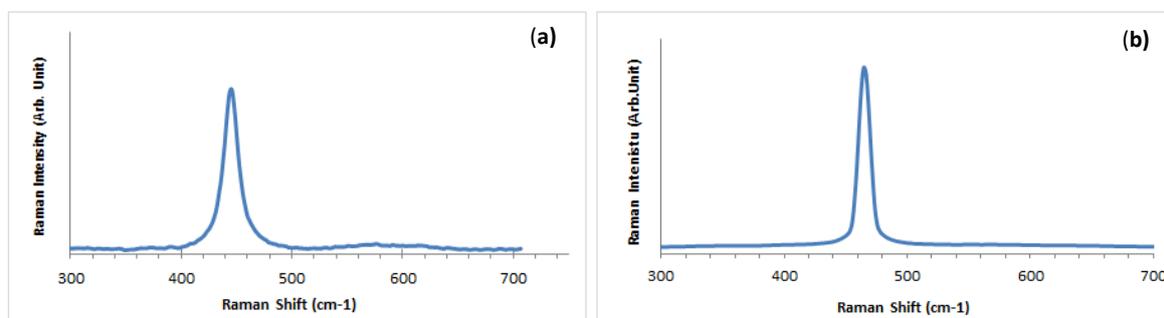


Fig. 3. Raman spectrum of (a) virgin UO₂ and (b) ThO₂ samples

Unfortunately, the initial Raman characterization of PuO₂ just after the heat treatment could not be carried out. The structure of PuO₂, identical to that of UO₂, is a fluorite one with an O_h space group and has previously been studied by Raman spectroscopy [19]. It has its T_{2g} phonon peak occurring at 477 cm⁻¹ [20].

3.3.2 Post-Irradiation

The Raman measurement was done on a cross-section of each irradiated half-disc at a distance of 106 μm from the surface. This corresponds to the region where maximum electronic energy was deposited during irradiation. The results obtained from this characterization are described below.

UO₂ & PuO₂ Spectra

Fig. 4(a) shows the Raman spectrum of irradiated UO₂ when analyzed with the 633 nm laser. It can be seen that along with the T_{2g}, there are the presence of three additional peaks in the 500-700 cm⁻¹ range denoted as U₁ (~ 527 cm⁻¹), U₂ (~ 572 cm⁻¹) and U₃ (~ 630 cm⁻¹). They have previously been observed and interpreted differently [3-5].

Raman theory states that if these three peaks were to represent a change in polarisability connected to the phonons of the compound, then they should still appear, either with a lower or higher intensity, even if a different laser for the analysis is used. Hence, the characterization was again carried out with the 532 nm green laser. Comparing the spectrum obtained from the 633 nm to that of the 532 nm laser, it can be observed that the peaks appear at the same position which confirms the origin as Raman signals for these UO₂ lines.

The obtained results for PuO₂ are similar with those previously observed after irradiating UO₂. Fig. 4(b) illustrates that the peaks remain at the same positions with different lasers and hence this also argues for a Raman origin of these lines in PuO₂.

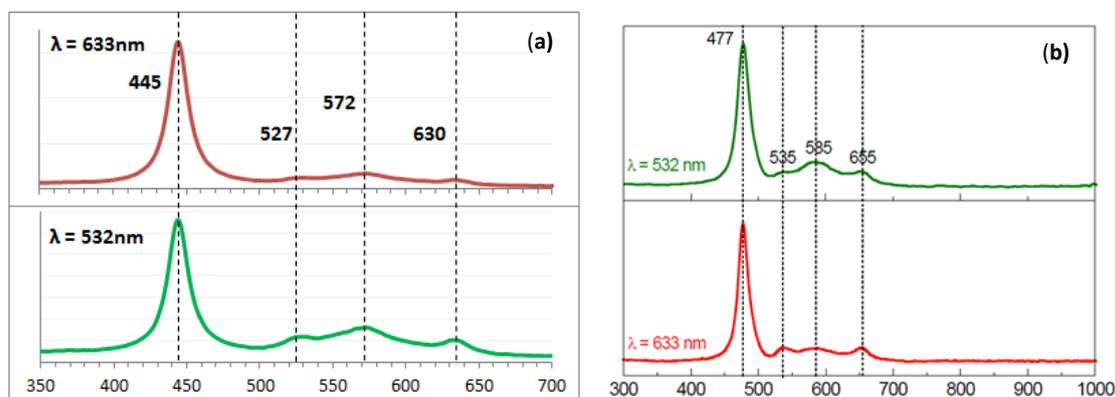


Fig. 4. Raman spectra of (a) UO_2 (b) PuO_2 when characterized using the 633 nm red & 532 nm green laser

ThO₂ Spectra

The Raman spectrum of ThO_2 (Fig. 5) shows that the 633 nm laser reveals the presence of four additional irradiation-induced peaks at 514 cm^{-1} , 539 cm^{-1} , 590 cm^{-1} and 622 cm^{-1} . However, the subsequent characterization with the 532 nm laser shows that they fall to different positions, and moreover with different intensities. These results help to state that the peaks in ThO_2 do not correlate to Raman scattering.

A luminescence mechanism can better explain the presence of these peaks. The energy of the photons from the laser is just sufficient to cause an absorption/emission process between different electronic energy levels in the material. The photons resulting from this process will be detected and appear as additional peaks on the Raman spectrum.

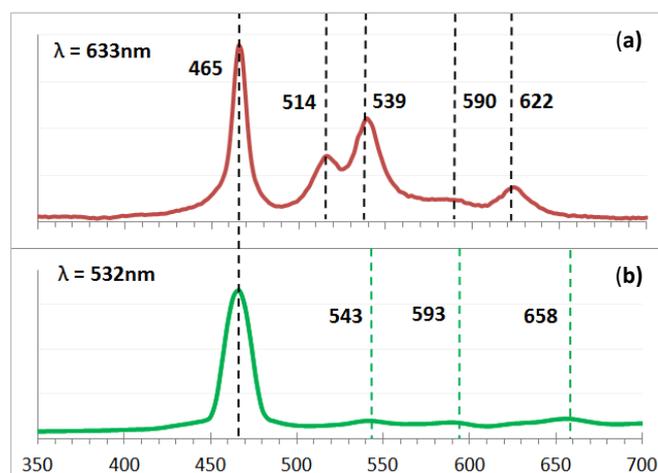


Fig. 5. Raman spectra of ThO_2 with 633nm and 532nm

4. Discussions

The results from the different experiments proved that ThO_2 behaves differently under irradiation and this can be translated as an occurrence of a different mechanism of irradiation defects compared to UO_2 and PuO_2 . Our study was mainly focused on the effect of electronic energy; therefore the stoichiometric nature of these compounds needs to be taken into account so as to propose a suitable mechanism of irradiation damage model in each case.

In UO_2 , U(III) and U(V) oxidation states can exist as well as the regular U(IV) [21]; while Pu(III) can also be formed in addition to Pu(IV) in PuO_2 [22]. Thorium tends to form stoichiometric oxides and the most stable oxidation state remains +4 [23]. Hence based on the stoichiometry difference of these actinides, the following possible irradiation defects models are proposed.

4.1 Mechanisms of Irradiation defects in ThO_2

According to the Bethe stopping power theory, around 11.55 keV is transferred to one electron during 21 MeV helium beam irradiation conditions. This energy is largely enough to enhance electronic defects, causing the excitation of an electron from the valence band up to the conduction band. Considering the stoichiometric nature of ThO_2 , the electron will tend to decay back and recombine with its associated hole. This decay evidences the existence of a radiative channel which is consistent with the Ion-Beam Induced luminescence signals. This process does not have any effect on the normal lattice vibrations, thereby explaining the absence of Raman signals. On the other hand, the electrons are also involved in the formation of colored defects associated with the blue color in the irradiated ThO_2 [24-25].

4.2 Mechanisms of Irradiation defects in UO₂/PuO₂

A unique mechanism could be held responsible for the irradiation defects in UO₂ and PuO₂. The initial stage can be described as similar to that of ThO₂ whereby electronic defects are involved. However, absence of the luminescence signals proves that no such radiative channel exists in these oxides. This can be correlated with the nature of these compounds since once in the conduction band and instead of trying to decay to retain stoichiometry, the electrons will tend to interact with the atoms in the lattice thereby causing an eventual change in the polarization of the phonons. This process is better defined as an electron-phonon coupling due to polarons [26, 27, 28-29] and will limit the light emission characteristics of UO₂. Moreover the similarity of the Raman triplet in UO₂ and PuO₂ sustains an intrinsic origin, due to the fluorine structure.

5. Conclusion

To better define the mechanism of irradiation defects in different actinides due to the effect of alpha-irradiation, UO₂ and ThO₂ were externally irradiated with helium beam using a Cyclotron facility. The pellets were then characterized using luminescence, visual and Raman methods. On the other hand the PuO₂, being too radioactive, had to be manipulated in a glove box and it was only possible to perform visual and Raman analyses.

This study shows that although ThO₂ shares similar characteristics as UO₂ and PuO₂, its mechanism of defect creation is totally different. Obtained results for the initially white ThO₂ showed that a blue color appeared in the irradiated areas and also the presence of luminescence signals during irradiation, while no new peaks were observed in its Raman spectrum. On the other hand, a Raman signature were observed during the post-mortem measurements of UO₂ and PuO₂, however no color change took place and the experimental set-up gave no indications of any luminescence signals. The results showed the existence of a radiative channel involving electronic defects in ThO₂ while in UO₂ and PuO₂ the presence of a coupling between electronic defects and phonons was confirmed due to the Raman signature.

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