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Tungsten oxide thin film exposed to low energy He plasma: evidence for a

thermal enhancement of the erosion yield

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**Abstract** 

Nanocrystalline tungsten oxide thin films (about 75 nm in thickness) produced by thermal

oxidation of tungsten substrates were exposed to low energy He plasma (≈ 20 eV/He) with a

flux of 2.5×10<sup>18</sup> m<sup>-2</sup> s<sup>-1</sup> at two temperatures: room temperature and 673 K. The structure and

morphology modifications which occur after this He bombardment and annealing treatments

was studied using Raman spectroscopy and transmission electron microscopy. Due to the low

fluence (4×10<sup>21</sup> m<sup>-2</sup>) and low ion energy, we have observed only few morphology modifications

after He plasma exposure at room temperature. On the contrary, at 673 K, a change in the layer

color is observed associated to an important erosion. Detailed analyses before/after exposure

and before/after annealing allow us to describe the He interaction with the oxide layer, its

erosion and structural modification at the atomic and micrometer scale.

Keywords: Tungsten oxide thin film, He plasma bombardment, erosion, plasma wall

interaction, Raman spectroscopy, transmission electron microscopy.

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#### 1. Introduction

Because of its favorable physical properties, such as a high melting point and an excellent erosion resistance, tungsten (W) is being used as plasma facing component (PFC) material for magnetic fusion devices such as the tokamaks ASDEX-Upgrade [1], ITER [2], JET [3] and WEST [4]. During the deuterium-tritium (D–T) fusion operation of ITER, significant helium (He) ash will be formed and will interact with the W divertor. Under plasma divertor conditions and even for impact energies below the physical sputtering threshold, which is estimated around 300 eV [5], He ions have been found to cause erosion and redeposition or codeposition as well as significant morphology changes in W [6-10]. It should be noted that He ion bombardment has been found to have high affinity to create bubbles, holes and nanostructures. These Damages will change W physical and mechanical properties and this will change their interaction with plasma (such as: erosion rate, implantation and diffusion depth).

Various types of mixed materials can be formed during redeposition and codeposition processes in presence of impurities [11]. Oxygen can exist as contamination in tokamak vacuum vessels and W PFCs can be oxidized, this process being facilitated by the high temperatures reached during operation of the machine. Tungsten oxide can therefore be formed at the surface of tokamak PFCs with various stoichiometries [11,12], the most stable being WO<sub>3</sub>, whose behavior in the tokamak environment is still not well known. Basically the tungsten oxide crystal WO<sub>3</sub> is built up from corner-sharing WO<sub>6</sub> octahedra, it easily forms substoichiometric shear phases (WO<sub>3-x</sub>) containing edge-sharing octaedra with various crystallographic structures.

Although many studies have been concerned with the consequences of He and D bombardments on W [5-10], only a few works have explored the behavior of WO<sub>3</sub>. The

presence of an oxide layer can reduce the sputtering energy threshold of W by a factor of 10, increase the erosion yield [13] and generate different hydrogen isotope retention mechanisms [14-16]. A. Pezzoli et al. [17] produced WO<sub>3</sub> layers by the pulsed laser deposition technique in order to simulate redeposited layer compounds that can be present on PFCs. They have shown that the layer morphology and stoichiometry can be controlled by changing background O<sub>2</sub> pressure during deposition. Surface modifications, such as nano-cracks, pinholes and bubbles, were observed after high flux (10<sup>24</sup> m<sup>-2</sup> s<sup>-1</sup>) D plasma exposure in Magnum–PSI at a layer temperature of 580 K [17]. Another study of tungsten oxide layers grown on polycrystalline ITER-grade surfaces by annealing in air for 10 min was carried out at various sample temperatures [18]. It has been shown that at low-energy (38 eV/D) and high flux (10<sup>22</sup> m<sup>-2</sup> s<sup>-1</sup>) D plasma exposure (fluence of 10<sup>26</sup> m<sup>-2</sup>), the formation of cracks appears for temperatures above 605 K.

In a previous study, we have observed a color change and a phase transition of WO<sub>3</sub> layers due to the formation of tungsten bronze ( $D_xWO_3$ ) after a low energy ( $\approx 20~eV$ ) and low fluence ( $4\times10^{21}~m^{-2}$ ) deuterium plasma exposure at room temperature [19]. This structural modification is due to D intercalation and deep diffusion into the WO<sub>3</sub> channels and is reversible after a few days at ambient conditions.

In this work, the structural changes of WO<sub>3</sub> layers under low energy ( $\approx 20$  eV) and low fluence (4×10<sup>21</sup> m<sup>-2</sup>) He plasma exposure at room temperature and 673 K was studied with the help of Raman spectroscopy and transmission electron microscopy. Our results emphasize an important effect of the sample temperature on these structural changes.

### 2. Experimental

## 2.1 WO<sub>3</sub> thin film production

Polycrystalline recrystallized tungsten (pc-W) samples produced by A.L.M.T. Corp., Japan (purity of 99.99 %), measuring  $7\times7\times0.3$  mm, were used as substrates for growing oxide layers. The W grain size was found to be a few tens of micrometers using scanning electron microscopy (SEM) (Microscope Philips XL30 SFEG, Schottky Field Emission Gun). The pc-W substrates were outgassed at  $10^{-7}$  torr and 673 K during 4 hours. Afterwards, thermal oxidation at 673 K on the pc-W substrates was performed at oxygen pressure of 590 torr during 90 min. Thanks to this procedure, we produced a set of identical  $\gamma$ -WO<sub>3</sub> (referred to WO<sub>3</sub> hereafter) samples whose structure was monoclinic and nanocrystalline, as confirmed by X-rays diffraction analysis (not shown in this paper). The oxide layer thickness, measured using a FIB cross-section technique (FEI Helios 600 NanoLab Dual Beam), was about  $75\pm4$  nm. The corresponding oxide layer color was blue.

Raman spectra were recorded using a Horiba-Jobin-Yvon HR LabRAM microscope ( $\lambda$  = 514 nm and 532 nm, laser spot size  $\approx 1~\mu m^2$ ). The laser power was kept at about 1 mW  $\mu m^{-2}$ . The Raman mapping mode was used to average 30 spectra collected over a 750  $\mu m^2$  area in order to increase the signal to noise ratio and to detect eventual inhomogeneities. Spectra were measured in the Raman shift range of 100 – 1250 cm<sup>-1</sup>. Transmission Electron Microscopy (TEM) (FEI Tecnai G2 – thermo-ionic LaB<sub>6</sub> electron source – accelerated voltage 200 kV) was used for the structural analysis of the oxide layer and oxide/tungsten interface on  $\sim$  100 nm slices cut perpendicular to the sample surface by Focused Ion Beam (FIB) technique.

#### 2.2 Plasma exposure stage

The University of Tennessee Nuclear Engineering (UTNE) ion exposure stage utilizes a PCS-ECR Plasma Cracker Source manufactured by SPECS Surface Nano Analysis GmbH, which is a compact electron cyclotron resonance (ECR) ion source (fig 1) [20]. This experiment is similar in design and purpose to the ion exposure stage utilized at Sandia National Laboratory, which demonstrated the capability to produce He bubble growth in heated W samples using a compact ECR ion source [21]. The UTNE source operates with helium or other inert gases and is equipped with a molybdenum extraction grid able to be biased up to -1000 V to accelerate the ions to the desired energy. The He ion flux used during exposures was kept at approximately  $2.5 \times 10^{18}$  m<sup>-2</sup> s<sup>-1</sup> at the sample face (with a corresponding current  $\approx 20$   $\mu$ A). The W samples were mounted on a pyrolytic boron nitride (PBN) plate heater manufactured by Momentive Performance Materials, which was capable of heating samples to temperatures up to approximately 1300 K.

The sample was held in place on the heater stage with the PBN plate that also controlled the exposed area (fig 2). The sample holder was mounted on a linear motion vacuum feedthrough in the sample loading chamber. The sample loading chamber could then be sealed and pumped down to several millitorr, allowing the gate valve separating the plasma exposure chamber and the sample loading chamber to be opened. The linear motion feedthrough then allowed the sample to be inserted into the ion exposure chamber and brought to within 4 cm of the exit aperture of the ion source.

The base pressure in the ion exposure chamber was about  $2.2 \times 10^{-5}$  torr before the exposure. In the case of the exposure at 673 K, the sample was heated up slowly during 90 min to reach 673 K with a pressure in the chamber around  $1.3 \times 10^{-5}$  torr, remaining at around  $9.4 \times 10^{-4}$  torr during the exposure at room temperature and 673 K. Table 1 summarizes the experimental

conditions the four identical WO<sub>3</sub> layers: (i) WO<sub>3</sub> as oxidized was used as a reference, (ii) WO<sub>3</sub>-673K was annealed in a vacuum chamber at 673 K during 2 hours at  $7.5\times10^{-6}$  torr, (iii) WO<sub>3</sub>-HeRT and (iv) WO<sub>3</sub>-He673K were exposed to 20 eV He plasma with a fluence of  $4\times10^{21}$  m<sup>-2</sup> (exposure time of 27 min) at room temperature and 673 K, respectively.

#### 3. Results and Discussions

#### 3.1 Annealing effect

Figure 3 shows the Raman spectra of as oxidized WO<sub>3</sub> (WO<sub>3</sub>, in blue) and WO<sub>3</sub> annealed at 673 K (WO<sub>3</sub>-673K, in red) in the 100–1100 cm<sup>-1</sup> region and normalized to the intensity of the main peak at 807 cm<sup>-1</sup>. In the literature, the monoclinic structure is identified by two bands peaking in the range 700-720 cm<sup>-1</sup> and 805-808 cm<sup>-1</sup> in the stretching mode region (500-900 cm<sup>-1</sup>) and in the range 128-135 cm<sup>-1</sup> and 268-275 cm<sup>-1</sup> in the bending mode region (100-500 cm<sup>-1</sup>) [22-24]. The features found here at 703 cm<sup>-1</sup>, 807 cm<sup>-1</sup> and 130 cm<sup>-1</sup>, 270 cm<sup>-1</sup> confirm the monoclinic structure of the sample as previously identified by XRD. The shoulder observed at 645 cm<sup>-1</sup> is attributed to oxide hydratation while the shoulder observed at 960 cm<sup>-1</sup> is attributed to terminal W=O stretching modes and reveals the presence of defects in the crystalline monoclinic structure. As the sample structure is nano-crystalline (see figure 6), these defects may be due to grain boundaries, in agreement with previous observations relating terminal W=O stretching modes to the presence of nano-grains [25]. Raman spectra of WO<sub>3</sub> as oxidized and WO<sub>3</sub> annealed are very similar, with no evidence for a phase reorganization: the monoclinic structure of the WO<sub>3</sub> layer is thus clearly not modified after annealing at 673 K. The slight decrease of the shoulder at 645 cm<sup>-1</sup> could be attributed to the dehydration of the WO<sub>3</sub> layer [26,27].

Figure 4 shows a TEM image of WO<sub>3</sub>-673K taken from a slice prepared by FIB. The oxide layer is compact and the tungsten/oxide interface is smooth and well defined. The layer thickness has not changed, still measured at  $75 \pm 4$  nm after annealing. Both the interface and the top surface (protected by a Pt coating during the FIB preparation) do not show any significant changes after annealing.

To conclude, Raman spectroscopy and TEM results show that neither the thickness nor the nano-crystalline structure of the initial WO<sub>3</sub> layer change after a thermal treatment at 673 K.

#### 3.2 He plasma exposure

Figure 5 shows photographs of WO<sub>3</sub>, WO<sub>3</sub>-HeRT and WO<sub>3</sub>-He673K samples. After He plasma exposure at room temperature (WO<sub>3</sub>-HeRT sample), the sample color has not changed while, on the contrary, after He exposure at 673 K (WO<sub>3</sub>-He673K sample), the sample color turned to orange, this color change being not reversible with time.

No color change was noticed on WO<sub>3</sub> heated at 673 K without bombardment, and therefore excluding an effect due to the temperature alone. As, due to interference phenomena, oxide layers with different thicknesses may have different colors [19], this change could be explained by a change in the thickness of WO<sub>3</sub>-He673K.

It is worthwhile to mention that SEM images (not shown here) show that the He bombardment has no effect on the surface of the oxide layer, without any evidence of morphological modifications such as nano-cracks or point-like features. This was expected as exposures were done at very low fluences ( $\approx 4 \times 10^{21} \text{ m}^{-2}$ ) while such modifications were observed in [17,18,28] with much higher fluences ( $\approx 10^{26} \text{ m}^{-2}$ ). In what follows, we detail TEM analyses aimed at studying the He bombardment effects in the sub-surface and in the bulk of the oxide layer.

Figure 6 shows TEM images of the WO<sub>3</sub> layer exposed at room temperature (WO<sub>3</sub>-RT sample). The surface is smooth and the oxide/tungsten interface is smooth and well defined as previously shown for WO<sub>3</sub>-673K (fig 4). The thickness of the oxide layer is measured at  $71 \pm 4$  nm (fig 6a). Comparing this value to the original thickness ( $75 \pm 4$  nm), we notice no significant change in thickness and hence no significant erosion.

On the contrary, TEM images of exposed WO<sub>3</sub> at 673K (WO3-He673K sample) shows significant damage and the formation of new features at the interface (fig 7). First, the oxide layer thickness is now measured at  $54 \pm 5$  nm (fig 7a): a decrease in thickness of about 21 nm which corresponds to an erosion rate of about 0.013 nm s<sup>-1</sup>. Assuming a layer density of 7.16 g cm<sup>-3</sup>, we thus estimate tungsten oxide sputtering yield  $Y_s$  at  $0.10 \pm 0.05$  at 673 K. Second, the presence of interference fringes in the TEM image of fig 7b shows that the bulk of the layer (beyond 10 nm) is still crystallized. Nevertheless, the oxide sub-surface (fig 7b) shows the presence of a 10 nm thick modified layer which appears to be darker on the image. This darker contrast reveals a higher density due to a loss of oxygen which can be preferentially sputtered. This has also been reported in [17] and has been already observed for oxide materials in other applications [29].

The oxide layer is not homogeneous anymore and in order to better identify the nature of this 10 nm surface layer we have performed Raman analysis using a UV laser wavelength ( $\lambda$ =325 nm) (figure 8). This allows only the damaged surface layer to be probed, without penetrating into the bulk. Figure 8 shows the normalized Raman spectra of WO<sub>3</sub>-HeRT (blue) and WO<sub>3</sub>-He673K (red) in the stretching mode region (500-1100 cm<sup>-1</sup>). The Raman spectrum of WO<sub>3</sub>-He673K is broadened and the two main peaks of the monoclinic  $\gamma$ -WO<sub>3</sub> are not separated anymore. This typical broad band feature is a signature of amorphous tungsten oxide (a-WO<sub>3</sub>) [17, 30].

Finally, fig 7a shows that the oxide/tungsten interface is highly damaged and reveals the formation of voids. These damages cannot be related only to a temperature effect as we have previously shown (fig 4) and are due to the He bombardment. This shows that He has diffused throughout the oxide layer, and that the clustering of trapped He has formed bubbles.

Raman analyses performed at  $\lambda = 532$  nm for which the layer is transparent allow us to deepen the understanding of the structural modification of the entire layer. In fig 9, the normalized Raman spectra of WO<sub>3</sub> (black), WO<sub>3</sub>-HeRT (blue) and WO<sub>3</sub>-He673K (red) are reported in the stretching mode region (500-1100 cm<sup>-1</sup>). Compared with WO<sub>3</sub>, the Raman intensity (integrated area) has decreased by a factor of 1.3 for WO<sub>3</sub>-HeRT and by a factor of 3.9 for WO<sub>3</sub>-He673K. The main peaks of the monoclinic structure are observed for the three samples although they are broadened after exposition, this broadening being larger at 673K than at room temperature, indicating the creation of defects and thus an increase of the local disorder. This disorder can be represented by the distortion of the WO<sub>6</sub> octahedrons which compose the monoclinic WO<sub>3</sub> structure. The creation of disorder is confirmed by the increase of the relative intensity of the 703 cm<sup>-1</sup> and 960 cm<sup>-1</sup> peaks with respect to that of the main peak at 807 cm<sup>-1</sup> and the increase of the relative intensity between bending and stretching mode bands (not shown here). The peak at 807 cm<sup>-1</sup> is blue-shifted to 810 cm<sup>-1</sup> and 826 cm<sup>-1</sup> for WO<sub>3</sub>-HeRT and WO<sub>3</sub>-He673K, respectively. This shift is consistent with a shortening of W-O bonds due to (i) the formation of oxygen vacancies, (ii) a compressive stress introduced by trapped He and (iii) the breaking of WO bonds.

Note that the Raman spectra show that all the entire layer is affected by these changes and therefore that there He diffuses throughout the oxide layer.

#### 3.3 Discussion

Annealing at 673 K does not produce significant structural modification, and in particular does not increase the lattice organization: this is not very surprising as this temperature was used for the oxide layer production and this shows that the sample structure is not altered even after several weeks under air.

Exposing the tungsten oxide layer at room temperature with a fluence of  $4\times10^{21}$  m<sup>-2</sup> of 20 eV He ions did not introduce any damage at the oxide/tungsten interface or erosion: this latter observation is in agreement with Roth et al. [13] which, when tungsten oxide is produced during exposition with He ions under background oxygen pressure of  $8\times10^{-5}$  torr, has extrapolated the sputtering energy threshold by He below 10 eV. In our case, the ion energy is 20 eV and the expected sputtering yield is about  $7\times10^{-4} \text{ WO}_3/\text{He}$  [13], which would correspond to a decrease in thickness of 0.15 nm, much less than our thickness measurement precision. No noticeable structural modification at the microscale is observed and the monoclinic nano-crystalline structure is still well characterized by Raman spectroscopy. Nevertheless, the introduction of local disorder is identified which could be represented by an increase of the disorder of the octahedral O structure surrounding W atoms generated by the formation of oxygen vacancies subsequently to collisions, by the compressive stress induced by the trapped He or by breaking of WO bonds.

Unlike this He bombardment, a previous study has shown the D bombardment of a similar oxide layer under similar plasma conditions [19] leads to the change of the layer color (blue to orange), this coloration effect being reversible. This D plasma chromic effect is similar to the extensively studied electrochromic effect of  $WO_3$  where the ion diffusion and intercalation is induced by the application of an electrical voltage. In both cases, a coloration is observed which is due to the formation of  $D_xWO_3$  bronze by  $D^+$  intercalation and diffusion throughout the  $WO_3$ 

layer. Even though if, as D, He ions diffuse, intercalate and induce local disorder, unlike D, He does not lead to chemical reactivity and to bronze formation.

On the contrary, major changes such as bubble or void formations at the oxide / tungsten interface are observed when exposing the oxide layer at 673 K at similar plasma conditions. In the case of pure tungsten, these morphology changes have been already observed for higher He energy on tungsten at surface temperature ranging 673-1273 K [6-10]. While still not completely understood, they are thought to result from near-surface He trapping at intrinsic or extrinsic defect sites. Due to low He solubility in tungsten, the trapped He atoms form clusters, which result in progressively higher lattice distortions, which relax by dislocation loop punching, producing He-filled cavities (bubbles) of increasing size. Here, both the diffusion of He throughout the film and the thermal activation of the cluster formation at the interface can explain the bubbles we observed at 673 K.

Contrary to what is observed at room temperature, a significant film thickness decrease by  $21 \pm 5$  nm is observed at 673 K that we attribute to erosion. This is the first time that such a sputtering effect is observed for helium at low energy. The only experimental study previously published on the sputtering of tungsten oxide by low energy He ions was that of Roth et al. [13] at room temperature. As an example of temperature effect, we can mention the sputtering of tungsten oxide by 150 eV - 10 keV oxygen ion beams studied in [31] where it was found that the yield increases by increasing the target temperature. In another study, the sputtering yield and the roughness of W layers irradiated by He ions at various energies and various temperatures have been studied by Yunfeng Wu et al. [32]. It was found that, for He ion energy of 300 eV, the W sputtering yield increases from  $4.06 \times 10^{-3} \text{ WO}_3/\text{He}$  at T = 293 K to  $1.44 \times 10^{-2} \text{ WO}_3/\text{He}$  at 973 K [32]. Although the sputtering mechanisms are not yet fully understood, our study clearly shows that the combination of He bombardment and surface heating induce a sputtering effect. A possible scenario is that He ion bombardment breaks bonds allowing

weakly bounded  $W_xO_y$  molecules to be produced in the sub-surface whose release into the gas phase is thermally driven. Assuming that this loss corresponds to a WO<sub>3</sub> sputtered layer whose density is 7.14 g/cm<sup>3</sup>, we estimate the sputtering yield of WO<sub>3</sub> at about 0.10  $\pm$ 0.05 WO<sub>3</sub>/He.

We have shown the formation of 10 nm damaged sub-surface layer which is deficient in oxygen and most likely highly disordered, and which may correspond to the region of collision cascades. We have demonstrated that this damaged layer is an amorphous tungsten oxide (a-WO<sub>3</sub>). An example of similar sub-surface damages exceeding the implantation zone due to low energy ion implantation was the amorphisation of a 30  $\pm$  10 nm carbon layer observed using similar methods (FIB/TEM) for the eroded graphite surface of the Tore Supra tokamak [33]. Here there is no clear sign that the modified layer has become completely amorphous but it is probably highly deficient in oxygen and probably a key region to explore to understand the sputtering mechanisms

#### 4. Conclusion

Nanocrystalline tungsten oxide WO<sub>3</sub> thin films produced by thermal oxidation on tungsten substrate were exposed to low energy He plasma ( $\approx 20 \, \text{eV/He}$ ) with a flux of  $2.5 \times 10^{18} \, \text{m}^{-2} \, \text{s}^{-1}$  at a fluence of  $4 \times 10^{21} \, \text{m}^{-2}$  at two different temperatures: room temperature and 673 K. In both cases, defect formation (local disorder introduced by oxygen vacancies, compressive stress or by WO breaking bonds) changes the Raman signature of the oxide film, whereas the nano-crystalline structure of the film remain unaltered. We have shown that, when activated by temperature, exposure of the tungsten oxide with low energy He ions leads to erosion: although no sputtering is observed at room temperature, we have measured a sputtering yield of  $0.10 \pm 0.05 \, \text{WO}_3/\text{He}$  at 673 K. We have put in evidence that the sub-surface was disordered and oxygen deficit can be likely explained by preferential sputtering of oxygen. We have observed

the formation of voids at the WO<sub>3</sub>/W interface which could be the consequence of He implanted in the sub-surface layer followed by thermal diffusion of He atoms throughout the film and their clustering. Further studies are necessary to analyse the role of temperature in the activation of the mechanisms at play in the erosion processes.

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### **Figure Captions**

Figure 1: UTNE ion exposure stage with the compact ECR ion source. Samples are loaded in the right chamber (Sample Loading Chamber) and then transferred through a gate valve to the left chamber (Ion Exposure Chamber) using a linear motion feedthrough for exposure to He plasma generated in the ECR ion source.

Figure 2: Square tungsten oxide sample mounted on the heater stage held in place with the polycrystalline boron nitride face plate.

Figure 3: Raman spectra of as oxidized WO<sub>3</sub> (blue) and annealed WO<sub>3</sub> at 673 K during 2 hours WO<sub>3</sub>-673K (red). Spectra were measured using  $\lambda$ =532 nm.

Figure 4: TEM image of a FIB cross-section of annealed WO<sub>3</sub> at 673 K during 2 hours (WO<sub>3</sub>-673K).

Figure 5: Photographs of tungsten oxide films: (a) WO<sub>3</sub>, (b) WO<sub>3</sub>-HeRT and (c) WO<sub>3</sub>-He673K.

Figure 6: (a) TEM image of WO<sub>3</sub> film exposed to He at RT (WO<sub>3</sub>-HeRT) and (b) high resolution TEM image of the zone indicated in the square in (a) showing the nanocrystalline structure of WO<sub>3</sub>-HeRT.

Figure 7: (a) TEM image of WO<sub>3</sub> film exposed to He at 673 K (WO<sub>3</sub>-He673K) and (b) high resolution TEM image showing the formation of a darker layer on the oxide surface.

Figure 8: Raman spectra of WO<sub>3</sub>-HeRT (blue) and WO<sub>3</sub>-He673K (red) samples. Spectra were measured using  $\lambda$ =325 nm (UV laser).

Figure 9: Raman spectra of WO<sub>3</sub> (black), WO<sub>3</sub>-HeRT (blue) and WO<sub>3</sub>-He673K (red) samples. Spectra were measured using  $\lambda$ =532 nm.

**Table 1:** Experimental conditions of the four tungsten oxide samples: reference oxide WO<sub>3</sub>, annealed oxide WO<sub>3</sub>-673K, oxide exposed to He plasma at 300 K WO<sub>3</sub>-HeRT, and oxide exposed to He plasma at 673 K WO<sub>3</sub>-He673K.

	Sample	Experimental conditions
Unexposed	$WO_3$	As oxidized
	WO <sub>3</sub> -673K	Annealed at 673 K for 2 hours
Exposed	WO <sub>3</sub> -HeRT WO <sub>3</sub> -He673K	Room Temperature $\approx 20 \text{ eV}$ $\approx 2.5 \times 10^{18} \text{ m}^{-2} \text{ s}^{-1}$ $4 \times 10^{21} \text{ m}^{-2}$

#### **FIGURES**

Figure 1

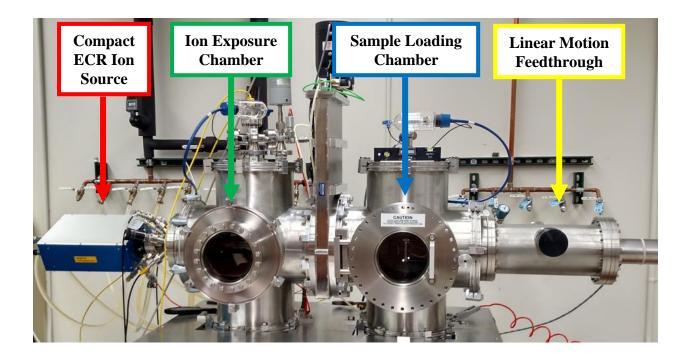


Figure 2

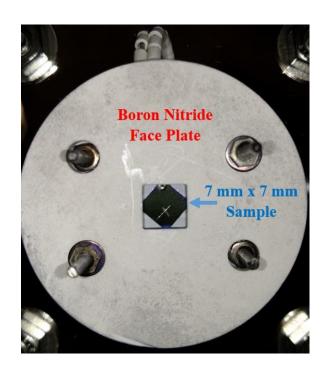


Figure 3

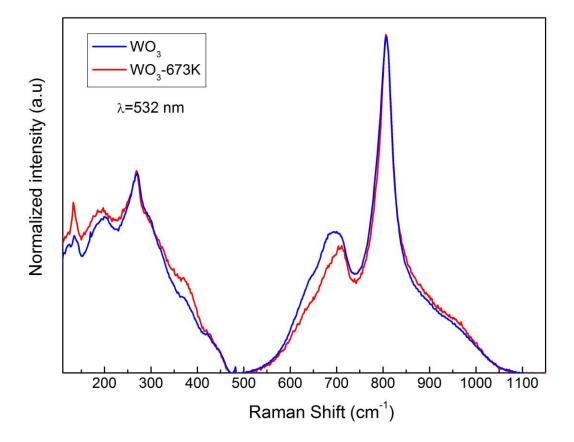


Figure 4

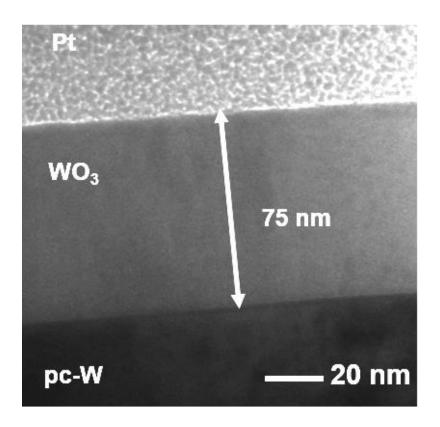


Figure 5

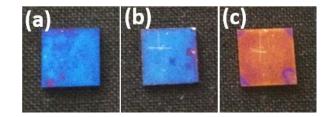


Figure 6

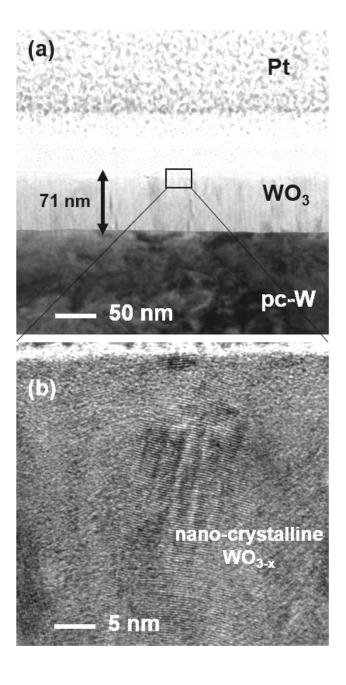


Figure 7

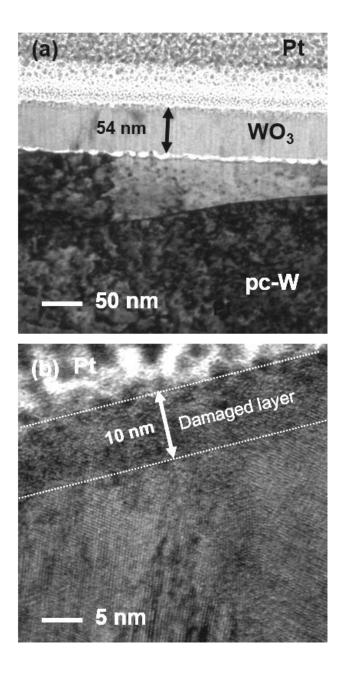


Figure 8

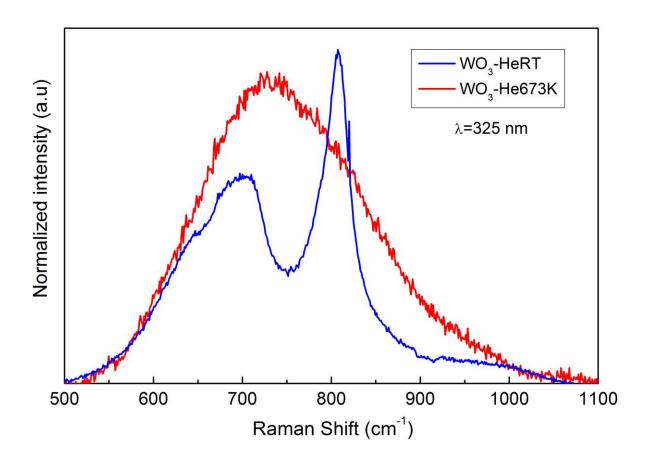


Figure 9

