

Combined study of titanium dioxide nanoparticle transport and toxicity on microbial nitrifying communities under single and repeated exposures in soil columns

Marie Simonin, Jean M. F. Martins, Gaëlle Uzu, Erwann Vince, Agnès

Richaume

► To cite this version:

Marie Simonin, Jean M. F. Martins, Gaëlle Uzu, Erwann Vince, Agnès Richaume. Combined study of titanium dioxide nanoparticle transport and toxicity on microbial nitrifying communities under single and repeated exposures in soil columns. Environmental Science and Technology, 2016, 50 (19), pp.10693-10699. 10.1021/acs.est.6b02415. hal-01604452

HAL Id: hal-01604452 https://hal.science/hal-01604452

Submitted on 15 Oct 2021

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers. L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.

1	A combined study of TiO ₂ nanoparticle transport and toxicity on microbial
2	nitrifying communities under single and repeated exposures in soil columns
3 4	Marie SIMONIN ^{1,2,3,4†} , Jean M.F. MARTINS ^{4,5} , Gaëlle UZU ^{4,6} , Erwann VINCE ^{4,5} and Agnès RICHAUME ^{1,2,3*}
5	¹ Université de Lyon, Lyon, France
6	² Université Claude Bernard Lyon 1, Villeurbanne, France
7	³ CNRS, UMR 5557, Microbial Ecology Lab, Université Lyon 1, Villeurbanne, France
8	⁴ Université Grenoble Alpes, LTHE UMR 5564, Grenoble F-38000, France
9	⁵ CNRS, LTHE, F-38000 Grenoble, France
10	⁶ IRD, LTHE, F-38000 Grenoble, France
11	
12	*Correspondence to: agnes.richaume@univ-lyon1.fr
13 14	[†] Present address: Department of Biology, Duke University, Durham, NC, USA; Center for the Environmental Implications of Nanotechnology, Duke University, Durham, NC, United States.
15	
16	
17	Révision Soumise le 25/07/2016
18	
19	
20	
21	
22	
23	

24 ABSTRACT

25 Soils are exposed to nanoparticles (NPs) due to their increasing use in many commercial 26 products. Adverse effects of NPs on soil microorganisms have been reported in several 27 ecotoxicological studies using microcosms. Although repeated exposures are more likely to 28 occur in soils, most of these previous studies were performed as a single exposure to NPs. 29 Contrarily to single contamination, the study of multiple NP contaminations in soils requires 30 the use of specialized setups. Using a soil column experiment, we compared the influence of 31 single and repeated exposures (one, two or three exposures that resulted in the same final concentration) on the transport of titanium dioxide NPs (TiO₂-NPs) through soil and the effect 32 33 of these different exposure scenarios on the abundance and activity of soil nitrifying microbial communities after a 2-month incubation period. The transport of TiO₂-NPs was very limited 34 under both single and repeated exposures and was highest for the lowest concentration 35 36 injected during the first application. Significant decreases in nitrification activity and 37 ammonia-oxidizing archaea and bacteria populations were observed only for the repeated 38 exposure scenario (three TiO₂-NP contaminations). These results show that under repeated 39 exposures, the transport of TiO₂-NPs to deep soil layers and groundwater is limited and that a 40 chronic TiO₂-NPs contamination is more harmful for the soil microbiological functioning than 41 a single exposure.

42

43 Key-words: TiO₂ nanomaterials, chronic and acute exposure, transport, microbial
44 ecotoxicology, nitrification.

46 **INTRODUCTION**

47 In recent years, several studies have highlighted deleterious effects of metal-oxide 48 nanoparticles (NPs) on soil (micro)organisms and, consequently, on soil functioning and 49 fertility^{1–3}. In most of these studies, soils were exposed to single homogeneous applications 50 of NPs into microcosms³. However, NPs are more likely to be released onto the soil surface 51 through repeated applications of biosolids, irrigation or through the use of nanofertilizers or nanopesticides^{4–6} and may be leached to further depths through soil porosity⁷. Therefore in 52 53 contaminated soils NPs are heterogeneously distributed, especially along vertical soil profiles⁸. 54 This heterogeneous distribution of NPs is controlled by the soil physicochemical properties 55 that are known to influence the mobility of colloids according to the colloidal filtration theory^{9–11} and also by NP properties, such as size and/or surface charge⁷. 56

To date, only a few studies were performed on repeated or chronic NP exposure in soil¹². The lack of studies dealing with chronic NP toxicity in soils is mainly due to the experimental difficulties associated with executing repeated exposures in microcosm experiments. Implementing more appropriate experimental designs that reflect realistic environmental conditions and consider the distribution of NPs according to soil depth are therefore required.

The aim of this work was to compare the effect of single and repeated NP applications on: (i) the transport of TiO₂-NPs through a soil profile, and (ii) their toxicity to the soil microbial communities by focusing on disruption of the nitrification process which is a critical step in the nitrogen cycle. Nitrification is a crucial microbiological process for soil fertility controlled by the activity of both of ammonia- and nitrite-oxidizers that convert ammonium to nitrate. This key step of the nitrogen cycle is known to be very sensitive to environmental stressors, especially chronic disturbances^{12–14}. Therefore, the abundance of ammonia-oxidizing archaea and bacteria (AOA and AOB, respectively) was selected for monitoring soil toxicity in this study
because this endpoint has been proposed as relevant ecological indicators for soil health^{15,16}.
TiO₂-NPs are the most produced NPs and it has been predicted that they may represent 50 %
of all engineered NPs retrieved in soil, especially because of the repeated application of
biosolids on agricultural soils^{17,18}. TiO₂-NPs are known to be toxic for microbes, even in the
absence of light like in conditions encountered in soils, due to a depolarization and loss of
membrane integrity translating in a cell osmotic stress^{19,20}

76 Our experimental design, based on saturated soil columns, enabled the study of the fate and 77 impact of both single and repeated TiO₂-NP exposures in soil under more realistic exposure 78 conditions than what can be achieved with classical microcosm experiments. In the context of 79 short-term lab-scale experiment, soil columns were contaminated using three exposure 80 scenarios (one, two, or three applications) resulting in the same total TiO₂-NP concentrations applied to soil columns (372 mg NPs kg⁻¹ dry soil). After the last NP application, four layers of 81 82 the contaminated soil columns corresponding to different depths were collected and 83 incubated separately for 2 months to evaluate the impact of TiO₂-NP exposure on the nitrification process along the soil profile. The nitrification enzyme activity (NEA) and the 84 abundance of AOA and AOB were measured. 85

86

87 MATERIALS AND METHODS

88 **Soil**

The upper 20 cm layer of a silty-clay soil (Cambisol, WRB, 2006) under a permanent pasture
was collected at Commarin (Côte d'Or, France). After soil sampling, roots and plant litter were

manually removed. The soil was sieved at 2 mm and homogenized before storage at 4 °C. The
main soil characteristics were analyzed by the Laboratoire d'analyse des sols (INRA Arras,
France) following NF ISO protocols: sand, 10 %; loam, 51 %; clay, 39 %; pH 7.7; organic matter
(OM), 7.9%; cation exchange capacity (CEC), 20.1 cmol kg⁻¹; water holding capacity (WHC), 51
%; and ionic strength, 1.37 mM.

96 TiO₂ nanoparticles

97 Uncoated titanium dioxide nanoparticles (mixture of anatase (80%) and rutile (20%) crystal 98 structure) were purchased from Sigma Aldrich (St Louis, USA) with at least 99.5 % purity. The 99 mean primary particle size of the TiO₂-NPs was 28.7 (± 7.1) nm in powder as measured with a 100 ZEISS Ultra 55 scanning electron microscopy–field emission gun (SEM-FEG) and energy 101 dispersive spectroscopy (EDS) with a SDD detector (BRUKERAXS-30 mm²).

102 The size and zeta potential of the TiO₂-NPs were characterized using Dynamic Light Scattering 103 (DLS) in the soil solution used as the background solution in column experiments. The soil 104 solution was prepared following the protocol described by Simonin et al.²¹. Briefly, 10 g of soil 105 dispersed in 50 mL of ultrapure water were shaken for 30 min at 180 rpm and 20 °C in a 106 refrigerated incubator shaker (New Brunswick – Eppendorf, Hamburg, Germany). The 107 solutions were then centrifuged for 20 min at 8000g and 20 °C (Centrifuge 5804R, Eppendorf, 108 Hamburg, Germany) to eliminate particles larger than 20 nm according to Stokes' law. The 109 supernatants were collected and stored at 4 °C. The resulting soil solution had a pH of 7.1, an 110 ionic strength of 1.6 mM and a dissolved organic carbon (DOC) concentration of 8.9 mg L⁻¹. 111 Measurements were performed in triplicate with three suspensions of TiO₂-NPs (50, 25 and 112 16.7 mg L⁻¹) after dispersion by ultrasonication (Sonication bath, Bioblock Scientific) for 5 min

to ensure the suspension homogeneity. For each sample, the mean of three measurementswas recorded.

115 Exposure of soil in columns to single and repeated TiO₂-NP applications

116 Three TiO₂-NP suspensions were prepared in soil solution for each exposure scenario: 50 mg 117 L⁻¹ for the single exposure experiment, 25 mg L⁻¹ for the two-exposures experiment, and 16.7 118 mg L⁻¹ for the three-exposures experiment. For the three exposure scenarios, the soil columns 119 received the same net amount of NPs, corresponding to a final concentration of 372 mg kg⁻¹ 120 dry soil, which is high compared to the predicted TiO₂ concentration (1.2 mg kg⁻¹ year⁻¹) in 121 European soils¹⁸. The choice of a high input concentration of TiO₂-NPs was made to ensure a 122 titanium concentration in the column effluents that was high enough to be distinguished from 123 the background concentration. Indeed, soils naturally contain very high concentrations of Ti²² (in average 4000-5000 mg kg⁻¹ and 4520 \pm 233 mg Ti kg⁻¹ dry soil in the studied soil). After 124 125 saturation and leaching with 100 mL of the soil solution, the column effluents before NP-126 spiking exhibited an average concentration of 0.33 \pm 0.1 mg Ti L⁻¹. Since we observed that 127 TiO₂-NP aggregation and surface charge were significantly different in water and soil 128 solution²¹, the suspensions of TiO₂-NPs were prepared in soil solution to achieve a more 129 realistic assessment of the fate and impact of NPs in the soil.

Soil exposures to NPs were performed in 1 x 10 cm glass columns (C10/10, GE Healthcare), homogeneously packed with soil (8 g equivalent dry soil), at 28 °C in the dark (see Figure S1 for schematic figure of the column experimental set up). Flow adaptors (AC 10, GE Healthcare) were adjusted on the top of the columns to ensure constant and similar soil heights during experiments. At the beginning of the experiment, soil columns were saturated and leached bottom-up with 100 mL of the pristine soil solution to reduce and stabilize the background Ti

concentration in the effluents. During the entire transport experiment, the solutions were injected in the column using a peristaltic pump at a flow rate of 0.1 mL min⁻¹, which corresponds to a Darcy flow of 0.15 cm h⁻¹. After this first step of saturation/leaching, 5 pore volumes (PV) (each pore volume was 6.7 mL) of the TiO₂-NPs suspension (either 50, 25 or 16.7 mg L⁻¹) were injected and then 10 PV of the soil solution free of TiO₂-NPs were immediately injected in the columns.

The single exposure experiment was performed as a one-time injection of 50 mg L⁻¹ TiO₂-NPs
into six saturated soil columns. Six control columns received the same PV of the soil solution
free of TiO₂-NPs.

Repeated exposure experiments were conducted as two or three applications of TiO₂-NPs with 25 and 16.7 mg L⁻¹, respectively in 6 columns for each treatment. Six additional control columns were used for each type of exposure as mentioned above. The successive applications of NPs were separated by a 15-day delay during which columns were maintained horizontally at 28 °C in the dark at constant moisture.

150 Transport of TiO₂-NPs in soil columns

151 For each exposure treatment, effluents from a randomly chosen column were continuously 152 sampled into 15 mL centrifuge tubes (1 mL sampled in 10 minutes) using a fraction collector 153 (Gilson, Minipulse 3). Ti concentrations in the spiking suspensions (C₀) and in the effluents (C) 154 were determined using a microwave assisted (Novawave, SCP Science) strong acid extraction 155 method (hydrofluoric acid + nitric acid) and were used to establish dimensionless TiO₂ 156 breakthrough curves (C/C_0) . Ti concentrations were measured by inductively coupled plasma 157 - optical emission spectrophotometry (ICP-OES; Varian 700-ES, Varian Inc. Scientific Instruments, Palo Alto, USA). Control columns leached with NP-free soil solution enabled the 158

determination of the background concentration of titanium in the effluents. In order to ensure
the quality of the titanium measurements, a certified reference material was measured along
with our samples: Sandy Loam 10, RTC-CRM027, lot HC027 certified reference material from
Sigma-Aldrich. The concentrations found were within 95–102 % of the certified values for all
measured elements.

164 Consistently with Nickel et al.²³, TiO₂-NP concentrations retained in the different soil layers 165 after contamination could not be accurately measured due to the naturally high titanium 166 content of the soil (4520 mg Ti kg⁻¹ dry soil) compared to the TiO₂-NP input concentration (372 167 mg kg⁻¹). Indeed, the standard error of the soil total Ti content measurements (233 mg kg⁻¹ 168 dry soil) was too close to the TiO₂-NP concentration added to the soil to expect reliable values 169 of TiO₂-NPs concentration in the different soil layers. Thus, our estimation of the final TiO₂-NP 170 concentration retained in soil is based on the measured Ti concentrations (mass balance) in 171 the column effluents and not a direct measurement of the Ti concentration in soil.

172

173 Impact of TiO₂-NPs on the nitrification process along the soil profile

174 The impact of TiO₂-NPs on soil nitrification activity and ammonia-oxidizing microorganisms 175 abundance as a function of soil depth was studied 2 months after the last exposure to NPs. 176 Immediately after the last exposure, soil columns were sliced in four layers according to the 177 distance from the inlet (0–2 cm, 2–4 cm, 4–6 cm and 6–8 cm). Each slice was then placed into 178 60 mL glass flasks and incubated at constant moisture for 2 months at 28 °C in the dark. This 179 experimental design resulted in 144 samples: 6 column treatments (three exposure scenarios 180 and three control treatments) x 4 depths x 6 replicates. At the end of the incubation time, 1.5 g of soil (equivalent dry weight, dw) were immediately used for the measurements of 181

182 nitrification enzyme activity (NEA) and 0.5 g of soil were stored at -20 °C before DNA
183 extraction.

184

185 DNA extraction and abundance of ammonia-oxidizers

DNA was extracted from 0.5 g of frozen soil using the Fast DNA[®] spin Kit for soil (MP Biomedicals, Solon, OH, USA), following the manufacturer's instructions and then DNA concentrations were determined using the Qubit dsDNA BR Assay (Invitrogen).

The abundance of ammonia-oxidizers (AOA and AOB) was measured by quantitative PCR using
a Lightcycler 480 (Roche Diagnostics, Meylan, France). The primers and thermal cycling
conditions used are described in Table 1.

192 Ammonia monooxygenase (amoA) gene abundance measurements for AOA and AOB 193 quantification were performed in a final reaction volume of 20 µL and contained (final 194 concentrations) 0.5 µM of each primer for the bacterial amoA or 0.75 µM of CrenamoA616r 195 and 1 µM of CrenamoA23f for the archaeal amoA, 2 % bovine serum albumin (BSA), 1X of 196 QuantiTect SybrGreen PCR Master Mix (Qiagen, Courtaboeuf, France) and 10 ng of soil DNA 197 extract or 10⁷–10² gene copies number of an in-house plasmid containing cloned bacterial 198 (Nitrosomonas europaea, GenBank accession number:L08050) and archaeal (54d9 fosmide 199 fragment²⁴) amoA genes. Melting curve analysis confirmed the specificity of amplification of 200 the two genes.

201

203 Nitrification enzyme activity

NEA was determined according to the protocol described by Dassonville et al. ²⁵. Sub-samples 204 205 of fresh soil (1.5 g equivalent dry weight) were incubated with 3 mL of a solution of (NH₄)₂SO₄ 206 in order to reach nitrogen-as-ammonia (N-NH4⁺) concentrations of 50 µg g⁻¹ dry soil. Distilled 207 water was added to each sample to achieve 12 mL of total liquid volume in glass plasma flasks. 208 The flasks were sealed with Parafilm® and incubated at 28 °C under 180 rpm constant 209 agitation. During the incubation, 1 mL of soil slurry was sampled at 2 h, 4 h, 6 h, 8 h and 10 h, 210 filtered (0.2 µm pore size) and transferred in vials stored at -20 °C. The analysis of NO3⁻ 211 concentrations was performed by ionic chromatography (DX120, Dionex, Salt Lake City, USA) 212 equipped with a 4 mm × 250 mm column (IonPac AS9 HC). NEA was expressed as µg nitrogenas-nitrate (N-NO₃⁻) $h^{-1}g^{-1}$ dry soil. 213

214

215 Statistical analysis

216 All results are presented as means (± standard error). A two-way analysis of variance (ANOVA) 217 and *post-hoc* Tukey's honest significant difference (HSD) were performed to test the effect of 218 TiO₂-NP exposures and soil depth on the nitrifying activity and ammonia-oxidizer abundance 219 for each type of exposure scenario separately. Data were log-transformed prior to analysis 220 when necessary to ensure conformity with the assumptions of normality and homogeneity of 221 variances. T-tests were conducted to compare the aggregation and zeta potential of TiO₂-NPs 222 in the different spiking suspensions. All statistical analyses were carried out using the R 223 statistical software 2.13.2²⁶.

225 **RESULTS**

226 Characteristics of TiO₂-NPs in the spiking soil solutions

TiO₂-NPs were characterized in the suspensions used to contaminate soil columns under the three exposure scenarios. The aggregation of TiO₂-NPs increased with the concentration (Table 2). At the lowest concentrations applied three times successively in soil columns (16.7 mg L⁻¹), the average TiO₂-NP hydrodynamic diameter was 111.2 (± 10.1) nm, while at the highest concentration used for the single application (50 mg L⁻¹), the hydrodynamic diameter was 153.9 (± 9.97) nm.

The surface charge of TiO₂-NPs assessed through zeta potential measurements (electrophoretic mobility) were extremely close for the three suspension concentrations (-15.4 to -16.6 mV; Table 2), even if it was significantly lower in the 25 mg L⁻¹ suspension (Table 2).

237 Effect of TiO₂-NP concentration and of the number of applications on their mobility in soil

The experimental design enabled the study of the influence of TiO₂-NPs concentration on their transport through soil after single exposures at 16.7, 25 or 50 mg L⁻¹ and after repeated exposures consisting of two or three applications of NPs in suspensions containing 25 and 16.7 mg L⁻¹, respectively (Table 3).

The mobility of TiO_2 -NPs decreased with increasing TiO_2 -NP concentration (Table 3). For an injected concentration of 16.7 mg L⁻¹, 10.66 % of TiO_2 -NPs were recovered in the effluents, whereas only 4.95 and 2.19 % were recovered when applying the suspensions containing 25 and 50 mg L⁻¹, respectively.

In the columns that received two applications of NPs, the mobility of TiO₂-NPs was greatly reduced compared to the mobility measured during the first application (Table 3). Only 0.9 % of the added TiO₂-NPs were recovered in the effluents for both 25 and 16.7 mg L⁻¹ exposure concentrations (Table 3). For the third NP application at 16.7 mg L⁻¹, the mobility of TiO₂-NPs was almost negligible (0.2%; Table 3).

251 We calculated the final TiO₂-NP concentration retained in the column on the basis of the mass 252 recovery calculations in the effluents after background subtraction (Table 3). Although the 253 same theoretical TiO₂-NPs concentration of 372 mg kg⁻¹ was applied to the soil, the final NPs 254 retained concentration differed in the three exposure scenarios (Table 3). The highest final 255 concentration of TiO₂-NPs (364 mg kg⁻¹) was measured in the columns exposed to a single 256 application at 50 mg L⁻¹. In the columns receiving two applications at 25 mg L⁻¹, the final TiO₂-NPs concentration was estimated at 350 mg kg⁻¹, which corresponds to a decrease of 6 % 257 compared to the single exposure. After three applications at 16.7 mg L⁻¹, only 328 mg kg⁻¹ of 258 259 TiO₂-NPs were retained in the soil, which corresponds to a decrease of 12 % and 6 %, 260 compared to the single and dual exposure scenarios, respectively.

261 Effect of single and repeated NP contaminations on nitrification enzyme activity and 262 ammonia-oxidizer abundance

After two months of incubation following the last application of NPs to soil columns, NEA, AOA and AOB abundances were measured. When considering the whole soil column (i.e. the four soil layers), the single exposure (i.e. single application at 50 mg L⁻¹) resulted in a significant increase of the AOA abundance (Figure 1; Table 4) but no effect was observed on AOB abundance or NEA (Figure 1; Table 4). After two successive NP applications at 25 mg L⁻¹, neither the NEA nor the ammonia-oxidizer abundances were affected by TiO₂-NPs contamination (Figure 1, Table 4). In contrast, NEA, AOA and AOB abundance were
 significantly decreased after three repeated applications of TiO₂-NPs at 16.7 mg L⁻¹ (Figure 1;
 Table 4).

A significant effect of soil depth on NEA was only observed after two and three exposures (*P* = 0.02 and *P* = 0.003, respectively; Table 4). After two soil exposures to NPs, the NEA differed significantly according to soil depth, especially at 2–4 cm (Figure 2). After three exposures, NEA was significantly more reduced by TiO₂-NPs in the first 2 cm than in the other soil layers (Figure 2). No significant statistical interaction between TiO₂-NPs contamination and soil depth was found (Table 4).

278

279 **DISCUSSION**

280 Consequences of single and repeated exposures on TiO₂-NPs transport in soil

281 Our experimental design based on soil columns dynamic contamination enabled coupled 282 mobility and toxicity studies of TiO₂-NPs under single and repeated exposures at different 283 input concentrations. The results showed that the transport of TiO₂-NPs is low (90 to 99% of 284 retention) and influenced by the concentration injected in the soil columns. After a single 285 application, TiO₂-NP retention in soil increased with increasing injected concentrations. 286 Similar behavior of silica nanoparticles has been reported in sand columns by Vitorge et al.²⁷, 287 who showed that NPs mobility increases at low concentration regimes. These results can be 288 explained by the increase of TiO₂-NP aggregation with increasing concentration, which can 289 favor the filtration, deposition and straining of these colloids^{7,28}. Although the significant 290 difference of TiO₂-NP aggregation between the three concentrations tested in soil solution (42 nm maximum, 28% of the highest hydrodynamic diameter measured) supports this hypothesis, the involvement of other factors such as heteroaggregation kinetics cannot be ruled out.

294 Previous studies on the mobility of uncoated TiO₂-NPs in soil reported contrasting results. Nickel et al. $(2015)^{23}$ observed no transport using an input concentration of 1 g TiO₂-NPs L⁻¹, 295 296 while significant transport was recorded in different soils exhibiting low clay contents and ionic strengths with several applied concentrations of TiO₂-NPs⁹. Here, we show that although 297 298 low, the transport of TiO₂-NPs can occur after a single exposure despite the fine texture of the 299 studied soil (39 % clay and 51 % loam), especially at the lowest concentration regime, which 300 is more likely to be observed in natural soils. In the case of an accidental spill for which high 301 NPs concentrations can be expected, the mobility of TiO₂-NPs will be likely very low, in 302 agreement with a classical blocking mechanism typically observed at high colloidal concentrations^{27,28} 303

304 For repeated exposure in soil columns, an extremely low mobility of TiO₂-NPs was observed 305 (< 1%) and no concentration effect was observed contrarily to what was observed with the 306 single exposures. The distribution of TiO₂-NPs along the soil profile could not be measured 307 without specific labeling²³. However, these results observed in only 8 cm long columns suggest 308 that in soils repeatedly exposed to TiO₂-NPs, these contaminants would accumulate in the top 309 soil, where, coincidentally, most of the biological activity occurs. In these conditions, it is likely 310 that NPs transport to deeper soil layers and groundwater is limited. In order to confirm this 311 assumption future studies are needed to assess TiO₂-NPs transport in longer soil columns and 312 in undisturbed soil cores.

314 Deleterious effects on nitrification after repeated exposure

315 Two months after the last application of TiO₂-NPs to the soil columns, significant decreases in 316 NEA and ammonia-oxidizer abundances were observed only in soil samples exposed three 317 times to the lowest TiO₂-NPs concentration (16.7 mg L⁻¹). In this case, the final TiO₂-NPs 318 concentration in soil was slightly lower than in the two other exposure scenarios. Surprisingly, 319 we have observed an increase of AOA abundance in the single exposure scenario but without 320 significant changes on NEA. A possible explanation is that a shift occurred in the AOA 321 community and that AOA populations with lower nitrifying activity were stimulated in these 322 conditions. We show in this short-term experiment that soil repeated exposure to TiO₂-NPs 323 has a more deleterious impact on soil microbial community and soil functioning than a single 324 exposure. This result highlights that the nitrifying population did not become more resistant 325 and resilient to TiO₂-NPs after multiple exposures. On the contrary, the nitrification process 326 and ammonia-oxidizers became more sensitive, probably because organisms stressed by the 327 previous exposures to NPs had less energy to cope with additional disturbances and/or because of a decrease or a shift of soil microbial diversity²⁹. It is well known that a microbial 328 329 community can become tolerant to various stressors over time through the selection of more 330 tolerant populations and/or the dispersal of specific resistances via mobile genetic 331 elements^{14,29,30}. Chronic exposures to pollutants occurring over a longer term will likely have 332 less consequences on nitrification because of potential microbial community adaptation over time, as it was observed by Mertens et al.³¹ after 2 years of exposure to zinc pollution. To gain 333 334 a better knowledge of NP impact on soil functioning under chronic contamination, it would be 335 necessary to study the effects on microbial activity, abundance and diversity after each exposure and before each new exposure, in order to monitor the resistance and resilience of 336 337 the microbial community in the long term. This approach could lead to new insights into

sensitive and tolerant microbial species among nitrifiers and thus on TiO₂-NP toxicity and
 adaptation mechanisms in soil.

340 After three NP applications, it is likely that most of TiO₂-NPs were retained close to the inlet 341 of the column through filtration and/or straining. Although a TiO₂ concentration gradient can 342 be expected along the soil profile according to the filtration theory³², no significant effect of 343 depth on ammonia-oxidizer abundance was observed. Nitrification was significantly more 344 diminished in the 0–2 cm and 6–8 cm layers than in the middle 2–4 and 4–6 cm soil layers. 345 With common soluble pollutants, the general assumption is that the highest concentrations 346 induce the most deleterious effects. In the case of NPs, concentration effects are more 347 complex because of aggregation processes and thus the reactivity, bioavailability and toxicity of these contaminants can vary in function of the spiked concentration^{21,33–35}. Such changes 348 in NPs aggregation with concentration are likely to induce nonlinear dose-effect 349 relationships³⁶. Our results suggest that microbial communities may be affected by TiO₂-NPs 350 351 at the depths of preferential NP accumulation (top of the soil column) but also at deeper 352 depths likely reached by NPs at lower concentration (bottom of the soil column).

353 The multidisciplinary approach developed in this study enabled the comparison of the impact 354 of single and repeated TiO₂-NPs exposures in water-saturated soil columns. The transport of 355 TiO₂-NPs in the studied silty-clay soil was evidenced but was limited (over 90% retention) for 356 both single and repeated applications in 8 cm long columns. TiO₂-NPs mobility was also shown 357 to increase at low NPs concentrations indicating an aggregation-controlled process. 358 Considering these results, the transport of TiO₂-NPs to deeper soil layers or groundwater 359 appears limited. Although the repeat exposures resulted in lower final TiO₂-NP concentrations 360 in soil, the impact was greater on the soil microbial community and on soil functioning (NEA)

compared to the single exposure condition. Future studies should address the fate and effects of TiO₂-NPs chronically added to soils in biosolid amendments, as it is one of the most relevant routes by which NPs enter the soil environment³⁷. Chronic exposure in short and long term experiments representing realistic scenarios of soil exposure to NPs are urgently needed to study the resistance and resilience of soil microbial communities to these emerging contaminants and their consequences on soil fertility.

367

368

369 TABLE CAPTIONS

370

371 Table 1 PCR primers and thermal cycling conditions used for quantification of ammonia-372 oxidizer abundances

373

Table 2 Size and zeta potential of TiO_2 -NPs in soil solution for the 3 tested concentrations measured with a Nano ZS (Malvern). Means and standard errors are presented (n = 3). Values labeled with the same letter do not differ at P < 0.05.

377

Table 3 TiO_2 relative mass recoveries in column effluents for the three applied NPs concentrations in the different exposure scenarios (one, two, or three applications). The final TiO_2 -NP concentration retained in the column was calculated from the relative mass recovery of TiO_2 -NPs in the effluents. The presented values have been corrected by the Ti background concentration in the effluents measured before NP application (0.33 mg Ti L⁻¹).

383

Table 4 P-values from an ANOVA analysis of the effects of TiO_2 -NPs, soil depth and their interaction on nitrification enzyme activity, AOA and AOB abundances for each exposure scenario. Values in bold are significant at P < 0.05.

387

388 FIGURE CAPTIONS

Figure 1 Effect of TiO₂-NPs on nitrification enzyme activity (NEA), AOA and AOB abundances in whole columns (average of the 4 soil depths). Data are expressed as the percentage of change relative to the control treatment for each exposure. Standard errors are presented (n = 6). Significant effects are indicated (***, P < 0.001; **, P < 0.01; *, P < 0.05).

393

Figure 2 Effect of TiO₂-NPs on nitrification enzyme activity (NEA) measured in the four soil column layers (depths 0-2 cm; 2-4 cm; 4-6 cm; 6-8 cm) after one, two or three soil exposures. Data are expressed as percentage of change from the control treatment for each exposure condition. Standard errors are presented (n = 6). For each exposure scenario, bars labeled with the same letter do not differ at P < 0.05.

399

400 ACKNOWLEDGMENTS

401 Marie Simonin was supported by a Ph.D grant from Rhône-Alpes Region – ARC 402 Environnement. This work was funded by a grant from the French National Program EC2CO 403 Microbien of CNRS. The authors thank the technical assistance of Sungeun Lee. Nitrification 404 enzyme activity measurements were performed at the AME platform (Microbial Ecology

- 405 UMR5557-USC1364, Lyon) and quantitative PCR at the DTAMB platform (IFR 41, University
- 406 Lyon 1). LTHE is part of the Labex OSUG@2020 (ANR10 LABX56).

- 408 **REFERENCES**
- 409 (1) Dinesh, R.; Anandaraj, M.; Srinivasan, V.; Hamza, S. Engineered nanoparticles in the soil
 410 and their potential implications to microbial activity. *Geoderma* 2012, 173–174, 19–27.
- 411 (2) Pan, B.; Xing, B. Applications and implications of manufactured nanoparticles in soils: a
 412 review. *Eur. J. Soil Sci.* 2012, *63* (4), 437–456.
- 413 (3) Simonin, M.; Richaume, A. Impact of engineered nanoparticles on the activity,
 414 abundance, and diversity of soil microbial communities: a review. *Environ. Sci. Pollut.*415 *Res.* 2015, 1–14.
- 416 (4) Brar, S. K.; Verma, M.; Tyagi, R. D.; Surampalli, R. Y. Engineered nanoparticles in
 417 wastewater and wastewater sludge Evidence and impacts. *Waste Manag.* 2010, *30*418 (3), 504–520.
- 419 (5) Liu, R.; Lal, R. Potentials of engineered nanoparticles as fertilizers for increasing
 420 agronomic productions. *Sci. Total Environ.* 2015, *514*, 131–139.
- 421 (6) Servin, A.; Elmer, W.; Mukherjee, A.; Torre-Roche, R. D. la; Hamdi, H.; White, J. C.;
 422 Bindraban, P.; Dimkpa, C. A review of the use of engineered nanomaterials to suppress
 423 plant disease and enhance crop yield. *J. Nanoparticle Res.* 2015, *17* (2), 1–21.
- 424 (7) Cornelis, G.; Hund-Rinke, K.; Kuhlbusch, T.; Brink, N. van den; Nickel, C. Fate and
 425 Bioavailability of Engineered Nanoparticles in Soils: A Review. *Crit. Rev. Environ. Sci.*426 *Technol.* 2014, 44 (24), 2720–2764.
- 427 (8) Navarro, D. A.; Banerjee, S.; Watson, D. F.; Aga, D. S. Differences in Soil Mobility and
 428 Degradability between Water-Dispersible CdSe and CdSe/ZnS Quantum Dots. *Environ.*429 Sci. Technol. 2011, 45 (15), 6343–6349.
- 430 (9) Fang, J.; Shan, X.; Wen, B.; Lin, J.; Owens, G. Stability of titania nanoparticles in soil
 431 suspensions and transport in saturated homogeneous soil columns. *Environmental*432 *Pollution* 2009, 157, 1101–1109.
- (10) Martins, J. M. F.; Majdalani, S.; Vitorge, E.; Desaunay, A.; Navel, A.; Guiné, V.; Daïan, J.
 F.; Vince, E.; Denis, H.; Gaudet, J. P. Role of macropore flow in the transport of
 Escherichia coli cells in undisturbed cores of a brown leached soil. *Environ. Sci. Process. Impacts* 2013, *15* (2), 347–356.
- 437 (11) Vitorge, E.; Szenknect, S.; Martins, J. M. F.; Barthès, V.; Auger, A.; Renard, O.; Gaudet,
 438 J.-P. Comparison of three labeled silica nanoparticles used as tracers in transport
 439 experiments in porous media. Part I: Syntheses and characterizations. *Environ. Pollut.*440 2014, 184, 605–612.
- 441 (12) Dalzell, D. J. B.; Alte, S.; Aspichueta, E.; de la Sota, A.; Etxebarria, J.; Gutierrez, M.;
 442 Hoffmann, C. C.; Sales, D.; Obst, U.; Christofi, N. A comparison of five rapid direct toxicity
 443 assessment methods to determine toxicity of pollutants to activated sludge.
 444 *Chemosphere* 2002, 47 (5), 535–545.

- 445 (13) Broos, K.; Mertens, J.; Smolders, E. Toxicity of heavy metals in soil assessed with various
 446 soil microbial and plant growth assays: A comparative study. *Environ. Toxicol. Chem.*447 **2005**, *24* (3), 634–640.
- 448 (14) Bissett, A.; Brown, M. V.; Siciliano, S. D.; Thrall, P. H. Microbial community responses to
 449 anthropogenically induced environmental change: towards a systems approach. *Ecol.*450 *Lett.* 2013, *16*, 128–139.
- 451 (15) Schloter, M.; Dilly, O.; Munch, J. C. Indicators for evaluating soil quality. *Agric. Ecosyst.*452 *Environ.* 2003, *98* (1–3), 255–262.
- 453 (16) Wessén, E.; Hallin, S. Abundance of archaeal and bacterial ammonia oxidizers Possible
 454 bioindicator for soil monitoring. *Ecol. Indic.* **2011**, *11* (6), 1696–1698.
- 455 (17) Keller, A. A.; McFerran, S.; Lazareva, A.; Suh, S. Global life cycle releases of engineered
 456 nanomaterials. *J. Nanoparticle Res.* **2013**, *15* (6), 1–17.
- 457 (18) Sun, T. Y.; Gottschalk, F.; Hungerbühler, K.; Nowack, B. Comprehensive probabilistic
 458 modelling of environmental emissions of engineered nanomaterials. *Environ. Pollut.*459 **2014**, *185*, 69–76.
- 460 (19) Sohm, B.; Immel, F.; Bauda, P.; Pagnout, C. Insight into the primary mode of action of
 461 TiO2 nanoparticles on Escherichia coli in the dark. *Proteomics* 2015, *15* (1), 98–113.
- 462 (20) Erdem, A.; Metzler, D.; Cha, D. K.; Huang, C. P. The short-term toxic effects of TiO2
 463 nanoparticles toward bacteria through viability, cellular respiration, and lipid
 464 peroxidation. *Environ. Sci. Pollut. Res.* 2015, *22* (22), 17917–17924.
- 465 (21) Simonin, M.; Guyonnet, J. P.; Martins, J. M. F.; Ginot, M.; Richaume, A. Influence of soil
 466 properties on the toxicity of TiO2 nanoparticles on carbon mineralization and bacterial
 467 abundance. *J. Hazard. Mater.* 2015, 283, 529–535.
- 468 (22) Aubert, H.; Pinta, M. *Trace Elements in Soils*; Elsevier, 1980.
- 469 (23) Nickel, C.; Gabsch, S.; Hellack, B.; Nogowski, A.; Babick, F.; Stintz, M.; Kuhlbusch, T. A.
 470 Mobility of coated and uncoated TiO 2 nanomaterials in soil columns–Applicability of
 471 the tests methods of OECD TG 312 and 106 for nanomaterials. *J. Environ. Manage.* 2015,
 472 157, 230–237.
- 473 (24) Treusch, A. H.; Leininger, S.; Kletzin, A.; Schuster, S. C.; Klenk, H.-P.; Schleper, C. Novel
 474 genes for nitrite reductase and Amo-related proteins indicate a role of uncultivated
 475 mesophilic crenarchaeota in nitrogen cycling. *Environ. Microbiol.* 2005, 7 (12), 1985–
 476 1995.
- 477 (25) Dassonville, N.; Guillaumaud, N.; Piola, F.; Meerts, P.; Poly, F. Niche construction by the
 478 invasive Asian knotweeds (species complex Fallopia): impact on activity, abundance and
 479 community structure of denitrifiers and nitrifiers. *Biol. Invasions* 2011, *13* (5), 1115–
 480 1133.
- 481 (26) R Core Team. R: A language and environment for statistical computing. R Foundation
 482 for Statistical Computing. Vienna, Austria 2015.
- 483 (27) Vitorge, E.; Szenknect, S.; Martins, J. M. F.; Gaudet, J.-P. Size- and concentration484 dependent deposition of fluorescent silica colloids in saturated sand columns: transport
 485 experiments and modeling. *Environ. Sci. Process. Impacts* 2013, *15* (8), 1590–1600.
- 486 (28) Solovitch, N.; Labille, J.; Rose, J.; Chaurand, P.; Borschneck, D.; Wiesner, M. R.; Bottero,
 487 J.-Y. Concurrent Aggregation and Deposition of TiO2 Nanoparticles in a Sandy Porous
 488 Media. *Environ. Sci. Technol.* 2010, 44 (13), 4897–4902.
- 489 (29) Griffiths, B. S.; Philippot, L. Insights into the resistance and resilience of the soil
 490 microbial community. *FEMS Microbiol. Rev.* 2013, *37* (2), 112–129.

- 491 (30) Allison, S. D.; Martiny, J. B. H. Resistance, resilience, and redundancy in microbial
 492 communities. *Proc. Natl. Acad. Sci.* 2008, *105* (Supplement 1), *11512–11519*.
- 493 (31) Mertens, J.; Broos, K.; Wakelin, S. A.; Kowalchuk, G. A.; Springael, D.; Smolders, E.
 494 Bacteria, not archaea, restore nitrification in a zinc-contaminated soil. *ISME J.* 2009, *3*495 (8), 916–923.
- 496 (32) Huber, N.; Baumann, T.; Niessner, R. Assessment of Colloid Filtration in Natural Porous
 497 Media by Filtration Theory. *Environ. Sci. Technol.* 2000, *34*, 3774–3779.
- 498 (33) Choi, O. K.; Hu, Z. Q. Nitrification inhibition by silver nanoparticles. *Water Sci. Technol.*499 **2009**, *59* (9), 1699.
- 500 (34) Simon-Deckers, A.; Loo, S.; Mayne-L'hermite, M.; Herlin-Boime, N.; Menguy, N.;
 501 Reynaud, C.; Gouget, B.; Carrière, M. Size-, Composition- and Shape-Dependent
 502 Toxicological Impact of Metal Oxide Nanoparticles and Carbon Nanotubes toward
 503 Bacteria. *Environ. Sci. Technol.* 2009, 43 (21), 8423–8429.
- 504 (35) Menard, A.; Drobne, D.; Jemec, A. Ecotoxicity of nanosized TiO2. Review of in vivo data.
 505 Environ. Pollut. 2011, 159 (3), 677–684.
- 506 (36) Bernhardt, E. S.; Colman, B. P.; Hochella, M. F.; Cardinale, B. J.; Nisbet, R. M.;
 507 Richardson, C. J.; Yin, L. An ecological perspective on nanomaterial impacts in the
 508 environment. *J. Environ. Qual.* 2010, *39* (6), 1954–1965.
- 509 (37) Pradas del Real, A. E.; Castillo-Michel, H.; Kaegi, R.; Sinnet, B.; Magnin, V.; Findling, N.;
 510 Villanova, J.; Carrière, M.; Santaella, C.; Fernández-Martínez, A.; et al. Fate of Ag-NPs in
 511 Sewage Sludge after Application on Agricultural Soils. *Environ. Sci. Technol.* 2016, *50* (4),
 512 1759–1768.
- 513