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▶ To cite this version:

Collet Pierre, Eglantine Flottes, Alain Favre, Ludovic Raynal, Hélène Pierre, et al.. Techno-economic and Life Cycle Assessment of methane production via biogas upgrading and power to gas technology. Applied Energy, 2017, 192, pp.282-295. 10.1016/j.apenergy.2016.08.181 . hal-01519912

HAL Id: hal-01519912 https://hal.science/hal-01519912

Submitted on 9 May 2017

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Techno-economic and Life Cycle Assessment of methane production via biogas upgrading and power to gas technology

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- 13
- 14 Abstract

15 To decrease the use of fossil fuels and face the energetic demand, the integration of renewable 16 energy is a necessary step. Part of this renewable energy can be supplied by the production of 17 electricity from photovoltaic panels and windfarms. The massive use of these intermittent energies will lead to overproduction periods, and there is consequently a need to convert this surplus of 18 19 electricity into a storable form of energy. Power-to-gas (PtG) technology consists in using electricity 20 to convert water into hydrogen by electrolysis, and then to synthetize methane from carbon dioxide 21 and hydrogen. Techno-economic and Life Cycle Assessment of methane production via the 22 combination of anaerobic digestion and PtG technology have been applied to sewage sludge 23 valorization. Process studies and equipment design have been addressed considering already 24 available technologies. Sensitivity analyses have been done on biogas upgrading technologies, 25 electricity prices, annual operation time and composition of the electricity mix with also a 26 comparison between PtG and direct injection. It appears that the more the electricity is expensive, the longer the operation time of the methanation process must be to be competitive with injection 27 of methane from biogas. Reduction of electricity consumption of the electrolysis step decreases 28 29 production costs. Even if the current context does not feature adapted conditions to ensure an 30 economically viable chain, the evolution of the energetic context in the next few years as well as the 31 expected technological improvements will contribute to overall cost reduction. From an 32 environmental point of view, continuous PtG generates more greenhouse gases than direct injection, but intermittent operation with use of renewable electricity can significantly reduce GHG emissions. 33 34 From an endpoint impacts perspective, impact from continuous PtG are higher than biogas 35 upgrading, but much lower than fossil energy. Future development of low electricity consumption of 36 the electrolysis process, and integration of renewable credits from CO₂ valorization can increase the 37 competitiveness of this technology.

38 Keywords: Life Cycle Assessment (LCA); Power-to-Gas; Methane; CO₂ valorization; Energy storage;
 39 Production costs

40 **1** Introduction

Even if fossils fuels accounted for 83% of total primary energy supply in 2008 (IPCC 2011), the use of 41 42 fossil energy faces many issues: decreasing reserves, emissions of greenhouses gas (GHG) and other 43 pollutants during their combustion, but also dependence to importations, with the associated risks. It 44 is consequently compulsory to develop alternative energy production technologies. Renewable 45 energy can play a central role in minimizing these issues, and can be partly supplied by electricity 46 from renewable sources, like photovoltaic panels and windfarms. This electricity emits less GHG 47 (Varun et al. 2009) and pollutants (Turconi et al. 2013; Turconi et al. 2014) than fossil fuels (except 48 nuclear electricity), and can be produced almost all over the planet (de Vries et al. 2007). 49 Nevertheless, renewable electricity production faces two major drawbacks: first its intermittency, 50 with a production which cannot be adjusted to electricity demand, and second its substitution to 51 high energy density fuels, especially in transport and heat (Sterner 2009).

52 While renewable electricity represents a small proportion of electricity production in Europe (24.7%) 53 and in France (16.5%) in 2014 (http://ec.europa.eu/eurostat/statistics-54 explained/index.php/Electricity production and supply statistics), recent publication of the 55 European Commission on energy production forecasts a large share of renewables (between 64% 56 and 97% depending on the scenarios) in the electricity mix in 2050 (European Parliament 2011), and 57 scenarios developed by the ADEME (ADEME 2015) go as far as to propose fully renewable electricity 58 mix.

59 To fulfill the needs of electricity demand with the integration of more renewables in the electricity mix, the installed capacity should be increased. This massive electricity production from renewable 60 61 energies will lead to overproduction periods, where produced electricity would not be totally 62 employed. Therefore there is a need to store this energy. Several electricity storage technologies are available like batteries, pumped hydropower plants, compressed air energy storage or hydrogen 63 64 storage technologies, with different prices, run times and storage capacities (Hadjipaschalis et al. 2009). Pumped hydropower technology is the most massive storage technology available today, and 65 66 is generally enough to balance the current electrical system. However it will be probably inadequate 67 to store large amounts of overproduced electricity (ADEME 2014) and it has already been largely 68 deployed. In this perspective, natural gas network offers a high storage capacity (for example 135 69 TWh in France, (DGEC (Direction générale de l'énergie et du climat) 2011)), and therefore conversion 70 of electricity into gas would offer an interesting leverage to valorize overproduced electricity. In this 71 perspective, production of methane (CH_4) by power-to-gas (PtG) technology can greatly increase the 72 total production of CH_4 from biogas by combining carbon dioxide (CO_2) contained in the biogas and 73 converted to CH_4 via methanation with CH_4 already present in the biogas (Mohseni et al. 2012; Reiter 74 & Lindorfer 2015; Zoss et al., 2016). PtG is defined here as using electricity to convert water into 75 hydrogen (H₂) by electrolysis, and then to synthetize methane from carbon dioxide (CO₂) and H₂ 76 through methanation (Sabatier reaction). H2 production from electrolysis plays a key role in 77 integrated energy system (Yan, 2016), as fuel for transportation for instance (Nistor et al., 2016). 78 Furthermore, the synthesis of methane through methanation can strongly contribute to large scale 79 energy storage, as CH₄ injection is not limited in the gas grid, contrary to hydrogen which also faces 80 process and safety management issues.

81 Both economic and environmental criteria are crucial to fully assess the relevance of a new technology. In this study we propose a model for CH₄ production from PtG that evaluates these two 82 83 dimensions of sustainability, by taking into account intermittent operation functioning. Economic assessment is done by calculating capital expenditures (CAPEX) and operational expenditures (OPEX) 84 85 for each analyzed configuration, as in de Boer et al., (2014). Economic assessment of methane 86 production from PtG have been proposed recently (Götz et al., 2016), with sometimes the use of 87 time dependent optimization approach (Rivarolo et al., 2014). Environmental evaluation is done with 88 Life Cycle Assessment (LCA), a standardized tool allowing to assess the environmental impacts of the 89 whole cycle of a process, from raw extraction to final wastes management (ISO 2006). It should be 90 noticed that it is an attributional study (and not consequential), which means that changes in the 91 techno-economic sphere induced by large-scale production of bioCH₄ from PtG are not considered 92 (Ekvall & Weidema 2004). In particular, this implies that the use of long-term marginal data to 93 characterize the electric mix is not in the scope of the study (Amor et al. 2014). Several LCAs of H₂ 94 production have been done in the past years (Dufour et al. 2012; Dufour et al. 2009; Patyk et al. 95 2013; Cetinkaya et al. 2012), with sometimes a focus on the use of electricity from renewables (Mori 96 et al. 2014). Nevertheless, only a few environmental studies have been conducted on CH_4 production 97 by PtG: GHG emissions of used electricity for electrolysis in Jentsch et al., (2011) or direct carbon 98 emissions (de Boer et al. 2014), and only one Life Cycle Approach to evaluate GHG emissions (Reiter 99 & Lindorfer 2015). To our knowledge, this is the first time that both economic and environmental 100 assessments are performed together to evaluate the production of bioCH₄ from PtG technology with 101 continuous and intermittent operations modes.

102 2 Methodology

103 2.1 Overview of the system

104 PtG consists in the conversion of electricity into H_2 and further into CH_4 , which can be stored without 105 restriction in the natural gas network. CH₄ can then be further used for different purpose: storage of 106 electricity, heat production, raw materials for chemical industry and transportation services via NGV 107 production. In this paper, we will focus on the production of heat from CH₄, as it is the main use of 108 natural gas, from two different sources: from biogas upgrading, with no valorization of the CO₂, or 109 from biogas upgrading and CO₂ conversion into CH₄ via methanation (Figure 1). Different 110 configurations are analyzed : CH4 from methanation after biogas upgrading, CH4 from methanation 111 of the biogas without upgrading (direct methanation), and finally CH4 from biogas upgrading without 112 methanation. A fossil reference system (natural gas) is also included in the LCA. The system assessed in the model includes all steps from biogas production to CH₄ combustion in a boiler: anaerobic 113 114 digestion of sewage sludge, biogas upgrading and compression, electrolysis, methanation, 115 completion, injection in the gas network, and combustion. The wastewater treatment plant of the 116 study is designed for a population equivalent of 300 000 inhabitants. The daily quantity of sewage 117 sludge processed is 16 440 kg of dry matter, which leads to a biogas production of 230 m³.h⁻¹. The 118 anaerobic digestion plant, as well as the electrolysis and the methanation installations are supposed 119 to be near the wastewater treatment plant. The inventory is based on figures derived from academic 120 resources, internal communications with industrials and processes described in the Ecoinvent 121 database (Frischknecht et al. 2007). The location of the system is in France; as a consequence the 122 electric mix is the French one, with a low carbon content. Infrastructures are included in the 123 economic assessment but are not considered in the LCA, in line with the assessment of renewable energy production in The Renewable Energy Directive (Parliament 2009). 124



Biogas upgrading without methanation

126 Figure 1: Overview of biomethane production with (a) and without (b) CH₄ synthesis via methanation

127 2.2 Anaerobic digestion step

125

The data used for the functioning of the anaerobic digestion plant are summed up in Table 1. The heat necessary for operating the anaerobic digestion plant is provided by burning part of the produced biogas in a boiler and by the recovery of wasted heat from methanation when it is functioning. Compositions of the biogas and the digestates come from literature survey and from Suez Environment internal data. Biogenic CH_4 emissions occur during biogas production and digestates storage, and NH_3 and N_2O emissions are linked with the spreading of the digestates on the

- 134 fields. Biogenic CO₂ fixation and emission are not taken into account, and it is supposed that the
- 135 carbon of the digestates is entirely reemitted in the environment.
- 136

Table 1: Main assumptions at the anaerobic digestion step

Description	Values	Reference
Inputs		
Sewage sludge	685 kg.h ⁻¹	
Electricity consumption	336 MJ.h ⁻¹	(Frank et al. 2012)
Heat consumption	1343 MJ.h ⁻¹	(Frank et al. 2012)
Outputs		
Biogas	230 m3.h ⁻¹	Present data
	0.220 L.g ⁻¹ SewSlu	
	CH ₄ : 65.63%mol / CO ₂ : 34.37%mol	
	T: 40°C / P: 1 bar	
Digestates	685 kg.h ⁻ 1	(ADEME 2011)
	N:55 g.kg ⁻¹ Dig / P: 48.5 g.kg ⁻¹ Dig /	
	K: 4.28 g.kg ⁻¹ Dig	
Methane (biogenic) emissions	3 135 gCH ₄ .h ⁻¹	(Amon et al. 2006)
N ₂ O emissions	$808 \text{ gN}_2 \text{O.h}^{-1}$	(Foley et al. 2010)
N-NH ₃ emissions	7 809 gN-NH ₃ .h ⁻¹	(Foley et al. 2010)

137 2.3 Biogas upgrading and compression steps

138 In the present study, the impurities contained in the biogas (especially sulphur and silicone derived 139 compounds) are removed via activated carbon adsorption, this latter step being considered as 140 integrated with the biogas production. Biogas is thus considered as composed of CH_4 and CO_2 and 141 biogas upgrading focus on CO₂ removal. Four different technologies are considered for biogas 142 upgrading, they are listed in Table 2 with their main characteristics. This table is based on different 143 publications (Ryckebosch et al. 2011; Kaparaju 2013; IEA 2009; IEA 2014), and gives gross ranges for 144 each of biogas upgrading technologies. The color code is the following: green is used for a 145 characteristic that gives an specific advantage to the technology, especially in the context of 146 downstream use of CO₂ in the methanation process, and on the contrary orange is used for 147 drawbacks. For instance CO₂ outlet pressure after biogas upgrading can be around 1.8 bar (amines),

148 which represents a relative benefit as there slightly less electricity is needed for CO2 compression 149 before methanation compared to the other upgrading technologies. While water scrubbing 150 technology is the favorite technology for biogas upgrading, one observes that the second preferred technology is the chemical absorption. For the latter process, the main drawback is the heat 151 152 requirement at the regeneration step, a well-known drawback in the post-combustion carbon 153 capture community (Steeneveldt et al. 2006). However, in the present case, where heat can be 154 supplied by methanation, this drawback can easily be overcome, and even turned into an advantage. 155 Similarly the methane loss of membrane technology is not anymore an issue since the CO_2 containing 156 some CH_4 will be converted into CH_4 at the methanation step. Furthermore, membrane technology is 157 considered as the easiest to operate which can be a major advantage for such small size units for 158 which maintenance and labor operating cost must be minimum (Makaruk et al. 2010).

Table 2: Comparison of upgrading technologies.

	Water Scrubber	Pressure Swing Adsorption	Chemical absorption (amines)	Membrane	
General characteristics					
Electricity demand (kWh.m ⁻³ _{BG})	0.2 -0.3	0.2 - 0.25	0.15	0.25	
Heat demand (°C)	no	no 120-160		no	
Operation pressure (bar)	5 - 10	4 - 7	0.1	5 - 10	
Methane recovery (%)	98	98	99.96	80-99.5	
Methane content in upgraded gas (%mol)	>97	>96	>99	>95	
Exhaust gas treatment	yes	yes	no	yes	
Water demand	yes (+)	no	yes	no	
Demand on chemicals	no	no	yes	no	
Ease of operation	complex	intermediate	complex	easy	
Total costs (€/m ³ _{CH4})	0.13	0.25	0.17 - 0.28	0.12 - 0.22	
Number of units					
number 2011	48	41	31	6	
number 2013	113	58	61	27	
Integration with methanation					
Heat integration	No	No Yes (similar ΔH)		No	
CO ₂ pressure	1	<1	1.8	1	

160

161 Amines scrubbing and membranes are thus the two different upgrading technologies that have been 162 assessed in the present study. Chemical (amines) scrubbing corresponds to the formation of chemical 163 bonds between the CO_2 and a solvent, in general an aqueous solution of alcanolamines. Regeneration of the solvent is endothermic, and required heat is provided by burning part of the 164 biogas or, preferentially in the present case, by the recovery of wasted heat from methanation when 165 166 it is functioning. Monoethanolamine (MEA) has been considered for solvent since it is a well-known 167 industrial solvent characterized as a highly reactive high energy requirement solvent. The first 168 characteristic is in favor of low capital expenditures (CAPEX) while the second is in favor of high 169 operational expenditures (OPEX) (Raynal et al. 2011). In the present case, since heat is available at 170 methanation step, CAPEX optimization was thought more important than OPEX optimization. 171 Ammonia and acetaldehyde emissions due to oxidative and thermal degradation of the solvents have 172 been taken into account. In membranes separation, pressurized biogas is passed through a 173 membrane which is selective for CO₂. It is important to notice that depending on the chosen 174 upgrading technology, CH₄ content in the CO₂ flow strongly varies. A scenario without biogas 175 upgrading (direct methanation) is also assessed in this study. In all the cases previously described, a 176 supplementary step of compression of the outputs of the upgrading process is needed before the 177 methanation step. All the assumptions for biogas upgrading and compression steps are presented in 178 Table 3.

179

Table 3: Main assumptions at the biogas upgrading and compression steps

Description	Values	Reference	
Amines scrubbing and CO ₂ compression			
Inputs			
Biogas	230 m ³ h ⁻¹		

159

Electricity consumption	94.3 MJ.h ⁻¹	Present data and (Starr et al. 2012)		
Heat consumption	421 MJ.h ⁻¹	Present data		
Water	126 kg.h ⁻¹	Present data		
MonoEthanolAmines	11.5 g.h ⁻¹	Present data and (Reiter & Lindorfer		
		2015)		
Outputs				
CO₂outAmines	$80 \text{ m}^3.\text{h}^{-1}$	Present data		
	CO ₂ : 97.27%mol / H ₂ O: 2.73%mol			
	T: 150°C / P: 13.3 bar			
CH ₄ outAmines	156 m ³ .h ⁻¹	Present data		
	CH ₄ : 96.79%mol / CO ₂ : 0.98%mol / H ₂ O: 2.23%mol			
	T: 50°C / P: 5.6 bar			
Ammonia	16 g.h ⁻¹	(Reiter & Lindorfer 2015)		
Acetaldehyde	423 g.h ⁻¹	(Reiter & Lindorfer 2015)		
	Membranes and CO ₂ compression			
Inputs				
Biogas	$230 \text{ m}^{3}\text{h}^{-1}$			
Electricity consumption	152 MJ.h ⁻¹	Present data		
Outputs				
CO ₂ outMb	$95.1 \text{ m}^3.\text{h}^{-1}$	Present data		
	CO ₂ : 81.95%mol / CH ₄ : 18.05%mol			
	T: 150°C / P: 13.3 bar			
CH₄outMb	135 m ³ .h ⁻¹	Present data		
	CH ₄ : 99.15%mol / CO ₂ : 0.85%mol			
	T: 50°C / P: 17 bar			
Direct: Biogas compression				
Inputs				
Biogas	$230 \text{ m}^{3}\text{h}^{-1}$			
Electricity consumption	108 MJ.h ⁻¹	Present data		
Outputs				
BiogasCompToMetha	$225.5 \text{ m}^{3}\text{h}^{-1}$	Present data		
	CH ₄ : 65.63%mol / CO ₂ : 34.37%mol			
	T: 144°C / P: 15.3 bar			
BiogasComp	$4.5 \text{ m}^3 \text{h}^{-1}$	Present data		
	CH ₄ : 65.63%mol / CO ₂ : 34.37%mol			
	T: 144°C / P: 15.3 bar			

180 2.4 Electrolysis step

181 Water electrolysis from renewable energy is a sustainable method for H₂ production. It consists in splitting water into H₂ and oxygen (O₂). Inputs and outputs for this reaction are given in Table 4. Two 182 183 main technologies are available on the market: alkaline or polymer electrolyte membrane (PEM) 184 electrolysis. PEM water electrolysis is much more expensive, due to high capital cost of the cell stack 185 compared to alkaline liquid electrolyte water electrolysis (Leng et al. 2012). On the other hand, 186 intermittent operation of alkaline electrolysis is challenging (Reiter & Lindorfer 2015). Nevertheless 187 in the latest alkaline electrolysis technologies, operating on intermittent renewable electricity was 188 found to be durable (Graves et al. 2011). Consequently we chose alkaline electrolysis in this study.

189

Table 4: Main assumptions at the electrolysis step

Description	Values	Reference
Inputs		
Tap water	252 kg.h ⁻¹ T: 15°C / P: 1 bar	
Electricity consumption	5634 MJ.h ⁻¹	Present data based on DNV, (2013), Wang et al., (2014) and Alphea, (2015)

Outputs		
H ₂ fromElectrolysis	312 m ³ .h ⁻¹	Present data based on DNV, (2013), Wang et al.,
	T: 40°C / P: 15 bar	(2014) and Alphea, (2015)
O ₂ fromElectrolysis	156.5 m ³ .h ⁻¹	Present data based on DNV, (2013), Wang et al.,
	T: 40°C / P: 15 bar	(2014) and Alphea, (2015)

190 **2.5 Methanation step**

- 191 Methanation consists in synthetizing CH₄ from CO₂ and H₂ (Sabatier reaction):
- 192 4 $H_2 + CO_2 \rightarrow CH_4 + 2 H_2O$.

193 In this study, H_2 comes from water electrolysis, and CO_2 is supplied by two ways: as almost pure CO_2 194 from upgrading step (by amine scrubbing or membrane separation), or mixed with CH₄ in the direct 195 methanation scenario. Methanation can be done by chemical or biological catalysis (Burkhardt & 196 Busch 2013). In this study, fixed-bed technology is used, with chemical catalysts mainly composed of nickel (Ocampo et al. 2009; Wang et al. 2011). Methanation reaction is exothermic, with a heat 197 198 production of 0.165 MJ.mol⁻¹ of input CO₂ (Hoekman et al. 2010). Produced heat is used to 199 regenerate the solvent in biogas upgrading by amines scrubbing and / or to heat the anaerobic digesters. Required data are presented in Table 5, and are in line with recent work of Reiter and 200 201 Lindorfer (2015).



Table 5: Main assumptions at the methanation step

Description	Values	Reference
	CO ₂ from amines scrubbing	·
Inputs		
CO ₂ outAmines	$80 \text{ m}^3.\text{h}^{-1}$	
Electricity consumption	27.2 MJ.h ⁻¹	Present data
H ₂ fromElectrolysis	$310.2 \text{ m}^3.\text{h}^{-1}$	Present data
Outputs		
CH₄outMethaAmines	$80 \text{ m}^3.\text{h}^{-1}$	Present data
	CH ₄ : 96.56%mol / H ₂ : 1.69%mol / H ₂ O:	
	1.33%mol / CO ₂ : 0.80%mol	
	T: 50°C / P: 9.5 bar	
HeatoutMethaAmines	615.9 MJ.m ⁻³ CO₂outAmines	Present data
	CO ₂ from membranes	
Inputs		
CO ₂ outMb	95.1 m ³ .h ⁻¹	
Electricity consumption	33.3 MJ.h ⁻¹	Present data
H_2 from Electrolysis 312 m ³ .h ⁻¹		Present data
Outputs		
CH₄outMethaMb	98 m ³ .h ⁻¹	Present data
	CH4: 96.7%mol / H ₂ : 1.2%mol / H ₂ O:	
	1.1%mol / CO ₂ : 0.3%mol	
	T: 50°C / P: 9.5 bar	
HeatoutMethaMb	721 MJ.h ⁻¹	Present data
	Biogas	
Inputs		
BiogasCompToMetha	225.5 m3.h ⁻¹	
Electricity consumption	106 MJ.h ⁻¹	Present data
H ₂ fromElectrolysis 310.2 m ³ .h ⁻¹		Present data
Outputs		
CH₄outMethaBiogas	230.9 m ³ h ⁻¹	Present data
_	CH ₄ : 97.4%mol / H ₂ : 1.6%mol / H ₂ O:	
	1.3%mol / CO ₂ : 0.4%mol	

	T: 50°C / P: 9.5 bar	
HeatoutMethaBiogas	586.3 MJ.m ⁻³ Biogas	Present data

203 **2.6 Completion, injection and combustion steps**

- 204 Before injecting the bioCH₄ in the natural gas network, a completion step is necessary in order to
- bring the dew point to existing specifications (here 40°C and 5 bar). Electricity consumptions of this
- 206 operation for the three proposed scenarios are given in Table 6.
- 207

Table 6: Main assumptions at the completion step

Description	Values	Reference			
CH ₄ from CO ₂ amines					
Inputs					
CH ₄ outMethaAmines + CH ₄ outAmines	80 m ³ h ⁻¹ + 156 m ³ h ⁻¹				
Electricity consumption for completion	47.2 MJ.h ⁻¹	Present data			
Outputs					
CH₄CompletionAmines	232 m ³ h ⁻¹	Present data			
	CH ₄ : 98.54%mol / H ₂ : 0.58%mol /				
	CO ₂ : 0.80%mol / H ₂ O: 0.08%mol				
	T: 40°C / P: 5 bar				
	CH ₄ from CO ₂ membranes				
Inputs					
CH ₄ outMethaMb + CH ₄ outMb	98 m ³ h ⁻¹ + 135 m ³ h ⁻¹				
Electricity consumption for completion	37.3MJ.h ⁻¹	Present data			
Outputs					
CH₄CompletionMb	232 m ³ h ⁻¹	Present data			
	CH ₄ : 98.58%mol / H ₂ : 0.68%mol /				
	CO ₂ : 0.66%mol / H ₂ O: 0.08%mol				
	T: 40°C / P: 5 bar				
	CH₄ from Biogas				
Inputs					
CH ₄ outMethaBiogas + CH ₄ outBiogas	230.9 $\text{m}^{3}\text{h}^{-1}$ + 4.5 $\text{m}^{3}\text{h}^{-1}$				
Electricity consumption for completion	40 MJ.h ⁻¹	Present data			
Outputs					
CH₄CompletionBiogas	$233 \text{ m}^{3}\text{h}^{-1}$	Present data			
	CH ₄ : 97.8%mol / H ₂ : 1.15%mol / CO ₂ :				
	0.97%mol / H ₂ O: 0.08%mol				
	T: 40°C / P: 5 bar				

208

209 At last, data for injection and combustion have been directly taken from the Ecoinvent 2.2 database,

210 with biogenic CO₂ emissions instead of fossil CO₂ emissions. .

211 2.7 Economic data

- In Table 7, operating expenditure (OPEX) and capital expenditure (CAPEX) for all the processes are presented. OPEX corresponding here to electricity consumption can be calculated from previous tables, and for different electricity prices. CAPEX of each process (upgrading, electrolysis, methanation and injection) have been evaluated by IFPEN and are specific results of this study. The capital cost estimations have been assumed to be in the range of -30% / + 50%.
 Since the biogas production process has not been modeled from an economic point of view, the
- 218 biogas production price has been estimated. We have supposed that biomethane injection in the

- network allows an internal rate of return (IRR) equal to 10% on 15 years. Given the feed-in tariffs of
 bioCH4 fixed by decree in France in 2014
- 221 (<u>http://www.legifrance.gouv.fr/affichTexte.do?cidTexte=JORFTEXT000024833895</u>) and taking into
- account the costs of the final steps biogas upgrading and injection this leads to a biogas
- production cost equal to 0.36 €.m-3.
- 224

Table 7: Main assumptions for the CAPEX and the OPEX

Drococc		ODEX	CADEV	Poforoncoc
Process		UPEA	CAPEA	References
Biogas		0.36 €.m ⁻³ Biogas	-	
Electrolysis		Electricity consumption +	1000 €.kW ⁻¹	Present data based on DNV, (2013), ADEME, (2014),
		100 k€.year ⁻¹ for maintenance		IEA, (2015) and Götz et al. (2016)
Amines	Separation	Electricity consumption +	1050 k€, i.e	Present data, in line with 5000 €/(m ³ .h ⁻¹) (Bauer et
		0.8 €.hour ⁻¹	4570 €/(m ³ .h ⁻¹)	al., 2013 and TUV, 2012) for a capacity of 250 m^3 . h^{-1}
	Methanation	Electricity consumption +	400 k€	Present data, in line with Götz et al. (2016)
		6.2 €.hour ⁻¹		
Membranes	Separation	Electricity consumption +	750 k€, i.e	Present data, in line with 3200 €/(m ³ .h ⁻¹) (Bauer et
		1.7 €.hour ⁻¹	3260 €/(m ³ .h ⁻¹)	al., 2013) for a capacity of 250 $m^3.h^{-1}$
	Methanation	Electricity consumption +	650 k€	Present data, in line with Götz et al. (2016)
		12.4 €.hour ⁻¹		
Direct methanation		Electricity consumption +	650 k€	Present data, in line with Götz et al. (2016)
		14.1 €.hour ⁻¹		
Injection		100 k€.year ⁻¹	130 k€	Present data based on DNV, (2013), ADEME, (2014)
-				and IEA. (2015)

225 **3 Results**

226 3.1 Economic results

227 Continuous operation

Annualized production costs of can be seen on Figure 2, for an electricity price of 30 €.MWh⁻¹ and 228 229 with an IRR equal to 10% on 15 years. There is an important difference of total annualized production costs with and without methanation: approximatively 1650 k€.year⁻¹ for upgrading with PtG, and 940 230 k€.year⁻¹ for standard upgrading. The largest contribution to the production costs is the biogas 231 production, (around 80% for upgrading technologies and 45% for PtG scenarios). This cost is the 232 same in the five scenarios assessed (757 k€.year⁻¹), but as total production costs are much higher in 233 234 the methanation scenarios, the contribution of biogas is lower. The higher costs of the methanation 235 scenarios are mainly due to the electrolysis step. It represents more than 35% of the production 236 costs, with a large share due to electricity consumption (70% of the electrolysis costs). At last for PtG production with CH₄ and CO₂ separation, upgrading and methanation costs are quite similar. 237





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Figure 2: Contribution analysis of the annualized production costs

240 The contribution analysis of the production costs for 1 kWh of bioCH₄ from PtG or upgrading are 241 shown on Figure 3, for the same electricity price and IRR rate than the Figure 2. It appears that production costs of the different options are quite the same: they range from a minimum of 0.096 242 243 €.kWh⁻¹ for PtG from biogas with direct methanation and with membrane separation, and a maximum of 0.104 €.kWh⁻¹ for biogas upgrading with membrane separation. In fact, the higher 244 production cost of the PtG technologies are counterbalanced by a higher bioCH₄ injection in the gas 245 network (41 m³.h⁻¹ on average for standard upgrading versus 138 m³.h⁻¹ on average for methanation 246 247 scenarios). The respective contributions of the different steps are the same than for the annualized 248 production costs.

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Figure 3: Contribution analysis of the production costs for 1 kWh injected in the gas network

252 In Figure 4, overall production costs of the different scenarios are presented for electricity prices ranging from 0 to 45 €.MWh⁻¹. Costs of CH₄ from upgrading technologies are almost insensitive to 253 254 electricity price, as they are mainly driven by the fixed biogas cost, and because the only other 255 relevant OPEX corresponds to the injection step which is not correlated with the electricity price (see 256 Table 7). On the contrary, PtG scenarios are very sensitive to the electricity price, with an increase of 257 approximatively 35% of the production costs, mainly because of the electricity consumption at the electrolysis step. Compared to production of CH_4 with the cheaper upgrading technology (amines 258 scrubbing), CH₄ from PtG is competitive for electricity price below 27 €.MWh⁻¹ for separation with 259 amines scrubbing, below 34 €.MWh⁻¹ with separation by membranes and below 38 €.MWh⁻¹ for 260 261 direct methanation.



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263 Figure 4: Productions costs for 1 kWh injected in the gas network depending on the price of electricity

264 Intermittent operation

265 As previously pointed out, PtG can be seen as a relevant way to store overproduced electricity, and 266 then intermittent operation should also be analyzed. According to Gahleitner (2013) the use of such 267 fluctuating power sources is satisfying, and is currently in use in a lot of PtG pilot plants. Economic 268 results for operation time varying between 2000 and 8760 hours (continuous operation) have been assessed for PtG scenarios and CH₄ from amines scrubbing (Figure 5), with an electricity price equal 269 to 20 €.MWh⁻¹. Biogas upgrading is insensitive to the operation time. This can be explained by the 270 important contribution of the biogas production, and because others OPEX are almost exclusively 271 272 driven by the fixed cost of the rental of the injection station. On the opposite, costs of direct methanation are very strongly connected with the operation time, with costs between 0.330 €.kWh⁻¹ 273 for 2000 hours of operating time, and 0.090€.kWh⁻¹ for continuous operation. In fact in this 274 275 configuration CH₄ contained in the biogas is not recovered and injected in the gas grid when 276 methanation is not in operation. Consequently, the volume of energy produced is drastically reduced 277 with the increase of the intermittency. On the other hand, fixed costs (biogas production, all the 278 CAPEX and rental of the injection unit) are unchanged. Therefore the direct methanation leads to a 279 dramatic increase of the production costs with an intermittent functioning, and so this technology is 280 not suitable for non-continuous operation mode. Finally, like in the direct methanation, production 281 costs decrease for membranes and amines technologies of PtG with the increase of the operation, 282 but to a much lesser degree. Depending of the duration of the operation time, and for a given price 283 of electricity, production costs can be higher or lower than CH₄ from biogas upgrading. For instance, for a electricity at 20 €.MWh⁻¹, operation time should be at least equal to 6160 hours for membranes 284 285 separation and to 7240 hours for amines separation to be competitive with biogas upgrading.



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287 288 289

Figure 5: Productions costs for 1 kWh injected in the gas network depending on the technology and the operating time

Combined effect of electricity price and operation time have also been assessed (Figure 6). 290 291 Depending on the price of electricity, the required operation time to be competitive with biogas 292 upgrading can strongly vary. For instance, for a free electricity, operation time should be at least 293 equal to 4445 hours for membranes separation and to 5232 hours for amines separation to be 294 competitive with biogas upgrading. It means that the methanation unit should be in operation at 295 least almost 4500 hours to counterbalance the fixed costs and the CAPEX. At last, with an electricity price equal to 45 €.MWh⁻¹, PtG technologies cannot compete with biogas upgrading. We also clearly 296 297 see on Figure 6 that the operation cost difference between PtG from amines and from membranes 298 increase with the operation time.







Figure 6: Productions costs for 1 kWh injected in the gas network depending on the price of electricity and 301 the operating time

302 Sensitivity analysis

303 As it has been underlined in the contribution analysis, electrolysis and more precisely the electricity 304 consumption of this step, is one of the main contributor to the production costs of CH₄ from PtG. Thus, we conduct a sensitivity analysis on the electricity consumption for H₂ production. We discuss a 305 favorable scenario, with a reduction of 25% of the electricity consumption (*i.e* 3.75 kWh.m⁻³ H₂). This 306 value is in line with predicted values of future electricity consumptions around 3 kWh.m⁻³ H₂ 307 308 (Udagawa et al. 2007; Zeng & Zhang 2010). The evolution of production costs with low electricity consumption scenarios and for different electricity prices is shown on Figure 7. As expected, the 309 310 more electricity is expensive, the more electricity consumption reduction has an important effect on the diminution of the production costs. For a price of electricity of 10 €.MWh⁻¹, the decrease of the 311 production cost is only equal to 0.001 €.KWh⁻¹, but with an electricity at 45 €.MWh⁻¹, the reduction of 312 the production cost is equal to 0.007 €.kWh⁻¹. Consequently, the minimum price of electricity so that 313 CH₄ from PtG can be competitive with CH₄ from upgrading increases: it reaches 35 €.MWh⁻¹ for 314 amines scrubbing and is around 45 €.MWh⁻¹ for membranes separation. 315

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Figure 7: Productions costs for 1 kWh injected in the gas network depending on the price of electricity and the electricity consumption for H₂ production

321 3.2 Environmental impacts

322 Environmental impacts of the study have been assessed within the methodological framework of the 323 LCA based on a "cradle to grave" inventory of emissions and resources consumption. The considered functional unit in the study is one MJ produced by combustion of CH₄ in a boiler, based on the lower 324 heating value of CH₄ (50.0 MJ.kg⁻¹). As one of the main objective is to reduce GHG emissions in the 325 326 atmosphere with respect to fossil fuels, we therefore consider the climate change impact. It has been 327 assessed with the characterization factors provided by the fourth IPCC report for a temporal horizon 328 of 100 years (IPCC 2007). It should be underlined that the characterization factor of CH₄ in case of 329 methane loss has to be adapted. Since the carbon of the CH4 comes from the sewage sludge (and previously from the atmosphere), the characterization factor must be the one of biogenic CH_4 , which 330 331 is slightly below the one of the fossil CH_4 (22.3 kg CO_2 eq.kg⁻¹ vs. 25 kg CO_2 eq.kg⁻¹).

332 As recommended in The Renewable Energy Directive (Parliament 2009), upstream impacts of sewage 333 sludge are not taken into account . The same approach has been applied on LCA of sewage sludge 334 valorization (Sadhukhan 2014; Yoshida et al. 2013). The substitution method has been used for digestates accounting, in accordance with the ISO guidelines, which suggest to choose the 335 336 substitution instead of the allocation when it is possible. It corresponds to an expansion of the 337 system boundaries in order to consider the impacts engendered by the by-products. Therefore, the use of digestates as fertilizers are supposed to save the equivalent amount of mineral fertilizers for 338 339 N, P and K. In Figure 8, Figure 9 and Figure 11, the legend "digestates substitution" corresponds to this environmental credit, and "digestates emissions" to the emissions at the biogas plant and at thefield.

342 Continuous operation

343 Principal contributors to GHG emissions for the different assessed technologies are shown on Figure 344 8. From a global perspective, GHG emissions from PtG are almost equivalent for the three technologies, and higher than biogas upgrading. Nevertheless, they are still lower than GHG 345 346 emissions from natural gas combustion. For the fossil scenario (natural gas burned in a boiler), most 347 of the emissions correspond to fossil CO₂ from gas combustion. The remaining emissions are linked 348 to resource extraction and transportation of the gas in the pipelines. In the scenarios with production 349 of energy from sewage sludge, main GHG emissions come from the management of the digestates. 350 Environmental burdens are associated with loss of CH_4 (40%) and emissions of N_2O (60%). On the 351 other hand, negative impacts are due to the substitution of the digestates to the production and the 352 use of mineral fertilizers. For PtG technologies, electricity consumption is also an important 353 contributor to climate change. As the carbon content of the French electricity is particularly low due 354 to the large part of the nuclear in the mix composition, a sensitivity analysis on the source of 355 electricity (and as a consequence on its carbon content) is done in a following paragraph.



356

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Figure 8: Contribution analysis of GHG emissions for 1 MJ injected in the gas network

358 Intermittent operation

According to simulations based on ADEME 2030 – 2050 scenarios (ADEME 2012), hours of electricity overproduction occurrence was estimated at 28% of the year for 2020, 36% in 2030 and 41% in 2050, which is consistent with the ADEME's study on methanation (ADEME 2014). The composition of electricity surplus is 80% renewable and 20% nuclear by 2050. Figure 9 presents results for intermittent methanation during overproduction period. It is supposed that when methanation is not functioning, the non-injected gas (CO_2 with 18% of CH_4 in membrane separation and biogas with 65% of CH_4 in direct methanation) is flared, in order to reduce direct CH_4 emissions to the atmosphere. In this scenario, the results for membrane and amines separation are almost the same, so only amine technology will be represented for the next environmental impact assessment results.

In direct technology configuration, biogas is not recovered and injected in the gas grid when methanation is not in operation, like in the economic assessment. This leads to GHG emissions which are much higher than continuous operation (and even higher than natural gas combustion) for the three time horizons analyzed. On the contrary, intermittent operation with electricity from renewable resources and amines separation leads to an important decrease compared to continuous operation with French average electricity mix (-37% in 2020 and -40% in 2050). For the three scenarios assessed, climate change impact of PtG is even lower than biogas upgrading.





Figure 9: GHG emissions for 1 MJ injected in the gas network depending on the price of electricity and the operating time

378 Sensitivity analysis

379 Change

Change in the electricity mix composition

Results of using two different types of electricity are presented on Figure 10 (French electricity mix and European electricity mix). The results indicate that amines PtG technology are more sensitive to the composition of the electricity than biogas upgrading. The change from French electricity mix to European electricity mix induces an increase of 290% of the GHG emissions of PtG technology, and 384 only 70% of the emissions from biogas upgrading. This is due to the fact that PtG technologies 385 consume much more electricity than upgrading technologies, combined with the different carbon 386 contents of the electricity mix. In fact GHG emissions of 1 MJ of European electricity is 0.156 CO_2eq , 387 and only 0.029 CO₂eq for French electricity, because French mix is mainly composed of nuclear electricity (78% according to the Ecoinvent database), with very low GHG emissions (0.00172 388 389 $CO_2eq.MJ^{-1}$). Consequently, with the use of a European electricity mix (or with another average 390 electricity mix with high GHG emissions, like China or United Kingdom), climate change impact of PtG 391 technology becomes higher than natural gas, with the highest contribution coming from the 392 electricity consumption. Thus electricity with low carbon content is a mandatory prerequisite to 393 achieve a sustainable production of CH₄ from CO₂.



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Figure 10: Contribution analysis of GHG emissions for 1 MJ injected in the gas network depending on the electricity mix

As pointed out by Reiter and Lindorfer, (2015), other impacts than climate change have never been assessed for biomethane produced from PtG. In this article, endpoint impacts have been assessed with the ReCiPe method with normalization values for Europe and the Hierarchist perspective (Goedkoop et al. 2009). This perspective is based on the most common policy principles with regard to time frame and other issues. The time horizon of the selected impact is in this case 100 years. The three assessed endpoint impacts are:

404 - Human health, defined as the sum of years of life lost and years of life disabled, and
405 expressed in years.

- Ecosystem diversity, based on the loss of species during a certain time in a certain area as,
 and expressed as the potentially disappeared fraction of species (PDF) integrated over area
 (PDF x year).
- 409 410

- Resources availability, corresponding to the marginal increase in costs due to the extraction of a resource, and expressed in \$.

This method is easy to interpret because endpoints are directly associated with an area of protection, unlike midpoints impacts (Reap et al. 2008). On the other hand, it should be noticed that their level of uncertainty is higher than midpoints indicators (Bare et al. 2000). Results are shown on Figure 11, and each impact is standardized with the value of the worst scenario for this impact.

As for climate change, impacts of PtG are lower than the ones of natural gas, but higher than biogas upgrading for continuous operation. Contributors of human health and ecosystem impacts are quite the same than for climate change. For resources impact, contributors associated with pollutants emissions are not in the scope of the impact calculation. This leads to negative impact for bioenergy production, as only resource consumption for electricity generation and positive effect of mineral substitution of the mineral fertilizers by the digestates are taken into account.





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Figure 11: Contribution analysis of endpoints impacts (Human Health, Ecosystem and Resources) for 1 MJ injected in the gas network

424 **4 Discussion**

425 It must be underlined that this is a prospective LCA of a process that does not yet exist. Consequently 426 the systems described in this study can be subjected to important modifications. From a technical 427 point of view, some changes could be proposed to improve the economic and environmental balance 428 of this process. As biogas is used to heat the anaerobic digesters, the CO_2 resulting from the 429 combustion of this biogas can also be converted into bioCH₄ by methanation. The auto-consumption 430 of the biogas ranges from 21% (membrane upgrading) to 28% (amines upgrading), so the added 431 amount of bioCH₄ that could be injected in the gas network is far from being marginal, and could 432 potentially improve the overall performances of the process. About economic assumptions, biogas 433 production costs are based in this paper on French feed-in tariffs and an expected IRR. These values may seem conservative, compared to data from literature (between 0.24 and 0.30 €.m⁻³ of biogas in 434 435 Patterson et al. (2011)). In the future, a more detailed approach could be investigated to consolidate 436 biogas production costs, for instance by including the biogas production process in the economic 437 assessment. However, as the same biogas production costs have been used for upgrading and PtG 438 technologies, it should be underlined that this inclusion would not change the general conclusion of 439 the economic study. It also has to be noticed that electricity price is strongly correlated with the 440 annual operation hour: average electricity price increases with the load factor (ENEA, 2016). In our 441 study, different electricity prices are proposed for different operation time. This has been done to 442 assess how the analyzed system responds to these two key parameters independently. Further work 443 could therefore integrate price duration curves for different locations and different time horizons in 444 order to map out more clearly potential prospective scenarios. To go deeper into details about the 445 economic feasibility of the combination of anaerobic digestion and PtG technology, the effect of the 446 introduction of an incentive for "green gas" production could be assessed. As underlined by Guandalini et al. (2015), there is a need of specific incentive for PtG technologies in order to push the 447 448 competitiveness of this energy production system. They propose a "green gas" incentive between 0 and 20 €.MWh⁻¹. As this incentive is only related to the CH₄ from CO₂ conversion by methanation, its 449 450 amount is function of the proportion of $bioCH_4$ from methanation. In our system, CO_2 conversion into 451 bioCH₄ represent approximatively 50% of the total injected bioCH₄ for the amines scrubbing and the 452 membrane upgrading technologies. Applied to the case study described in section 3.1 (continuous operation and an electricity price of 30 €.MWh⁻¹), an incentive of 20 €.MWh⁻¹ leads to a reduction of 453 454 the production costs of almost 10%. At last learning curves (or experience curves) could be used to 455 quantify the future reduction of investment costs for the different assessed technologies. Progress 456 ratios (PR, defined as the rates of unit cost decline with each doubling of cumulative production) are 457 expected to be around 0.88 (Junginger et al., 2006) for biogas plant and 0.82 for H_2 production by 458 water electrolysis (Schoots et al., 2008). It is in line with Anandarajah et al. (2013) who report PR 459 around 15-20% for new technologies. Other authors suggest more globally that investments costs 460 could be divided by 2 in 20 years for thermochemical processes and in 30 years for biochemical 461 processes (due to more important sources of process locks for biological manipulations). From the 462 environmental point of view, conversion of sewage sludge into bioCH₄, using methanation to valorize the CO_2 of the biogas, can also be seen as a waste treatment process. Many LCAs have been 463 464 conducted in the past years to assess the environmental impacts of sewage sludge management 465 (Hospido et al. 2010; Sadhukhan 2014; Yoshida et al. 2013). A comparison with the most used 466 technology of sewage sludge valorization in France, i.e the production of electricity and heat by 467 cogeneration, could therefore be an interesting perspective.

468 **5** Conclusion

In this study a techno economic and environmental assessment of bioCH₄ production via biogas 469 470 upgrading and PtG technology has been performed. Five different scenarios have been analyzed: two 471 biogas upgrading scenarios (with amines and membrane technologies), and three scenarios with the 472 use of PtG technologies to produce bioCH4. These scenarios have been compared to a fossil 473 reference scenario of methane production from natural gas. Sensitivity analyzes have been done, 474 both on the economic and environmental assessments. From the economic point of view, PtG 475 technologies are competitive with upgrading ones for an average electricity price equal to 38 €.MWh 476 ¹ for direct methanation and separation by membranes. The lower competitiveness of amines 477 scrubbing is due to a lower volume of bioCH₄ produced, mainly because of the biogas auto-478 consumption to produce heat for the amines regeneration. For intermittent operation, competitive 479 prices of electricity for methanation can be higher than the upgrading scenarios, depending on the 480 operation time. It should be pointed out that direct methanation cannot be used in intermittent 481 operation, as the produced energy is drastically reduced because of the biogas lost when 482 methanation is not functioning. A reduction of 25% of the electricity consumption at the electrolysis 483 step causes an important decrease of the production costs, especially for scenarios with high 484 average electricity prices. From an environmental perspective, impacts of PtG are higher than biogas 485 upgrading for continuous operation. Intermittent operation can severely lower the climate change 486 impact for non-direct methanation scenarios, as the consumed electricity is mainly based on 487 renewable sources. The change in the electricity mix composition can also have a strong influence on 488 GHG emissions. This work is a first attempt to assess both the economic and the environmental 489 impacts of PtG technologies, and highlights the main bottlenecks in this production. Here we focus 490 on a simplified process where only CO₂ from biogas was converted into BioCH₄, but the optimum

- 491 from both environmental and economic points of view may consist in a more complex process where
- 492 CO₂ from internal biogas combustion should also be turned into bioCH₄.

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494 References

- 495 ADEME, 2014. Etude portant sur l'hydrogène et la méthanation comme procédé de valorisation de l'électricité
 496 excédentaire., pp.1 –238.
- 497 ADEME, 2011. Qualité agronomique et sanitaire des digestats. , pp.1–250.
- 498 ADEME, 2015. Vers un mix électrique 100% renouvelable en 2050. , pp.1–119.
- 499 ADEME, 2012. Vision 2030 2050 : L'exercice de prospective de l'ADEME. , pp.1–297.
- Alphea, 2015. Fabricants d'électrolyseurs alcalins pour la production d'hydrogène : caractéristiques des
 produits et informations sur les constructeurs.
- Amon, B. et al., 2006. Methane, nitrous oxide and ammonia emissions during storage and after application of
 dairy cattle slurry and influence of slurry treatment. *Agriculture, Ecosystems and Environment*, 112(2-3),
 pp.153–162.
- Amor, M. Ben et al., 2014. Implications of integrating electricity supply dynamics into life cycle assessment: A
 case study of renewable distributed generation. *Renewable Energy*, 69, pp.410–419. Available at:
 http://linkinghub.elsevier.com/retrieve/pii/S0960148114002389 [Accessed July 25, 2014].
- 508 Anandarajah, G. et al., 2013. Decarbonising road transport with hydrogen and electricity: Long term global
- technology learning scenarios. *International Journal of Hydrogen Energy*, 38(8), pp. 3419–3432.
- Bare, J.C. et al., 2000. Midpoints versus endpoints: The sacrifices and benefits. *The International Journal of Life Cycle Assessment*, 5(6), pp.319–326.
- Bauer, F. et al., 2013. Biogas upgrading Review of commercial technologies. (SGC Rapport; Vol. 270). Svenskt
 Gastekniskt Center AB, pp. 1-84.
- de Boer, H.S. et al., 2014. The application of power-to-gas, pumped hydro storage and compressed air energy
 storage in an electricity system at different wind power penetration levels. *Energy*, 72, pp.360–370.
- Burkhardt, M. & Busch, G., 2013. Methanation of hydrogen and carbon dioxide. *Applied Energy*, 111, pp.74–79.
 Available at: http://linkinghub.elsevier.com/retrieve/pii/S030626191300384X [Accessed November 6, 2013].
- Cetinkaya, E., Dincer, I. & Naterer, G.F., 2012. Life cycle assessment of various hydrogen production methods.
 International Journal of Hydrogen Energy, 37(3), pp.2071–2080. Available at: http://linkinghub.elsevier.com/retrieve/pii/S036031991102430X [Accessed June 2, 2014].
- 522 DGEC (Direction générale de l'énergie et du climat), 2011. Sommaire Rapport sur l'industrie pétrolière et 523 gazière en 2011. , pp.1–85.
- 524 DNV Kema, 2013. System Analysis Power to Gas Deliverable 1 : Technology Review. , pp.1–70.
- 525 Dufour, J. et al., 2012. Life cycle assessment of alternatives for hydrogen production from renewable and fossil
 526 sources. International Journal of Hydrogen Energy, 37(2), pp.1173–1183. Available at:
 527 http://linkinghub.elsevier.com/retrieve/pii/S0360319911022828 [Accessed June 10, 2014].
- Dufour, J. et al., 2009. Life cycle assessment of processes for hydrogen production. Environmental feasibility
 and reduction of greenhouse gases emissions. *International Journal of Hydrogen Energy*, 34(3), pp.1370–
 1376. Available at: http://linkinghub.elsevier.com/retrieve/pii/S0360319908016078 [Accessed May 28,
 2014].
- Ekvall, T. & Weidema, B.P., 2004. System boundaries and input data in consequential life cycle inventory
 analysis. *The International Journal of Life Cycle Assessment*, 9(3), pp.161–171. Available at:
 http://www.springerlink.com/index/10.1007/BF02994190 [Accessed August 30, 2012].
- 535 ENEA, 2016. The potential of power of to gas, pp. 1-51.
- 536 European Parliament, 2011. Energy Roadmap 2050.
- Foley, J. et al., 2010. Comprehensive life cycle inventories of alternative wastewater treatment systems. *Water research*, 44(5), pp.1654–66. Available at: http://www.ncbi.nlm.nih.gov/pubmed/20022351 [Accessed
 July 11, 2014].
- Frank, E.D. et al., 2012. Methane and nitrous oxide emissions affect the life-cycle analysis of algal biofuels.
 Environmental Research Letters, 7(1), p.014030. Available at: http://stacks.iop.org/1748-9326/7/i=1/a=014030?key=crossref.ae62cbb3318881630d5db3d9be522465 [Accessed May 28, 2013].
- 543 Frischknecht, R. et al., 2007. Overview and Methodology, , final report ecoinvent v2.0 No. 1, Swiss Centre for 544 Life Cycle Inventories.,

- 545 Gahleitner, G., 2013. Hydrogen from renewable electricity: An international review of power-to-gas pilot plants
 546 for stationary applications. *International Journal of Hydrogen Energy*, 38(5), pp.2039–2061. Available at:
 547 http://dx.doi.org/10.1016/j.ijhydene.2012.12.010.
- 548 Goedkoop, M. et al., 2009. ReCiPe 2008.
- Götz, M. et al., 2016. Renewable Power-to-Gas: A technological and economic review. *Renewable Energy*, 85, pp.1371–1390.
- Graves, C. et al., 2011. Sustainable hydrocarbon fuels by recycling CO2 and H 20 with renewable or nuclear
 energy. *Renewable and Sustainable Energy Reviews*, 15(1), pp.1–23. Available at:
 http://dx.doi.org/10.1016/j.rser.2010.07.014.
- Guandalini, G., Campanari, S. & Romano, M.C., 2015. Power-to-gas plants and gas turbines for improved wind
 energy dispatchability : Energy and economic assessment. *Applied Energy*, 147, pp.117–130. Available at:
 http://dx.doi.org/10.1016/j.apenergy.2015.02.055.
- Hadjipaschalis, I., Poullikkas, A. & Efthimiou, V., 2009. Overview of current and future energy storage
 technologies for electric power applications. *Renewable and Sustainable Energy Reviews*, 13(6-7),
 pp.1513–1522.
- Hoekman, S.K. et al., 2010. CO2 recycling by reaction with renewably-generated hydrogen. *International Journal of Greenhouse Gas Control*, 4(1), pp.44–50. Available at:
 http://linkinghub.elsevier.com/retrieve/pii/S1750583609001005 [Accessed November 8, 2013].
- Hospido, A. et al., 2010. Environmental assessment of anaerobically digested sludge reuse in agriculture:
 potential impacts of emerging micropollutants. *Water research*, 44(10), pp.3225–33. Available at:
 http://www.ncbi.nlm.nih.gov/pubmed/20347114 [Accessed July 13, 2014].
- 566 IEA, 2014. Biogas Country Report. , pp.1–50.
- 567 IEA, 2009. Biogas upgrading technologies developments and innovations. , pp.1 20.
- 568 IEA, 2015. Technology roadmap hydrogen and fuel cells. , pp.1 81.
- IPCC, 2007. Climate Change: The Physical Science Basis. Contribution of Working Group I to the Fourth
 Assessment Report of the Intergovernmental Panel on Climate Change. Cambridge University Press,
 Cambridge, UK and New York, NY, USA., pp.1 1007.
- 572 IPCC, 2011. Renewable Energy Sources and Climate Change Mitigation, Special Report of the
 573 Intergovernmental Panel on Climate Change., pp.1 1088.
- ISO, 2006. 14044:2006 Environmental management—life cycle assessment—requirements and guidelines.
 International Standards Organization.
- Jentsch, M., Trost, T. & Sterner, M., 2011. Evaluation of power-to-gas as long-term storage concept regarding
 electricity and CO 2 sources.
- 578 Junginger, M. et al., 2006. Technological learning in bioenergy systems. *Energy Policy*, 34, pp. 4024–4041.
- Kaparaju, P., 2013. Biogas Upgrading Scenarios in Europe Status & Prospects. In International Workshop on
 Promotion of Biogas Upgrading and Bottling in India and EU August 22 24, 2013, IIT Delhi.
- Leng, Y. et al., 2012. Solid-state water electrolysis with an alkaline membrane. *Journal of the American Chemical Society*, 134(22), pp.9054–9057.
- Makaruk, A., Miltner, M. & Harasek, M., 2010. Membrane biogas upgrading processes for the production of
 natural gas substitute. *Separation and Purification Technology*, 74(1), pp.83–92. Available at:
 http://dx.doi.org/10.1016/j.seppur.2010.05.010.
- Mohseni, F. et al., 2012. Biogas from renewable electricity Increasing a climate neutral fuel supply. *Applied Energy*, 90(1), pp.11–16. Available at: http://linkinghub.elsevier.com/retrieve/pii/S0306261911004697
 [Accessed November 6, 2013].
- Mori, M. et al., 2014. Life-cycle assessment of a hydrogen-based uninterruptible power supply system using
 renewable energy. *The International Journal of Life Cycle Assessment*. Available at:
 http://link.springer.com/10.1007/s11367-014-0790-6 [Accessed September 1, 2014].
- Nistor, S. et al., 2016. Technical and economic analysis of hydrogen refuelling. *Applied Energy*, 167, pp.211–
 220.
- 594 Ocampo, F., Louis, B. & Roger, A.-C., 2009. Methanation of carbon dioxide over nickel-based Ce0.72Zr0.2802
 595 mixed oxide catalysts prepared by sol-gel method. *Applied Catalysis A: General*, 369(1-2), pp.90–96.
- Parliament, E., 2009. Directive 2009/28/EC of the European Parliament and of the Council of 23 April 2009 on
 the promotion of the use of energy from renewable sources.
- Patterson, T. et al., 2011. An evaluation of the policy and techno-economic factors affecting the potential for
 biogas upgrading for transport fuel use in the UK. *Energy Policy*, 39(3), pp.1806–1816. Available at:
 http://dx.doi.org/10.1016/j.enpol.2011.01.017.
- 601 Patyk, A., Bachmann, T.M. & Brisse, A., 2013. Life cycle assessment of H2 generation with high temperature

- 602 electrolysis. International Journal of Hydrogen Energy, 38(10), pp.3865–3880. Available at:
 603 http://linkinghub.elsevier.com/retrieve/pii/S0360319913001493 [Accessed November 8, 2013].
- Raynal, L. et al., 2011. From MEA to demixing solvents and future steps, a roadmap for lowering the cost of
 post-combustion carbon capture. *Chemical Engineering Journal*, 171(3), pp.742–752. Available at:
 http://dx.doi.org/10.1016/j.cej.2011.01.008.
- Reap, J. et al., 2008. A survey of unresolved problems in life cycle assessment. *The International Journal of Life Cycle Assessment*, 13(4), pp.290–300. Available at: http://www.springerlink.com/index/10.1007/s11367 008-0008-x [Accessed October 26, 2012].
- Reiter, G. & Lindorfer, J., 2015. Global warming potential of hydrogen and methane production from renewable
 electricity via power-to-gas technology. *The International Journal of Life Cycle Assessment*. Available at:
 http://link.springer.com/10.1007/s11367-015-0848-0 [Accessed January 22, 2015].
- 613 Rivarolo, T. et al., 2016. Hydrogen and methane generation from large hydraulic plant: Thermo-economic 614 multi-level time-dependent optimization. *Applied Energy*, 113, pp.1737–1745.
- Ryckebosch, E., Drouillon, M. & Vervaeren, H., 2011. Techniques for transformation of biogas to biomethane. *Biomass* and *Bioenergy*, 35(5), pp.1633–1645. Available at:
 http://dx.doi.org/10.1016/j.biombioe.2011.02.033.
- Sadhukhan, J., 2014. Distributed and micro-generation from biogas and agricultural application of sewage
 sludge: Comparative environmental performance analysis using life cycle approaches. *Applied Energy*,
 122, pp.196–206. Available at: http://linkinghub.elsevier.com/retrieve/pii/S0306261914000804
 [Accessed September 3, 2014].
- Schoots, K. et al., 2008. Learning curves for hydrogen production technology: An assessment of observed cost
 reductions. *International Journal of Hydrogen Energy*, 33(11), pp. 2630–2645.
- Starr, K. et al., 2012. Life cycle assessment of biogas upgrading technologies. *Waste management (New York, N.Y.)*, 32(5), pp.991–9. Available at: http://www.ncbi.nlm.nih.gov/pubmed/22230660 [Accessed July 24,
 2012].
- Steeneveldt, R., Berger, B. & Torp, T.A., 2006. Closing the Knowing Doing Gap. *Chemical Engineering Research and Design*, 84(9), pp.739–769.
- Sterner, M., 2009. Bioenergy and renewable power methane in integrated 100% renewable energy systems. ,
 pp.1–230.
- Turconi, R. et al., 2014. Environmental impacts of future low-carbon electricity systems: Detailed life cycle
 assessment of a Danish case study. *Applied Energy*, 132(x), pp.66–73. Available at:
 http://linkinghub.elsevier.com/retrieve/pii/S0306261914006606 [Accessed July 23, 2014].
- Turconi, R., Boldrin, A. & Astrup, T., 2013. Life cycle assessment (LCA) of electricity generation technologies:
 Overview, comparability and limitations. *Renewable and Sustainable Energy Reviews*, 28, pp.555–565.
 Available at: http://dx.doi.org/10.1016/j.rser.2013.08.013.
- 637 TUV, 2012. Biogas to biomethane technology review, pp. 1-15.
- Udagawa, J., Aguiar, P. & Brandon, N.P., 2007. Hydrogen production through steam electrolysis: Model-based
 steady state performance of a cathode-supported intermediate temperature solid oxide electrolysis cell.
 Journal of Power Sources, 166(1), pp.127–136.
- Varun, Bhat, I.K. & Prakash, R., 2009. LCA of renewable energy for electricity generation systems-A review.
 Renewable and Sustainable Energy Reviews, 13(5), pp.1067–1073.
- de Vries, B.J.M., van Vuuren, D.P. & Hoogwijk, M.M., 2007. Renewable energy sources: Their global potential
 for the first-half of the 21st century at a global level: An integrated approach. *Energy Policy*, 35(4),
 pp.2590–2610.
- Wang, M. et al., 2014. The intensification technologies to water electrolysis for hydrogen production A
 review. *Renewable and Sustainable Energy Reviews*, 29, pp.573–588. Available at:
 http://dx.doi.org/10.1016/j.rser.2013.08.090.
- Wang, W. et al., 2011. Recent advances in catalytic hydrogenation of carbon dioxide. *Chemical Society reviews*,
 40(7), pp.3703–3727.
- 651 Yan, J. 2016. Integrated Energy Systems. *Applied Energy*, 167, pp.155–157.
- 652Yoshida, H., Christensen, T.H. & Scheutz, C., 2013. Life cycle assessment of sewage sludge management: A653review. Waste management & research : the journal of the International Solid Wastes and Public654CleansingAssociation,ISWA,31(11),pp.1083–101.Availableat:655http://www.ncbi.nlm.nih.gov/pubmed/24061046 [Accessed November 14, 2013].
- Zeng, K. & Zhang, D., 2010. Recent progress in alkaline water electrolysis for hydrogen production and
 applications. *Progress in Energy and Combustion Science*, 36(3), pp.307–326. Available at:
 http://dx.doi.org/10.1016/j.pecs.2009.11.002.

659Zoss, T. et al., 2016. Modeling a power-to-renewable methane system for an assessment of power grid660balancing options in the Baltic States' region. Applied Energy, 170(x), pp.278–285.