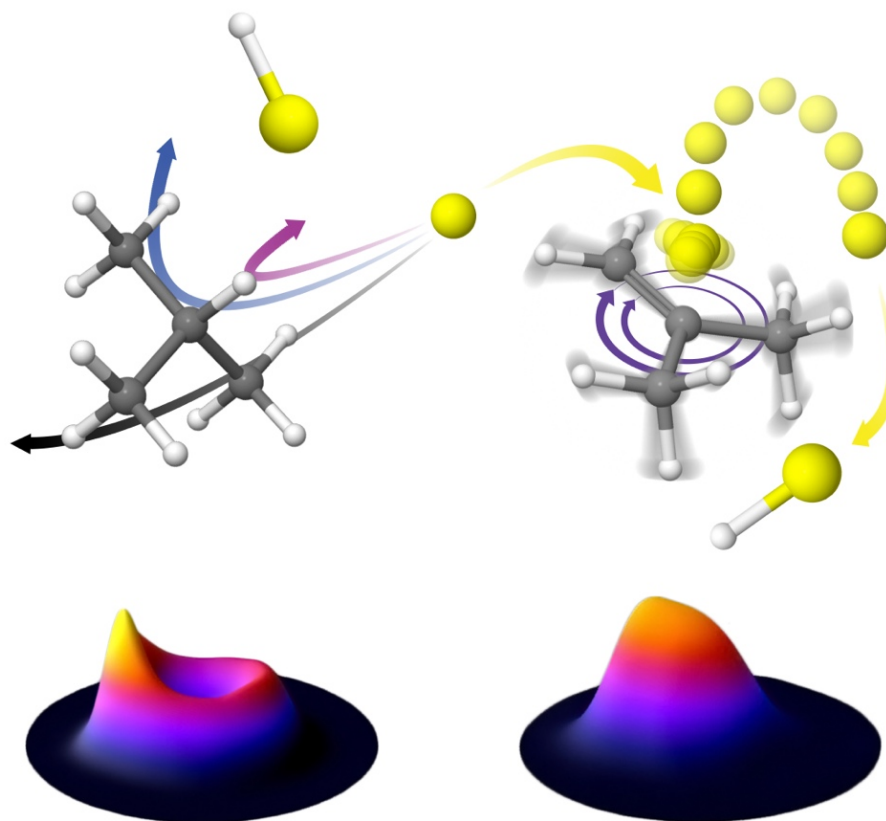


Crossed-Beam Slice Imaging of Cl Reaction Dynamics with Butene Isomers



¹ WAYNE STATE UNIVERSITY

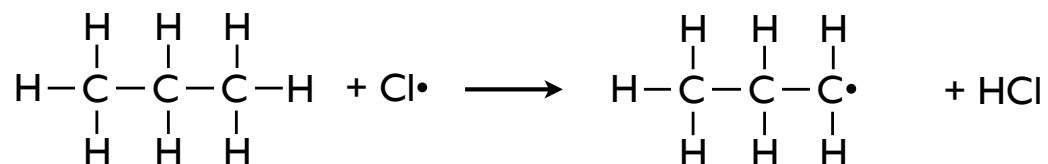
Baptiste Joalland¹

²  **FIU**
FLORIDA INTERNATIONAL UNIVERSITY

Yuanyuan Shi¹, Nitin Patel¹, Richard Van Camp¹, Alexander M. Mebel², Arthur G. Suits¹

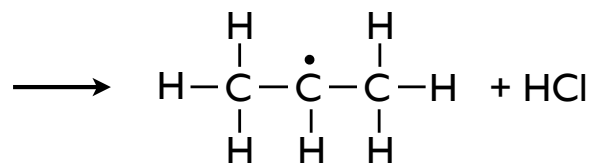
Crossed-beam and Imaging

Measuring differential cross sections: the case of the Cl+propane reaction



primary abstraction

-2 kcal/mol

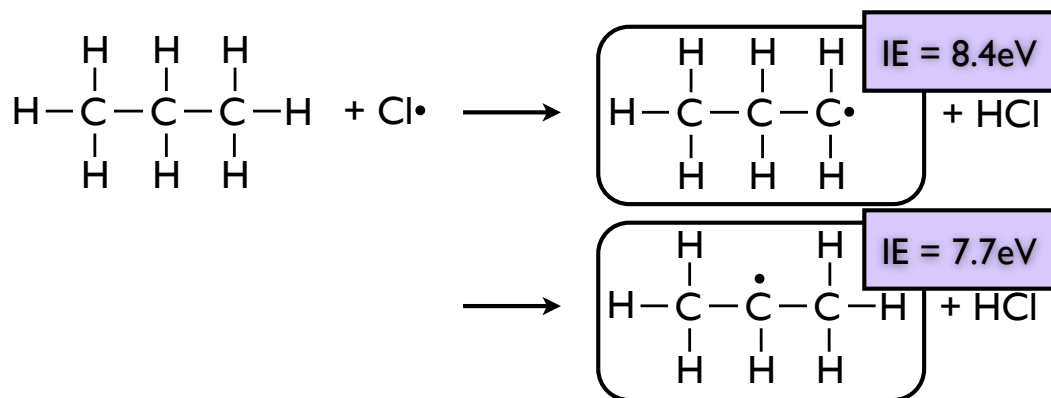


secondary abstraction

-6 kcal/mol

Crossed-beam and Imaging

Measuring differential cross sections: the case of the Cl+propane reaction



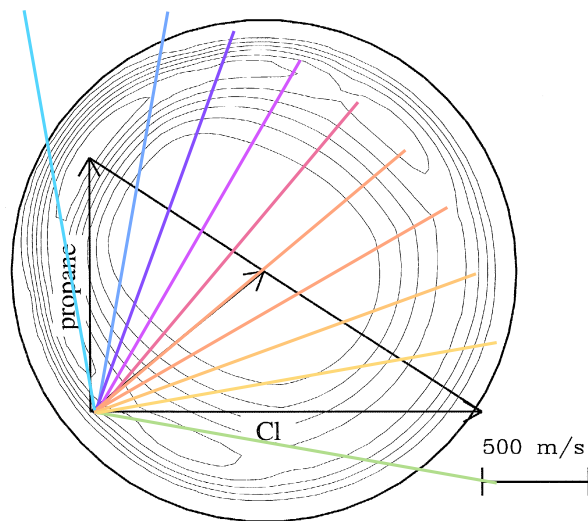
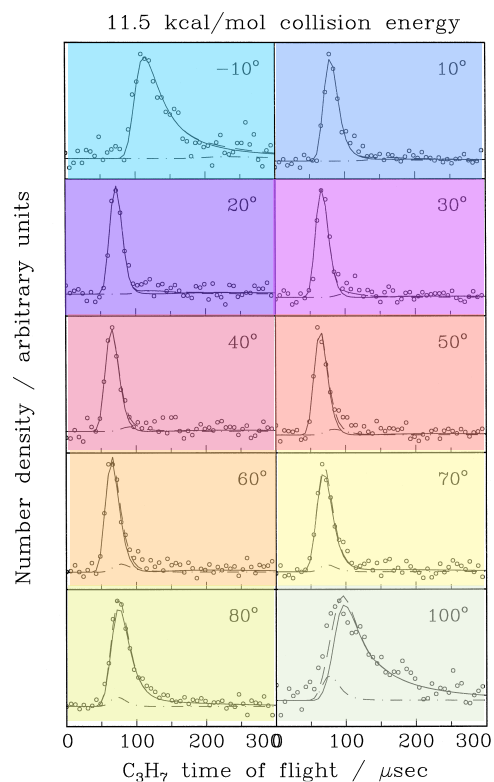
primary abstraction

-2 kcal/mol

secondary abstraction

-6 kcal/mol

Probing the radical product



DCSs = convolution fits from laboratory measurements

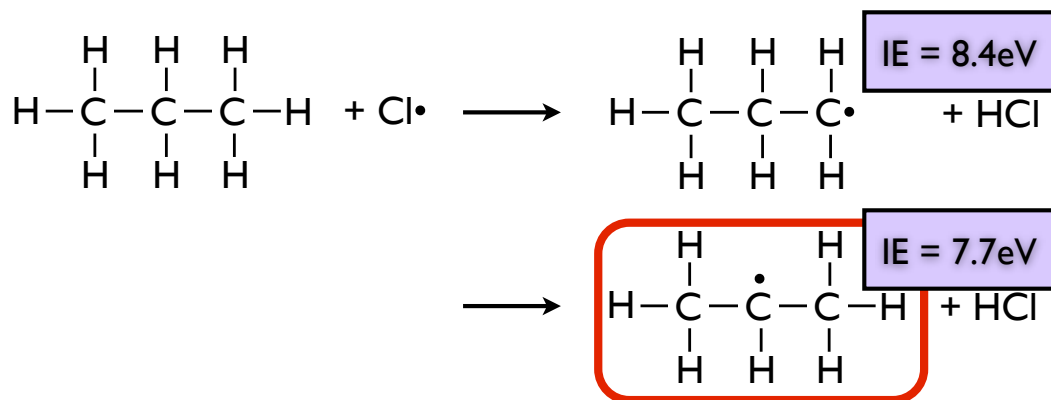
Probe: $h\nu = 9.5\text{eV}$

\Downarrow

No disentanglement between competing abstraction sites

Crossed-beam and Imaging

Measuring differential cross sections: the case of the Cl+propane reaction



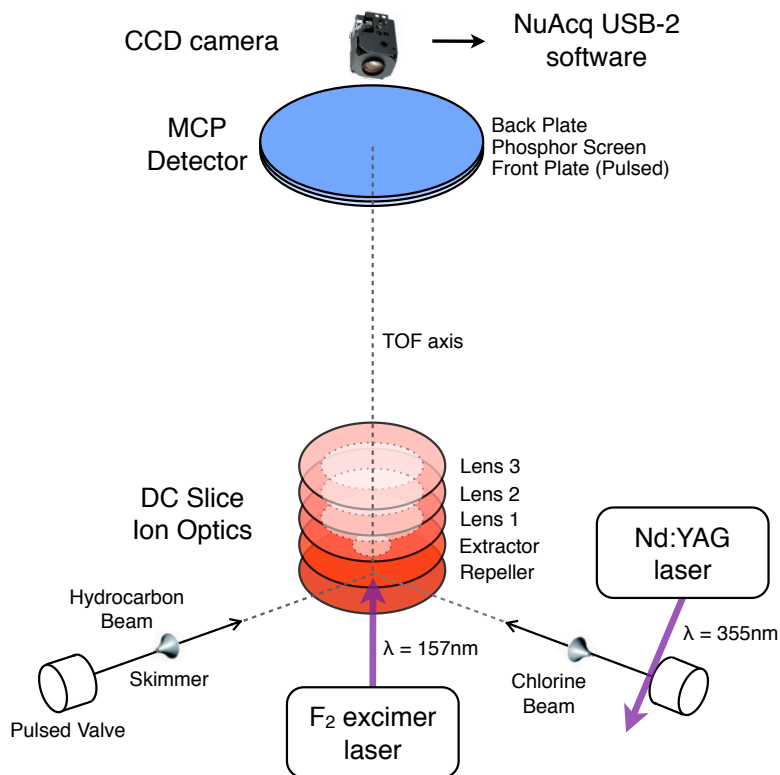
primary abstraction

-2 kcal/mol

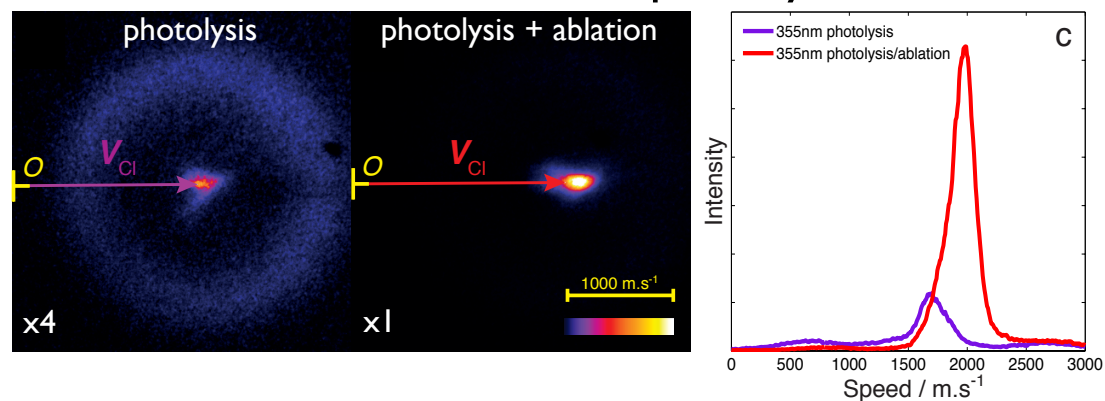
secondary abstraction

-6 kcal/mol

Probing the radical product



Chlorine atom source: 355nm photolysis + ablation

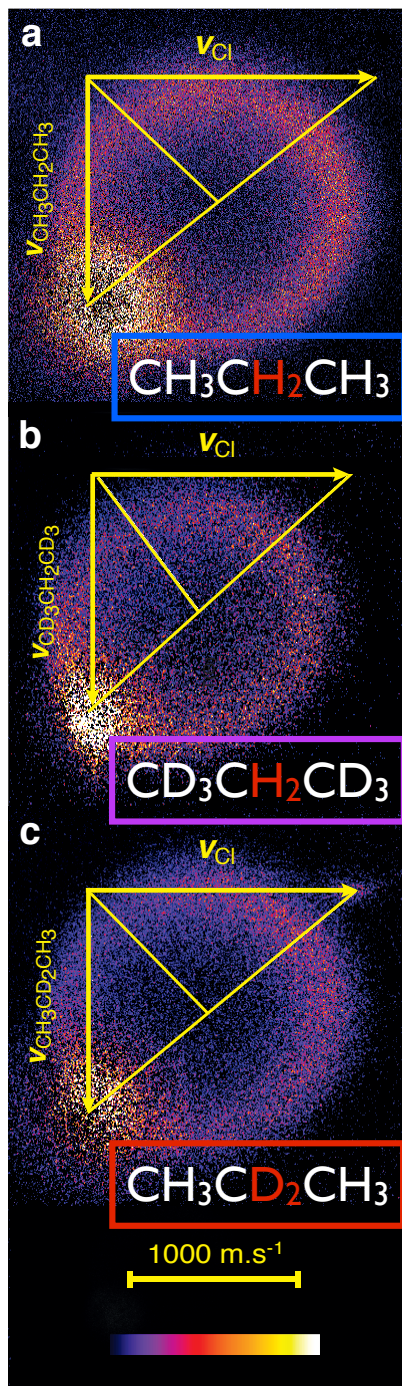


Probe: $h\nu = 7.9\text{eV}$ & Unfocused

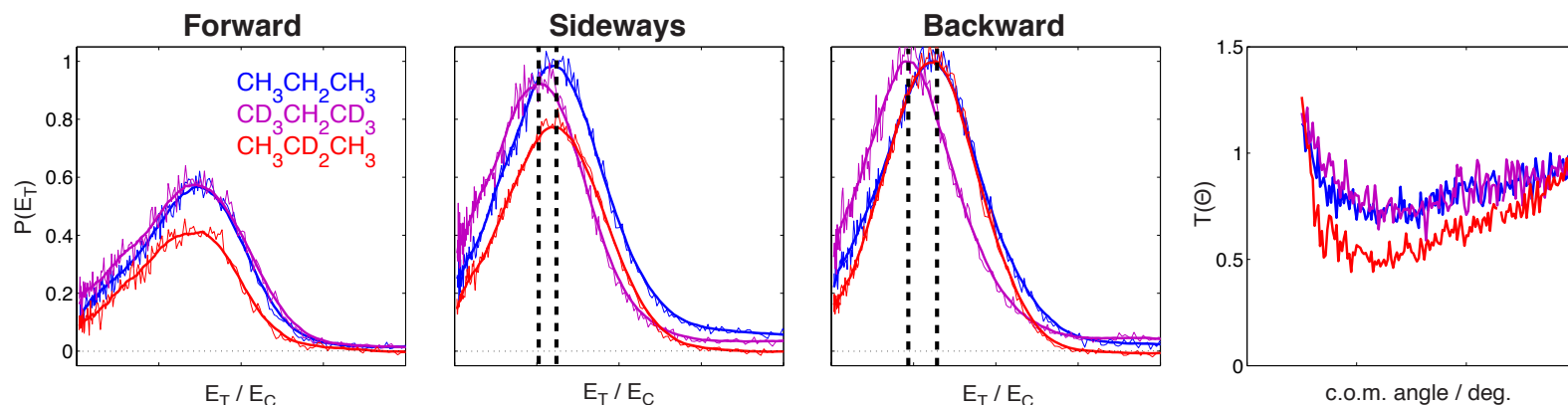
\Rightarrow No multiple photon ionization

\Rightarrow Product selectivity

Cl + propane



differential cross sections for H/D secondary abstractions at 12 kcal/mol

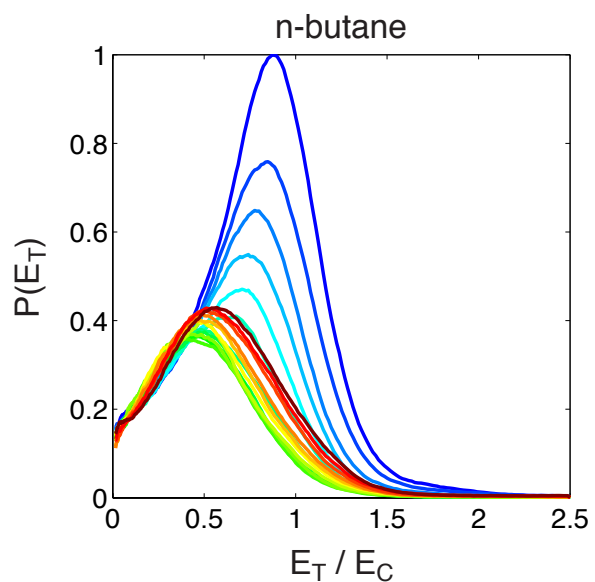


$P(E_T)$ in BW/SW directions of $\text{CD}_3\text{CH}_2\text{CD}_3$:
=> a more effective energy disposal into the propyl product
=> vibrational excitation \neq impulsive model

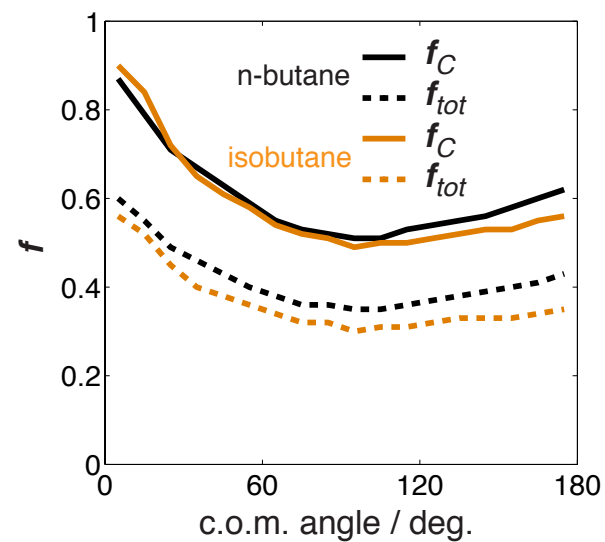
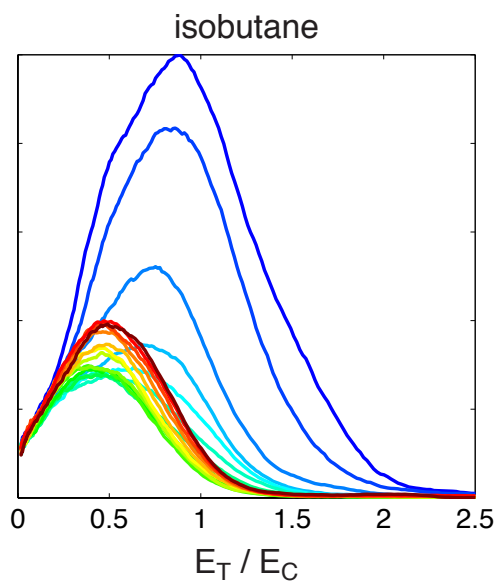
Different $T(\Theta)$ for H and D abstractions:
=> a kinematic effect

Cl + butane isomers

secondary abstraction



tertiary abstraction



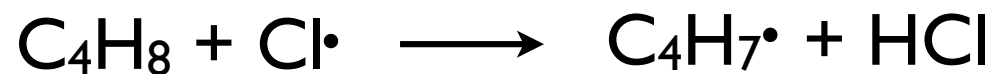
Isobutane in the FW direction:

- Broad E_T distribution
- Sharp angular distribution

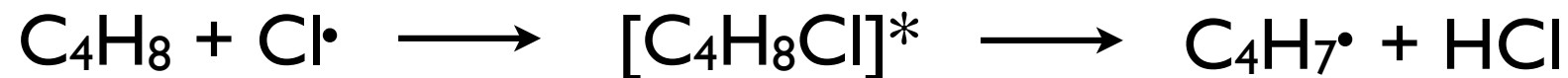
=> HCl($v'=1$) or steric hindrance effect?

Cl + butane isomers

Direct abstraction

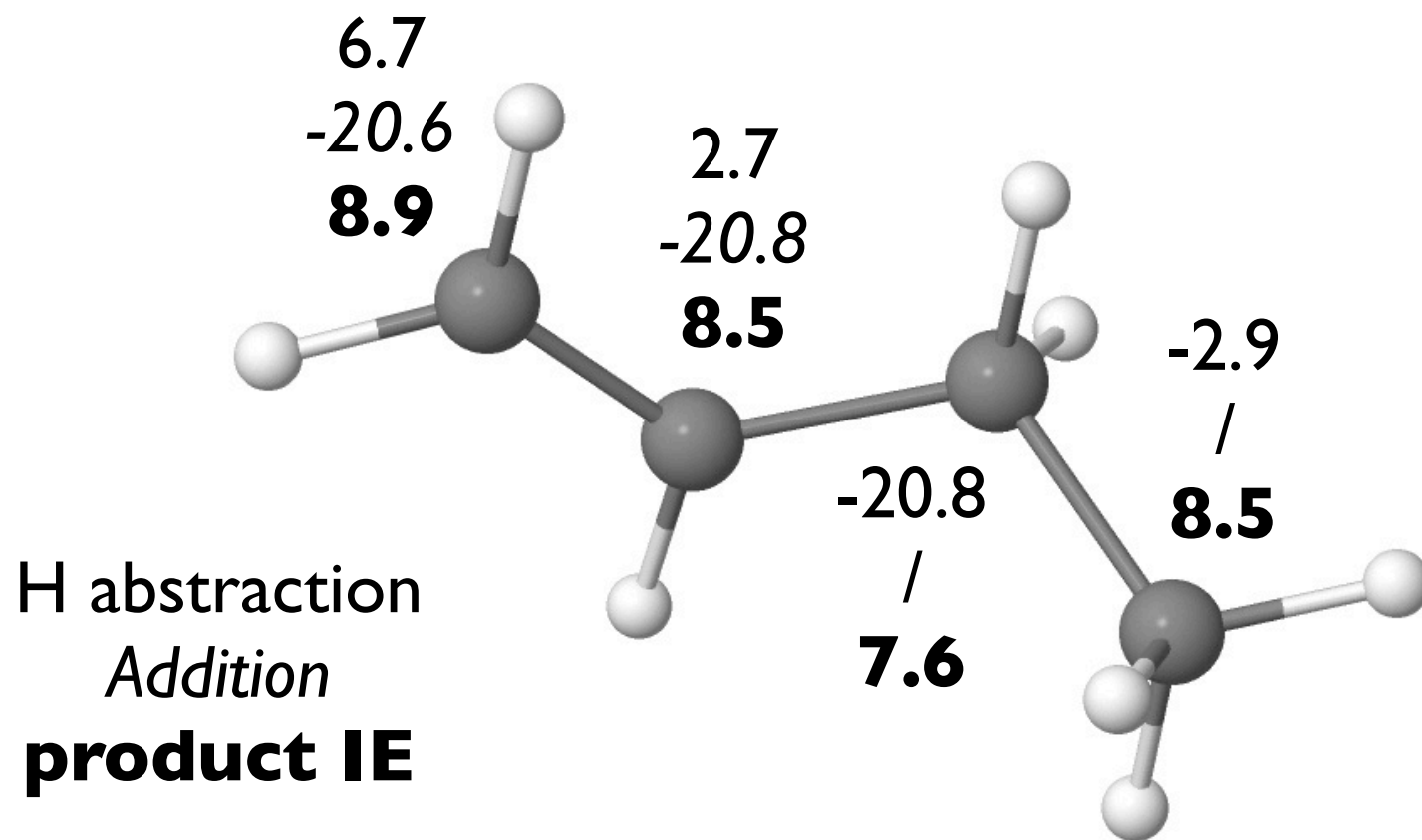


Addition/Elimination



Cl + butene isomers

energetics



H abstraction
Addition
product IE

gauche-1-butene

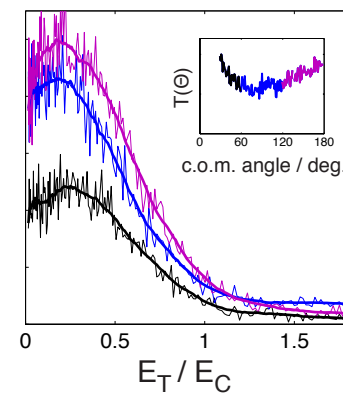
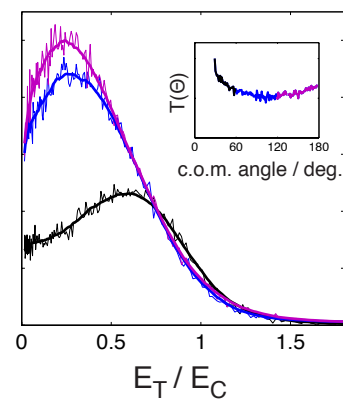
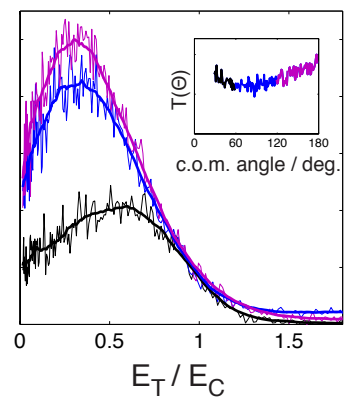
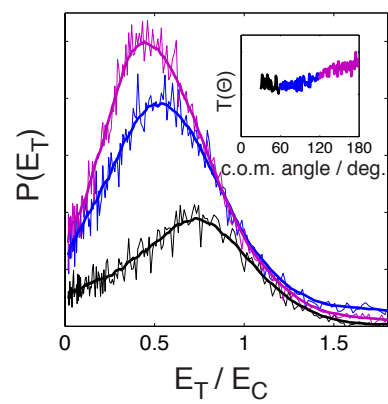
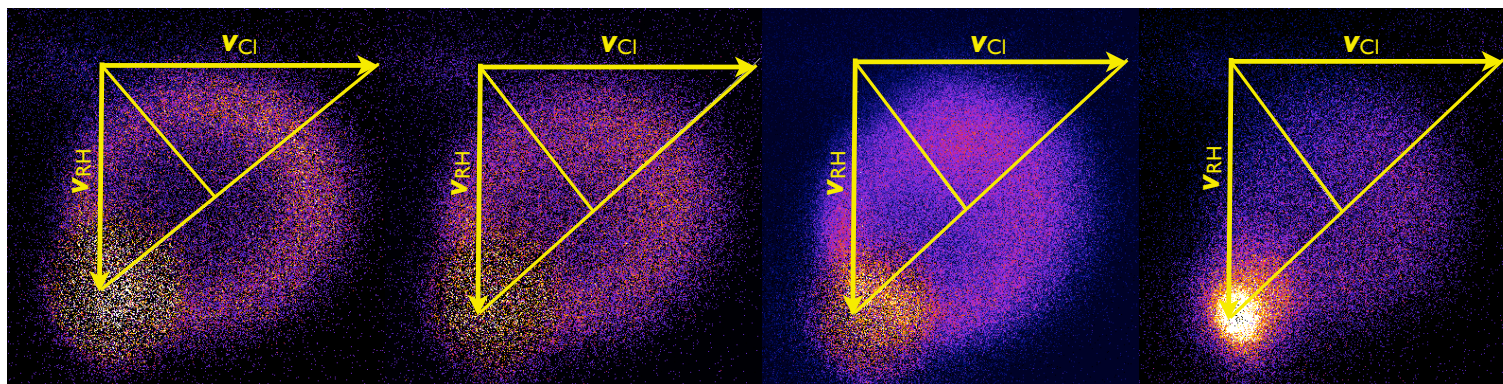
Cl + butene isomers

1-butene

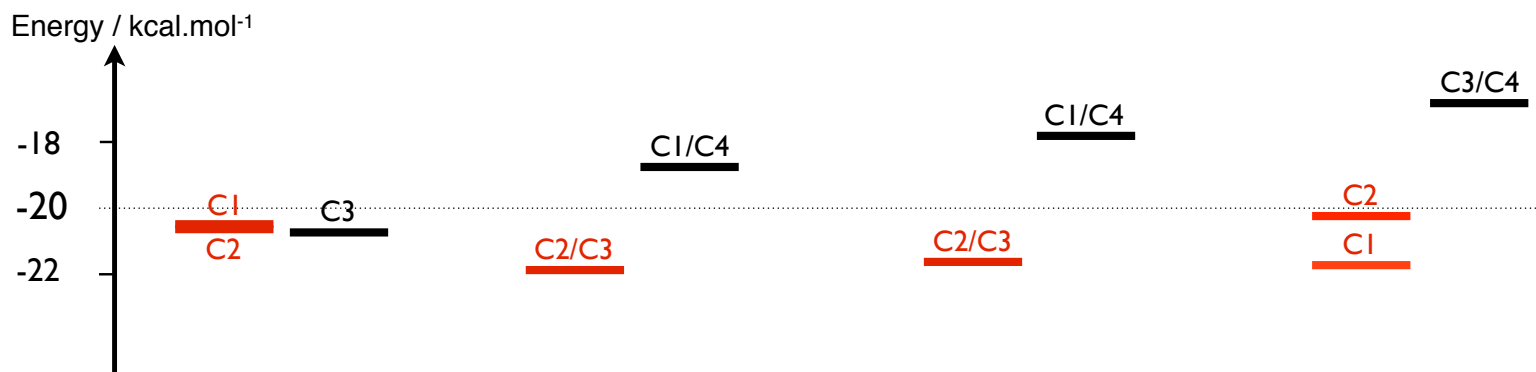
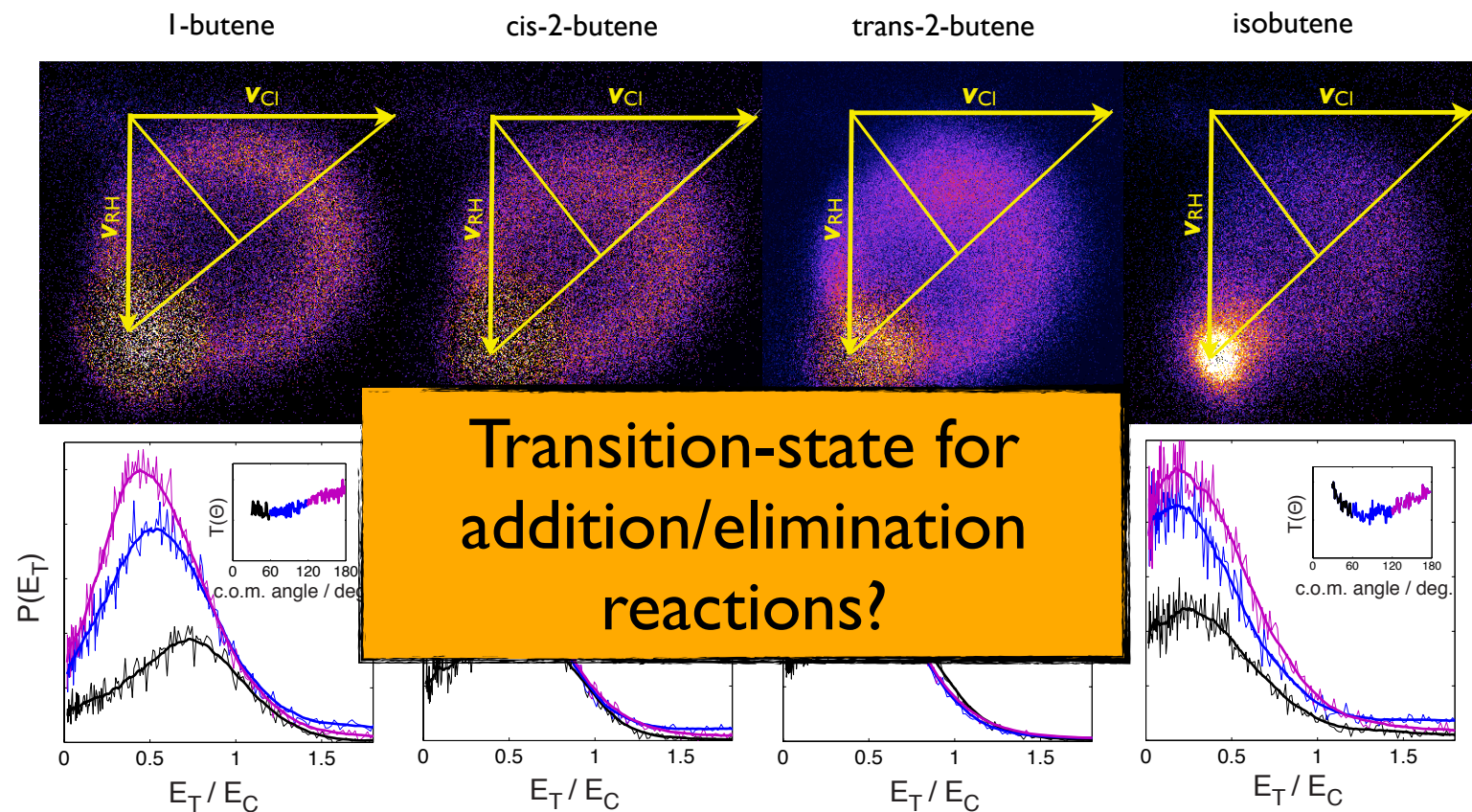
cis-2-butene

trans-2-butene

isobutene

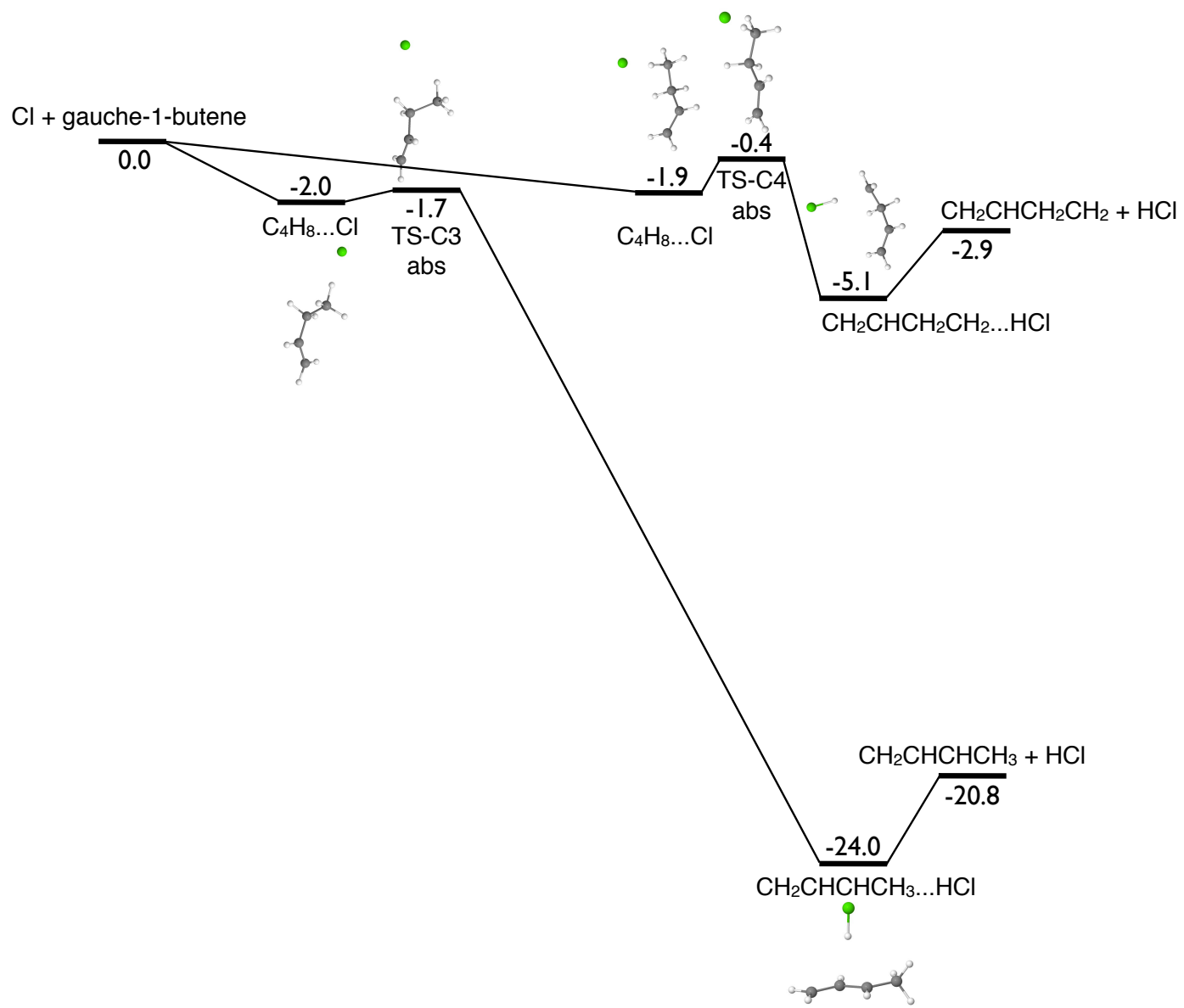


Cl + butene isomers

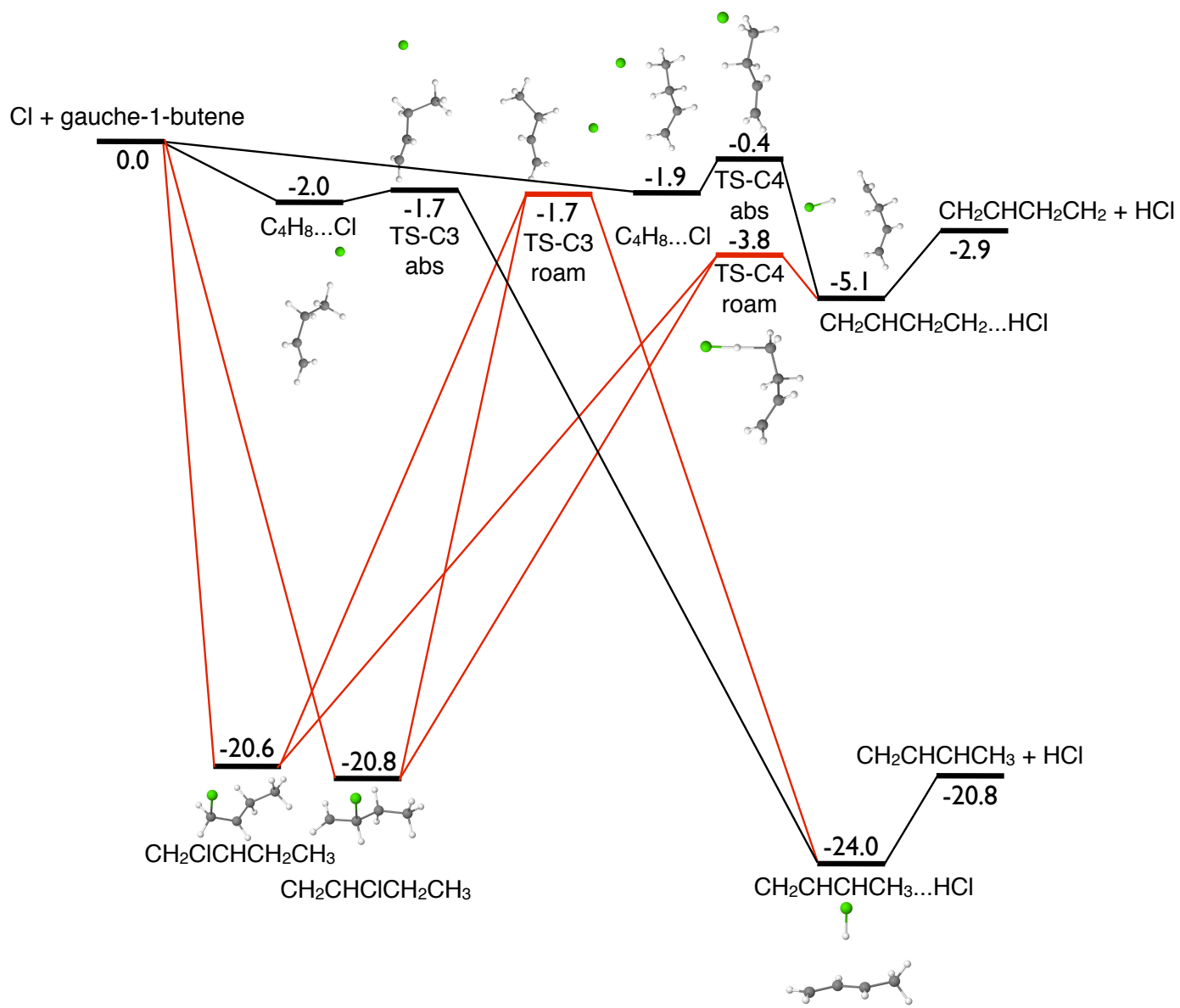


Cl addition vs. H abstraction

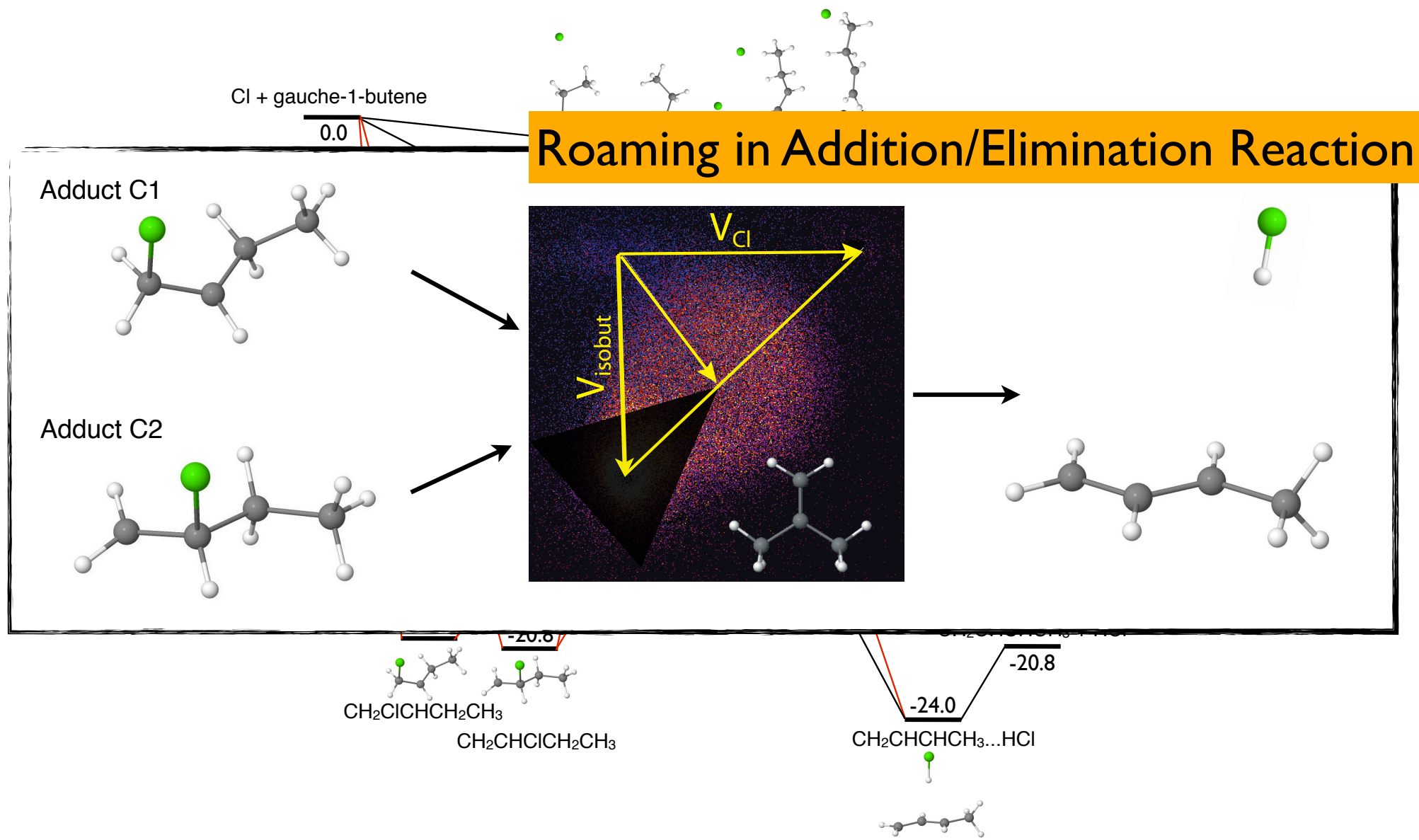
Cl + butene isomers



Cl + butene isomers



Cl + butene isomers



Conclusion/Perspectives

Cl + propane/butane

- Selective measurements of secondary and tertiary abstractions
- C3/C4 hydrocarbons = the last step before convergence to large system energy recoil
- the role of vibrational density-of-state

Cl + butene isomers

- Distinct dynamical behaviors reflecting the competition between direct abstraction and addition/elimination
- ROAMING in addition/elimination

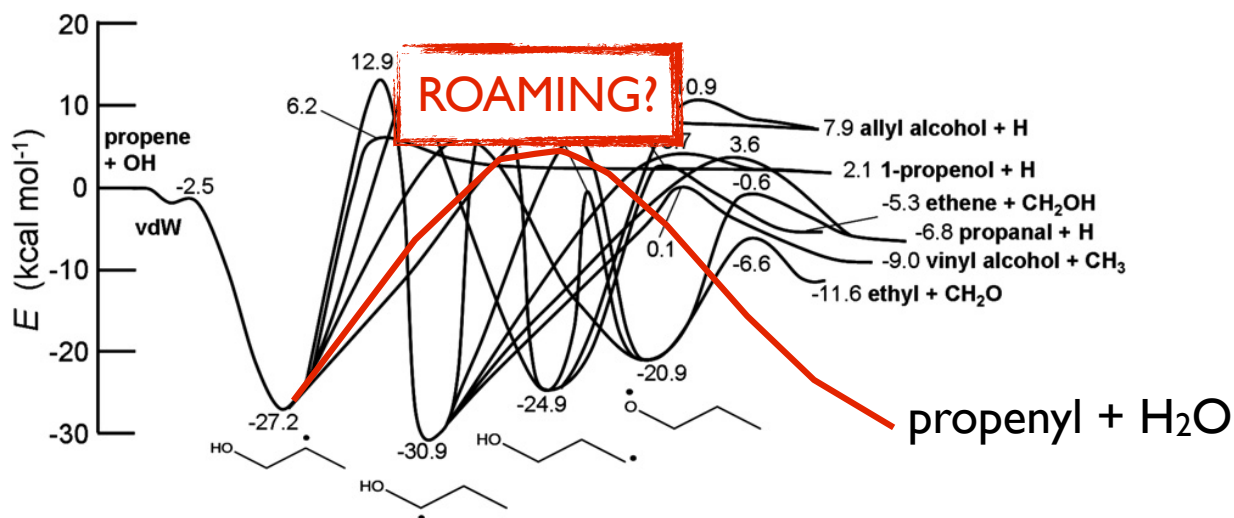
What about Kinetic Isotope Effect?

addition step = an inverse KIE

roaming-abstraction step = normal KIE

Food for thought:

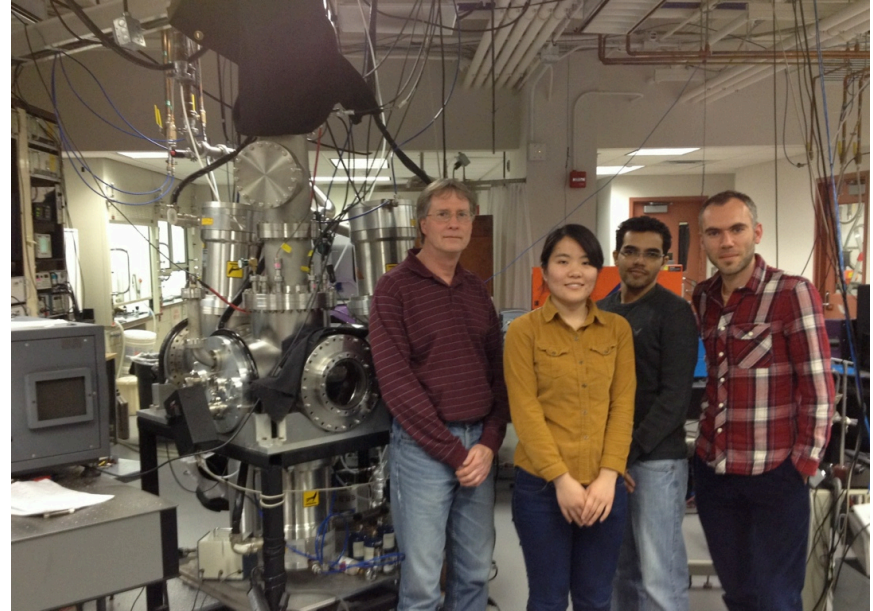
Dynamics of OH + Alkene Reactions



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Florida International University



WAYNE STATE
UNIVERSITY



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Rick, Yuanyuan, Nitin

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