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Luminescence properties of micrometric structures induced by direct laser writing in silver containing phosphate glass

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ABSTRACT

Luminescent structures have been implemented by direct laser writing using high repetition rate femtosecond laser in a silver ion containing photosensitive glass. The optical and luminescence properties have been investigated and compared to those measured after irradiation by a nanosecond laser followed by a thermal treatment. Analogous signals were obtained, demonstrating that the femtosecond direct laser writing combines in one single operation both a photomodification of the silver oxidation states, and a strong local concentration of silver clusters thanks to the high repetition rate. The comparative study confirms the remarkable stability of the photo induced species with respect to the temperature.

1. Introduction

Ultra short pulsed lasers, e.g. femtosecond (fs) lasers, have been demonstrated since more than 10 years to be powerful tools for surface and in depth modification of transparent materials [1,2]. The high spatial precision of these induced modifications has allowed for the fabrication of three dimensional (3D) micro optical devices and photonic structures, such as micro lenses [3], optical memories [4,5], waveguides [6,7], or optical filters [1]. High repetition rate (> 200 kHz) ultra short laser pulses cause heat accumulation effects [8]. Thermal energy is generally considered as a negative factor for precise laser processing, because the heat diffusion enlarges the modified volume. The influence of the temperature has been investigated by Shimizu et al., who related the size of fs laser induced ring structures presenting a high refractive index change to the local temperature gradient [9]. However the thermal effects may be considered as positive for activating ionic migration or for allowing phase separation mechanisms [2]. The local precipitation of crystals within the glass [10] or the modification of the chemical composition distribution around the laser focal volume [11,12] can be monitored to control the physico chemical properties in glass in 3D. Most of the time, the modification takes place on a micrometric scale. Developing new composite materials with innovating functions is the current big challenge of fs

laser structuring. One path is to introduce photosensitive ions inside the glass in which optical properties can be tailored by the interaction with the fs laser [2].

Recently our group showed that near infrared (NIR) fs laser structuring in silver ions containing photosensitive glass at a 1 to 10 MHz repetition rate resulted in fluorescent pipes without any significant variation of the refractive index ($\Delta n < 10^{-4}$) [13,14]. The section of the pipes is organized as a ring structure. The photo chemical reactions leading to the formation of highly fluorescent silver clusters are initiated by free electrons generated through multi photon absorption and activated by the thermal effects [15]. The proposed model for the fluorescent ring shaped formation [14] strongly suggested that threshold effects were mostly responsible of the spatial distribution of the silver clusters. A 3D control of the fluorescence intensity of silver clusters was achieved by controlling the NIR fs laser parameters [16]. We showed that the emission spectra are dependant on the irradiation conditions. However the influence of the temperature on the formation of the photo induced silver species is still under debate. The major difficulty is to separate the thermal effects from modifications induced by the radiation absorption. The characterization of the luminescence properties is not an easy task since the affected zone has sub micrometric dimensions.

In this study, the absorption and luminescence properties of the photo induced structures are investigated using an adapted irradiation protocol in order to access to the luminescence properties. The goal of the study is to compare the optical signatures (absorption and luminescence) of the glass after fs or nanosecond (ns) laser irradiation followed or not by a heat treatment. The effect and the role of hole and electron traps as well as those of the temperature on the

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formation of the photo induced luminescent structures in both irradiation regimes are discussed.

2. Experimental section

Glasses with the composition $40\text{P}_2\text{O}_5$ 55ZnO $1\text{Ga}_2\text{O}_3$ $4\text{Ag}_2\text{O}$ (mol%) were prepared by melt quenching as previously described in [14,15]. The pristine glass exhibits an emission band at around 370 nm for an excitation band centered at 260 nm.

A Yb:KGW 500 fs ultrafast laser (t pulse 500, Amplitude Systemes) operating at 1030 nm with a repetition rate of 10 MHz and a maximum pulse energy of 600 nJ was used for the irradiation. The laser mode is TEM_{00} with $M^2 = 1.2$. The laser beam was focused with a microscope objective ($\text{NA} = 0.75$) into the glass sample mounted on a computer controlled 3D translational stage with a 10 nm resolution. The local structuring mechanisms of the photosensitive silver containing glass have been described elsewhere [14,16]. The photo induced structure has a cylindrical shape, in which the walls are composed of photo produced luminescent species (Fig. 1a) [14,15]. The displacement of the sample along a transverse direction leads to the writing of parallel luminescent planes, in the longitudinal direction, since the center of the Gaussian beam erases the structure and its edges maintain the luminescent features (Fig. 1b) [16]. Parallel structures have been written to obtain an area of $400 \times 400 \mu\text{m}^2$ perpendicular to the propagation axis and formed of parallel planes. A stage velocity of $100 \mu\text{m} \cdot \text{s}^{-1}$ has been used with an irradiance of $6 \text{ TW} \cdot \text{cm}^{-2}$.

The transmission spectra were collected on a $50 \times 50 \mu\text{m}^2$ area of the “squared” pattern. The micro transmission spectra were collected using a CRAIC micro spectrometer equipped with a Xe lamp, a monochromator and a CCD detector. The spectral resolution was 0.7 nm. The reference transmission spectrum has been collected on a virgin area of the glass.

Glass samples were also exposed to a UV ns laser emitting at 355 nm (a mode locked Nd:YAG laser Surelite Continuum L10 equipped with BBO nonlinear crystals) pumped by flash lamps. The pulse duration was 5–7 ns at a 10 Hz repetition rate for an 80 mJ pulse energy. The beam diameter on the glass sample was 5 mm. The fluence was $320 \text{ J} \cdot \text{cm}^{-2}$ and the irradiance $67 \text{ MW} \cdot \text{cm}^{-2}$. The absorption spectra were measured on 2 mm thick samples with a double beam spectro photometer (CARY 5000 UV VIS) over the 200–800 nm spectral range. A heat treatment at 400 °C during 10 min was conducted after the ns laser irradiation.

The luminescence spectra (emission and excitation) on the fs and ns irradiated glasses were recorded on a SPEX Fluorolog 2 spectrofluorimeter (Horiba Jobin Yvon). The excitation source was a 450 W Xe lamp enabling continuous excitation from 200 nm to 800 nm. The

signal was detected and amplified by a Hamamatsu R298 photo multiplier tube cooled by Peltier effect.

3. Results

The transmission spectra of the “squared” pattern written by fs laser are reported in Fig. 2a. Two intense absorption bands peaking at 285 nm and 325 nm can be observed. A long tail of the absorption is noticed above 330 nm showing that absorption features are present up to 500 nm. The strongest absorption band at around 325 nm leads to a transmission value of around 50%, which underlines the strong absorbing character of the photo induced pattern. The lines do not fully fill the entire plane and the height of the structure is estimated to be $7 \mu\text{m}$ FWHM. It means that the absorption coefficient associated to the structure is high enough to absorb most of the incident light. The photo induced structures exhibit an intense white emission under excitation at 325 nm, as shown in Fig. 2b. The emission band maximum is centered at 525 nm, accompanied by an intense shoulder at around 400 nm. The excitation spectra for the emission at 700 nm give rise to two main bands (Fig. 2b) at 275 nm and 325 nm, with a shoulder around 260 nm.

Fig. 3a presents the absorption spectra of the virgin glass and the glass after UV ns laser irradiation, followed or not by a heat treatment at 400 °C for 10 min. In both cases, three intense absorption bands are present at around 285 nm, 325 nm and 380 nm. After the heat treatment, the intensity of these bands is strongly reduced. Absorption features can only be distinguished below 360 nm. For an excitation at 325 nm, the ns irradiated zone presents a red luminescence with a maximum at 630 nm and a shoulder at 450 nm (Fig. 3b). The heat treatment strongly modifies the emission spectrum with a strong decrease of the main band at 630 nm, accompanied by the increase of the band at around 500 nm and the appearing of a band at around 400 nm. The excitation spectra before heat treatment, for an emission wavelength at 700 nm, present excitation bands at 275 nm, 325 nm and 380 nm (Fig. 3c).

After the heat treatment, the excitation band at around 325 nm is blue shifted while a new strong excitation band appears, in relative intensity, at around 275 nm. The excitation at around 380 nm is no longer present.

4. Discussion

4.1. Phosphate glass irradiation and silver luminescent centers

Despite the numerous works dedicated to the study of the photoluminescence of photosensitive glasses under various irradiation (gamma, infrared), the luminescence mechanisms are not fully

Fig. 1. a) Single structure and b) “squared” pattern composed of parallel lines induced by femtosecond direct laser writing within the photosensitive glass.

Fig. 2. a) Transmission and b) emission and excitation spectra of the squared pattern ($50 \times 50 \mu\text{m}^2$ area) written by fs laser.

elucidated. An emission band from 560 nm up to 650 nm is obtained systematically by several authors in phosphate glass [17–23] under excitation in a wide spectral range between 270 and 400 nm. The photo induced silver species are generally described as electron centers (Ag^0), hole centers (Ag^{2+}) or clusters (Ag_m^{x+}). The Ag^0 centers absorb in the 350–500 nm range and are not luminescent [18,20,21]. The absorption of Ag^{2+} centers have been reported at between 280 and 310 nm and the emission is reported from 560 to 660 nm depending on the authors [22,23]. These species result from the transfer of a hole trapped on a phosphate group towards the $\text{dx}^2 - \text{y}^2$ orbitals of an Ag^+ ion, absorbing at around 280 nm [18,20]. The emission of Ag^{2+} centers is typical of defects close to phosphate groups [24,25]. The Ag_m^{x+} clusters of low nuclearity (typically Ag_2^{x+} or Ag_3^{x+}) absorb in the 270–320 nm range [18,20,21]. The theoretical absorption characteristics of various clusters were determined by Ershov et al. [26]. The authors calculated that the Ag^{2+} species have absorption features at around 310–320 nm and 360–400 nm and that the Ag_3^{2+} clusters exhibit intense absorption bands near 260–285 nm and 540–720 nm, and a lower one at 275–320 nm. However, no clear attribution of a specific luminescence has been provided up to now for such clusters [27–31].

We have conducted a systematic study combining EPR, optical absorption and luminescence spectroscopies to provide the assignment of the different absorption and luminescence features to the hole and the electron centers and to the clusters [32].

Fig. 3. a) Absorption spectra before, after ns-laser irradiation and after heat treatment (400 °C, 10 min), b) emission spectra for $\lambda_{\text{ex}} = 325 \text{ nm}$, and c) excitation spectra at $\lambda_{\text{em}} = 700 \text{ nm}$ after ns-laser irradiation, after ns laser irradiation and heat treatment and after fs-laser irradiation.

4.2. Ns laser irradiation

After ns irradiation, absorption bands at 380 and 325 nm are observed (Fig. 3a) and are attributed respectively to the Ag^0 and to the Ag^{2+} centers. An excitation at 325 nm gives rise to a main emission band at around 630 nm (Fig. 3b). This emission band is attributed to the Ag^{2+} centers. Moreover, regarding the emission at 630 nm (Fig. 3d), the main excitation bands at 325 nm and the shoulders

around 280 nm and 380 nm may be considered as directly connected to these centers.

4.3. *Ns laser irradiation followed by heat treatment*

After the heat treatment, the decrease of the intensity of the absorption spectrum, matching almost the absorption spectrum of the pristine glass, indicates that most of the species have disappeared (Fig. 3a). This decrease of the absorption spectrum is accompanied by a strong decrease of the emission band at 630 nm (Fig. 3b), related to the Ag^{2+} centers, and by the existence of a main emission at around 500 nm, which are most probably attributable to small clusters composed of a few silver atoms and ions such as Ag_2^+ or Ag_3^+ [33,19].

The excitation spectrum also shows significant modifications. The excitation band at 380 nm has disappeared, indicating most likely the disappearance of Ag^{2+} species, as suggested by the absorption spectrum. The band at 325 nm is blue shifted and a main band in relative intensity is seen at 275 nm. A cross analysis between the emission signatures at around 380 nm and 500 nm, for an excitation at 325 nm, and the excitation spectrum indicates that the formation of silver clusters is related to the thermal treatment above the glass transition temperature, which allows ionic and charge migrations.

Moreover after the thermal treatment, a lifetime of the excited state less than 10 ns is clearly measured as compared to a lifetime of about 8 μs measured prior to thermal treatment. This measurement is in accordance with the presence of silver clusters exhibiting large emission cross sections. Syutkin et al. assumed that the cluster formation was governed by the diffusion of Ag^+ ions [19,34], and requires the transfer of an electron coming from a phosphate group towards a Ag^+ ion to initiate the clustering reactions [34]. Even if the thermal treatment tends to show that the formation of silver clusters occurs, it is not possible to rule out the fact that clusters exist prior the thermal treatment. More investigations are necessary to separate the strong emission from the hole trapping centers Ag^{2+} , and the emission from the silver clusters, if they are present.

4.4. *Comparison between the different irradiations*

The emission spectrum of the “squared” pattern written by fs laser (Fig. 3b) presenting two main bands at around 380 nm and 500 nm shows strong similarities with the ns UV laser irradiated and heat treated sample. The transmission spectrum in Fig. 2a shows two main bands at 285 nm and 325 nm. The excitation spectra for an emission at 700 nm are also analogous with the ones of the heat treated sample after ns UV laser irradiation. The only clear difference is the shift of the band at around 325 nm for the ns UV laser irradiated and heat treated sample. Such effect could be related to the fact that in the heat treated sample, the hole trapping centers Ag^{2+} have most likely completely disappeared, leading to the extinction of the shoulder at 380 nm. The high repetition rate fs laser writing process leads to luminescent features comparable with ns UV laser irradiation followed by thermal treatment. The fs laser irradiation allows the formation of high concentrations of silver clusters, without significant resulting concentrations of unstable hole and electron traps such as Ag^{2+} and Ag^0 . This observation is in agreement with the stability of the fs laser induced luminescent structures through pulse after pulse heat accumulation, and is an important aspect in terms of future possible applications, such as long term 3D optical data storage [35].

5. Conclusions

In conclusion, new insights for the identification of the luminescent species induced by high repetition rate femtosecond laser in photosensitive silver containing glasses have been provided in this paper. The spectral (absorption, excitation and emission) properties

of such structures have been characterized and compared to those of nanosecond laser irradiated samples after annealing at a temperature above the glass transition temperature. Analogous signatures have been pointed out, indicating that the femtosecond laser irradiation combines the advantages of both the photo ionization and a local micrometric heat treatment in a one step procedure. The reported data bring new elements for the understanding of the remarkable stability of the photo induced structures through thermal treatments.

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