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ON-LINE DETERMINATION OF AGGREGATE SIZE AND MORPHOLOGY IN SUSPENSIONS
Michel Cournil, Frédéric Gruy and Patrick Cugniet

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Abstract. Information about the aggregation state of fine particles is an important element for process control, product quality monitoring and fundamental understanding in many cases of industrial slurries. When aggregates are small or fragile objects, their withdrawal is difficult and off-line characterization may be a source of error. This work deals with the application of different in-line methods to the characterization of silica aggregate size and morphology. These methods are based on turbidimetry. One of them consists of the analysis of the turbidity fluctuations and is operated on a commercial instrument. The other one uses the aggregate settling velocity which is determined by turbidimetry too, however with a home-made apparatus. This work gives us the opportunity to define morphological models for small aggregates and to calculate their drag coefficient. Thanks to these models, the aggregate morphological characteristics and the number of their constituting particles can be derived from the experimental results. Agreement between the different methods is examined and discussed.

INTRODUCTION

Contrary to agglomerates which are commonly cemented by crystalline bridges which confer them a rigid and solid structure, aggregates are most of the time small and fragile objects. In particular, in most experiments, their size does not exceed a maximum value which results either from the dynamic balance between aggregation and fragmentation or from zero aggregation efficiency beyond a certain size [1-2]. For fundamental reasons as well as for process monitoring purpose, real time knowledge of the aggregate size and morphology may be very useful. For these determinations, both on-line and in-line methods can be envisaged. Off-line methods, however, do not ensure isokinetic withdrawal, particularly, for small particles, and may damage fragile aggregates. In situ techniques could be certainly ideal; however, only very few exist and they need to be validated. Turbidimetry has been proved to be a particularly efficient method for in situ particle size measurements [3], particularly in aggregating systems [1]. Turbidimetry is based on light scattering by suspensions and rests on the Mie theory [4]. However, the scattering properties of small aggregates of micronic particles, for instance, is not completely known and requires additional research. Since the works of Wesely et al. [5], analysis of turbidity fluctuations has been shown as particularly interesting for the determination of particle number and size; this principle has given rise to a commercial instrument (Aello 4000).

The present work is devoted to the characterization of aggregates of silica formed in a stirred vessel. Aggregation itself has been studied by in situ turbidimetry [6-7]. Final aggregates are characterized both using turbidity fluctuations and settling velocity measurements. Interpretation of these data requires preliminary theoretical tasks: morphological modelling of the aggregates and calculation of their light scattering section and drag coefficient. Comparison between the morphology and size respectively given by each method is presented and commented on.

EXPERIMENTAL PART

1. Materials and techniques and procedure
Aggregation experiments are performed on samples of monodisperse silica spheres (1.5 µm in diameter, Geltech Inc products). Aggregation is studied in water at pH values 2 to 4.
The reactor used for this study of aggregation is a stirred tank the diameter of which is 120 mm (Figure 1). This reactor is equipped with four baffles. Liquid depth in the vessel is equal to diameter. Bottom part of the tank is rounded. Agitation is ensured by a four bladed 45° Teflon impeller of diameter 60 mm. Temperature is kept constant at 25.00°C ± 0.01°C by a double-wall jacket. The reactor is fitted with an optical system to measure in situ the suspension turbidity in the wavelength range 350 nm-800 nm. Details can be found in [6].

![Figure 1. Schematic representation of the aggregation reactor](image)

Turbidity $\tau$ expresses the extinction phenomenon of an incident light beam due to light scattering by solid particles; it is defined by relation:

$$\tau = \frac{1}{\lambda_0} \frac{\ln \left( \frac{I_0}{I} \right)}{L_{opt}}$$  \hspace{1cm} (1)$$

in which $\lambda_0$ is the wavelength, $I_0(\lambda_0)$ the intensity of the incident beam and $I(\lambda_0)$ the intensity of the transmitted beam after an optical path of length $L_{opt}$.

Turbidity depends on the mean diameter $d_p$ particle density function $f$ according to the integral:

$$\tau = \frac{1}{\lambda_0} \int C_{ext} \, d_p \, f \, d_p \, d\sigma$$  \hspace{1cm} (2)$$

Particle extinction section $C_{ext}$ is derived from the Mie theory [4]; its calculation is relatively easy for spherical, compact, or large particles, however much more delicate for small, non compact aggregates [8]. From equation (2), it appears that variations in the particle density function $f$ result in turbidity variation. For instance, in the case of 1.5 μm silica particles in water, turbidity decreases with aggregation, whereas it increases in the case of 0.5 μm particles. When particles leave the measurement cell, due to settling for instance, turbidity decreases obviously. These characteristics will be exploited later on.
Aello 4000 measurement cell is external to the reactor and located on a by-recirculation loop; thus measurements are in line and not in situ as previously. This apparatus delivers the turbidity signal of the suspension located in its measurement cell, however with a special interest in the signal fluctuations around its mean value. From these data, two parameters are calculated, mean extinction section $C_{\text{ext}}$ and particle number in the cell $n$. Compared to our in situ determinations, this measurement procedure is liable to damage the aggregates, however this effect is certainly reduced because shear stress in the loop is considerably lower than in the reactor and pumping conditions are relatively smooth. This is confirmed by the identical turbidity levels observed in the two experimental systems.

Aggregation of a 0.246 g silica sample is performed in the previous reactor at given stirring rate. However the experimental conditions, turbidity reaches a constant value after a few tens of minutes at the maximum; this plateau is generally interpreted as a steady state of the aggregate size distribution in the system and is characterized by a maximum aggregate size. After two hours, this means during this steady state, agitation is stopped; particle settling is clearly observed and turbidity decrease is recorded. In similar experiments, suspension is pumped to the Aello 4000 cell and characterized during the aggregation process and at its end.

2. Principle of interpretation of the sedimentation experiments
The turbidity probe is vertically located at the two-thirds of the vessel radius halfway between two baffles, and mounted at 4.8 cm from the upper surface of the liquid (Figure 2)

![Figure 2. Aggregate settling as detected by the turbidity sensor](image)

![Figure 3. Turbidity variation during aggregate sedimentation. a: large monodisperse aggregates; b: small monodisperse aggregates; c: polydisperse aggregates.](image)

As soon as agitation is stopped, aggregates start settling and gradual decrease in turbidity is observed. In the ideal case of monodisperse aggregates, the turbidity signal keeps constant for a while and sharply decreases only when the aggregates initially located near the liquid...
surface have crossed the measurement window. In the case of polydisperse aggregates, settling results in a classification of the aggregates according to their size and turbidity drop is not as sharp. Thus, according to the aggregate population nature, different turbidity plots against time are observed.

In Figure 3 we have represented the initial turbidity decrease due to aggregation in the stirred vessel in the time interval \([0, t_0 = 2 \text{ hours}]\) then the turbidity decrease due to sedimentation. Cases a and b are relative to monodisperse aggregates; case c, which concerns polydisperse aggregates, is the most commonly observed.

Settling velocity of a silica spherical particle in water is given by Stokes law:

\[ v_i = \frac{2}{9} \rho \frac{a_i g}{\mu} \tag{3} \]

in which \(a_i\) is the sphere radius, \(g\), the gravity, \(\mu\), the liquid dynamic viscosity and \(\rho\) the density difference between silica and water.

In the case of fractal-like aggregates, previous relation becomes:

\[ v = v_i \frac{i}{\beta} \tag{4} \]

in which \(i\) is the number of primary particles in the aggregate, \(\Omega\) a corrective drag coefficient, \(a_i\), the aggregate outer radius, \(v\), its settling velocity. \(D_f\), the aggregate fractal dimension, is defined by this relation. \(S\), structure factor, depends on \(D_f\) and is obtained from simulations.

In turbulent aggregation, \(D_f\) is most of the time equal to 2.4 and \(S = 0.79\) [8].

From experimental curves similar to plots of Figure 3a and 3b, we can easily calculate \(v\) by dividing the sensor mean depth from the top of the liquid (here 4.8 cm) by the settling duration. Then \(a_i\) is deduced from Eqn 3. This has been done for silica particles in conditions of non aggregation (pH = 8) and good agreement has been found. For experimental plots similar to Figure 3c, we have to choose a characteristic settling time to be able to determine the aggregate size. This will be discussed further.

3 Experimental results
3.1 Determination of the corrective drag coefficient

As no model was available to determine drag coefficient \(\Omega\) particularly for small aggregates, we performed specific experiments. In order to use handy objects, we did experiments on 1 mm glass beads aggregates that we prepared ourselves [6]. Different sizes (2 to 80 particles) and morphologies were obtained so. To keep Stokes settling conditions, we studied sedimentation in glycerol. Sedimentation time, which is relatively long, as determined by direct observation of the settling objects, thus settling velocity as easily obtained. To apply Eqns 4 and 5, we needed to know \(a_i\), we took it equal to the radius of the sphere which had the same projected area as the aggregate. As the aggregates were not necessarily isotropic, we obtained their projected surface area either from standard image analysis procedures of their photograph, or by direct calculation on the geometrical objects, knowing the coordinates of the primary particles. Then \(\Omega\) could be derived from Eqn 4. In all cases, \(\Omega\) as found practically equal to 1 (with the specified choice of \(a_i\)).
3.2 Settling of 1.5 \( \mu m \) silica particles

Using procedure described in section 2, we obtain the turbidity variation shown in Figure 4. Agitation was interrupted in the reactor after two hours aggregation. Plots of Figure 4 have the same shape as plot of Figure 3c. As aforesaid in section 2, characteristic time of settling should be chosen. Taken into account the existence of a main sedimentation wave on the different plots of Figure 4 we chose characteristic times corresponding to the end of this wave.

![Figure 4: Settling of 1.5 \( \mu m \) silica in water (pH 3, \( \lambda = 550 \) nm)](image)

Table 1 shows the different settling times and normalized settling velocity \( v/v_1 \) obtained for different stirring rates.

**Table 1: Normalized experimental settling velocity of silica aggregates formed at different stirring rates**

<table>
<thead>
<tr>
<th>Stirring rate (rpm)</th>
<th>Settling time(s)</th>
<th>( v/v_1 )</th>
</tr>
</thead>
<tbody>
<tr>
<td>200</td>
<td>6000</td>
<td>5.8</td>
</tr>
<tr>
<td>400</td>
<td>8000</td>
<td>4.4</td>
</tr>
<tr>
<td>600</td>
<td>13000</td>
<td>2.7</td>
</tr>
<tr>
<td>800</td>
<td>15400</td>
<td>2.3</td>
</tr>
</tbody>
</table>

3.3. Aello 4000 determinations

As aforesaid, the Aello 4000 equipment allowed us to obtain \( C_{ext} \) and number \( n \) of particles in the cell at any time, in particular, total number of silica particles \( n_0 \) is known. This is the initial number of particles which is measured at time zero prior to aggregation. Dividing \( n_0 \) by \( n \)
gives us \( L \) the mean number of silica particles per aggregate. Data relative to the aggregation-fragmentation steady state are reported in Table 2 for different stirring rates.

### 3.4. Image analysis

Attempts of aggregate removal after 2 hours aggregation have been made. After their withdrawal with a pipette, the samples have been carefully dried, then observed with a microscope coupled with an image analyser (Zeiss Axioskop microscope, magnification: \( x \) 500; video camera JVC KY-F58; Leica - in software). Photographs of the removed samples have been processed in order to determine the number of primary particles per aggregate. Results are reported in Table 2.

**Table 2: Estimation of the mean number of silica particles per aggregate**

<table>
<thead>
<tr>
<th>Stirring rate (rpm)</th>
<th>200</th>
<th>400</th>
<th>600</th>
<th>800</th>
</tr>
</thead>
<tbody>
<tr>
<td>( C_{\text{ext}} ) ( \text{m}^2 )</td>
<td>23</td>
<td>14</td>
<td>6</td>
<td>5</td>
</tr>
<tr>
<td>( n )</td>
<td>210</td>
<td>400</td>
<td>1000</td>
<td>1500</td>
</tr>
<tr>
<td>( n_0/n ) (( L ))</td>
<td>28</td>
<td>15</td>
<td>6</td>
<td>4</td>
</tr>
<tr>
<td>Particle number (from image analysis)</td>
<td>46</td>
<td>…</td>
<td>25</td>
<td>20</td>
</tr>
<tr>
<td>Particle number (from settling data))</td>
<td>20</td>
<td>15</td>
<td>5</td>
<td>5</td>
</tr>
</tbody>
</table>

**Table 3: Calculated values of normalized \( (\nu/v_1) \) settling velocity of aggregates for different fractal dimensions and particle numbers**

<table>
<thead>
<tr>
<th>Fractal dimension ( D_f )</th>
<th>Particle number</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>2</td>
</tr>
<tr>
<td>8</td>
<td>3.1</td>
</tr>
<tr>
<td>16</td>
<td>4.3</td>
</tr>
<tr>
<td>24</td>
<td>5.3</td>
</tr>
<tr>
<td>32</td>
<td>6.0</td>
</tr>
</tbody>
</table>

### DISCUSSION

#### 1. Settling velocity calculations

Using Equations (3) to (5) of section 2 and experimental determinations of the corrective drag coefficient (section 3.1), we can predict the normalized settling velocity of aggregates of given number of primary particles and given fractal dimension. These calculations are reported in Table 3.
Now this table can be used as follows:
i) the respective experimental normalized settling velocities of the aggregates formed at
different stirring rates are extracted from Table 1;
ii) the respective closest values of normalized settling velocity are searched for in the columns
$D_f = 2.4 – 2.5$ which is the likely fractal dimension of the aggregates. In fact, the fractal
dimension could have been obtained by image analysis of experimental samples, however,
only in the case of very large aggregates of small particles. This is not the case here. Thus, we
selected the present values because they were commonly found from previous experiments
and simulations.
iii) the corresponding particle number in the aggregate is found after interpolation on data of
Table 3 and then reported in last row of Table 2.

We can now compare the estimations of particle number in small silica aggregates which
were derived using the different methods.
i) Aello 4000 measurements and determinations from settling velocities are in good
agreement and indicate that this particle number is very low. Aello equipment is supposed to
give reliable results for particles ranging between 1 and 250 micrometers in equivalent
diameter;
ii) direct measurements on withdrawn samples are very different and certainly non
representative. They confirm the well known difficulty of isokinetic removal of small objects.
Similar studies were performed on 0.5 m silica particles and lead us to the same conclusions
[6]. The Aello 4000 apparatus still provided us with correct results in spite of the low value of
the particle diameter.

2. Maximum aggregate size
As mentioned in the introduction of this paper, the existence of maximum aggregate size
can be due to two main reasons: fragmentation or collision efficiency becoming zero beyond
a critical size.

2.1. Fragmentation
The occurrence of breakage depends on the balance between the disaggregation effects due to
the action of the fluid and the overall cohesion of the aggregate due to the interactions
between primary particles. The hydrodynamic effects are of different nature according that the
aggregate is larger or smaller than the Kolmogorov microscale. Only the latter case is
compatible with the experimental conditions of this study. It corresponds to a shear stress
originating from the local velocity gradient $\gamma$ and acting on the aggregate. The breakage rate
or the fragmentation kernel $K_f$ is generally assumed to be proportional to $\gamma e^{\frac{\sigma}{\tau_s}}$, where $\sigma$
is the mean mechanical strength of the aggregate and $\tau_s$ is the mean shear stress. Thus, the breakage
rate depends on the hydrodynamic conditions of the flow, via $\gamma$ and on the characteristics of
the aggregates: outer radius, fractal dimension, primary particle radius and cohesion force
between primary particles (typically van der Waals force).

In recent works [2, 7-8] on aggregation of titanium dioxide or silica in similar conditions,
however, we came to the conclusion that such fragmentation kernels were not able to interpret
the maximum size observed in our experiments, thus we adopted the following theoretical
approach.

2.2. Zero collision efficiency
This approach was especially developed by Brakalov [9]. It is known that the collision
efficiency between equally sized spherical particles decreases with the particle size. The
decrease is sharper as the particles (aggregates) are porous. Other wise, the aggregate, which
results from two smaller aggregates, can be too loose to survive. Brakalov shows that it exists a maximum value for the aggregate size. From different experimental works, it appears that the maximum particle size $a_L$ depends on shear rate, according to relation:

$$\frac{a_L}{a_i} = \gamma^{c'}$$  \hspace{1cm} (6)

In recent works on aggregation in the same reactor \cite{2, 6, 7}, we found values from 0.22 to 0.60 for exponent $c'$. Moreover, after integrating the assumption of zero aggregation efficiency beyond this size in our models, we obtained good agreement between predicted and observed dynamics.

We recall that the gradient velocity as a function of the turbulent energy dissipation rate $\varepsilon_m$ and kinematic viscosity $\nu$:

$$\gamma = \frac{\varepsilon_m^{1/2}}{\nu}$$  \hspace{1cm} (7)

Mean value of $\varepsilon_m$ is given by the ell-knon equation:

$$\varepsilon_m = N_p \omega^3 D_s^5 \frac{V}{V'}$$  \hspace{1cm} (8)

in which, $N_p$ is the power number, $D_s$ the stirrer diameter, $\omega$ the rotation rate of the stirrer and $V$ the volume of the suspension.

Using the results of the present study (respectively Aello measurements and settling data from Table 2) and equations (5-8), we can now verify relation (6) in our case and determine exponent $c'$. Good agreement is found, which validates the mathematical form of Equation (6) and respective values of $c'$ are: 0.24 and 0.19. These values are of the same order of magnitude as previous results obtained on titanium dioxide aggregation \cite{2}.

CONCLUSION

This study clearly proves the interest of in situ or in line particle size determinations, especially based on turbidimetry, for aggregate or agglomerate characterization. Thanks to this procedure, small and fragile objects can be characterized in situ without taking the risk of damaging or non isokinetic withdrawals. Another interesting aspect of the Aello 4000 measurements is the possibility of quasi-instantaneous characterizations, allowing in this to characterize rapidly changing media or objects. From a fundamental point of view, this work has allowed us to model the settling velocity of small ramified objects, which are not known so far. The dependence on the aggregate particle number has been determined according to the stirring rate. Thanks to these data progress can be expected in the understanding of the aggregate fragmentation mechanisms.

REFERENCES

\[7\] Cournil M., Gruy F., Cugniet, P., Congrès Français de Génie des Procédés (2003), Saint-Nazaire, submitted.