Rationale and conditions for the requirement of chiral bioanalytical methods in bioequivalence studies

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Abstract

Purpose The aim of the present work was to assess the need for chiral bioanalytical methods in bioequivalence studies.

Methods The samples from a bioequivalence study of two ibuprofen 2% oral suspensions that had shown bioequivalence for AUC and $C_{\max}$, but not for $t_{\max}$ (medians of 2.0 and

0.75 h) with a non-chiral method were assayed with a chiral method to investigate whether there was an actual difference in the rate of absorption within the limits of $C_{\max}$ and AUC bioequivalence.

Results The non-chiral method and the sum of concentrations of both enantiomers obtained with the chiral method gave a similar outcome (90% CI $C_{\max}$ non-chiral: 82.77–96.09, sum of enantiomers: 82.19–98.23; 90% CI AUC_t non-chiral: 107.23–115.49, sum of enantiomers: 105.73–121.35). However, the chiral method showed differences in AUC and $C_{\max}$ that resulted in non-bioequivalence for the individual enantiomers (90% CI $C_{\max}$ *S*-ibuprofen: 76.05–91.36, *R*-ibuprofen: 87.84–113.05; 90% CI AUC_t *S*-ibuprofen: 96.67–105.86, *R*-ibuprofen: 118.86–142.24). The differences in the pharmacokinetics of each enantiomer, and thus in the enantiomer concentration ratio, were dependent on the rate of absorption.

Conclusions Due to the fact that in bioequivalence studies the rate of absorption of the new product is unknown, chiral bioanalytical methods should be employed for chiral drugs, such as ibuprofen, whose enantiomers exhibit different pharmacodynamic characteristics and whose enantiomer concentration ratio might be modified by the rate of absorption, irrespective of whether the enantiomer is the minor enantiomer or the similarity of the pharmacokinetics of the enantiomers at a given rate of absorption.

Keywords Ibuprofen · Bioequivalence · Enantiomers · Chirality · Regulatory requirements

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Introduction

The need for chiral bioanalytical methods in bioequivalence studies is a controversial issue that has not been agreed upon

by different regulatory agencies [1, 2]. The FDA Guidance for Industry [2] recommends the measurement of the racemate using a non-chiral assay for bioequivalence studies. Measurement of individual enantiomers in bioequivalence studies is only recommended when all of the following conditions are met: (1) the enantiomers exhibit different pharmacodynamic characteristics, (2) the enantiomers exhibit different pharmacokinetic characteristics, (3) primary efficacy/safety resides with the minor enantiomer, and (4) non-linear absorption is present (as expressed by a change in the enantiomer concentration ratio with change in the input rate of the drug) for at least one of the enantiomers. In such cases, bioequivalence criteria should be applied to the enantiomers separately.

It is obvious that if enantiomers do not exhibit different pharmacodynamic characteristics, chiral bioanalytical methods are unnecessary. However, the other three conditions are debatable or difficult to understand.

It has been shown previously that even when the active (eutomer) and inactive enantiomer (distomer) exposure ratio is near unity, the non-chiral bioanalytical method may be biasing the results [3]. Obviously, if the eutomer were the minor enantiomer, the bias would be greater, but it does not mean that in other cases the bias is nonexistent.

The concept of non-linear absorption (expressed as a change in the enantiomer concentration ratio with a change in the input rate of the drug) has been confused frequently with non-linear pharmacokinetics (e.g. as required in the CPMP Notes for Guidance [1, 4]). Boni et al. [5] illustrated with etodolac that this non-linear absorption may occur in drugs with linear kinetics both for the racemic and the individual enantiomers.

In addition, it must be taken into account that for those drugs where a change in the rate of absorption produces a change in the enantiomer concentration ratio, it could be understood that these enantiomers exhibit different pharmacokinetic characteristics, at least at some rates of absorption.

The need for chiral bioanalytical methods in bioequivalence studies has been discussed from a theoretical point of view [6], or experimentally, changing the input rate artificially [5]. However, actual examples would be convenient to confirm if chiral bioanalytical methods are necessary in bioequivalence studies.

Ibuprofen is a chiral non-steroidal anti-inflammatory drug with the *S*-enantiomer possessing most of the beneficial anti-inflammatory activity [7]. Although enantiomer exposure ratio is near unity, the drug exhibits a complex and marked stereoselectivity in its pharmacokinetics [8]. In addition, according to the Biopharmaceutics Classification System (BCS) [9, 10], ibuprofen is a highly soluble at neutral pH class II drug with complete oral absorption, but whose rate of absorption is highly dependent on formulation and manufacturing factors [11, 12] or even on the interaction

with antacids [13]. In other words, due to its low solubility at acidic pH, different formulations may show different in vivo dissolution rates, which lead to different rates of absorption [14]. For these reasons it was chosen as a model drug in a previous [3] and the present work.

The aim of the present work was to challenge the validity of the present regulatory requirements on chiral bioanalytical methods in bioequivalence studies. For this purpose, the samples from a bioequivalence study of two ibuprofen 2% oral suspensions that had shown bioequivalence for AUC and $C_{\max}$, but not for $t_{\max}$, with a non-chiral method, were assayed with a chiral bioanalytical method to investigate an actual difference in rate of absorption within the limits of $C_{\max}$ and AUC bioequivalence.

Materials and methods

Study design

A randomized, open-label, single dose, balanced, two-way crossover study was conducted with the approval of the Ethical Committee (CEIC-IMAS) and the Spanish Agency for Medicines and Health Care Products. Written informed consent was obtained from subjects after explaining the nature and purpose of the study. The 18 volunteers (nine males and nine females) were non-smokers; were determined to be in good health by physical examination, complete blood count, urinalysis, and serum test on hepatic and renal function; and refrained from taking any drug including OTC products for at least 1 week prior to or during the study. After fasting overnight for 8 h, a dose of 600 mg of ibuprofen 2% suspension was orally administered with 220 mL of water, either as a new test product (Laboratorios Gelos, Spain; batch: R-002, content: 102%), or as the reference product (Dalsy, Laboratorios Knoll, Spain; batch: R-50, content: 103%). Blood was drawn before dosing and at 0.25, 0.5, 0.75, 1, 1.5, 2, 2.5, 3, 4, 6, 8, 10, 12 and 24 h after dosing through an in-dwelling catheter placed in an antecubital vein in the forearm. The blood samples were centrifuged, and the plasma was collected and stored at -20°C until assayed. Seven days after this first administration, the alternative formulation was given to complete the sequence.

Assay procedure

Ibuprofen plasma concentrations were determined by the sponsor following a validated non-chiral method already described [15]. Subsequently, samples were assayed for ibuprofen enantiomers using a chiral AGP- $\alpha 1$ column of 100×4 mm ID (Chrom Tech, Sweden) according to the method previously published [3]. Ibuprofen enantiomers were extracted from 0.6 mL of plasma (to which 0.15 mL

of 3 N HCl was added) with 5 mL of cyclohexane. After 15 min of vortexing, the samples were centrifuged at 1,200 *g* for 5 min, and 3 mL of the organic phase was separated. The organic extracts were evaporated to dryness in a Savant Speedvac concentrator (Savant Instrument, USA) at 45°C under vacuum. The residue was then reconstituted with 1.5 mL of mobile phase and an aliquot of 100 μ L was injected into the HPLC.

The assay was performed with a modular Jasco (Japan) liquid chromatograph equipped with a DG-2080-53 3-line degasser, a LG 2080-02 ternary gradient unit, a PU-1580 pump, an AS2050 Plus autosampler (100 μ L sampling loop) and a UV-1575 UV-visible detector. Integration of the peaks was performed with the program Borwin 1.5 for PC (JMBS Developments). Separation and determination of the proportions of *R*- and *S*-ibuprofen enantiomers were achieved with a chiral AGP column and UV detection at 220 nm. The mobile phase was 990 mL phosphate buffer with orthophosphoric acid (pH 6.9)/10 mL acetonitrile/204 μ L N, N-dimethyloctylamine at a flow rate of 1.5 mL/min.

Under these conditions *R*- and *S*-ibuprofen enantiomers had retention times of 2.2 and 3.5 min respectively. Resolution determined by the USP 29 method was of 2.1. Figure 1 shows the chromatograms of standards of racemic ibuprofen (Fig. 1a), pure *S*-ibuprofen (Fig. 1b) and a representative test sample (Fig. 1c). Linearity of the method was studied in the range between 0.25 and 30 μ g/mL of *R*- and *S*-ibuprofen (0.25, 0.5, 1, 5, 10, 15, 20 and 30 μ g/mL) using as reference compounds racemic ibuprofen and pure *S*-ibuprofen supplied by Shasun (India). The results were analysed by linear regression, and correlation coefficients obtained were always at least 0.98 for both enantiomers. The lower quantification limit was 0.25 μ g/mL for both enantiomers with relative standard deviation of 7.4 and 7.0% for the *R*- and *S*-ibuprofen enantiomers respectively. Accuracy (tested at 1, 15 and 30 μ g/mL) ranged from 96.2 to 102.4%. Inter-day precision during in-study validation (tested at 0.75, 7.5 and 22.5 μ g/mL) was not higher than

4.5 and 5.6% for the *R*- and *S*-enantiomers respectively. Ibuprofen concentration data reported as racemic were obtained as the sum of the two enantiomers.

Pharmacokinetics

The estimation of the pharmacokinetic parameters was performed from the plasma concentration-time data by non-compartmental analysis using WinNonlin Professional, version 2.0 (Pharsight, USA). $C_{\max}$ and $t_{\max}$ were obtained directly from the plasma concentration-time data. The AUC to the last time with measurable concentration (C_t) of the analyte (AUC_{0-t}) was estimated by using the linear trapezoidal rule. The apparent elimination rate constant (k_e) was calculated by least squares log-linear regression from data of the last sample time points of each plasma concentration-time curve. The AUC extrapolated to infinity ($AUC_{0-\infty}$) was determined by adding the extrapolated area (C_t/k_e) to the AUC_{0-t} .

Statistical analysis

The two-way analysis of variance (ANOVA) for crossover design was used to assess the effect of formulations, periods, sequences and subjects within sequence on logarithmically transformed data of AUC_{0-t} , $AUC_{0-\infty}$ and $C_{\max}$ [16]. The parametric analyses of the data were performed with WinNonlin Professional. The comparison of enantiomers within each formulation was carried out with a similar methodology, considering *S*-enantiomer as test and *R*-enantiomer as reference.

Results

Figures 2 and 3 show the plasma concentration-time profile of total ibuprofen (sum of *S*- and *R*-enantiomers) and that of the *S*- and *R*-enantiomers of both formulations respectively.

Fig. 1 HPLC chromatograms of standard racemic ibuprofen (**a**), *S*-ibuprofen (**b**) and a representative test sample (**c**)

Fig. 2 Mean plasma concentration-time curve ($n=18$) of racemic ibuprofen after oral administration of 600 mg of ibuprofen in the reference and test formulations. Vertical bars show standard error of the mean

The bioequivalence analyses based on the original non-chiral bioanalytical method, the sum of the enantiomers and the individual *S*- and *R*-enantiomers are summarised in Table 1.

The test product showed bioequivalence to the reference product for $C_{\max}$ and AUC when considering either the non-chiral concentrations or the sum of enantiomers. However, the $C_{\max}$ of the eutomer (*S*-ibuprofen) and the AUC of the distomer (*R*-ibuprofen) were not bioequivalent.

The mean relative bioavailability of the sum of enantiomers of the test product was 13% higher for AUC and 10% lower for $C_{\max}$ compared to the reference one, and these differences were statistically significant since their 90% confidence intervals did not include 100%. The mean relative bioavailability of the eutomer of the test product was 1% higher for AUC and 17% lower for $C_{\max}$ than the reference one, and the difference was significant for $C_{\max}$. Conversely, the mean relative bioavailability of the distomer of the test product was 30% higher for AUC and 0.4% lower for $C_{\max}$, and the difference in AUC was significant. Obviously, the point estimates of AUC and $C_{\max}$ of the racemate are located in between those of the enantiomers.

Fig. 3 Mean plasma concentration-time curve ($n=18$) of ibuprofen enantiomers after oral administration of 600 mg of ibuprofen in the reference and test formulations. Vertical bars show standard error of the mean

The ratio of *S*-enantiomer to *R*-enantiomer in each product for the main pharmacokinetic parameters, presented in Table 2, shows that *S*-ibuprofen is the overall prevalent enantiomer in plasma and those ratios were higher with the reference product. Moreover, the *S/R* enantiomer ratio changed over time and showed a different profile between products, with higher eutomer/distomer ratio in the formulation with the faster rate of absorption (Fig. 4).

Discussion and conclusions

The test product showed a slower rate of absorption, with more than 30 min delay in $t_{\max}$, despite a similar $C_{\max}$, and its marketing authorisation was rejected since, for ibuprofen, onset of action is considered clinically important. However, this criterion is somewhat subjective, and it might have been approved since AUC and $C_{\max}$ were bioequivalent, and alternative metrics such as partial AUC [17] are not used due to their high variability. Furthermore, if this drug had been one for which onset of action was not clinically relevant, the test product would have been approved based solely on AUC and $C_{\max}$. Figure 3 and Table 1, however, show that the difference in $C_{\max}$ of the eutomer is larger and its 90% confidence interval lies outside the bioequivalence limits, which supports the rejection of the application. Importantly, this illustrates that even when $C_{\max}$ is equivalent there is room for large differences in rate of absorption because $C_{\max}$ is not a sensitive metric for detecting differences in rate of absorption [18, 19]. These differences in rate of absorption may produce very different enantiomer ratios in products containing drugs with non-linear absorption as defined by the FDA [2].

Interestingly, the AUCs of the eutomer (*S*-ibuprofen) were more similar between products, but the AUCs of the distomer (*R*-ibuprofen) were more different when compared with the original non-chiral analysis or the sum of enantiomers. This could be considered as clinically irrelevant but shows a different biopharmaceutical quality.

The logical observation that the point estimates of the sum of enantiomers is between the point estimates of the enantiomers demonstrates again [3] that the results of chiral assay may be biased even when the eutomer is not the minor one. The difference in eutomer exposure is detected with less accuracy by non-chiral assay. This 6.5% shift in the $C_{\max}$ point estimate changed the conclusion of the study despite the fact that the eutomer is the enantiomer with a higher exposure (20% in case of the generic and 50% in case of the reference since it is formulation dependent). The failure to conclude equivalence does not seem to be due to a higher variability since the range of the 90% CI of $C_{\max}$ is similar. In summary, data generated using non-stereospecific assays may not be

Table 1 Results of the bioequivalence analyses of test and reference formulations obtained by non-chiral bio-analytical method and chiral bio-analytical method (*S*- and *R*-ibuprofen)

	Non-chiral	Sum of enantiomers	<i>S</i> -enantiomer	<i>R</i> -enantiomer
AUC _{0–t_s} , point estimate (%)	111.29	113.27	101.16	130.03
AUC _{0–t_s} , 90% CI (%)	107.23–115.49	105.73–121.35	96.67–105.86	118.86–142.24 ^a
AUC _{0–∞} , point estimate (%)	111.39	113.32	101.57	129.06
AUC _{0–∞} , 90% CI (%)	107.29–115.65	105.98–121.17	97.18–106.16	118.19–140.92 ^a
C _{max} , point estimate (%)	89.18	89.85	83.36	99.65
C _{max} , 90% CI (%)	82.77–96.09	82.19–98.23	76.05–91.36 ^a	87.84–113.05
t _{max} of the test (h)	2.0	2.5	2.5	2.25
t _{max} of the reference (h)	0.75	0.875	1	0.75

Test/reference ratios for AUC and C_{max} based on log-transformed data are expressed as percentages

^a Non-bioequivalent results

extrapolated to explain the exposure of the individual enantiomers [20].

Table 2 illustrates the concept of an enantiomer exposure ratio that is formulation dependent or absorption rate dependent. The *S*/*R* AUC ratio showed a 34% difference (154 vs. 120%) and the *S*/*R* C_{max} ratio a 20% difference (121 vs. 101%) between products. In our previous study [3] differences were smaller, 10% in AUC (123 vs. 114%) and 6% in C_{max} (112 vs. 106%).

This study reminds that it is important to distinguish between non-linear kinetics (dose- or time-dependency) and non-linear absorption, as defined by the FDA [2], which may occur even in cases of linear kinetics, since ibuprofen shows non-linear kinetics due to saturation of protein binding only at doses greater than 600 mg. However, the inversion of *R*-ibuprofen to *S*-ibuprofen [21] is responsible of this non-linear absorption. This is consistent with the observations by Boni et al. [5], who showed that non-linear absorption may occur in a drug with linear pharmacokinetics in both the racemate and the individual enantiomers such as etodolac.

It could be considered that ibuprofen enantiomers show similar pharmacokinetic characteristics based on selected literature data. For example, in our previous work the *S*/*R* ratio was shown to be bioequivalent in that reference formulation (Neobrufen 600 mg tablet) [3]. However, in the present study, the reference product (Dalsy 100 mg/5 mL oral suspension) showed a point estimate of 121% for C_{max} and 154% for AUC, suggesting that the kinetics was not similar between the two enantiomers. Therefore, the similarity of pharmacokinetics between enantiomers should not be considered a compelling justification to use non-chiral bioanalytical methods, unless this similarity is thoroughly proven at a large range of different absorption rates.

Based on the results of this investigation it can be concluded that only two of the four FDA conditions have to be fulfilled to require a chiral bioanalytical method in bioequivalence studies. Measurement of individual enantiomers in bioequivalence studies should be recommended when (1) the enantiomers exhibit different pharmacodynamic characteristics, and (2) non-linear absorption is present, as expressed by a change in the enantiomer concentration ratio

Table 2 AUC and C_{max} data comparison regarding the enantiomer and drug formulation

	<i>S</i> / <i>R</i> ratio in test	<i>S</i> / <i>R</i> ratio in reference
AUC _{0–t_s} , point estimate (%)	120.71	154.39
AUC _{0–t_s} , 90% CI (%)	113.51–128.37	146.13–163.12
AUC _{0–∞} , point estimate (%)	122.38	154.58
AUC _{0–∞} , 90% CI (%)	115.24–129.95	146.28–163.35
C _{max} , point estimate (%)	101.05	121.38
C _{max} , 90% CI (%)	95.48–106.93	114.95–128.18
t _{max} of <i>S</i> -enantiomer (h)	2.50	1.00
t _{max} of <i>R</i> -enantiomer (h)	2.25	0.75

Fig. 4 Mean *S*/*R* enantiomer plasma concentration ratio-time curve ($n=18$) after oral administration of 600 mg of ibuprofen in the reference and test formulations (calculated as mean of the individual ratios per time-point). Vertical bars show standard error of the mean

with change in the input rate of the drug, for at least one of the enantiomers.

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Conflict of interest None

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