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Supercritical fluid extraction of 4(5)-methylimidazole (4MeI) and 2-acetyl-4(5)-(1,2,3,4)tetrahydrohydroxybutylimidazole (THI) from grounded coffee with liquid chromatographic - electrospray mass spectrometric (HPLC/ESI-MS) quantification

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Supercritical fluid extraction (SFE) of 4(5)-methylimidazole (4MeI) and 2-acetyl-4(5)-(1,2,3,4-tetrahydroxybutyl)-imidazole (THI) from ground-coffee with liquid chromatographic-electrospray mass spectrometric quantification (HPLC/ESI-MS)

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Abstract

Two polar analytes, 4(5)-methylimidazole (4MeI) and 2-acetyl-4(5)-(1,2,3,4-tetrahydroxybutyl)-imidazole (THI) were extracted with supercritical CO₂ modified with aqueous methanol. The method was applied to a roasted coffee powder with good recoveries. Method efficiency was compared with that of solid phase extraction using SCX Disc cartridges and validated for spiked solid matrix. The analytes were determined using isocratic liquid chromatography – mass spectrometry (LC/MS) on an Atlantis HILIC Silica column (150 x 2.1 mm, 3 µm) with 80 % methanol and 20 % 0.01 mol.l⁻¹ ammonium formate as mobile phase. The limit of quantification was around 1.5 pg for 4MeI and 2.0 pg for THI. The linearity of the calibration curves was satisfactory as indicated by correlation coefficients of > 0.999. The coefficient of variation for the intra-day and the inter-day precision was < 4% (n = 6). The accuracy was in the range 98 % - 101% and recovery was ≥ 98% and ≥ 99% for THI and 4MeI, respectively. Several samples of Arabica coffee from various locations and commercially available “shelf” coffee products (Arabica/Robusta mixtures) were analyzed to test the method.

Keywords: supercritical fluid extraction, solid phase extraction, 4(5)-methylimidazole, 2-acetyl-4(5)-(1,2,3,4-tetrahydroxybutyl)-imidazole, 4MeI, THI, LC/MS, roasted coffee

Introduction

Coffee is well known as a stimulant of a central nervous system and as a diuretic. Its world production is about 4 millions tons per annum. Small, evergreen woods of *Coffea arabica*, *C. liberica* and *C. canephora* (*Rubiaceae*) are planted in many varieties in tropic areas all over the world. They bear coffee seeds (*Semen Coffeae*) containing 0.3 – 2.5 % of caffeine and traces of theophylline – important natural drugs - purine derivatives originated from xanthine methylation (Tomko 1999, Reeve et al. 1993). Ripe coffee seeds are rid of the pulp, endocarp and seed coat in so-called wet or dry way. Finally the selected grains are roasted at 200 – 250 °C to get typical coffee aroma and taste.

During the roasting process, undesirable imidazole compounds – a neurotoxic 4(5)-methylimidazole (4MeI) and an immunosuppressive 2-acetyl-4(5)-(1,2,3,4-tetrahydroxybutyl)-imidazole (THI) – are formed by heating saccharides with nitrogenous compounds. THI has a lymphopenic effect and impairs immunity of rats and mice by modifying splenocyte function, in a histamine-like receptor site (Vávrová et al. 2003). THI also restrains contact hypersensitivity. Mice fed with THI have less CD4+ and CD8+ T cells – THI probably inhibits a contact hypersensitivity response by preventing the recruitment of CD4+ T cells in the draining lymph nodes (Gugasyan et al. 1995).

It is impossible to avoid completely the THI and 4MeI formation during the coffee roasting and other food industry processes, but the concentration of both compounds should be kept as low as possible in foods and beverages. For example, the European Commission specified the legal limit of 10 and 200 mg.kg⁻¹ for THI and 4MeI, respectively, for Class III caramel colour.

Several procedures have been published for separate determination of either 4MeI or THI in 1986 – 95 period. They have been based on LC/UV and LC/MS (Thomsen and Willumsen 1995, Moravcová et al. 1992, Müller et al. 1998, Wagner et al. 1993, Lawrence and Charbonneau 1987, Ding et al. 1991), capillary electrophoresis (Ong et al. 1994), and GC with derivatisation (Kröplien 1986, Fernandes and Ferreira 1997). Quantification of 4(5)-methylimidazole in coffee was performed via ion-pair extraction with bis-2-ethylhexylphosphate (BEHP), isobutylchloroformate derivatisation and GC/MS (Casal et al. 2002). These methods are sensitive and precise, but limited to one or the other of the two compounds. The first successful attempt to separate and determine 4MeI and THI in a single run via LC/MS was published in 2003 (Klejdus et al. 2003). The same research group also investigated solid phase extraction as an instant tool for simultaneous separation of both

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2 68 analytes from liquid matrices (Klejdus et al. 2006). Both analytes can be separated and
3 69 determined in a short time using a single-run analysis.

5 70 Solid matrices present another challenge in THI/4MeI separation. Roasted coffee can
6 71 be, of course, turned to liquid “coffee drink” matrix. A set of spike experiments showed, that
7 72 although the recovery of spiked analytes from liquid solutions is quantitative, spiking the
8 73 standards to roasted coffee beans and subsequent treatment with hot water brought less
9 74 favourable results. SPE is able to extract all analytes from the liquid phase, but the hot water
10 75 soaking seemingly does not remove all spiked analytes from solid coffee matrix (Klejdus et
11 76 al. 2006).

15 77 Thus, supercritical fluid extraction was selected as a method for THI and 4MeI
16 78 isolation from powdered solid coffee matrix. Due to low viscosity and high diffusivity of
17 79 supercritical fluid, analytes can be extracted from the matrix most effectively. Apart from
18 80 that, analytes are carried out of the system during the dynamic extraction. The balance in the
19 81 extraction cell shifts and it causes faster analyte removal from the matrix. Extraction
20 82 efficiency and precision was validated in a series of spiked sample experiments. Reminders
21 83 of coffee samples after supercritical fluid extraction were soaked in hot water and extracted
22 84 with SPE to verify SFE efficiency. Also, several different varieties and consumer coffee
23 85 mixtures were tested to exclude possible matrix effect.

28 86 Together with new isolation method, a new LC/MS method was developed for the
29 87 simultaneous determination of THI and 4MeI. Method used in previous experiments (Klejdus
30 88 et al. 2003, 2006) works well. But it is quite costly, as the pH of analyses is a limiting pH
31 89 value of the MetaChem Polaris C-18A column applicability. The column is exposed to a
32 90 series of serious stresses and after a limited number of analyses, separation performance is
33 91 decreased: retention times are shifted and peak width broadened. Because of that, brand new
34 92 method for Atlantis™ HILIC Silica column was developed for a mildly acidic mobile phase
35 93 in pH range 5 - 6, with excellent precision and reproducibility.

40 94 A hydrophilic interaction chromatography (HILIC) is a variant of a normal phase
41 95 chromatography, where polar stationary phase is used with a mobile phase, which contains
42 96 high concentration of organic solvent and low concentration of aqueous solvent. In HILIC,
43 97 organic portion of mobile phase is a weak solvent, the aqueous portion is a strong solvent and
44 98 the compound elution is in the order of increasing hydrophobicity. HILIC is also referred to as
45 99 “aqueous normal phase” or “reverse reversed phase”, since the elution order is similar to that
46 100 of normal phase and used solvents are similar to those of reversed phase chromatography.

HILIC columns retain highly polar analytes with volatile mobile phases that are ideal for compound ionization by ESI – MS. This leads to good sensitivity and low limits of detection.

Materials and methods

Chemicals

4(5)-methylimidazole (4MeI) was purchased from Sigma-Aldrich (St Louis, USA). The standard of 2-acetyl-4(5)-(1,2,3,4-tetrahydroxybutyl)-imidazole (THI) was obtained from Dr. D. V. Myers (Coca Cola Co., USA). HPLC grade methanol and ammonium formate were from Merck (Darmstadt, Germany). Standards were prepared by dissolution in reverse-osmosis Milli-Q water (Millipore, Bedford, USA) at a stock concentration of 10 µg.ml⁻¹. The solutions were stable for at least one week when stored in dark at 4 °C. All solutions were filtered through a 0.45 µm Teflon membrane filters (MetaChem, Torrance, USA) prior to HPLC analysis. All other chemicals were of analytical grade.

Samples and sample preparation

The samples of coffee varieties were obtained from the Oxalis (import comp., Zlín, Czech Republic). The „Brazilia-Santos“ was the well-known 100 % brazil Arabica. The „Haiti 5X“ contained extra large coffee beans from the Haiti region Pilate. The „Kenya“ originated from the Mount Kenya plantations with altitude over 2000 m. The „Mexico Altura“ was a high mountain Arabica harvested at mexican coffee plantations in altitude 1000 – 1600 m. The „Boncampo“ was obtained from Migros-Genossenschafts-Bund (Zürich, Switzerland). It was a commercially available coffee mixture containing Arabica from Central and South America and also from India/Indonesia, and Robusta from Africa and Asia – the most complex mixture available at the market. Except the Boncampo mixture (Arabica/Robusta), all samples were pure Arabica. All samples were used as the whole beans and ground in an analytical grinder IKA Yellow Line A10 (IKA Works Inc., Wilmington, USA) to assure similar size of particles in all samples.

In supercritical fluid extraction, sample amount was 1 g. For the solid-phase extraction (Klejduš et al 2006), 5 g of coffee was extracted with hot water (90 ml) at 90 °C for 5 min. The mixture was then cooled and solids were filtered off. The clear filtrate was transferred into a 100 ml volumetric flask and adjusted with water up to the mark. An aliquot (1000 µl) was diluted with water (3000 µl) and extracted.

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134 *Supercritical fluid extraction*

135 The standard solutions of THI and MeI were prepared by mixing 1 mg of each compound in
136 100 ml of ultra-pure water. A given volume of THI/4MeI solution was injected into the
137 middle of a matrix bed (glass wool). As for real samples, 1.00 g of roasted coffee powder was
138 weighted into the extraction cell. The coffee samples spiked with the THI/4MeI standard
139 solutions were left to equilibrate for at least 10 min (5 days usually) at laboratory conditions.
140 The procedure avoided immediate dissolution of freshly spiked analytes in too soon added *in-*
141 *situ* modifier. An attempt was made to use higher amounts of *in-situ* modifier; but at higher
142 temperatures the solvent caused some pressure in the extraction cell, leaved through the
143 restrictor and caused slow extraction of ballast compounds, impending the restrictor plugging.
144 When 200 µl of the *in-situ* modifier was added to the 0.7 ml cartridge, it was immediately
145 sealed and closed in the extraction cell of a heating block to avoid evaporation of the *in-situ*
146 modifier.

147 The extraction cells were cleaned in ultra-pure H₂O in sonication bath and rinsed with
148 MeOH. Inner space of extraction cell of the heating block was also washed with water and
149 methanol, since not all experiments had quantitative recovery during the extraction conditions
150 optimisation. This eliminated or minimised cross-contamination dragged by the extraction
151 phase modifier during de-pressurization after the extraction step. Blank samples were
152 prepared by extracting the same amounts of the matrix. Neither the analytes were found in
153 blank matrices, nor THI/4MeI residues were found in the instrument after the extraction cell
154 heating block cleaning.

155 All extractions were carried out using an SE-I instrument (SEKO-K, Ltd., Brno, Czech
156 Republic) equipped with a valve for continuous modifier addition (VALCO, Switzerland,
157 loop 500 µl) according to the method described by Ventura (Adam et al. 2000, Eisner et al.
158 2002). The extraction cell was pressurized at 50 to 150 °C up to 10 - 40 MPa using SFC/SFE
159 grade CO₂ (Siad, Brno, Czech Republic) with 10 % v/v of modifier mixture added via
160 VALCO valve to the instrument piston pump.

161 The modified supercritical fluid went through the extraction cartridge filled with the
162 sample for 45 min (dynamic extraction). The cell was de-pressurized through a fused silica
163 restrictor (12 cm, 50 µm i.d., flow rate of 750 - 850 ml.min⁻¹, Supelco) into a liquid trap - a
164 glass vial containing 30 ml trapping solvent at the laboratory temperature. Much smaller
165 amount of the trapping solvent would be sufficient for the quantitative analyte trapping; but
166 due to 800 ml/min flow rate a 100 ml flask with a flexible foil seal was used as a trapping vial

167 to prevent the solvent splashing out of a vial. Thus a larger volume was needed to achieve
168 sufficient solvent column height.

169 Prior to the analyses, the extracts were evaporated to dryness in a rotary vacuum
170 evaporator IKA RV 05-ST with a water bath HB 4 (all from IKA – Werke GmbH and Co.,
171 Staufen, KG, Germany), dissolved in 500 μl of ultra-pure water and injected directly into the
172 HPLC/MS system (spiked samples) or filtered through the 0.45 μm membrane filter
173 (MetaChem) before the injection (real samples, spiked real samples) to protect the HPLC
174 system.

176 *Solid-phase extraction*

177 Ansys SPEC SCX Disc 15mg/3ml (Ansys Technologies, Torrance, USA) SPE cartridges,
178 tried-and-true in previous analyses (Klejduš et al. 2003), were used for all solid-phase
179 extractions. The cartridges were conditioned with 500 μl methanol and 500 μl water. Usually
180 3 ml samples with 20 μl 0.1 mol.l^{-1} HCl passed through the cartridge. Impurities were washed
181 out with 1000 μl methanol. Retained imidazoles were eluted from discs with 2 ml of methanol
182 acidified with 5 mol.l^{-1} HCl v/v (3:1) and extracts were evaporated to dryness in the rotary
183 vacuum evaporator at 39 °C. The residues were dissolved in 500 μl of Milli Q water and 10 μl
184 volumes were injected directly onto the reverse phase chromatographic column.

186 *High-performance liquid chromatography*

187 The HPLC chromatographic system HP 1100 (Hewlett Packard, Waldbronn, Germany)
188 equipped with a vacuum degasser (G1322A), a binary pump (G1312A), an autosampler
189 (G1313A), a column thermostat (G1316A) and a diode array detector (model G1315A) was
190 coupled on-line to a mass selective HP MSD detector (G 1946A, Hewlett Packard, Palo Alto,
191 USA). The ChemStation software (Rev. A 07.01) controlled the whole liquid
192 chromatographic system. A hydrophilic interaction chromatographic column Atlantis HILIC
193 Silica (150 x 2.1 mm, 3 μm) was employed. The mobile phase was methanol (80 %) and 0.01
194 mol.l^{-1} ammonium formate in Milli-Q water (20 %). Isocratic mode was used for the
195 separation. The flow rate was 0.2 mL.min^{-1} and the temperature of the column oven was set at
196 25 °C. The column effluent was monitored with the diode array detector at 290 nm for THI
197 and at 215 nm for 4MeI.

199 *Mass spectrometry*

The eluent from the liquid chromatograph was directly introduced into the quadrupole mass spectrometer operated in the positive ESI mode. The nebulizer gas pressure was 50 psi, the drying gas was nitrogen at the 12 L.min⁻¹ and at the temperature of 350 °C. The capillary voltage was 1 500 V. The gain was 1. The spectra were recorded from m/z 50 to 250 for full scan mode. All experiments were performed at the fragmentor voltage 35 eV for molecular ions. For qualitative analysis, ion ratio experiments were performed for standard solutions and real samples at the fragmentor voltage 65 eV for parent ions and m/z 231/213, m/z 231/195 and 231/153 for THI and at the fragmentor voltage 100 eV for the parent ion and m/z 83/56 for 4MeI. The dwell time for each analyte was 199 ms. The mass spectrometer was calibrated with an ESI tuning solution obtained from Hewlett Packard (Palo Alto, CA, USA).

A sequence 231 – 213 – 195 appears in the mass spectrum of both THI tautomers that corresponds to stepwise liberation of a single molecule of water from the side chain. The fragment 195 liberates ketene from the double bond of the acetyl group forming an ion 153 as can be seen in both spectra. The sequence is the same for both isomers. In addition the tautomer liberates another water molecule from the fragment 195 forming an ion 176. Because of the water liberated preferentially through six-member cyclic intermediate, it looks like tautomer depicted in Figure 1 (right side).

Accuracy, precision and recovery

Accuracy, precision and recovery of the determination of imidazole analytes were evaluated with the powdered coffee beans spiked with THI/4MeI standards (concentrations varying from 0.5 µg.g⁻¹ to 3 µg.g⁻¹). Coefficients of variation % C.V. of intra-day assay were performed in 6 spiked extracts. Inter-day precision was determined by analysing six spiked extracts during a 6-day period. Analyte concentrations were calculated from calibration curves. Accuracy (in %) was evaluated by comparing the estimated concentration with the known concentrations of the individual imidazoles.

Results and discussion

High-performance liquid chromatography – mass spectrometry

A new LC/ESI/MS analytical method, using Atlantis HILIC Silica column and mildly acidic medium (pH 5 – 6) was developed for unambiguous identification and determination of THI and 4MeI. Inherent pH and thermal instability of silica-based column, like MetaChem Polaris C-18A used in previous paper (Klejduš et al. 2003) significantly limits the range of

chromatographic conditions, under which such columns can be used, especially under reversed phase conditions. In the THI/4MeI analyses, the column was operated at edge conditions of pH applicability that decreased its lifetime and led to retention times shift and peak width broadening.

The optimisation was carried out by flow injection analysis (FIA) injecting several solutions containing the individual compounds at the concentration of $1 \text{ ng} \cdot \mu\text{l}^{-1}$. The analytes produced protonated molecules $[\text{M}+\text{H}]^+$ at m/z 231 for THI and at m/z 83 for 4MeI. The tandem mass spectrometry was therefore used in order to obtain additional structural information by detecting diagnostic product-ions obtained by in-source collision-induced dissociation (CID) of the precursor ion. Three qualifying fragment ions (m/z 231/213, m/z 231/195 and m/z 231/153) for THI and one qualifying fragment ion (m/z 83/56) for 4MeI were used, according to the selected-ion monitoring (SIM) technique.

Validation of the LC/ESI-MS method

Several experimental parameters such as limit of detection (LOD), limit of quantification (LOQ), linearity, recovery, stability and repeatability were evaluated for validation of the HPLC/ESI-MS method. The separated HPLC zones were detected by the selected ion-monitoring (SIM) mode (m/z 231/213, m/z 231/195 and m/z 231/153 for THI and m/z 83/56). The limit of detection (LOD) was estimated to be around 1.5 pg for 4MeI and at 2.0 pg for THI with calculated $\text{S/N} = 3$ (see Table 1), while the limit of quantification (LOQ) was 4.4 pg for 4MeI and 6.3 ng ml^{-1} for THI, respectively ($\text{S/N} = 10$). Molecular protonated ions were used for LOD and LOQ determinations. The calibration curves were linear in the concentration ranges $1 - 70 \text{ ng ml}^{-1}$ for 4MeI and $1 - 60 \text{ ng ml}^{-1}$ for THI with correlation coefficient $r > 0.9993$ for 4MeI and $r > 0.9996$ for THI, respectively. Very good results were obtained for run-to-run repeatability of retention times, peak areas and concentrations of 4MeI and THI (see Table 1). The data shown in Table 2 were achieved by injecting the standard solution of 4MeI and THI to the column in six replicates, in order to test the intra-day variation.

The inter-day variation of the method was assessed by analysing of the standard mixtures containing 1 and $3 \text{ } \mu\text{g} \cdot \text{g}^{-1}$ of both 4MeI and THI injected independently during 6 days in six replicates. The inter-day accuracy for the standard solution at the concentration $1 \text{ } \mu\text{g} \cdot \text{g}^{-1}$ was found to be 101.5 and 100.6 %, for 4MeI and THI, respectively, whereas that of the standard solution at the concentration of $3 \text{ } \mu\text{g} \cdot \text{g}^{-1}$ was found to be 99.1 and 100.2 %, respectively. The precision of the inter-day analysis, expressed as RSD%, was calculated as

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2 268 1.66 and 0.60 % for 4MeI and THI, respectively, in the 1 $\mu\text{g.g}^{-1}$ standard solutions, and 1.57
3 269 and 0.7 %, respectively, in the 3 $\mu\text{g.g}^{-1}$ standard solutions (see Table 2).
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5 270 The recovery of the method was verified analysing roasted coffee beans samples after
6 271 addition of two known amounts of 4MeI and THI standard solution and subjected to the
7 272 previously described SPE extraction after equilibration (ageing for 5 days). The theoretical
8 273 concentrations of the spiking 4MeI and THI standards were 1 and 3 $\mu\text{g. g}^{-1}$. The mean
9 274 recovery and the standard deviations for six replicates are given in Table 3.
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14 276 ***Supercritical fluid extraction***

15 277 *Optimisation of extraction conditions for imidazoles*

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17 278 Both analytes, THI and 4MeI, are quite polar compounds for an SFE context (see Fig. 2).
18 279 Quantitative extraction of polar molecules requires long extraction time or very high flow-rate
19 280 to ensure sufficient mass transfer. Also polarity of a supercritical fluid must be shifted by
20 281 means of a modifier addition, because pure CO_2 has solvation power close to hexane and is an
21 282 excellent solvent for non-polar compounds.
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24 283 Several sets of spiked sample experiments were performed to find optimal extraction
25 284 conditions for both analytes. An instrument adaptation for high CO_2 flow-rate, developed for
26 285 isoflavone extraction (Klejdus et al. 2003) was used, providing the gaseous phase flow-rate of
27 286 c. 750 – 850 ml.min^{-1} and using liquid trapping into foil-sealed 100 ml vial with 30 ml of
28 287 trapping solvent. Trapping efficiency and expected stripping effect was further tested.
29 288 Fortunately, non-volatile analytes are trapped quantitatively and stripping effect was not
30 289 observed in the first 45 min of any extraction. RSDs of all spiked analyte experiment were in
31 290 the range 1,2 % – 4,7 %.

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33 291 The best extraction conditions were examined in several sets of experiments. First,
34 292 recovery vs. extraction time was examined at 40 MPa and 150 $^{\circ}\text{C}$ (see Figure 3). As can be
35 293 seen the 55 % of THI and 74 % of 4MeI were extracted in the first 10 min and the 30 min
36 294 extraction was nearly quantitative (98.4 % and 98.1 % for THI and 4MeI, respectively).
37 295 Recoveries increased slightly up to 98.9 % and 99.8 % for THI and 4MeI in the 45th min,
38 296 respectively (see Figure 3). Fractions were taken also for 60, 75 and 90 min, but they did not
39 297 bring any measurable improvement of the results.

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41 298 Study of extraction kinetics was repeated with a real sample of roasted coffee powder
42 299 due to expectation of matrix effects. Courses of extraction kinetic curves were similar. In fact
43 300 the real sample extractions were faster a little bit, partly due to the lower analyte content in
44 301 real samples and, mostly, thanks to better-tuned modifier composition.
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With the optimal extraction time selected, extraction temperature and pressure were optimised (see Figure 3). Series of extraction was performed at 40 MPa for different temperatures and at 150 °C for different pressures. As for THI, the higher the temperature is, the better (150 °C is an instrument limiting value, higher temperatures were not examined). The decrease of extraction efficiency was observed for 4MeI at 75 and 100 °C. The highest recovery was also obtained at 150 °C in this case. Recoveries vary from 86 % to 99 % for THI and from 69 % to 89 % at 150 °C, according to extraction pressure applied (with modifier composition best suitable for both analytes, see Figure 3). The best recovery values were obtained at 40 MPa in both cases.

Ultra-pure water, methanol, ethanol, acetonitrile and their combinations were tested in preliminary experiments to find the best modifier composition (2, 5 and 10 % (v/v) of each were added on-line). An aqueous methanol with 10 % (v/v) of water gave the best results thus the concentrations close to that amount were examined more carefully (see Figure 3). The proper modifier composition for THI was aqueous methanol with MeOH/H₂O 5:1 (v/v) ratio or with higher percentage of water (see Figure 3). Lower percentage of water in aqueous methanol, i.e. MeOH/H₂O 9:1 (v/v), is favourable for 4MeI. The recovery of the extraction decreased with the higher amount of water. However, the modifier composition MeOH/H₂O 5:1 (v/v) provided recoveries 101.3 % and 89.4 % for THI and 4MeI, respectively. These values could be considered sufficient for their simultaneous extraction (quantitative recoveries of polar analytes are difficult to achieve with SFE and nearly 90 % recovery at conditions optimal for another compound seems to be quite satisfactory). THI is also extractable quite well (with recovery 86.9 %) under the conditions optimised for the 4MeI quantitative extraction (MeOH/H₂O 9:1 (v/v), recovery of 4MeI 98.9 %). The optimised composition MeOH/H₂O 5:1 (v/v) was selected for the simultaneous extractions of THI and 4MeI for further experiments due to its higher total recoveries. In experiments, where 4MeI is the more important analyte, 9:1 (v/v) concentration will be more useful.

Another set of extractions was performed to find out the best volumes of on-line modifier. The initial amount of the modifier added to the extraction cartridge was 200 µl. The higher volumes caused troubles during the extraction cell heating and leaved the cell trough the restrictor together with some ballast compounds. The concentration of a modifier in the extraction fluid depends on the amount of a liquid modifier added to the extraction phase during every filling of the extractor pump. Up to 1 ml of modifier can be added through the Valco valve loop. That is c. 10 % of the piston micro-pump volume. The modifier percentage is viable only for some substances and their combinations. Ultra-pure water, for example

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2 336 tends to tarry in the piston micro pump. It is accumulated there and decreases piston micro
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4 337 pump inner space. To avoid this, the modifier addition must be skipped for each n^{th} pump
5 338 filling (according to the modifier type and its volume). Thus the maximum concentration of a
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7 339 modifier in an extraction fluid is lower. However, the modifier concentration MeOH/H₂O 5:1
8 340 (v/v) can be added in 10 % (v/v) amount without any troubles and it is also best composition
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10 341 for 4MeI (see Figure 3). For THI, 5 % (v/v) is sufficient and any higher concentration is
11 342 suitable.
12 343 Last but not least was the determination of a suitable trapping solvent (see Figure 3).
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14 344 Methanol is not good for THI trapping (recovery 85.6 %); acetonitrile is an excellent solvent
15 345 for both analytes (99.4 % and 98.2 %, for THI and 4MeI, respectively). Ultra-pure water is
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17 346 very good solvent for 4MeI (100.2 %), but poor for THI (38.7 %).

Thus, spiked imidazoles are extractable very well with the modified supercritical fluids with sufficient recoveries in an acceptable extraction time. Of course, all the optimised parameters were not known at the beginning of this study. Most of the experiments were repeated in a “second run” under the “optimised conditions”. In Figure 3, only the results of the optimised experiments are given. Temperature optimisation, trapping solvent selection and the study of extraction kinetics were performed separately for the modifier composition MeOH/H₂O 5:1 (v/v) for THI and MeOH/H₂O 9:1 (v/v) for 4MeI. All the other studies were performed together only with MeOH/H₂O 5:1 (v/v) modifier. The composition was used also for all the real sample extractions. Recovery verification of SFE was also performed for the both modifier concentrations to authenticate the best conditions for both compounds. The values for 4MeI extracted with MeOH/H₂O 5:1 (v/v) modifier were also calculated and found out to be satisfactory.

Real samples extraction and method validation

SFE is a matrix-dependable method, where every new procedure must be verified also by application to a set of different real samples. The optimised extraction conditions were applied to the samples of roasted coffee powder, containing THI and 4MeI. A solid phase extraction was used as a comparative method for the neurotoxins determination (Klejdus et al. 2003, 2006). The both analytes were isolated and quantified in the concentrations high above their detection limits. Despite the expected matrix effects the method seems to work well with the acceptable reproducibility. The samples of fine-grained coffee (1.00 g) were homogeneous enough to avoid measurable differences between individual SFE samples. Any differences in their extractability from various matrices were not observed.

The efficiency of the supercritical fluid extraction was tested in two ways: via comparison with validated SPE extraction and via post-SFE liquid extraction of the “extracted” samples. Comparison of the results of SFE and SPE is presented in Table 4. Noteworthy, the results of both extraction methods are in the good coincidence for THI, although SPE was reported non-quantitative with spiked solid samples. For post-SFE tests, five used samples of coffee matrix (c. 5 g) were combined in one beaker and “coffee” was prepared in the same way like for the classical SPE. No traces of the analytes were found in any test solution.

For validation purposes, the coffee samples were spiked with the standard solution adding 0.5, 1, 2 and 3 µg of the analytes per 1 g of the sample (details given in experimental

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2 380 section). All the samples were extracted under the optimised conditions for both analytes. The
3 381 values obtained for the more suitable modifier are given for each substance (see Table 5).
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5 382 Slightly higher selectivity was achieved with SFE, but the conditions necessary for the
6 383 quantitative extraction of the analytes are quite hard, and many ballast compounds were co-
7 384 extracted. Because of that, quick test of available clean-up procedures was performed. Finally,
8 385 0.45 µm Teflon filter was selected for the extracts clean up, as it didn't hold part of analytes.
9 386 Although the goal to avoid any extract clean up was not achieved, viable method was found
10 387 that can also be usable for SPE extracts. Thus, both SFE and SPE can be used for the
11 388 quantitative extraction of THI/4MeI from roasted coffee solid samples (SFE) and lixiviates
12 389 (SPE).
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18 391 **Conclusions**

19 392 The SFE method was optimised for the THI and 4-MeI separation from solid coffee
20 393 samples. Both analytes are quantitatively extracted and it is also possible to co-extract one
21 394 analyte together with the other with acceptable recovery (101.3 % THI and 89.4 % 4MeI or
22 395 98.9 % 4MeI and 86.9 % THI). Extraction time is moderate – 45 min – and both, the precision
23 396 and the recovery of analytical method are sufficient. The method is suitable for all matrices
24 397 and the spiked analytes recovery is quantitative.
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29 398 Thus, an alternative method exists now for THI and 4MeI determination from real
30 399 coffee samples. Although the method lacks some merits of a standard “low flow-rate” SFE –
31 400 30 ml extracts must be pre-concentrated and even a clean-up procedure is necessary in the
32 401 case of real samples – it is relatively fast, reliable and works well with spiked solid samples.
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35 402 An attempt was made to enhance environmental friendliness of the method by using
36 403 ultra-pure water as a trapping solvent, but acetonitrile proved to be the only quantitative
37 404 trapping solvent. THI trapping in water is insufficient.
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39 405 Still, the solid phase extraction represents, in fact, an instant separation, while SFE
40 406 takes at least 30 min for extraction and some extra time for pre-concentration. Our results
41 407 indicate, that SPE of real samples is as efficient as SFE. However, SFE was successfully
42 408 validated for spiked solid samples now. With Teflon filter clean up, both SFE and SPE are
43 409 viable methods. Perhaps, SFE present an advantage for small amount extractions and for
44 410 samples, from which quantitative lixiviation of analytes with common solvents would be
45 411 difficult.
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49 412 Although imidazole concentrations in coffee are very low, coffee imidazoles are one
50 413 part of a total food imidazole puzzle. They are formed when sugars are heated together with
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nitrogen containing compounds. Thus, all baked products could contain some imidazoles in their dark parts (Casal et al., 2002). In some cases, baking tins are sprinkled with flour or desiccated coconut (although probably not because of this reason) and the burned flour/coconut layer, containing most of harmful compounds, should stay in the baking tin. In addition to coffee, baked pastries, caramel sweets, and dark beers participate in the puzzle partly while cola-drinks quite seriously, because of both, larger concentrations of imidazoles and lower respect of producers to standard concentration limits (40-times higher than limits for imidazoles were found in several cola-drinks in our lab). Thus, each source of imidazoles should be carefully examined.)

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2 479 CAPTIONS TO FIGURES
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5 481 **Fig. 1** Probable chemical structures of THI tautomers
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8 483 **Fig. 2** HPLC/MS chromatograms (SIM mode for m/z 231 THI, for m/z 83 4MeI) of real
9 484 samples A, standards B. Peak identification 1 - THI, 2 - 4MeI. For chromatographic
10 485 conditions see text.
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14 487 **Fig. 3** Optimisation of extraction parameters. All extraction performed from spike samples
15 488 under following conditions: T = 150 °C, p = 40 MPa, t = 45 min, c (modifier) = 10 %
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17 489 (v/v), *in situ* modifier: 200 µl, trapping: 30 ml acetonitrile, ACF: modifier THI -
18 490 MeOH/H₂O 5:1 (v/v), 4MeI MeOH/H₂O 9:1(v/v), BDE: modifier MeOH/H₂O 5:1 (v/v)
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20 491 for both analytes. RSDs were in the range 1,2 % - 4,7 %.

Table 1: Retention time (R.T.), peak area (P.A.) and their repeatabilities (RSDs given in brackets, n = 6) and calibration data and LODs for 4MeI and THI obtained under optimised separation conditions using UV and MS detection (injection volume 5 µl)

	4MeI		THI	
	UV ^a	SIM	UV ^a	SIM
R.T. (min) ^b	11.25(0.09) ^b		6.94(0.14) ^c	
P.A. (x 10 ⁵) ^c	4.45(1.12) ^d		2.23(2.24) ^e	
Concentration range (ng.ml ⁻¹)	40 – 650	1 – 70	22 – 850	1 - 60
Correlation coefficient (n = 6)	0.9998	0.9993	0.9999	0.9996
LOD (ng.ml ⁻¹)	14.5 ^f	0.1	8.2 ^g	0.2
LOQ (ng.ml ⁻¹)	48.3 ^f	0.4	27.4 ^g	0.6

^a cit. Moravcová et al. 1999, ^b range 6.94 – 6.95, ^c range 2.12 – 2.30, ^d range 11.23 – 11.26, ^e range 4.39 – 4.53, ^f 215 nm, ^g 290 nm

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Table 2: HPLC-MS validation: determination of THI and 4MeI in real coffee extracts, standard solutions and coffee extracts spiked with known concentrations of analytes (n=6)

Compound	THI	4MeI	THI	4MeI	THI	4MeI	THI	4MeI
C [µg/g]	0.5		1		2		3	
Standard solutions of known analyte concentration								
C [µg/g]	0.505	0.498	0.993	0.991	1.973	2.021	3.048	3.030
SD [µg/g]	0.009	0.006	0.015	0.020	0.027	0.054	0.078	0.062
RSD [%]	1.76	1.19	1.51	2.04	1.38	2.69	2.55	2.05
R [%]	100.93	99.67	99.32	99.12	98.67	101.03	101.59	101.00
Ground roasted coffee extract spiked with known analyte concentration								
C [µg/g]	0.536	1.662	1.014	2.156	2.034	3.150	3.017	4.138
SD [µg/g]	0.005	0.019	0.012	0.024	0.010	0.005	0.032	0.026
RSD [%]	0.91	1.15	1.18	1.13	0.47	0.16	1.05	0.63
R [%]	100.48	101.06	98.14	100.50	100.04	100.17	99.44	99.83
Real sample (coffee Mexico)								
C [µg/g]	0.034	1.145						
SD [µg/g]	0.0005	0.0073						
RSD [%]	1.57	0.64						

[illegible]

Compound	THI	4MeI	THI	4MeI	THI	4MeI	THI	4MeI
	1 (intraday)		1 (interday)		3 (intraday)		3 (interday)	
Standard solution								
C [μg/g]	0.994	1.013	1.006	1.015	3.000	2.985	3.01	2.97
SD [μg/g]	0.010	0.011	0.006	0.012	0.022	0.029	0.02	0.05
RSD [%]	1.01	1.04	0.60	1.16	0.72	0.98	0.60	1.57
R [%]	99.40	101.28	100.60	101.46	99.99	99.51	100.17	99.08
Spiked coffee extract								
C [μg/g]	1.343	2.087	1.329	2.099	3.318	4.157	3.377	4.138
SD [μg/g]	0.008	0.014	0.007	0.027	0.046	0.032	0.039	0.053
RSD [%]	0.59	0.69	0.52	1.28	1.39	0.76	1.16	1.29
R [%]	100.33	99.21	99.28	99.76	99.39	101.29	101.17	100.85
Extract of coffee								
C [μg/g]	0.338	1.104						
SD [μg/g]	0.002	0.015						
RSD [%]	0.48	1.33						

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Table 4: Quantification of imidazoles in real sample (ground roasted coffee).

COFFEE	THI [µg/g] (RSD [%])		4MeI [µg/g] (RSD [%])	
	SFE (solid sample)	SPE (liquid coffee)	SFE (solid sample)	SPE (liquid coffee)
Brazil	0.0165 (4.29)	0.0154 (1.72)	1.34 (2.41)	0.60 (3.22)
Kenya	0.0025 (3.82)	0.0021 (2.25)	0.77 (3.65)	0.41 (2.59)
Haiti	0.0262 (3.04)	0.0250 (2.61)	1.45 (4.47)	0.77 (1.66)
Mexico	0.0337 (2.32)	0.0332 (1.63)	1.11 (3.70)	0.39 (2.48)
Boncampo	0.0088 (2.66)	0.0086 (2.08)	1.01 (4.29)	0.35 (2.70)

Table 5: SFE validation: the extraction of THI and 4MeI from real coffee sample, sea sand model matrix and real coffee sample spiked with known concentrations of analytes (n=5)

Compound	THI				4MeI			
C [$\mu\text{g/g}$]	0.05	0.10	0.15	0.20	1	2	3	4
Sea sand (p.a.) matrix spiked with known analyte concentration								
C [$\mu\text{g/g}$]	0.0509	0.103	0.147	0.199	0.992	2.038	2.949	4.088
SD [$\mu\text{g/g}$]	0.0019	0.005	0.005	0.009	0.047	0.105	0.132	0.230
RSD [%]	3.78	4.57	3.35	4.73	4.75	5.14	4.48	5.62
R [%]	101.71	102.55	98.19	99.30	99.17	101.92	98.30	102.19
Ground roasted coffee matrix spiked with known analyte concentration								
C [$\mu\text{g/g}$]	0.085	0.131	0.184	0.231	2.076	3.033	4.065	4.975
SD [$\mu\text{g/g}$]	0.004	0.006	0.008	0.011	0.064	0.096	0.176	0.195
RSD [%]	4.71	4.81	4.43	4.90	3.09	3.18	4.32	3.91
R [%]	101.97	97.78	100.07	100.23	98.61	97.51	98.92	97.36
Real sample (coffee Mexico)								
C [$\mu\text{g/g}$]	0.034	1.110						
SD [$\mu\text{g/g}$]	0.001	0.041						
RSD [%]	2.32	3.70						

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Figure 1