



HAL
open science

Toward a better quantitative understanding of polar stratospheric ozone loss

K. Frieler, M. Rex, R.J. Salawitch, T. Canty, M. Streibel, R.M. Stimpfle, K. Pfeilsticker, M. Dorf, D.K. Weisenstein, Sophie Godin-Beekmann

► **To cite this version:**

K. Frieler, M. Rex, R.J. Salawitch, T. Canty, M. Streibel, et al.. Toward a better quantitative understanding of polar stratospheric ozone loss. *Geophysical Research Letters*, 2006, 33, pp.L10812. 10.1029/2005GL025466 . hal-00083562

HAL Id: hal-00083562

<https://hal.science/hal-00083562>

Submitted on 6 Aug 2020

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.

Toward a better quantitative understanding of polar stratospheric ozone loss

K. Frieler,¹ M. Rex,¹ R. J. Salawitch,² T. Canty,² M. Streibel,³ R. M. Stimpfle,⁴ K. Pfeilsticker,⁵ M. Dorf,⁵ D. K. Weisenstein,⁶ and S. Godin-Beekmann⁷

Received 12 December 2005; revised 20 March 2006; accepted 31 March 2006; published 27 May 2006.

[1] Previous studies have shown that observed large O₃ loss rates in cold Arctic Januaries cannot be explained with current understanding of the loss processes, recommended reaction kinetics, and standard assumptions about total stratospheric chlorine and bromine. Studies based on data collected during recent field campaigns suggest faster rates of photolysis and thermal decomposition of ClOOCl and higher stratospheric bromine concentrations than previously assumed. We show that a model accounting for these kinetic changes and higher levels of BrO can largely resolve the January Arctic O₃ loss problem and closely reproduces observed Arctic O₃ loss while being consistent with observed levels of ClO and ClOOCl. The model also suggests that bromine catalysed O₃ loss is more important relative to chlorine catalysed loss than previously thought.

Citation: Frieler, K., M. Rex, R. J. Salawitch, T. Canty, M. Streibel, R. M. Stimpfle, K. Pfeilsticker, M. Dorf, D. K. Weisenstein, and S. Godin-Beekmann (2006), Toward a better quantitative understanding of polar stratospheric ozone loss, *Geophys. Res. Lett.*, 33, L10812, doi:10.1029/2005GL025466.

1. Introduction

[2] Proper quantitative understanding of polar stratospheric O₃ destruction, including the relative contributions of bromine and chlorine species, is a prerequisite for reliable predictions of future polar ozone. The topic has become especially exigent as many independent studies indicate measured O₃ loss is consistently underestimated by current stratospheric chemistry models, particularly during cold Arctic Januaries [e.g., Hansen *et al.*, 1997; Becker *et al.*, 1998; Deniel *et al.*, 1998, Rex *et al.*, 2003, Rex *et al.*, 2004]. The discrepancies lie outside of uncertainties related to heterogeneous chlorine activation [e.g., Rex *et al.*, 2003]. This suggests the possibility of unknown O₃ loss processes, especially under cold mid-winter conditions, or that estab-

lished catalytic processes are more efficient than currently thought.

2. Tools

[3] We use Match results for the cold Arctic winters 1994/95, 1995/96, 1999/00, 2000/01, 2002/03 and the Antarctic winter 2003 to test our quantitative understanding of measured O₃ loss. We have not reprocessed data for the Arctic winter 1991/92, because only a rudimentary Match analysis was carried out and these data are generally less reliable. Following Rex *et al.* [2003], a photochemical box model is run along trajectories identical to those studied within the Match campaigns. To quantify chemical O₃ loss independent from transport effects, the Match technique is based on a regression analysis of pairs of ozonesonde measurements of the same air parcel sampled at different times.

[4] Here, we use the model system described by Rex *et al.* [2003] to calculate the amount of ClO_x (ClO + 2 × ClOOCl) (“necessary ClO_x”) needed to reproduce the measured O₃ loss rate. The model has two updates. First, by specifying HO₂ concentrations as a function of SZA [Hanisco *et al.*, 2002], the model now accounts for O₃ loss due to cycles limited by ClO + HO₂ and BrO + HO₂, in addition to loss by the ClO + ClO, BrO + ClO and O + ClO cycles. These two new cycles contribute ~2% to total O₃ loss. Second, the integration scheme has changed [e.g., Canty *et al.*, 2005]. While the old model version represented the diurnal variation of all species assuming a zonal flow (balance of 24 hour avg. production and loss), the new model fully integrates the chemical equations along air mass trajectories. The resulting difference in calculated O₃ loss rates is less than 5%. Thus, the updated version of the model is consistent with results given by Rex *et al.* [2003].

[5] Since Rex *et al.* [2003], a number of atmospheric observations have appeared suggesting an alternative description of ClO-ClOOCl kinetics and increased levels of stratospheric bromine. Here, we quantify the effect of these observations on chemical O₃ loss for the Arctic and Antarctic vortices using four different sets of model parameters. The “reference run” uses JPL 2002 kinetics [Sander *et al.*, 2003] and a BrO_x (BrO + BrCl) profile for the Arctic vortex, Feb. 2000, found by the AER 2D model assuming only CH₃Br + halons supply stratospheric bromine. The BrO_x profile ranges from 11 to 13 pptv at the 450 to 525 K potential temperature (Θ) levels, respectively, and is similar to a profile calculated by the 3D CTM Mimoso-CHIM (see auxiliary material¹).

¹Research Department, Alfred Wegener Institute for Polar and Marine Research, Potsdam, Germany.

²Jet Propulsion Laboratory, California Institute of Technology, Pasadena, California, USA.

³European Ozone Research Coordinating Unit, Department of Chemistry, University of Cambridge, Cambridge, UK.

⁴Department of Chemistry and Chemical Biology, Harvard University, Cambridge, Massachusetts, USA.

⁵Institut für Umweltphysik, University of Heidelberg, Heidelberg, Germany.

⁶Atmospheric and Environmental Research, Inc., Lexington, Massachusetts, USA.

⁷Service d'Aéronomie-CNRS, Université Pierre et Marie Curie, Paris, France.

[6] The “new kinetics” run uses values of the ClOOCl absorption cross section (σ_{ClOOCl}) and the ClO-ClOOCl equilibrium constant (K_{EQ}) based on recent atmospheric observations of [ClO] and [ClOOCl]. *Stimpfle et al.* [2004], using daytime observations during cold Arctic conditions, showed that dimer photolysis (J_{ClOOCl}) may be occurring significantly faster than calculated using JPL 2002 values of σ_{ClOOCl} . Assuming the JPL 2002 rate constant for formation of ClOOCl, they demonstrated consistency between atmospheric observations and J_{ClOOCl} found using σ_{ClOOCl} from the laboratory study of *Burkholder et al.* [1990]. Nighttime measurements of [ClO] and [ClOOCl] were used to suggest that K_{EQ} is smaller than the JPL 2002 recommendation and is in better agreement with the *Cox and Hayman* [1988] value, leading to a higher [ClO] to [ClOOCl] ratio during darkness. Lower values of K_{EQ} are also supported by stratospheric observations provided by *von Hobe et al.* [2005] and *Berthet et al.* [2005] and a recent laboratory measurement by *Plenge et al.* [2005]. For the “new kinetics” simulation, we use values of σ_{ClOOCl} from *Burkholder et al.* [1990] extrapolated to 450 nm as described by *Stimpfle et al.* [2004] and K_{EQ} from *Cox and Hayman* [1988].

[7] We show results for a BrO_x profile based on a DOAS measurement of BrO over Kiruna, Sweden (68°N) on Feb. 18, 2000 [*Fitzenberger, 2000; Dorf, 2005*]. The BrO_x profile ranges from 19 to 21 ppt between 450 and 500 K. This is nearly 50% larger than both model BrO_x profiles that assume supply of Br_y from CH₃Br + halons, likely reflecting a ~6 pptv contribution to stratospheric inorganic bromine (Br_y) from short lived bromocarbons and tropospheric BrO [e.g., *Pfeilsticker et al., 2000*]. Higher values of BrO, similar to the Arctic DOAS profile used here, have been reported by independent aircraft and balloon observations of BrO [*Salawitch et al., 2005*]. Further discussion of BrO and details of how we have accounted for small temporal variations in Br_y are given in the auxiliary material. For clarity, results using JPL 2002 kinetics and the DOAS BrO_x profile are not shown, since they differ only slightly from the “new kinetics” run. Results are shown for a “combined run” that uses “new kinetics” and the DOAS BrO_x profile.

3. Improvements with Respect to the “January Ozone Loss Problem”

[8] The “reference run” (Figure 1, black lines) shows that, for January of all presented Arctic winters and July of the Antarctic winter, the amount of “necessary ClO_x” to match observed chemical O₃ loss rates (top half of each box) exceeds ~3.7 ppbv, the maximum amount of total stratospheric chlorine [*World Meteorological Organization (WMO), 2003*]. For 3 of the 6 winters, O₃ loss rates calculated on the basis of ClO_x = 3.7 ppbv (“maximum possible ozone loss”, shown in lower half of each box only for the time period when nearly complete chlorine activation might be expected) underestimate the observed O₃ loss by more than the 1 σ uncertainty of the measurements. This suggests our current representation of known O₃ loss cycles is insufficient to account for observed loss rates.

[9] Figure 1 shows that changes to J_{ClOOCl} and K_{EQ} (“new kinetics”) increase “maximum possible ozone loss”

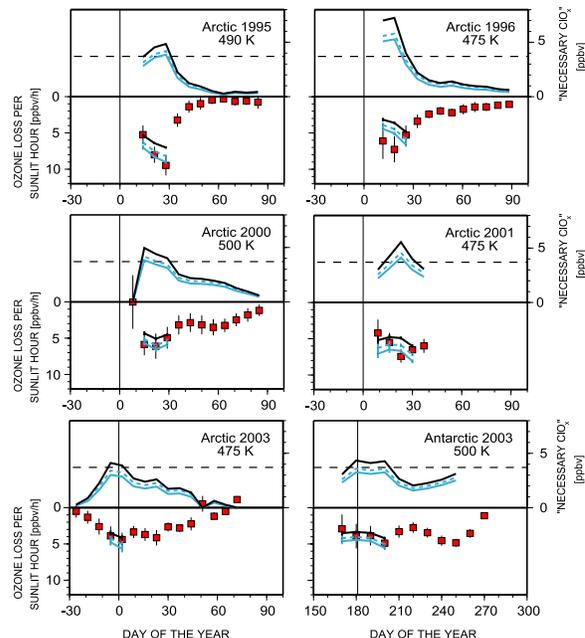


Figure 1. Chemical O₃ loss rate in the polar vortex based on Match (red boxes; error bars are 1 σ uncertainty). The abundance of ClO_x necessary to account for the measured O₃ loss (“necessary ClO_x”) and the modeled O₃ loss assuming ClO_x = 3.7 ppbv (“maximum possible ozone loss”) are shown in the upper and lower parts of each plot, respectively. The dashed line in the upper part of each plot marks the level of 3.7 ppbv ClO_x. Maximum possible ozone loss is shown only for the time periods where nearly complete chlorine activation is likely to occur. Black lines: reference run (JPL 2002 kinetics + AER BrO_x); dashed blue lines: “new kinetics” (J_{ClOOCl} from *Burkholder et al.* [1990] and K_{EQ} from *Cox and Hayman* [1988]) + AER BrO_x; solid blue lines: “new kinetics” + BrO_x derived from DOAS BrO measurements.

by ~20% (blue dashed lines). This effect is due to higher J_{ClOOCl} ; the extrapolation of the ClOOCl cross sections longward of 410 nm plays no significant role in this result. The overall effect of the new K_{EQ} is a slight decrease in modeled O₃ loss (~2%). The impact remains small if K_{EQ} is taken from “Fit 2” of *von Hobe et al.* [2005], the lowest reported value of K_{EQ} .

[10] Results from a model run that uses “new kinetics” and DOAS BrO_x (“combined run”) are shown by solid blue lines in Figure 1. This simulation leads to a ~30–40% increase in ozone loss relative to the reference run and largely resolves the discrepancy between maximum possible ozone loss and measured loss. Present uncertainties in BrO_x derived from DOAS BrO lead to a ~7% uncertainty in “maximum possible ozone loss”.

[11] The selection of different Θ levels for Figure 1 is driven by the availability of reliable Match data between 475 and 500 K. For Arctic 2002/03 and Antarctic 2003 we also modeled the 500 and 475 K levels, respectively. Results are similar to those shown in Figure 1. The “combined run” is not sufficient to fully resolve the discrepancy for 1991/92 reported by *Rex et al.* [2003] (not shown). A discrepancy at the 1 σ to 2 σ level remains for two points during that year.

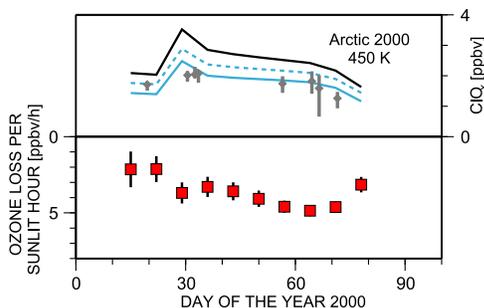


Figure 2. Chemical O₃ loss rate in 1999/2000, 450K, based on Match (red boxes; error bars are 1 σ uncertainty). The lines represent the necessary ClO_x to account for measured O₃ loss, where the colors/line types correspond to the same model runs as described in Figure 1. Gray diamonds: mean value of all ClO_x measurements at 450 \pm 10 K, for the 8 ER-2 flights that remained entirely inside the vortex, vertical bars represent the maximum and minimum of the considered ClO_x data.

However, the overall picture is that modeled maximum possible ozone loss rates for the “combined run” are within the statistical and systematic uncertainties of the observed rates for nearly all time periods considered.

[12] These results should not be interpreted as evidence for complete, quantitative understanding of polar O₃ loss because the calculations are based on upper limits of ClO_x. A more meaningful comparison of modeled and measured ozone loss rates is described in the following section, which considers measurements of ClO_x at the 450 K level of the Arctic winter 1999/2000. Similar measurements of ClO_x are not available for higher Θ levels considered in Figure 1.

4. Arctic Winter 1999/00, 450 K: “Necessary ClO_x” in Comparison to Measured ClO_x

[13] A stringent test of our quantitative understanding of chemical O₃ loss is provided by comparing measured ClO_x to calculated “necessary ClO_x” needed to account for observed ozone loss rates. Here, we use a time series of ClO_x based on measurements of [ClO] and [ClOOCl] from *Stimpfle et al.* [2004], for Arctic winter 1999/00. We compare measurements near $\Theta = 450$ K (e.g., ClO_x collected between 440 and 460 K) to results from the Match analysis at 450 K. We selected data from 8 flights within the Arctic vortex, during the period Jan. 20 to Mar. 12, 2000. For each individual flight, the average of all ClO_x measurements at 450 \pm 10 K is compared to the “necessary ClO_x” derived from measured ozone loss rates. The air masses encountered by the aircraft are representative of vortex conditions sampled by Match (see auxiliary material).

[14] Figure 2 shows that to explain the measured O₃ loss rates, the “reference run” based on JPL 2002 kinetics and BrO_x from CH₃Br + halons needs significantly more ClO_x (black line) than was observed. During late Jan./early Feb., the discrepancies are larger than the 20% uncertainty of the ClO_x measurements [*Stimpfle et al.*, 2004]. The amount of ClO_x needed to account for measured O₃ loss on the basis of the “new kinetics” (dashed blue line) is also higher than the observations, but is just within the uncertainty range. Results using DOAS BrO_x and JPL 2002 kinetics are almost

indistinguishable from the “new kinetics” run. Nearly perfect agreement between measured ClO_x and “necessary ClO_x” is found for the “combined run” (solid blue line). Figure 2 demonstrates that use of a faster photolysis rate for ClOOCl (consistent with atmospheric observations of [ClO] and [ClOOCl]) and higher bromine loading (consistent with a \sim 6 pptv source of Br_y from species other than CH₃Br + halons) provides an overall good quantitative explanation of Arctic ozone loss rates.

[15] *Rex et al.* [2003] noted the tendency for models to underestimate observed chemical loss of Arctic O₃, particularly during cold Januaries. They speculated that changes to ClO-ClOOCl kinetics and BrO_x, similar to those considered here, could largely resolve the discrepancy. The calculations shown in this paper are given greater credence by numerous studies related to ClO-ClOOCl kinetics [*Stimpfle et al.*, 2004; *Berthet et al.*, 2005; *Plenge et al.*, 2005; *von Hobe et al.*, 2005] and stratospheric Br_y [*WMO*, 2003; *Canty et al.*, 2005; *Dorf*, 2005; *Salawitch et al.*, 2005] that have appeared since *Rex et al.* [2003] was completed. Hence, this work builds upon and advances the speculation of our earlier study.

5. Relative Importance of the BrO + ClO Cycle

[16] To calculate the relative importance of each O₃ loss cycle in the model, we use all Match measurements between Jan. 10 to Feb. 10 (Arctic, all years) and Jul. 10 to Aug. 10 (Antarctic, 2003) to calculate O₃ loss per sunlit hour for these winter periods. The model was run along trajectories, with ClO_x adjusted to reproduce the measured O₃ loss rates for the three combinations of kinetic parameters and BrO_x considered above (Figure 3, top). We calculate O₃ loss per sunlit hour for the 5 catalytic cycles and then derive the relative fraction of each cycle to the total modeled loss rate

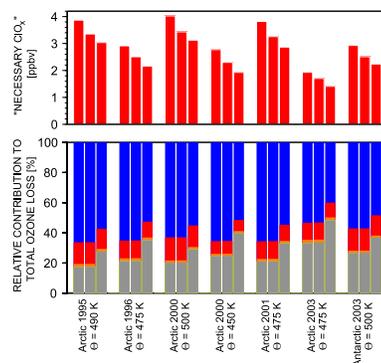


Figure 3. (top) Level of ClO_x needed to account for observed O₃ loss for the “reference run” (JPL 2002 kinetics, AER BrO_x) (1st bar of each grouping), the “new kinetics run” (2nd bar), and the “combined run” (3rd bar). Each grouping of columns represents results of model runs executed along the Match trajectories available between Jan. 10 and Feb. 10 (Arctic) and Jul. 10 and Aug. 10 (Antarctic), respectively. (bottom) Relative importance of the ClO + ClO cycle (blue), the BrO + ClO cycle (gray), the ClO+O cycle (red), and the HOCl cycle (orange) to total O₃ loss. The contribution from the HOBr cycle was calculated but is too small to represent. Groupings are the same as for Figure 3 (top).

(Figure 3, bottom), which by definition equals the measured O₃ loss rate.

[17] The kinetics changes considered here do not alter the relative importance of the BrO + ClO cycle (the increase in J_{ClOOC1} , which dominates the “new kinetics” run, results in faster O₃ loss by all cycles). Even using K_{EQ} of von Hobe *et al.* [2005], the effect on the relative importance of BrO + ClO is negligible. In contrast, use of DOAS BrO_x increases the relative importance of the BrO + ClO cycle by ~10–15%. Hence, the BrO+ClO contribution to total O₃ loss ranges from ~27 to 48% for the “combined run”, in comparison to 17 to 33% for the “reference run”. The relative importance of the BrO + ClO cycle increases with decreasing ClO_x. This explains the differences between results for winter 2000/01 compared to winter 2002/03, where chlorine activation derived from measured O₃ loss rates is much lower.

6. Concluding Remarks

[18] We show that use of a faster value for J_{ClOOC1} and abundances of BrO_x derived from a measured BrO profile increase calculated polar O₃ loss rates by up to 40%, resulting in overall consistency between observed O₃ loss rates and upper limits for O₃ loss based on an assumption of a completely activated vortex (e.g., ClO_x ≈ 3.7 ppbv). Levels of ClO_x needed to account for observed O₃ loss rates at Θ = 450 K are shown to be in remarkably good agreement with measurements of ClO_x for the Arctic winter of 999/00. The faster value of J_{ClOOC1} is based on a laboratory study [Burkholder *et al.*, 1990] and is consistent with measured partitioning of ClO and ClOOC1 [Stimpfle *et al.*, 2004] and the JPL 2002 rate constant for the ClO self reaction. The DOAS BrO_x profile is ~6 pptv higher than a model profile based on supply of Br_y from only CH₃Br + halons, consistent with other estimates of contributions to Br_y from short lived bromocarbons and tropospheric BrO [Pfeilsticker *et al.*, 2000; Salawitch *et al.*, 2005]. Further laboratory observations of the ClOOC1 cross section, extending to 450 nm, are needed to assess the accuracy of J_{ClOOC1} . Also, further atmospheric observations of BrO are needed to better define levels of BrO_x in the polar vortices.

[19] This study suggests that previous discrepancies between measured and modeled polar O₃ loss rates might be resolved by greater efficiency of known catalytic cycles, rather than by the introduction of new chemical loss processes. These results could represent an important “bridge” between analysis of atmospheric observations and computationally expensive 3D model descriptions of climate chemistry interactions used to predict future levels of polar ozone.

[20] **Acknowledgments.** We thank R. Lehmann for helpful discussions. Research at the Jet Propulsion Laboratory, California Institute of Technology, and at the Atmospheric and Environmental Research, Inc., is performed under contract with the National Aeronautics and Space Administration. Work at AWI was supported by the BMBF (project AFO 2000/07ATC08) and EC DG, Research under the projects QUOBI (EVK2-2001-00129) and SCOUT-03 (505390-GOCE-CT-2004). Work at the Univ. of Heidelberg was supported by the BMBF (project 50FE0017).

References

- Becker, G., R. Müller, D. S. McKenna, M. Rex, and K. S. Carslaw (1998), Ozone loss rates in the Arctic stratosphere in the winter 1991/92: Model calculations compared with Match results, *Geophys. Res. Lett.*, **25**, 4325–4328.
- Berthet, G., P. Ricaud, F. Lefèvre, E. Le Flochmoën, J. Urban, B. Barret, N. Lauté, E. Dupuy, J. De La Noë, and D. Murtagh (2005), Nighttime chlorine monoxide observations by the Odin satellite and implications for the ClO/Cl₂O₂ equilibrium, *Geophys. Res. Lett.*, **32**, L11812, doi:10.1029/2005GL022649.
- Burkholder, J. B., et al. (1990), Ultraviolet absorption cross sections of Cl₂O₂ between 210 and 410 nm, *J. Phys. Chem.*, **94**, 687–695.
- Canty, T., et al. (2005), Nighttime OCIO in the winter Arctic vortex, *J. Geophys. Res.*, **110**, D01301, doi:10.1029/2004JD005035.
- Cox, R. A., and G. D. Hayman (1988), The stability and photochemistry of dimers of the ClO radical and implications for Antarctic ozone depletion, *Nature*, **332**, 796–800.
- Deniel, C., R. M. Bevilacqua, J. P. Pommereau, and F. Lefèvre (1998), Arctic chemical ozone depletion during the 1994–1995 winter deduced from POAM II satellite observations and the REPROBUS three-dimensional model, *J. Geophys. Res.*, **103**, 19,231–19,236.
- Dorf, M. (2005), Investigations of inorganic stratospheric bromine using balloon borne DOAS measurements and model simulations, doctoral dissertation, Univ. of Heidelberg, Heidelberg, Germany.
- Fitzenberger, R. (2000), Investigation of the stratospheric inorganic bromine budget for 1996–2000: Balloon borne measurements and model comparison, doctoral dissertation, Univ. of Heidelberg, Heidelberg, Germany.
- Hanisco, T. F., J. B. Smith, R. M. Stimpfle, D. M. Wilmouth, J. G. Anderson, E. C. Richard, and T. P. Bui (2002), In situ observations of HO₂ and OH obtained on the NASA ER-2 in the high-ClO conditions of the 1999/2000 Arctic polar vortex, *J. Geophys. Res.*, **107**(D20), 8283, doi:10.1029/2001JD001024.
- Hansen, G., T. Svenøe, M. P. Chipperfield, A. Dahlback, and U. Hoppe (1997), Evidence of substantial ozone depletion in winter 1995/96 over Northern Norway, *Geophys. Res. Lett.*, **24**, 799–802.
- Pfeilsticker, K., W. T. Sturges, H. Bösch, C. Camy-Peyret, M. P. Chipperfield, A. Engel, R. Fitzenberger, M. Müller, S. Payan, and B.-M. Sinnhuber (2000), Lower stratospheric organic and inorganic bromine budget for the arctic winter 1998/99, *Geophys. Res. Lett.*, **27**, 3305–3308.
- Plenge, J., et al. (2005), Bond strength of chlorine peroxide, *J. Phys. Chem. A*, doi:10.1021/jp044142h.
- Rex, M., R. J. Salawitch, M. L. Santee, J. W. Waters, K. Hoppel, and R. Bevilacqua (2003), On the unexplained stratospheric ozone losses during cold Arctic Januaries, *Geophys. Res. Lett.*, **30**(1), 1008, doi:10.1029/2002GL016008.
- Rex, M., R. J. Salawitch, P. von der Gathen, N. R. P. Harris, M. P. Chipperfield, and B. Naujokat (2004), Arctic ozone loss and climate change, *Geophys. Res. Lett.*, **31**, L04116, doi:10.1029/2003GL018844.
- Salawitch, R. J., D. K. Weisenstein, L. J. Kovalenko, C. E. Sioris, P. O. Wennberg, K. Chance, M. K. W. Ko, and C. A. McLinden (2005), Sensitivity of ozone to bromine in the lower stratosphere, *Geophys. Res. Lett.*, **32**, L05811, doi:10.1029/2004GL021504.
- Sander, S. P., et al. (2003), Chemical kinetics and photochemical data for use in atmospheric studies, *Evaluation No. 14, JPL Publication 02–25*, Jet Propul. Lab., Pasadena, Calif.
- Stimpfle, R. M., D. M. Wilmouth, R. J. Salawitch, and J. G. Anderson (2004), First measurements of ClOOC1 in the stratosphere: The coupling of ClOOC1 and ClO in the Arctic polar vortex, *J. Geophys. Res.*, **109**, D03301, doi:10.1029/2003JD003811.
- von Hobe, M., et al. (2005), A re-evaluation of the ClO/Cl₂O₂ equilibrium constant based on stratospheric in situ observations, *Atmos. Chem. Phys.*, **5**, 693.
- World Meteorological Organization (WMO) (2003), Scientific assessment of ozone depletion: 2002, *Global Ozone Research and Monitoring Project—Report No. 47*, 498 pp., Geneva, Switzerland.
- T. Canty and R. J. Salawitch, Jet Propulsion Laboratory, California Institute of Technology, Pasadena, CA 91109, USA.
- M. Dorf and K. Pfeilsticker, Institut für Umwelphysik, University of Heidelberg, D-69120 Heidelberg, Germany.
- K. Frieler and M. Rex, Research Department, Alfred Wegener Institute for Polar and Marine Research, Telegrafenberg A45, D-14401 Potsdam, Germany. (kfrieler@awi-potsdam.de)
- S. Godin-Beekmann, Service d’Aéronomie-CNRS, Université Pierre et Marie Curie, F-75252, Paris Cedex 05, France.
- R. M. Stimpfle, Department of Chemistry and Chemical Biology, Cambridge, MA 02138, USA.
- M. Streibel, European Ozone Research Coordinating Unit, Department of Chemistry, University of Cambridge, Cambridge CB2 1EW, UK.
- D. K. Weisenstein, Atmospheric and Environmental Research, Inc., Lexington, MA 02421–3136, USA.