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From colloidal-silica sols to aerogels and xerogels

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Abstract -

By destabilization of quasi-monodisperse silica sols, fractal clusters can be obtained. Solvent evacuation gives a porous solid with fractal structure. Structural investigations of these colloidal aerogels by small-angle neutron scattering are presented, and compared to similar observations in concentrated sols and xerogels.

An investigation of the vibrational modes of these materials by inelastic neutron scattering gives clear evidence of a strong non-Debye contribution in the low-frequency regime.

Introduction.

Silica gels and aerogels are known to be excellent model-materials for the study of fractal inhomogeneous media [1]. They can be prepared under various physical and chemical conditions leading to strongly different microstructures. Small-angle scattering techniques using neutrons or X-rays have proven to be well suited for investigating these differences and to characterize the gel structure [2,3]. In this paper we present small-angle neutron scattering (SANS) data on gels derived from the destabilization of colloidal silica sols. The structural differences of a broad set of samples (aerogel, xerogel, sols) are investigated quantitatively by comparing their scattering profiles. Preliminary results of an investigation of the vibrational dynamics by inelastic neutron scattering are also presented for one of these "colloidal" aerogels.

Sample preparation and characterization.

Commercial silica solutions (Ludox manufactured by E.I. Dupont de Nemours and Co) are destabilized by decreasing the pH-value of the aquasol. After gelation and subsequent solvent exchange, wet gels are either supercritically dried to obtain aerogels or dried in ambient conditions to form xerogels. The supercritical drying, which preserves the fine structure of the original wet gel, allows to produce highly porous samples with densities in the range of 100-300 kg.m^{-3} depending on the initial silica concentration. In
contrast, the classical drying leads to samples with a more compact structure
and a macroscopic density of about half the silica density. Two kinds of sol
which differ by their particle size, have been studied. In this paper, they are
referred to by their commercial name TM and LS. The average radius $r_0$
with a dispersity $\Delta r_0$ of the spherical colloidal particles have been measured from
transmission electron microscopy experiments [4]. Polydispersity appears much
larger for small particles.

Colloidal gels result from the connection of these silica particles into
aggregated clusters. The precise knowledge of the structure at the particle
scale makes these samples a simple model system for the interpretation of
elastic scattering experiments.

Experimental results and discussion.

SANS studies were performed on the PAXE spectrometer at the Laboratoire
Léon Brillouin in Saclay, France. A combination of two incident wavelengths and
two sample-to-detector distances in order to cover the q-range from $3 \times 10^{-2}$
to 0.3 nm$^{-1}$, was used.

Figure 1 shows, in a log-log plot, the comparison of SANS profiles for
colloidal solutions, xerogels and aerogels prepared from the TM and LS sols. At
large q ($q > \pi/r_0$), the scattering originates from the surface of the
particles. The scattered intensity $I(q)$ then follows the well-known Porod law
$I(q) \sim q^{-4}$ for all samples. In addition, smeared oscillations are superimposed
to the power law. Their amplitude and width vary with the nature of the sol. In
contrast, in the low-q range scattering curves are strongly sample dependent.
Indeed, for $q$-values smaller than $\pi/r_0$ scattering is dominated by correlation
effects between particles.

For the diluted sols the quantitative description of the whole
scattering curve can be made by assuming that the silica particles scatter
neutrons independently. Thus $I(q)$ is the sum of the intensities scattered by
each particle, the so-called form factor $P(q)$, depending only on the shape and
the size of the scattering particle [5].

Introducing a dispersity of particle size as a Gaussian distribution
of $r_0$ with a standard deviation $\delta r_0$, allows to fit the upper curves of
Fig. 1. The best fits are shown on the figure as solid lines. The same
calculated curves, arbitrarily shifted along the vertical axis to overlap the
high-q regime of the xerogels and the aerogels are also shown in Fig. 1. The
excellent adjustment in this regime, for all samples in a given serie
demonstrates that the shape and the size distribution of the particles remains
unchanged from the colloidal solutions to the dry gels.

At low q, a small departure from the calculated curves occurs in the
diluted sols. This is related to interference effects between adjacent
particles. For xerogels, these interferences are enhanced due to the high
particle number density. As a consequence a strong drop of the scattered
intensity at small q is observed, resulting in the appearance of a maximum
in $I(q)$. 

Theoretically, such interferences can be expressed by the Fourier
Transform $S(q)$ of the pair correlation function $g(r)$ of individual
scatterers. The function $g(r)$ is defined as the probability of finding two
Fig. 1 - A log-log plot of the scattered intensity $I(q)$ vs $q$ for colloidal solutions xerogels and aerogels prepared from the TM and LS sols. Solid lines are fits to the form factor as explained in the text.

Particles separated one from the other by a distance $r$. The intensity scattered by an assembly of $N$ correlated particles can then be written as $[6]$

$$I(q) = N \cdot P(q) \cdot S(q)$$

The correlation peak observed for the xerogels reveals that the particles are not arranged at random but exhibit some short-range ordering.

For the aerogels of low density, the scattered intensity at small-$q$ follows the same power law $I(q) \sim q^{-1.8}$ for the two samples, as indicated by the straight lines in Fig. 1. This behavior is the signature of a mass fractal structure, with a fractal (Hausdorff) dimension $D = 1.8$. The spatial arrangement of the particles in such fractal aggregates is well described by a pair correlation function with the form $[7]$

$$g(r) \propto r^{D-3} \exp(-r/\xi)$$

Here $\xi$ is a fractal persistence length relative to the crossover from fractal structure to homogeneity at large length-scale. For the two low-density aerogels
Fig. 2 - Normalized inelastic structure factor for a base catalyzed aerogel (B) of density 70 kg.m\(^{-3}\) and a LS colloidal aerogel of density 100 kg.m\(^{-3}\). Solid line is the instrumental resolution function taken from Vanadium run at room temperature.

Presented here, $\xi$ is outside the length scale investigated experimentally by SANS. Using the proper normalization of $g(r)$ and taking into account the interference effects due to the finite size of the scattering particles allows to fit simultaneously the fractal and Porod regimes and their crossover. Such a fit procedure is developed and discussed in Ref. 4. In this process the fractal dimension $D$ is an adjustable parameter and is found equal within the accuracy of the method to the slope of the straight line $(D = 1.8)$ for the two aerogels.

These colloidal aerogels have thus a fractal dimension similar to that previously observed in aerogels prepared from organosilicate precursors under base catalysis [3]. The main differences between the two kinds of samples arise from the size of their elementary particles and their connectivities.

In order to get a better insight on the physical properties of such fractal solids, it is worthwhile to study their vibrational dynamics. This has been done extensively [8] for materials prepared from organosilicates. Vibrational excitations obviously depend on both the structure and the connectivity of the materials. The influence of the sample connectivity can be investigated by comparing the properties of materials having the same fractal dimension, but exhibiting differences in the way the particles are connected at the microscopic and mesoscopic scales.

As shown previously [9], the particle size has a strong influence on the low-frequency spectra in inelastic light scattering. Raman spectra in the 5-50 cm\(^{-1}\) range are dominated by a strong peak whose position and width are
directly related to the size and the polydispersity of particles, respectively.

We present here preliminary inelastic neutron scattering spectra recorded on the backscattering spectrometer IN10 at the Institut Laue-Langevin in Grenoble. Inelastic structure factors \( S(q,\omega) \) obtained from the LS colloidal and base-catalysed organosilicate aerogels are shown in Fig. 2.

In this frequency-wavevector domain, it can be shown \( [10] \) that

\[
S(q,\omega) \propto G(\omega)/\omega^2
\]

where \( G(\omega) \) is the vibrational density of states. The experimental observation clearly excludes an assignation of the scattering to phonons, as the latter would lead to a frequency independent \( S(q,\omega) \). As measured in Raman scattering the lowest-frequency vibrational mode of particles occurs at frequencies of about 100 GHz. The above discussion strongly suggests that the observed scattering originates from fracton modes. A comparison of the two curves indicates large differences in the fracton dynamics on the two samples of different connectivity.

References