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A COMPARISON OF EXPERIMENT AND THE THEORY OF CONTINUOUS ORDERING

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and

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Abstract. — The Fourier transform of the Helmholtz Free Energy $F(k)$, for composition fluctuations of wave vector $k$, has been obtained for several ordering alloys above $T_c$ by measuring the absolute X-ray diffuse scattering intensity due to local order, at high temperatures, along lines in reciprocal space. The single crystals examined were Cu$_3$Au, Cu-23 at% Au, Cu-18.5 at% Au, and CoPt$_3$. The values obtained are compared to the theory of continuous transformations and it is found that the linear theory is satisfactory but the nearest-neighbour approximation is inadequate. In particular, the gradient energy is a strong function of $k$. This is also indicated from measured relaxation times in different crystallographic directions; these vary less with direction than predicted when the dependence of the gradient energy on $k$ is ignored. It is possible to measure the activation energy for interdiffusion in volume in all the alloys at low temperatures by measuring the absolute intensity due to local order, at high temperatures, along lines in reciprocal space. The single crystals examined were Cu$_3$Au, Cu-23 at% Au, Cu-18.5 at% Au, and CoPt$_3$. The results are compared to the theory of continuous transformations and it is found that the linear theory is satisfactory but the nearest-neighbour approximation is inadequate. In particular, the gradient energy is a strong function of $k$. This is also indicated from measured relaxation times in different crystallographic directions; these vary less with direction than predicted when the dependence of the gradient energy on $k$ is ignored. It is possible to measure the activation energy for interdiffusion in volume in all the alloys at low temperatures by measuring the absolute intensity due to local order, at high temperatures, along lines in reciprocal space. 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continuum ordering. The quantity \( F(\mathbf{k}) \) can also be investigated by measuring the equilibrium X-ray intensity due to local order \([1]\). In Laue units:

\[
I_{E(\mathbf{R})}(\mathbf{k}) = \frac{N_e k_B T}{c(1 - c)} F(\mathbf{k})
\]  

(2)

where \( k_B \) is Boltzmann's constant, \( N_e \) is the number of atoms per unit volume and \( c \) is the atomic concentration of solute.

In its most simplified form, that is, when the basic diffusion equation is solved in the linear nearest-neighbour approximation \([3]\):

\[
F(\mathbf{k}) = f'' + 2 \eta^2 M(\mathbf{k}) + 2 KB^2(\mathbf{k})
\]  

(3)

The term \( K \) is the gradient energy coefficient which is negative for ordering systems (as they prefer a large gradient in composition). The second term is the elastic energy:

\[
\eta = \frac{1}{a} \frac{\partial a}{\partial \mathbf{k}},
\]

where \( a \) is the lattice parameter, and \( M(\mathbf{k}) \) is an effective elastic modulus. For the short waves involved in ordering this needs to be calculated from a force-constant model (see Appendix). For higher-neighbour interactions the gradient energy becomes a function of \( \mathbf{k} \) \([4, 5, 6]\) and Eq. (3) becomes:

\[
F(\mathbf{k}) = \frac{N_e k_B T}{c(1 - c)} + 2 V(\mathbf{k}) + 2 \eta^2 M(\mathbf{k})
\]  

(4a)

The first term in Eq. (4a) is the entropy of mixing of an ideal solution. For crystal structures with a center of symmetry:

\[
V(\mathbf{k}) = \sum_p V_p - \sum_p V_p' [1 - \cos (\mathbf{k} \cdot \mathbf{r}(p))]
\]  

(4b)

The term \( p \) represents a sum over interatomic vectors with components \( p_1, p_2, p_3 \). The pair energy is \( V_p' \). The second sum in Eq. (4b) is the more exact gradient energy term (which divided by \( B^2(\mathbf{k}) \) we shall call \( K(\mathbf{k}) \)) whereas the first is part of \( f'' \) in Eq. (3).

As Cook has pointed out \([1]\), most studies of continuous transformations have been concerned with spinodal decomposition and very little is known about continuous ordering. Even for spinodal decomposition, although the wavelength dependence of \( \tau \) (Eq. (1a)) has been examined in many papers as well as the temperature dependence, and values of \( K/f'' \) have been determined based on the nearest-neighbour approximation, the dependence of \( \tau \) on \( \mathbf{k} \) has not been studied. Furthermore, in only one case \([7]\) were absolute intensities employed so that \( K \) and \( f'' \) could be separately measured in one experiment. Even less information is available for ordering systems. Paulsen \([8]\) employed films of alternate layers of Cu and Au to study the decay of waves in this system. In particular, the wavelength and temperature (but not the crystallographic) dependence were examined.

Naumora \etal \([9]\) employed bulk Fe-Al alloys but examined only quenched specimens to study kinetics, and did not correct the measured diffuse scattering for effects due to atomic displacements.

Here, studies are reported of the absolute equilibrium short-range order intensity vs. temperature and the kinetics in different crystallographic directions and for different wavelengths, when the temperature of a single crystal is changed \( \approx 25 \) °C from various temperatures (above \( T_c \)). This situation corresponds to such a small change in local environment that the linear theory should be applicable. This will allow a test of the dependence of the gradient energy on \( \mathbf{k} \). All data was taken at high temperatures. The diffuse intensity due to local order was separated from that due to size effects. The results indicate that the \( \mathbf{k} \) dependence of the gradient energy (which comes from higher-neighbour interactions) is important.

This study also indicates that measurements of the diffuse scattering from bulk alloy single crystals can be employed (as well as thin films) to quantitatively examine the theory and to obtain information on diffusivities at low temperatures.

2. Experimental methods. — 2.1 Materials. — Single crystals of Cu-33.3Au, Cu-23 at pct Au, Cu-18.5 at pct Au, CoPt-3, and Fe-29.2 at pct Al were studied. The compositions were known to \( \pm 0.1 \) at pct or better. Their growth, characterization, and preparation for X-ray measurements has been described elsewhere \([10-14]\). These were all in the form of plates \( 3 \times 10^{-4} \cdot 5 \times 10^{-4} \) mm thick. The effect of surface roughness was measured by examining the fluorescence by MoK\( \alpha \) vs. 2θ. These X-ray data was corrected for this effect. Transition temperatures were defined by examining the intensity of a superlattice reflection. A sharp discontinuity occurs at \( T_c \) for Cu-33.3Au as the order-disorder transition is first order. For CoPt-3 and Cu-18.5 at pct Au there is a two-phase region and the change in slope of the intensity vs. temperature was employed. The Fe-Al alloy undergoes a higher order transition and the inflection point was employed to crudely define the transition.

2.2 X-ray System. — The diffractometer was a GE XRD-5 system modified to utilize a doubly bent graphite incident beam monochromator \([17]\). A specimen was held on a eucentric quarter circle goniometer in the \( \chi = 90^\circ \) configuration on a resistance heated metal stage covered with a water-cooled Be hemisphere \( 7.5 \times 10^{-4} \) m thick \([18]\). The system was evacuated to \( 2 \times 10^{-5} \) mm Hg or less. Temperature was controlled to \( \pm 0.25 \) °C or better with a feedback controller system. Two thermocouples were placed at different points under a single crystal to check for any gradients. A receiving slit and scatter slit were employed; total horizontal
divergence was 0.75 to 1.20° 2θ. Total vertical divergence was 2 to 2.70° 2θ (measured with a perfect silicon crystal [17]). The range of a diffuse scattering peak was 3 to 7° 2θ at half height, depending on the crystal.

The radiation was CoKα at 50 kV, 7 mA. The half-wavelength component was 0.1 pct. All intensities were measured for 10^5-10^6 counts (in a monitor detector) from a V₂O₅ film placed in the exit beam from the monochromator, thereby minimizing effects of changes in air density, X-ray tube output and any slight misalignment of the monochromator during the measurements.

The entire system was controlled by a 16K Digital Equipment Corporation PDP8-E minicomputer with a twin tape unit (to store computed angles 2, 3 and data). This system carried out the 2 and 3 motions of the detector, counting to a statistical error of at least 0.5 pct. Some 20 minutes were required to equilibrate well away from Tc, but times up to a few days were needed near the transition.

Data were corrected for surface roughness, measured polarization factor [21], and air scattering was subtracted (determined with a Pb beam trap replacing the specimen). The results were then placed on an absolute scale by determining the intensity of the direct beam. This was done by integrating peaks from an Al powder compact, correcting for thermal diffuse scattering and comparing the result to calculated values [20]. Calculated Compton scattering was then subtracted and the data reduced to Laue units per atom by dividing by Nc c (1 - c) fα - fβ^2. The scattering factors employed in these calculations can be found in reference [22]. (For the case of Cu₃Au these were corrected for Debye-Waller factors, but in the other alloys, these factors were not known.)

For the Cu-Au alloys and CoPt₃ the ordering transformation is from an Al (fcc) structure to L1₂; superstructure reflections appear below Tc at 001, 011, etc., in reciprocal space. For Fe-29.2 pct Al the change is from B2 (CsCl) to a DO₃ cell, with superstructure reflections at 1/2 1/2 1/2, etc. Accordingly, in the former case the diffuse scattering due to local order was examined along <00h> and <0hh> lines in reciprocal space, whereas in the latter it was examined only along the <hhh> direction. The intensity due to local order needed to determine I(k) in Eq. (2) was separated from that due to scattering from average and mean-square atomic displacements from the average lattice sites, following the method of Borie and Sparks [22-24]. For Cu-Au alloys and CoPt₃ (Al structure), with the total elastic diffuse intensity in Laue units per atom written as I^U,

\[ I^U_{550}(00h) = I^U(00h) - \frac{h}{2} [I^U(0, 2, 0) - I^U(2 - h, 0, 2)] + \frac{(h^2 - 2h)}{8} \times \]
\[ \times [I^U(2, 0, 4 - h) - 2 I^U(2 - h, 0, 2) + I^U(h, 0, 2)], \quad (5a) \]
\[ I^U_{550}(h00) = I^U(0, h00) - h[I^U(0, h00) - I^U(2 - h, h, 0)] + \]
\[ + \frac{h^2 - 2h}{4} [I^U(4 - h, h, 0) - 2 I^U(2 - h, h, 0) + I^U(h00)] \]
\[ + \frac{h^2}{4} [I^U(h - 1, h - 1, 1) - 2 I^U(3 - h, h - 1, 1) + I^U(3 - h, 3 - h, 1)], \quad (5b) \]

with h taken from 1 to 3/2. For Fe-29.2 pct Al (B2 structure) with h from 1/2 to 1:
\[ I^U_{558}(hhh) = I^U(0, hhh) + \frac{6h - 9h^2}{8} I^U(2 - h, h, h) + \frac{3h^2}{4} I^U(2 - h, 2 - h, h) + \]
\[ + \frac{3h^2 - 6h}{8} I^U(2 + h, h, h). \quad (5c) \]

Similar separation equations are available for the various displacement terms [22]. These equilibrium measurements were made vs. temperature at the position of the superlattice reflection which occurs below Tc, with sufficient counts so that the statistical error after separation was 1.5% or better. Measurements were also made along complete lines for each alloy at fixed temperatures; in this case, after separation the error was 8 pct or less.

### 2.4 Kinetic studies.

The value of τ(k) can be determined at various positions k in reciprocal space, in various crystallographic directions, and at various temperatures, by examining the kinetics
of the change in intensity due to local order when the temperature is changed:

$$\ln \left[ I_{\text{SR0}}^{u}(k, t) - I_{\text{SR0}}^{u}(k, \infty) \right] =$$

$$= \ln \left[ I_{\text{SR0}}^{u}(k, 0) - I_{\text{SR0}}^{u}(k, \infty) \right] - t/\tau(k). \quad (6)$$

Because ratios are employed, the absolute intensity is not required. Furthermore, background and Compton scattering are independent of time, so that these cancel in the differences. The measured intensity is merely corrected for angular factors (surface roughness, polarization factor and $f_{s} - f_{0}$). The separation equations, (5a-5c) were utilized to remove displacement effects, with $I_{\text{SR0}}^{u}$ replaced by $\Delta I_{\text{RAW}}$.

$$\Delta I_{\text{RAW}} = \left( \frac{I_{\text{RAW}}(k, t) - I_{\text{RAW}}(k, \infty)}{\text{ANGULAR FACTORS}} \right).$$

Eq. (1a) may be rewritten as:

$$[\tau(k) F(k)]^{-1} = \frac{2 B^2(k)}{f''} D_{0} e^{-Q/k_{B} T}. \quad (7a)$$

Taking logarithms of both sides of Eq. (7a):

$$\ln \left[ (\tau(k))^{-1}/F(k) \right] =$$

$$= \ln 2 B^2(k) - \ln f'' + \ln D_{0} - Q/k_{B} T. \quad (7b)$$

From the slope of the left hand side of Eq. (7b) vs. $1/T$ the activation energy for diffusion, $Q$ may be obtained. From the intercept and with $f''$ determined from equilibrium intensities (see Eq. (2)) or from thermodynamic data, $D_{0}$ can be measured.

A specimen was brought to one temperature level and equilibrated, checking the total counts ($10^{4}-10^{5}$) for fifteen minutes or more. The temperature was then changed 15-25°C (except for Fe-Al, see below) and data was sampled at one position (k) every 1-20 seconds depending on the temperature and crystal. (The monitor was not employed.) The specimen was then brought back to the original temperature level, re-equilibrated and the process repeated for another point needed in the separation equations, cycling again for the other points needed. The entire process was repeated until $1.8 \times 10^{5}$ counts were collected at each point. This operation, including cycling, was carried out by the minicomputer system. Typical data are shown in figure 1a. This data was fitted with smooth curves and separated, resulting in the filled circles in figure 1b; the slope yields $\tau$. (The crosses in this figure represent unsmoothed differences; without the smoothing the relaxation times were the same, but the errors were ± 5% of $\tau$, rather than the ± 0.3% after smoothing.)

All measurements except those carried out on Cu$_{3}$Au, were made in a vacuum in the high-temperature attachment previously described. The furnace was within 0.25°C of equilibrium in 40 seconds after a ~ 25°C change, and the relaxation times of the change in intensity due to local order when the temperature is changed:

$$\ln \left[ I_{\text{SR0}}^{u}(k, t) - I_{\text{SR0}}^{u}(k, \infty) \right] =$$

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All measurements except those carried out on Cu$_{3}$Au, were made in a vacuum in the high-temperature attachment previously described. The furnace was within 0.25°C of equilibrium in 40 seconds after a ~ 25°C change, and the relaxation times...
(for the intensity to fall to $1/e$ of its original value) were longer. For Cu$_3$Au, the relaxation times were short ($\approx 10$ s) and a static atmosphere of purified helium was introduced into the camera, reducing the total time for a change in temperature of $25^\circ$C to 13 seconds.

3. Results and discussion. — 3.1 Equilibrium Studies. — 3.1.1 Temperature Dependence. — The term $T/\Theta_k^{U0}$ at $k^*$, the position of a superlattice reflection that would occur below $T_c$, was found to be linear vs. $T$ (except near $T_c$ for Cu$_3$Au, where heterophase fluctuations appear [25]). Thus $F(k^*)$ (from Eq. (2)) can be written as $\Delta S_m(T-T_{\text{instability}})$ where $T_{\text{instability}}$ is the temperature for continuous ordering obtained by extrapolating $F(k^*)$ to zero. The slopes, $\Delta S_m$, which correspond to the second derivative of the entropy of mixing are given in table I. If the system can be described in the linear, nearest-neighbour approximation previously mentioned (see Eq. (3)) and also if the elastic free energy and gradient energy are essentially independent of temperature over the range studied it is easy to show that:

$$\Delta S_m' = \frac{N_x k_B}{c(1-c)} \left[ 1 + \frac{d \ln \gamma}{d \ln c} \right], \quad (8)$$

where $\gamma$ is the activity coefficient of the solute. Hence, comparisons with measured thermodynamic data are also shown in the table. The poor comparisons are the first indications in this study that the simple nearest-neighbour approximation is inadequate.

3.1.2 $k$ Dependence. — For the nearest-neighbour approximation, from Eqs. (2) and (3):

$$\frac{N_x k_B T}{c(1-c) \Theta_k^{U0}(k)} = 2 \eta^2 M(k) = f'' + 2 KB^2(k). \quad (9)$$

A typical set of data for this equation is shown in figure 2. Note particularly that along the $\langle hh0 \rangle$ direction (filled circles) the points do not fall along a straight line. (This was also the case for Cu-25 at pct $\omega$ CoPt$_3$.) Ignoring this point for the moment, and performing a least squares fit to Eq. (9) yields the values of $f''$ and $K$ in table II. Comparisons of these values with data from thermodynamic studies (for $f''$) and with values calculated from pair potentials (for the gradient energy $K$) are fair. But if the data along the two lines is treated separately for the three alloys, $f''$ varies by as much as a factor of four, and $K$ by as much as a factor of two [22]. As $f''$ is a scalar and $K$ is a second rank tensor [29], the values must be the same in the two directions for this cubic material. The fit is not improved if a quasi-three shell model is employed for the elastic modulus, rather than a near neighbor model (see the Appendix). Furthermore, if $K(k)$ is calculated for Cu$_3$Au and pair potentials up to the sixth shell [21], this term varies by 30 pct from $h = 0.5$ to $h = 1.0$ [22].

![Figure 2](image-url)

**TABLE I**

<table>
<thead>
<tr>
<th>Alloy</th>
<th>Measured (±)</th>
<th>Estimated from Data in Ref. [26]</th>
<th>Temperature Range (°C)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Cu$_3$Au</td>
<td>1.006 (±0.002)</td>
<td>3.3</td>
<td>388.25-42.50</td>
</tr>
<tr>
<td>Cu$<em>{81.5}$Au$</em>{18.5}$</td>
<td>1.203 (±0.008)</td>
<td>2.7</td>
<td>331.00-369.30</td>
</tr>
<tr>
<td>CoPt$_3$</td>
<td>0.528 (±0.004)</td>
<td>2.7</td>
<td>683.50-720.30</td>
</tr>
<tr>
<td>Fe$<em>{70.8}$Al$</em>{29.2}$</td>
<td>1.90 (±0.18)</td>
<td>4.2</td>
<td>540.50-575.00</td>
</tr>
</tbody>
</table>

It appears much more satisfactory to take into account the variation of the gradient energy term with $k$. This is done in table III for Cu$_3$Au, the only alloy studied for which pair potentials are available [21, 28] as well as the necessary thermodynamic studies [26]. (The measured pair potentials actually include the elastic free energy term $2 \eta^2 M(k)$ and Langer's correction [30]). The fit along both lines seem reasonable. In table III a comparison is made of the calculated and measured gradient energy coefficient and the agreement is excellent.

It appears then that the gradient energy must be considered as a function of $k$ for alloys with only short range order, even at high temperatures. It
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Experimentally determined and estimated \( f'' \) and \( K \) for various alloys

\[
f'' \text{ (erg/cm}^3) \times 10^{11} \quad K \text{ (erg/cm)} \times 10^{-6} \]

<table>
<thead>
<tr>
<th></th>
<th></th>
<th></th>
<th></th>
<th></th>
</tr>
</thead>
<tbody>
<tr>
<td>Cu_3Au</td>
<td>1.04 (± 0.04)</td>
<td>1.37</td>
<td>- 7.8 (± 1.4)</td>
<td>{ - 5.29 (*), - 7.58 (*) }</td>
</tr>
<tr>
<td>Cu_{81.5}Au_{18.5}</td>
<td>2.05 (± 0.03)</td>
<td>1.12</td>
<td>- 12.2 (± 1.3)</td>
<td>not available (*)</td>
</tr>
<tr>
<td>CoPt_3</td>
<td>0.61 (± 0.03)</td>
<td>2.35 (*)</td>
<td>- 5.6 (± 0.8)</td>
<td>not available (*)</td>
</tr>
<tr>
<td>Fe_{70.8}Al_{29.2}</td>
<td>0.31 (± 0.01)</td>
<td>2.93 (*)</td>
<td>- 1.0 (± 0.4)</td>
<td>not available (*)</td>
</tr>
</tbody>
</table>

(*) Equation used for calculation is: \( K = [V_1 + 2V_3 + 2V_4]a^2 \) and \( V_i \)'s were obtained from Bardhan [21].

(\*) \( V_i \)'s obtained from reference [28].

(\*) Quantitative data for \( V_i \)'s are not available.

(\*) \( f'' \) was estimated using only the excess partial molar free energy but not the excess partial molar entropy; it was not available.

Table III

\( F(k) \) and \( K(k) \) at various positions \((k)\) in reciprocal space for Cu_3Au at 420 °C.
The measured \( F(k) \) and \( K(k) \) obtained directly from \( I_{SRO}(k) \) are shown for comparison

\[
f'' \text{ (erg/cm}^3) \quad K(k) \text{ (erg/cm)} \times 10^{-6} \quad B^2(k) \text{ (erg/cm}^2) \times 10^{17} \quad F(k) = f'' + 2K(k)B^2(k) \text{ (erg/cm}^3) \times 10^{11} \\
\]

<table>
<thead>
<tr>
<th>( k )</th>
<th>( f'' ) from Ref. [26, 27] (\times 10^{11})</th>
<th>( K(k) ) Estimated (*)</th>
<th>Measured</th>
<th>( B^2(k) ) Estimated</th>
<th>Measured</th>
</tr>
</thead>
<tbody>
<tr>
<td>(0, 0, 1)</td>
<td>1.37</td>
<td>- 5.29</td>
<td>- 6.00</td>
<td>0.1114</td>
<td>0.19</td>
</tr>
<tr>
<td>(0, 0, 0.9)</td>
<td>1.37</td>
<td>- 5.24</td>
<td>- 5.36</td>
<td>0.1087</td>
<td>0.23</td>
</tr>
<tr>
<td>(0, 0, 0.8)</td>
<td>1.37</td>
<td>- 5.07</td>
<td>- 5.03</td>
<td>0.1007</td>
<td>0.35</td>
</tr>
<tr>
<td>(0, 0, 0.7)</td>
<td>1.37</td>
<td>- 4.78</td>
<td>- 4.96</td>
<td>0.0884</td>
<td>0.53</td>
</tr>
<tr>
<td>(0, 0, 0.6)</td>
<td>1.37</td>
<td>- 4.35</td>
<td>- 5.47</td>
<td>0.0729</td>
<td>0.74</td>
</tr>
<tr>
<td>(0, 1, 1)</td>
<td>1.37</td>
<td>- 5.29</td>
<td>- 5.96</td>
<td>0.1114</td>
<td>0.19</td>
</tr>
<tr>
<td>(0, 0.9 0.9)</td>
<td>1.37</td>
<td>- 5.27</td>
<td>- 5.89</td>
<td>0.1113</td>
<td>0.20</td>
</tr>
<tr>
<td>(0, 0.8, 0.8)</td>
<td>1.37</td>
<td>- 5.20</td>
<td>- 5.74</td>
<td>0.1104</td>
<td>0.22</td>
</tr>
<tr>
<td>(0, 0.7, 0.7)</td>
<td>1.37</td>
<td>- 5.09</td>
<td>- 5.61</td>
<td>0.1066</td>
<td>0.29</td>
</tr>
<tr>
<td>(0, 0.6, 0.6)</td>
<td>1.37</td>
<td>- 4.86</td>
<td>- 6.01</td>
<td>0.0981</td>
<td>0.42</td>
</tr>
<tr>
<td>(0, 0.5, 0.5)</td>
<td>1.37</td>
<td>- 4.44</td>
<td>- 5.98</td>
<td>0.0835</td>
<td>0.63</td>
</tr>
</tbody>
</table>

(*) Calculated with a six shell approximation, Eq. (4b) and pair potentials in reference [21]. Using Moss' potentials (Ref. [28])

\[ K(001) = - 6.55 \times 10^{-8} \text{ erg/cm} \quad K(0, 0, 0.7) = - 4.99 \times 10^{-8} \text{ erg/cm} \quad K(0, 0.7, 0.7) = - 5.62 \times 10^{-8} \text{ erg/cm} \]

is, therefore, to be expected that this dependence will be important in kinetic studies, as local order will increase during a quench.

3.2 KINETIC STUDIES. — 3.2.1 \( k \) Dependence. —
The temperature changes were selected to give a total change in intensity of < 25 pct so that the total change in local concentration was the order of 13 pct or less. With these changes it was possible to make precise kinetic measurements at several \( k \) values, particularly for Cu-18.5 at pct Au and Cu-23 at pct Au; for these alloys the temperatures above \( T_c \) are sufficiently low that the relaxation times were considerably longer than that of the heating stage. Some results are given in table IV where the dependence of \( \tau \) on \( k \) is clearly indicated. Curiously, the relaxation time for scattering from mean square atom displacements \( (\tau_{\text{ms}}) \) is less than that for local...
The wavelength and crystallographic dependence of \( \tau \) is illustrated in figure 3a where the experimental data is compared to the predictions of the linear nearest-neighbor theory. The agreement is poor; considerable improvement is achieved if the calculation of \( \tau(k) \) is made with the measured \( F(k) \), figure 3b. Once again it is clear that the gradient energy is a strong function of \( k \).

It can be shown [22] that the reason investigators [31, 32] find a single relaxation time when employing resistivity measurements to obtain relaxation times during changes in local order is due to the fact that this measurement heavily weights \( \tau \) near the superlattice positions.

3.2.2 Temperature dependence. — Measurements were made only at the positions where superlattice reflections would occur below \( T_c \). No separation of the data was made into \( I_{\text{SKRO}} \) and intensity due to atomic displacements, because at those positions the latter is 5 pct or less than the total and tests indicated that the relaxation times were unaffected.

3.2.2.1 Cu-18.5 at pct Au. — According to Eq. (7b), if \( F(k) \) is linear vs. \( T \) with the form

\[
\Delta S_m^a(T - T_{\text{instability}}),
\]

then \( \ln \left[ \tau^{-1}(001)/(T - T_{\text{instability}}) \right] \) vs. \( 1/T \) should be linear (1). This appears to be the case in figure 4, where \( D_0 \) and the activation energy for diffusion obtained from Eq. (7b) are also shown. The error in \( D_0 \) includes the estimated errors in \( \Delta S^a, f^a, B^2(001) \).

\[\text{FIG. 3a.} \quad \text{The relaxation time} \; \tau(h) \; \text{vs.} \; h \; \text{for} \; 
\text{Cu}_{81.5}\text{Au}_{18.5} \; \text{at} \; 355 \; ^\circ\text{C} \; \to \; 330 \; ^\circ\text{C} \; \text{along} \; \langle 001 \rangle \; \text{and} \; \langle h0l \rangle \; \text{directions.} \; \text{The following values were used for calculations:}
\]

\[
a = 3.717 \; \text{Å}, \quad D = 0.034 \times 10^{-4} \; \text{cm}^2/\text{s}, \quad f^e = 2.05 \times 10^{14} \; \text{erg/cm}^3, \quad K = 12.2 \times 10^{-8} \; \text{erg/cm};
\]

i.e., the values in table II. The elastic free energy \((2 \pi^2 M(k))\) was calculated as indicated in the Appendix. The experimentally determined \( \tau \)'s normalized at the 001 position to the predicted curve are also shown in the figure.

\[\text{FIG. 3b.} \quad \text{The relaxation time estimated with the measured Fourier representation of total free energy, } F(k) \; \text{(see Table III). Experimental results are also shown.}
\]

| Table IV |
|------------------|------------------|
| Relaxation times at various \( k \) positions |

| \( \text{Cu}_{81.5}\text{Au}_{18.5} \) |
| \( T(\text{C}) \text{(*)} \) | \( k \) | \( \tau_{\text{SKRO}} \) (s) | \( \tau_{\text{BS}} \) (s) |
|------------------|------------------|
| 355 \to 330      | (0, 0, 1)        | 166.74 (± 0.56) | 145.24 (± 1.52) |
| 355 \to 330      | (0, 0, 1, 1)     | 98.49 (± 0.87)  | 91.15 (± 2.33)  |
| 355 \to 330      | (0, 1, 1)        | 165.05 (± 0.86) | 144.00 (± 2.21) |
| 355 \to 330      | (0, 1, 1, 1)     | 130.36 (± 1.27) | 120.50 (± 2.98) |
| 355 \to 330      | (0, 1, 2, 1, 2)  | 68.03 (± 1.50)  | 50.25 (± 7.12)  |

(*) Arrows indicate the direction of temperature change.
(**) \( \tau_{\text{BS}} \) at 0, 1.2, 1.2 was too fast to be measured.

and the error in the intercept of the least squares fit of figure 4.

The results are compared to other data for this system in table V. The values reported here are in good agreement with tracer studies [35] but not with the work on thin films [8].

3.2.2.2 Cu_{71}Au. — The kinetics are so rapid in this alloy that measurements could only be made at the superlattice position, and within 4-5 °C of \( T_c \). Because of this small temperature range no activation...
FIG. 4. - \( \ln \left[ \frac{1}{T} - \frac{1}{T_{\text{trans}}(\text{def})} \right] \times 10^3 \) vs. \( \frac{1}{T} \) for Cu-18.5 at \( \text{pct Au} \), quenched from 25-30 \(^\circ\)C above to the indicated temperatures.

**Table Va**

Interdiffusion coefficients for Cu-Au alloys

<table>
<thead>
<tr>
<th>Alloy</th>
<th>( Q ) (kcal/mole)</th>
<th>( D_o ) (cm(^2)/s)</th>
<th>Temperature Range ((^\circ)C)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Cu-18.5 at ( \text{pct Au} ) (*)</td>
<td>34.0 (± 3.3)</td>
<td>0.004 (± 0.005)</td>
<td>665-800</td>
</tr>
<tr>
<td>Cu-10.5 at ( \text{pct Au} ) (*)</td>
<td>41.6 (± 3.9)</td>
<td>0.107</td>
<td>665-800</td>
</tr>
<tr>
<td>Cu-1.5 at ( \text{pct Au} ) (*)</td>
<td>44.4 (± 2.9)</td>
<td>0.034 (± 0.001)</td>
<td>330-345.5</td>
</tr>
<tr>
<td>Cu-0.5 at ( \text{pct Au} ) (*)</td>
<td>33.8 (± 5.2)</td>
<td>0.001</td>
<td>200-260</td>
</tr>
<tr>
<td>Cu-0.0 at ( \text{pct Au} ) (*)</td>
<td>34.4 (± 8.7)</td>
<td>0.001 8</td>
<td>665-800</td>
</tr>
</tbody>
</table>

**Table Vb**

Self diffusivity and tracer diffusivity in Cu-Au alloys

<table>
<thead>
<tr>
<th>Self diffusivity</th>
<th>( Q ) (kcal/mole)</th>
<th>( D_o ) (cm(^2)/s)</th>
<th>Temperature Range ((^\circ)C)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Au in Cu ( ^{33} )</td>
<td>41.70 (± 0.3)</td>
<td>0.091 (± 0.001)</td>
<td></td>
</tr>
<tr>
<td>Cu in Cu ( ^{34} )</td>
<td>47.12 (± 0.3)</td>
<td>0.200 (± 0.030)</td>
<td></td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Tracer diffusivity</th>
<th>( Q ) (kcal/mole)</th>
<th>( D_o ) (cm(^2)/s)</th>
<th>Temperature Range ((^\circ)C)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Au in Cu ( ^{33} )</td>
<td>42.6 (± 1.0)</td>
<td>0.030 (± 0.002)</td>
<td>400-1050</td>
</tr>
<tr>
<td>Cu in Au ( ^{34} )</td>
<td>40.65</td>
<td>0.105</td>
<td>700-906</td>
</tr>
</tbody>
</table>

(*) Data obtained from reference [8].

(1) Current result.

From measurements of relaxation obtained by following changes in Young's modulus obtained by Weisberg and Quimby [37], \( \tau \) was 14.1 ± 0.3 s at \( T_c + 0.5 \(^\circ\)C \), in good agreement with the results of this study, 11.7 ± 0.1 s.

3.2.2.3 CoPt\(_3\). — The data for determining \( D_o \) and \( Q \) are shown in figure 6. The activation energy is slightly larger than that measured for antiphase domain growth below \( T_c \) [38], 63 kcal/mole. The values for self-diffusion in Co and Pt are 68 and 65 kcal/mole, respectively [39].

3.2.2.4 Fe-29.2 at \( \text{pct Al} \). — Measurable relaxation times could be obtained only 100 \(^\circ\)C below \( T_c \) (525 \(^\circ\)C). At higher temperatures, the reaction was too rapid. Kinetic measurements were therefore made between 307 and 416 \(^\circ\)C. Both quenching and heating were employed, and there was little difference. The results are shown in figure 7. (The term \( D_o \) could not be obtained, since \( K, \Delta S_m \), etc., were not available.) No data on the diffusivities in the ordered phase are available, but the results are much closer to the activation energy for self-diffusion in Al than that in Fe [34].
4. Conclusions. — 4.1 Results are reported here for the first time on the values of the transform of the Helmholtz free energy vs. wavelength of composition fluctuations in alloys with local order, and on the variation in relaxation time of these fluctuations vs. crystallographic direction. These results clearly indicate that the nearest-neighbour approximations in the theory of continuous ordering are inadequate. The gradient energy is a strong function of the wavelength and crystallographic direction.

4.2 A technique has been demonstrated for measuring the interdiffusion coefficient at low temperatures in bulk alloy crystals. This method involves measurements of the absolute intensity due to local order vs. temperature in the range of interest, and measurement of the changes in diffuse intensity with time for small temperature changes in this range.

4.3 A study similar to this one but above a miscibility gap would be of considerable interest as a more quantitative test of the theory of spinodal decomposition than has heretofore been obtained.

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The authors would especially like to thank Prof. F. Kayser, Iowa State University for the loan of a single crystal of Cu-18.5 at pct Au, and Dr. L. Guttman for crystals of Fe-29.2 at pct Al. Mr. Robert Lloyd designed the computer interface for monitoring the temperature. Profs. L. H. Schwartz, J. E. Hilliard, H. Cook and Dr. L. Guttman provided valuable discussions.

Appendix. — Following Cook and de Fontaine [5] the elastic free energy of a crystal may be written as follows, starting with a harmonic expansion in terms of transforms of deviations $q$ and $u$ from the average composition and the average lattice, respectively:

$$ F_e = \frac{N}{2} \sum_{k \in k_R} F_e(k) Q^*(k) Q(k), $$

(A.1a)

with:

$$ F_e(k) = \Phi(k) - \sum_{ij} \Phi_i(k) \Phi_j(k)^*, $$

(A.1b)

where $\Phi(k)$, $\Phi_i(k)$ and $\Phi_j(k)$ are Fourier transforms of solute ($\varphi(r, r')$), solute-lattice ($\varphi(r, r')$) and lattice-
A COMPARISON OF EXPERIMENT AND THEORY OF CONTINUOUS ORIÉKING

The product, $Q^*(k)$ $Q(k)$, can be interpreted as the solute intensity related to $I_{SO}(k)$, per atom. $F_e(k)$ is normally a function of wavelength as well as a function of the crystallographical direction.

Following the nomenclature used by Born [41], the appropriate Fourier transformations of coupling constants for both fcc and bcc materials are listed in table A.1, employing shorthand notations for the C.P.'s as defined for example by Walker [42]. (C.P.'s beyond the third coordinate shell are assumed here to be zero.)

Then, an effective elastic modulus $M(k)$ is defined for cubic crystals as:

$$M(k) = \frac{F_e(k)}{2\Omega^2},$$

where $\eta$ is the linear expansion coefficient per unit composition change, $\eta = \frac{\partial \ln a}{\partial c}$ and $\Omega$ is the atomic unit volume.

1) FIRST-SHELL MODEL. — a) fcc Structure. - - In the derivation, only nearest-neighbour interactions are considered, such that only $\alpha_1, \alpha_1, \alpha_1, \beta_1$, and $\gamma_1$ (Table A.1) are non-zero. Therefore, with the long wavelength relationships (Table A.2) the Fourier transforms of the coupling parameters can be written as follows, with the continuous variables in reciprocal space, $h_1, h_2, h_3$:

$$\Phi_{11}(h_1, h_2, h_3) = ac_{11}(2 - \cos \pi h_1 \cos \pi h_2 - \cos \pi h_1 \cos \pi h_3) + 2 a c_{44} - \frac{1}{2} c_{11} \cos \pi h_1 \cos \pi h_3,$$

$$\Phi_{12}(h_1, h_2, h_3) = \Phi_{21}(h_1, h_2, h_3) = a(c_{12} + c_{44}) \sin \pi h_1 \sin \pi h_2,$$

$$\Phi_{13}(h_1, h_2, h_3) = i \left[ \frac{1}{4} a^2 \eta(c_{11} + 2 c_{12}) \sin \pi h_1(\cos \pi h_2 + \cos \pi h_3) \right].$$

TABLE A.1

(a) Fourier transforms of bcc coupling parameters (*)

$$\Phi_{11}(h_1, h_2, h_3) = 8 \alpha_1(1 - \cos \pi h_1 \cos \pi h_2 \cos \pi h_3) + 2 \alpha_2(1 - \cos 2 \pi h_1) +$$

$$+ 2 \beta_2(2 - \cos 2 \pi h_2 - \cos 2 \pi h_3) + 4 \gamma_1(1 - \cos 2 \pi h_2 \cos 2 \pi h_3) +$$

$$+ 8 \alpha_3(2 - \cos 2 \pi h_1 \cos 2 \pi h_2 \cos 2 \pi h_3) \]$$

$$\Phi_{12}(h_1, h_2, h_3) = 8 \gamma_1 \sin \pi h_1 \sin \pi h_2 \sin \pi h_3 + 4 \gamma_3 \sin 2 \pi h_2 \sin 2 \pi h_3,$$

$$\Phi_{13}(h_1, h_2, h_3) = i[8 \alpha_1 \sin \pi h_1 \cos \pi h_2 \cos \pi h_3 + 2 \alpha_2 \sin 2 \pi h_1 + 4 \alpha_3 \sin 2 \pi h_1(\cos 2 \pi h_2 + \cos 2 \pi h_3)]$$

$$\Phi_{13}(h_1, h_2, h_3) = 8 \alpha_1(1 + \cos \pi h_1 \cos \pi h_2 \cos \pi h_3) +$$

$$+ 4 \beta_3(3 + \cos 2 \pi h_1 \cos 2 \pi h_2 \cos 2 \pi h_3) +$$

$$+ 2 \beta_3(3 + \cos 2 \pi h_1 + \cos 2 \pi h_2 + \cos 2 \pi h_3).$$

(b) Fourier transforms of fcc coupling parameters

$$\Phi_{11}(h_1, h_2, h_3) = 4 \alpha_1(2 - \cos \pi h_1 \cos \pi h_2 - \cos \pi h_1 \cos \pi h_3) + 4 \beta_1(1 - \cos \pi h_2 \cos \pi h_3) +$$

$$+ 2 \beta_2(1 - \cos 2 \pi h_1) + 2 \beta_2(2 - \cos 2 \pi h_2 - \cos 2 \pi h_3) + 8 \alpha_3(1 - \cos 2 \pi h_1 \cos \pi h_2 \cos \pi h_3) +$$

$$+ 8 \beta_3(2 - \cos \pi h_1 \cos 2 \pi h_2 \cos 2 \pi h_3) \]$$

$$\Phi_{12}(h_1, h_2, h_3) = 4 \gamma_1 \sin \pi h_1 \sin \pi h_2 + 8 \gamma_3 \sin \pi h_1 \sin \pi h_2 \cos 2 \pi h_3 +$$

$$+ 8 \gamma_5 \cos \pi h_3(\sin 2 \pi h_1 \sin \pi h_2 + \sin \pi h_1 \sin 2 \pi h_3)$$

$$\Phi_{13}(h_1, h_2, h_3) = i[4 \alpha_1 \sin \pi h_1(\cos \pi h_2 + \cos \pi h_3) + 2 \alpha_1 \sin 2 \pi h_1 + 8 \alpha_5 \sin 2 \pi h_1 \cos \pi h_2 \cos \pi h_3 m +$$

$$+ 8 \beta_3 \sin \pi h_1(\cos \pi h_2 \cos 2 \pi h_3 + \cos \pi h_3 \cos 2 \pi h_2) \]$$

$$\Phi_{13}(h_1, h_2, h_3) = 4 \alpha_1(3 + \cos \pi h_1 \cos \pi h_2 + \cos \pi h_1 \cos \pi h_3 + \cos \pi h_2 \cos \pi h_3) +$$

$$+ 2 \alpha_3(3 + \cos 2 \pi h_1 + \cos 2 \pi h_2 + \cos 2 \pi h_3) +$$

$$+ 8 \alpha_3(3 + \cos \pi h_1 \cos \pi h_2 \cos 2 \pi h_3 + \cos \pi h_1 \cos 2 \pi h_2 \cos \pi h_3 + \cos 2 \pi h_1 \cos \pi h_2 \cos \pi h_3).$$

(*) C.P.'s beyond the third shell are assumed to be zero.
\[ \Phi(h_1, h_2, h_3) = 4 \alpha_1 (3 + \cos \pi h_1 \cos \pi h_2 + \cos \pi h_3 \cos \pi h_1 + \cos \pi h_2 \cos \pi h_3). \]

The rest of the terms appearing in Eq. (A.1b) can be obtained by permutations among \( h_1, h_2 \) and \( h_3 \). The shorthand notation for solute coupling constants, \( \alpha_1 \), can be expressed in terms of the lattice coupling constants using Eq. (41) in reference [5], which, for cubic crystals is:

\[ \varphi(r) = \eta \varphi_i(r) x_i(r) + \frac{1}{4} S(ii) \eta^2 \varphi_i(r) x_i(r), \quad (A.5) \]

where:

\[ S(ii) = \begin{cases} 1 & \text{if } i = j \\ \frac{2c_{ii}}{c_{ii} + c_{jj}} & \text{if } i \neq j, \end{cases} \]

and \( x_i(r) \) are the components of an interatomic vector in real space. Then, with the shorthand notations for the C.P.'s defined by Walker [42]:

\[ \phi\left(\frac{1}{2}, 0, \frac{1}{2}\right) = \alpha_1 = \frac{1}{2} \eta \left[ \varphi_{11} \left(\frac{1}{2}, 0, \frac{1}{2}\right) + \varphi_{12} \left(\frac{1}{2}, 0, \frac{1}{2}\right) \right] + \frac{1}{4} \eta^2 \left[ \varphi_{11} \left(\frac{1}{2}, 0, \frac{1}{2}\right) + \varphi_{12} \left(\frac{1}{2}, 0, \frac{1}{2}\right) \right] + \frac{1}{4} \eta^2 \frac{c_{12}}{c_{12} + c_{44}} (\frac{1}{2}, 0, \frac{1}{2}) = a \eta \alpha_1 - \frac{1}{8} \eta^2 \alpha_1 - \frac{1}{4} \frac{c_{12}}{c_{11} + c_{44}} a^2 \eta^2 \gamma_1. \quad (A.6a) \]

and:

\[ \alpha_1 = \frac{1}{32} a^3 \eta^2 (c_{11} + 2 c_{12}). \quad (A.6b) \]

The effective modulus can now be expressed along three crystallographic directions. For the \( \langle h00 \rangle \) direction with \( h_1 = h \) and \( h_2 = h_3 = 0 \), Eq. (A.4) becomes:

\[ \begin{align*}
\Phi_{11}(h00) &= 2 ac_{11}(1 - \cos \pi h), \\
\Phi_{22}(h00) &= \Phi_{33}(h00) = 2 ac_{44}(1 - \cos \pi h), \\
\Phi_{12}(h00) &= \Phi_{13}(h00) = \Phi_{31}(h00) = \Phi_{23}(h00) = \Phi_{22}(h00) = 0, \\
\Phi_1(h00) &= \alpha_1 = \frac{1}{2} a^2 \eta (c_{11} + 2 c_{12}) \sin \pi h, \\
\Phi_2(h00) &= \Phi_3(h00) = 0, \\
\Phi(h00) &= \frac{1}{4} a^3 \eta^2 (c_{11} + 2 c_{12}) (2 + \cos \pi h). \quad (A.7) \end{align*} \]

The inverse matrix \( \{ \Phi_i(h00) \} \) is obtained from the matrix \( \{ \Phi_i(h00) \} \):

\[ \{ \Phi_i(h00) \} = \begin{bmatrix}
1 & 0 & 0 \\
0 & 2 ac_{11}(1 - \cos \pi h) & 0 \\
0 & 0 & 2 ac_{44}(1 - \cos \pi h)
\end{bmatrix} \]

Therefore, the elastical free energy \( F_e(h00) \) in its Fourier representation is obtained from Eq. (A.1b):

\[ F_e(h00) = \frac{1}{8} a^3 \eta^2 (c_{11} + 2 c_{12}) \left[ 3 - 2 \frac{c_{12}}{c_{11}} + \left( 1 - 2 \frac{c_{12}}{c_{11}} \right) \cos \pi h \right]. \quad (A.8) \]
and subsequently the effective elastical modulus $M(h00)$ is:

$$M(h00) = \frac{F_1(h00)}{2\Omega^2} = \frac{1}{2} (c_{11} + 2c_{12}) \left[ \frac{3}{2} - \frac{c_{11}}{c_{11}} + \left( \frac{1}{2} - \frac{c_{11}}{c_{11}} \right) \cos \pi h \right].$$  \hfill (A.9)

Similarly for the $\langle hh0 \rangle$ direction:

$$M(hh0) = \frac{1}{4} c_{11} \left[ \left( 1 + 2 \frac{c_{11} + 2c_{12}}{c_{11}} \right) (3 + 2 \cos \pi h + \cos^2 \pi h) - \frac{\left( 1 + \frac{2c_{11} + 2c_{44}}{c_{11}} \right)^2 (1 + \cos \pi h)^3}{2c_{44} + \left( c_{44} + \left( c_{11} + c_{44} \right) (1 + \cos \pi h) \right)} \right].$$  \hfill (A.10)

For the $\langle hhh \rangle$ direction:

$$M(hhh) = \frac{3}{4} (c_{11} + 2c_{12}) (1 + \cos^2 \pi h) + \frac{3}{2} \frac{(c_{11} + 2c_{12})^2}{(c_{12} + c_{44})} \left[ 3 + \left( \frac{c_{11} + 2c_{44}}{c_{12} + c_{44}} \right)^2 \right] \sin^2 \pi h \cos^2 \pi h \left[ 1 - \left( \frac{c_{11} + 2c_{44}}{c_{12} + c_{44}} \right)^2 \right].$$  \hfill (A.11)

b) bcc Structure. — Assuming all the nearest-neighbour coupling constants plus $\beta_2$ are non-zero, the Fourier transforms of the coupling constants for a bcc system (Table A.1) are given by:

$$\Phi(h_1, h_2, h_3) = 8 a_1 \sin \pi h_1 \cos \pi h_2 \cos \pi h_3,$$

$$\Phi_4(h_1, h_2, h_3) = 8 i \tilde{a}_1 \sin \pi h_1 \cos \pi h_2 \cos \pi h_3,$$

$$\Phi_1(h_1, h_2, h_3) = 8 a_1 (1 - \cos \pi h_1 \cos \pi h_2 \cos \pi h_3) + 2 \beta_2 (2 - \cos \pi h_2 - \cos 2 \pi h_3),$$

and

$$\Phi_2(h_1, h_2, h_3) = 8 \gamma_1 \sin \pi h_1 \sin \pi h_2 \sin \pi h_3.$$  \hfill (A.12)

From the long wavelength relationships (Table A.2):

$$\alpha_1 = \frac{1}{2} a c_{11}, \quad \beta_2 = \frac{1}{2} a(c_{44} - c_{11}), \quad \gamma_1 = \frac{1}{4} a(c_{44} + c_{12}),$$

$$\tilde{a}_1 = \frac{1}{8} a^2 \eta(c_{11} + 2c_{12}), \quad \text{and} \quad \tilde{a}_1 = \frac{3}{32} a^2 \eta^2(c_{11} + 2c_{12}).$$  \hfill (A.13)

The effective modulus will again be calculated along three simple crystallographic directions. For the $\langle h00 \rangle$ direction:

$$M(h00) = \frac{1}{4} (c_{11} + 2c_{12}) (3 - c_{11} - 2c_{12}) (1 + \cos \pi h).$$  \hfill (A.14)

For the $\langle hh0 \rangle$ direction:

$$M(hh0) = \frac{3}{4} (c_{11} + 2c_{12}) (1 + \cos^2 \pi h) - \frac{(c_{11} + 2c_{12})^2}{c_{11} + c_{12} + 2c_{44}} \cos^3 \pi h.$$  \hfill (A.15)

For the $\langle hhh \rangle$ direction:

$$M(hhh) = \frac{3}{4} (c_{11} + 2c_{12}) \left[ 1 + \cos^2 \pi h \frac{(c_{11} + 2c_{12}) \sin^2 \pi h \cos^2 \pi h}{c_{11}(1 - \cos \pi h) + (c_{11} + c_{12} + c_{44}) \sin^2 \pi h \cos \pi h} + (c_{44} - c_{11}) \sin^2 \pi h \right].$$  \hfill (A.16)

The elastic constants employed in this work are given in table A.3.
Long wavelength relations for coupling parameters.

(i) bcc
\[ 2 \alpha_1 + 2 \beta_2 + 4 \alpha_3 + 4 \beta_3 = aC_{11} \]
\[ 4 \gamma_1 + 8 \gamma_3 = a(C_{11} + 2 C_{12}) \]
\[ 4 \alpha_1 + 2 \alpha_2 + 8 \alpha_3 = \frac{1}{2} a^2 \eta(C_{11} + 2 C_{12}) \]

(ii) fcc
\[ 4 \alpha_1 + 4 \alpha_2 + 16 \alpha_3 + 8 \beta_3 = aC_{11} \]
\[ 2 \alpha_1 + 2 \beta_1 + 4 \beta_3 + 4 \alpha_3 + 20 \beta_3 = aC_{44} \]
\[ 4 \gamma_1 + 8 \gamma_3 + 32 \delta_3 = a(C_{44} + 2 C_{12}) \]
\[ 8 \alpha_1 + 4 \alpha_2 + 16(\alpha_3 + \beta_3) = \frac{1}{2} a^2 \eta(C_{11} + 2 C_{12}) \]

where \( C_{11}, C_{12} \) and \( C_{44} \) are the usual elastic constants.

<table>
<thead>
<tr>
<th>System</th>
<th>( C_{11} )</th>
<th>( C_{12} )</th>
<th>( C_{44} )</th>
</tr>
</thead>
<tbody>
<tr>
<td>( Cu_3Au )</td>
<td>( 1.590 \times 10^{12} )</td>
<td>1.246</td>
<td>0.549</td>
</tr>
<tr>
<td>( Cu_{81.5}Au_{18.5} )</td>
<td>1.795</td>
<td>1.355</td>
<td>0.589</td>
</tr>
<tr>
<td>( CoPt_3 )</td>
<td>3.077</td>
<td>2.193</td>
<td>0.735</td>
</tr>
<tr>
<td>( Fe_{70.8}Al_{29.2} )</td>
<td>1.664</td>
<td>1.226</td>
<td>1.310</td>
</tr>
</tbody>
</table>

\( \eta = M(k) \) (units are dyne/cm²)

<table>
<thead>
<tr>
<th>System</th>
<th>( \eta )</th>
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<td>1.664</td>
</tr>
</tbody>
</table>

Quasi-three-shell model

Therefore, a quasi-three-shell model (which assumes that \( \alpha_1, \alpha_2, \alpha_3, \beta_1, \beta_2, \beta_3, \gamma_1, \gamma_3, \delta_3 \), plus \( \tilde{\alpha}_1 \) and \( \tilde{\alpha}_2 \) are non-zero) may be assumed to obtain the effective elastic modulus along the \( \langle h00 \rangle \) and \( \langle hh0 \rangle \) lines for an fcc structure. Hence:

\[ M(h00) = \frac{2}{a^3 \eta^2} \left[ 8 \tilde{\alpha}_1(2 + \cos \pi h) - \frac{8 \tilde{\alpha}_1 \sin \pi h}{(8 \alpha_1 + 16 \alpha_3)(1 - \cos \pi h) + (2 \alpha_2 + 8 \alpha_3)(1 - \cos 2 \pi h)} \right] \quad \text{(A.18)} \]

and:

\[ M(hh0) = \frac{2}{a^3 \eta^2} \times \left[ 4 \tilde{\alpha}_1(3 + 2 \cos \pi h + \cos^2 \pi h) - \frac{8 \tilde{\alpha}_1 \sin \pi h(1 + \cos \pi h)}{4 \alpha_1(2 - \cos \pi h - \cos^2 \pi h) + 4 \beta_1(1 - \cos \pi h) + 2 \alpha_2 + 2 \beta_2)(1 - \cos 2 \pi h) + 8 \alpha_3(1 - \cos \pi h \cos^2 \pi h) + 16 \delta_3 \sin \pi h \sin 2 \pi h + 4 \beta_3(1 - \cos 2 \pi h - \cos^2 \pi h) + (4 \gamma_1 + 8 \gamma_3) \sin^2 \pi h} \right] \quad \text{(A.19)} \]

An attempt was made by Gragg and Bardhan [47] to evaluate the force constants using the X-ray diffuse scattering pattern due to local atomic order and static atomic displacements. However, the analysis indicated large errors.
References