

Carbon isotope evidence for sedimentary miliacin as a tracer of *Panicum miliaceum* (broomcorn millet) in the sediments of Lake le Bourget (French Alps).

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Abstract

We here report on the determination of the carbon isotopic composition of miliacin (olean-18-en-3 β ol methyl ether), extracted from the sediments of Lake le Bourget (French Alps). It is compared to the $\delta^{13}\text{C}$ of miliacin extracted from *Panicum miliaceum* (broomcorn millet, a C4 plant) and *Chionochloa sp.* (a C3 plant). The $\delta^{13}\text{C}$ of sedimentary miliacin (-21.5 ‰) is very close to that of miliacin extracted from bran (-23 ‰) and seeds (-23.5 ‰) of *P. miliaceum* and significantly different from that of *Chionochloa sp.* (-33 ‰). These results provide additional support for the use of sedimentary miliacin as a tracer of broomcorn millet, a C4 cereal cultivated since the Bronze Age around Lake le Bourget. These findings illustrate the potential of this compound to reconstruct past agriculture from lake sediment archives. Finally, considering the high abundances of miliacin in the bran of *P. miliaceum* this compound could have been wind-transported to the sediment during threshing and winnowing on the lake shore.

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1. Introduction

The development of novel tracers permitting unravelling possible interactions between past climate and human activities is of direct relevance within the context of future Global Change (Dearing, 2006). We recently proposed that miliacin (olean-18-en-3 β -ol methyl ether or germanicol methyl ether, Figure 1) detected in the sediments of Lake le Bourget (French Alps) could be a specific tracer for the former cultivation of *Panicum miliaceum* (broomcorn millet) in the catchment (Jacob et al., 2008). This assumption was based on the fact that miliacin is the sole pentacyclic triterpene methyl ether (PTME) found in Lake le Bourget sediments. Because of the peculiar resistance of PTMEs to diagenesis (Jacob et al., 2005), this implies that miliacin was certainly the sole PTME produced in large amounts in the catchment. Few plants are reputed to synthesize miliacin as the sole PTME. These consist of *Microstegium vimineum*, *Syntherisma sanguinalis*, *Glyceria acutiflora*, *Panicum sp.* and *P. miliaceum* (Jacob et al., 2005). By comparing this list with an archaeobotanical inventory of plants that developed around the lake during the Bronze Age (Bouby and Billaud, 2001), we conclude that *P. miliaceum* is the most probable biological source of miliacin in the sediments. Therefore, miliacin in the sediments of Lake Le Bourget attest to the large development of *P. miliaceum* in the catchment. Since this plant is not indigenous to the study area, this implies that this plant was intentionally introduced (for agriculture) and that sedimentary miliacin attests to the cultivation of *P. miliaceum* in the watershed. As a matter of fact, the first appearance of miliacin in the sediments around 1700 BC slightly precedes the expected introduction of *P. miliaceum* in the region (Marinval, 1995). In addition, the drastic decrease of miliacin concentrations around 800 BC coincides with several evidences of human abandonment of lake shore dwellings due to a phase of climatic deterioration (Magny, 2004; Billaud and Marguet, 2005).

Despite these arguments, the certification of *P. miliaceum* as the source of miliacin in the sediments of Lake le Bourget requires additional investigation. *P. miliaceum* is a C4 plant imported from Asia whereas plants growing naturally around in the Alps have a C3 metabolism. By comparing the $\delta^{13}\text{C}$ of miliacin extracted from these sediments with that of miliacin extracted from *P. miliaceum* and *Chionochloa sp.*, we aim to confirm the use of sedimentary miliacin as a tracer of *P. miliaceum*, and by extension, of human cropping.

2. Samples and Methods

2.1. Isolation of miliacin

Sediment extraction was described previously (Jacob et al., 2008). The resultant neutral lipid fractions of eleven 1 cm-thick sediment samples (22 g in total) from Lake le Bourget core LDB04 (at depths corresponding to the 1225 - 770 BC time interval; Jacob et al., 2008) were combined in order to obtain enough miliacin for $\delta^{13}\text{C}$ measurements. This extract was dissolved in 0.5 ml of hexane and then fractionated by flash chromatography on silica first activated at 110 °C for 48 h and then deactivated with 5% water. After the successive elution of aliphatic hydrocarbons (fraction F1 obtained with 2 ml hexane) and aromatics (fraction F2 obtained with 1 ml hexane and 2 ml hexane:toluene 3:1), miliacin was found in fractions F3 and F4 obtained with 2 ml hexane:toluene (1:1) and 2 ml hexane:ethyl acetate (19:1), respectively. Fractions F3 and F4 were combined and dried under nitrogen.

4.3 g of *P. miliaceum* seeds and 2.4 g of *P. miliaceum* bran were extracted with the same protocol as described in Jacob et al. (2008). The total extracts were dissolved in 2 ml of hexane, and 0.5 ml of each extract was fractionated as above. Purified miliacin was obtained from Pr. R. Smith (Loughborough University, UK) and Pr. H.E. Connor (Christchurch University, NZ). This compound was purified from *Chionochloa sp.* extracts (Russell et al., 1976; Connor and Purdie, 1976; Connor, 2004).

2.2. GC-MS analyses

In order to verify the purity of miliacin in the combined F3+F4 fraction of *P. miliaceum* seeds and bran as well as in F3+F4 fraction of LDB04 sediment extracts, these fractions were analysed by GC-MS on a TRACE-PolarisGCQ. The gas chromatograph was fitted with a Rtx-5MS capillary column (30 m, 0.25 mm i.d., 0.25 μm film thickness) with 5 m of guard column. The GC operating conditions were: temperature held at 40 $^{\circ}\text{C}$ for 1 min, then increased from 40 to 120 $^{\circ}\text{C}$ at 30 $^{\circ}\text{C min}^{-1}$, 120 to 300 $^{\circ}\text{C}$ at 3 $^{\circ}\text{C min}^{-1}$, with a final isothermal hold at 300 $^{\circ}\text{C}$ over 30 min. The sample was injected splitless, with the injector temperature set at 280 $^{\circ}\text{C}$. Helium was the carrier gas. The mass spectrometer was operated in the electron ionisation (EI) mode at 70 eV ionization energy and scanned from 50 to 600 Da. Concentrations of miliacin were estimated by using pure miliacin isolated from *Chionochloa sp.* as external standard.

2.3. GC-C-IRMS analyses

Miliacin standard, miliacin extracted from *P. miliaceum* bran and the miliacin-containing fraction from Lake le Bourget sediments were analysed by GC-C-IRMS, in a continuous flow mode. $\delta^{13}\text{C}$ values were determined with a HP5860 gas chromatograph coupled to a Micromass Optima (GVI) isotope ratio mass spectrometer via an Isochron III combustion interface heated at 850 $^{\circ}\text{C}$. The gas chromatograph was fitted with a BPX 5 column (60m, 0.32 mm i.d., 0.25 μm film thickness). The GC operating conditions were: temperature held at 50 $^{\circ}\text{C}$ for 2 min, then increased from 40 to 200 $^{\circ}\text{C}$ at 30 $^{\circ}\text{C min}^{-1}$, 120 to 350 $^{\circ}\text{C}$ at 5 $^{\circ}\text{C min}^{-1}$, with final isothermal hold at 350 $^{\circ}\text{C}$ over 20 min. The samples were injected in splitless mode, with the injector temperature set at 280 $^{\circ}\text{C}$. Helium was the carrier gas. Ultra high purity carbon dioxide gas with known $\delta^{13}\text{C}$ value was pulsed three times at the

beginning and end of each analysis. The accuracy of carbon isotope measurements was evaluated using a standard of three n-alkanes and dodecanoic methyl ester with offline measured $\delta^{13}\text{C}$ values (VGmix from Micromass). All $\delta^{13}\text{C}$ values are reported in standard ‰ notation relative to VPDB.

2.4. TC/EA-IRMS analyses

Standard miliacin crystals purified from *Chionochloa sp.* extracts were analysed by Thermal Conversion/Elemental Analyzer-Isotope Ratio Mass Spectrometry (TC/EA-IRMS) in dual inlet mode. $\delta^{13}\text{C}$ values were determined with a CARLO ERBA NA 1500 elemental analyser coupled to a Micromass SIRA 10, via a triple trap purification interface.

3. Results and discussion

Partial Total Ion Chromatograms (TIC) of the F3+F4 fraction obtained from Lake le Bourget sediments, *P. miliaceum* seeds and bran extracts and pure miliacin extracted from *Chionochloa sp.* are displayed in Figure 2. All TIC traces show relatively pure miliacin that elutes at 40.75 min in our temperature programme. The absence of any other fragment than those typical for miliacin mass spectra in the peaks eluting at 40.75 min with our temperature programme allowed us to exclude any significant coelution in all analyses (Figure 3). Concentrations in miliacin of the different samples are reported in Table 1. *P. miliaceum* seeds and bran contain 0.3 and 0.16 mg.g⁻¹ (plant dry mass) of miliacin, respectively. The average concentration of miliacin in the selected samples from Lake le Bourget sediments reaches 0.49 µg.g⁻¹ sediment.

The first record of miliacin in the literature was provided by Itô (1934) in a paper dealing with the chemical composition of some gramineae oils. This author found miliacin in

P. miliaceum and *Syntherisma sanguinalis* (hairy crabgrass). Miliacin was also found in several species of *Chionochloa*, sometimes associated with other PTMEs such as arundoin, lupeol methyl ether, β -amyrin methyl ether, parkeol and cycloartenol methyl ethers (Russell et al., 1976; Connor and Purdie, 1976; Connor, 2004). Miliacin is the sole PTME detected in *Glyceria acutiflora*. It also occurs in *Eragrostis ferruginea* where it is associated with β -amyrin methyl ether (see Ohmoto et al., 1970). As a conclusion, miliacin is found principally in gramineae with either a C3 or C4 metabolism.

Offline and online measurements of purified miliacin extracted from *Chionochloa sp.* are in rather good agreement (-31.2 and -33 ‰, respectively) and are consistent with a C3 metabolism. The difference between these two values might arise from a compound present in low amounts in the “purified” extract.

Miliacin extracted from *P. miliaceum* seeds and bran have $\delta^{13}\text{C}$ values of -23.5 ‰ and -23 ‰, respectively, i.e. typical for a lipid derived from C4 plants. The $\delta^{13}\text{C}$ of miliacin extracted from Lake le Bourget sediments (-21.5 ‰) is close to that of miliacin extracted from *P. miliaceum*, as opposed to the $\delta^{13}\text{C}$ of miliacin extracted from *Chionochloa sp.* (-33 ‰). This indicates that miliacin detected in the Lake le Bourget sediments was produced by a C4 plant. Since most of plants growing naturally in the Alps are C3 plants, miliacin extracted from Lake le Bourget sediments was probably produced by a C4 producer plant foreign to the study area, such as *P. miliaceum*. This is a further support for miliacin as a tracer specifically for *P. miliaceum* around Lake le Bourget.

These results not only provide further evidence on the origin of sedimentary miliacin but also allow the formulation of new assumptions on the transportation of this molecule from the plant to the sediment. We previously assumed that miliacin is present in plant epicuticular waxes (Jacob et al., 2008), as is the case for other pentacyclic triterpenes. High concentrations

of miliacin in *P. miliaceum* bran suggest that this part of the plant could also be an important contributor of sedimentary miliacin, for example by wind transport during winnowing.

Conclusion

The carbon isotopic composition of miliacin detected in the sediments of Lake le Bourget provides further evidence that *P. miliaceum*, a C4 plant, was the biological source of this molecule in this context. This result strengthens the confidence with which miliacin can be used to unravel the history of millet cultivation in sedimentary archives. In addition, the unique source of miliacin makes this biomarker a suitable target to unravel the evolution of environmental parameters from its carbon and hydrogen isotopic compositions. In a larger way, these findings also illustrate the possibilities of the molecular biomarker approach to better understand past interactions between climate, environment and human societies.

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Table captions

Table 1: Concentrations and $\delta^{13}\text{C}$ values of miliacin extracted from *P. miliaceum* seeds and bran, *Chionochloa sp.* and the sediments of Lake le Bourget. Uncertainties are standard errors of the mean for 2 or 3 measurements. Where no uncertainty is given, values represent a single measurement.

Sample	Weight extracted	Miliacin concentration	Miliacin $\delta^{13}\text{C}$ (‰ VPDB)	
			EA-IRMS	GC-IRMS
<i>P. miliaceum</i> seeds	4.3 g	0.3 mg/g		-23.5±0.3
<i>P. miliaceum</i> bran	2.4 g	0.16 mg/g		-23.0±0.3
LDB04 sediment samples	22 g	0.49 µg/g		-21.5±0.3
<i>Chionochloa sp.</i>	Purified compound		-31.2	-33.0±0.3

Figure captions

Figure 1: Structure of miliacin (olean-18-en-3βol methyl ether or germanicol methyl ether).

Figure 2: Partial GC-MS TIC traces of: (i) *Chionochloa sp.* purified miliacin; (ii) F3+F4 fractions of lipid extract from *Panicum miliaceum*; (iii) F3+F4 fractions from Lake le Bourget sediments.

Figure 3: Mass spectra of (a) miliacin extracted from Lake le Bourget sediments, (b) miliacin extracted from *P. miliaceum* bran and (c) purified miliacin from *Chionochloa sp.*

Figure 1

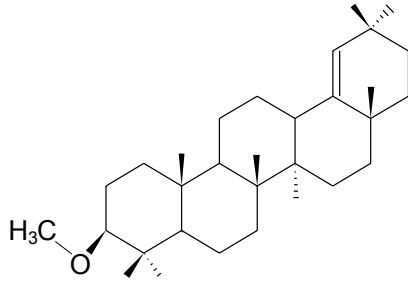


Figure 2

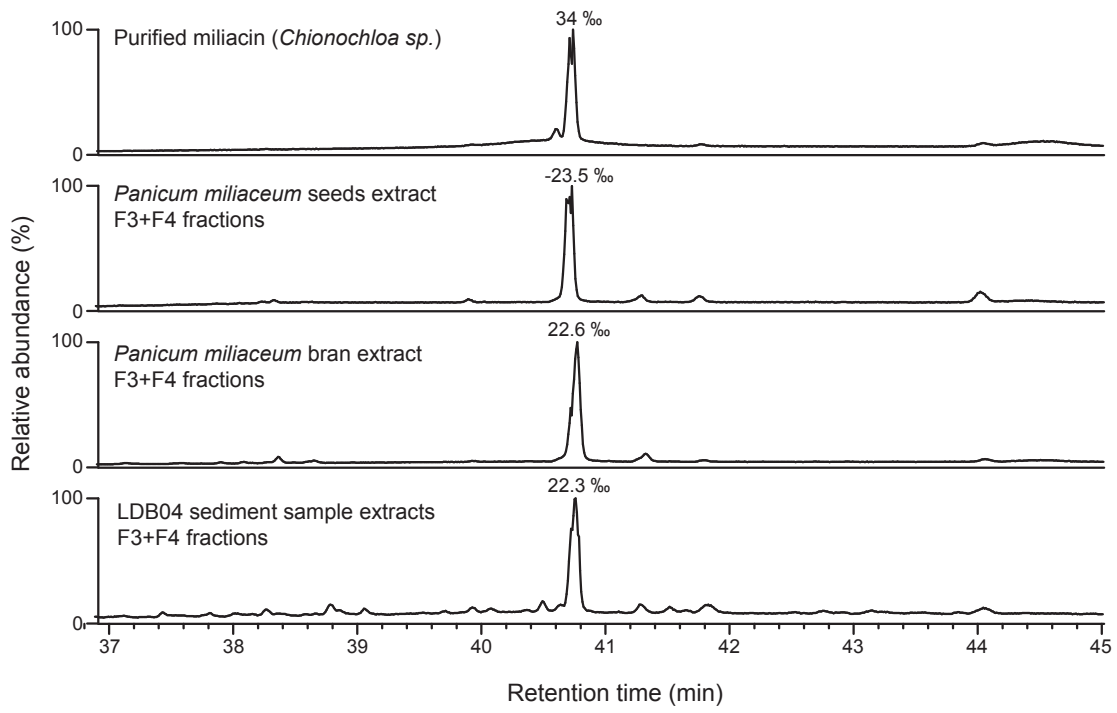


Figure 3

